Electronic supplementary information (ESI)

- 3 Ag-AgBr/TiO₂/RGO nanocomposite for visible-light photocatalytic
- 4 degradation of penicillin G
- 5 Penghua Wang a,b, Yuxin Tang c, Zhili Dong c, Zhong Chen c, Teik-Thye Lim b,b*
- 6 ^a School of Civil and Environmental Engineering, Nanyang Technological University, 50 Nanyang Avenue,
- 7 Singapore 639798, Singapore

1

2

12

13

14

- 9 Loop, CleanTech One, Singapore 637141, Singapore
- ^c School of Materials Science and Engineering, Nanyang Technological University, 50 Nanyang Avenue,
- 11 Singapore 639798, Singapore

Synthesis of graphene oxide (GO)

- 15 Millipore Co. MilliQ (MQ) water with resistivity of 18.2 M Ω cm was used throughout the
- study unless otherwise stated. GO was synthesized via the modified Hummers method. ¹ In a
- typical synthesis, 1.0 g of natural graphite (SP1, Bay Carbon Company, USA) and 0.5 g of
- NaNO₃ (99%, Sigma-Aldrich) were added into 23 mL of H₂SO₄ (98%, Sigma-Aldrich),
- followed by magnetic stirring for 1 h at <5 $^{\circ}$ C in an ice–water bath. Then, 3.0 g of KMnO₄
- 20 (99%, Sigma-Aldrich) was slowly introduced to the above suspension, followed by
- 21 continuous magnetic stirring for another 1 h at <5 °C. Subsequently, the ice–water bath was
- replaced by an oil–bath. The suspension was gradually heated to 35 \pm 1 $^{\circ}$ C and maintained at
- 23 this temperature for 2 h. After that, 46 mL of MQ water was added slowly to the suspension,
- 24 and the temperature was raised to 98 ± 1 °C and kept for 30 min. After the suspension was

E-mail address: cttlim@ntu.edu.sg (T.-T. Lim).

^{*} Corresponding author. Tel.: +65 6790 6933; fax: +65 6791 0676.

32

33

cooled to room temperature, 140 mL of MQ water and followed by 10 mL of H₂O₂ (30%, Sigma–Aldrich) were added to stop the reaction. Finally, the suspension was repeatedly centrifuged and washed with 5% HCl (37%, Sigma–Aldrich) and MQ water till neutral pH. The product was vacuum–freeze dried at –50 °C for one week to obtain GO powder.

W. S. Hummers and R. E. Offeman, *J. Am. Chem. Soc.*, 1958, **80**, 1339-1339.

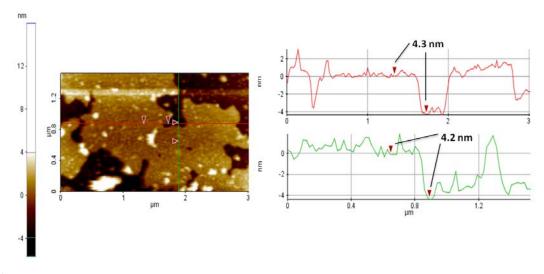


Fig. S1. AFM image of GO with height profiles taken in two different locations.