

## **Supporting information**

### **Table of contents**

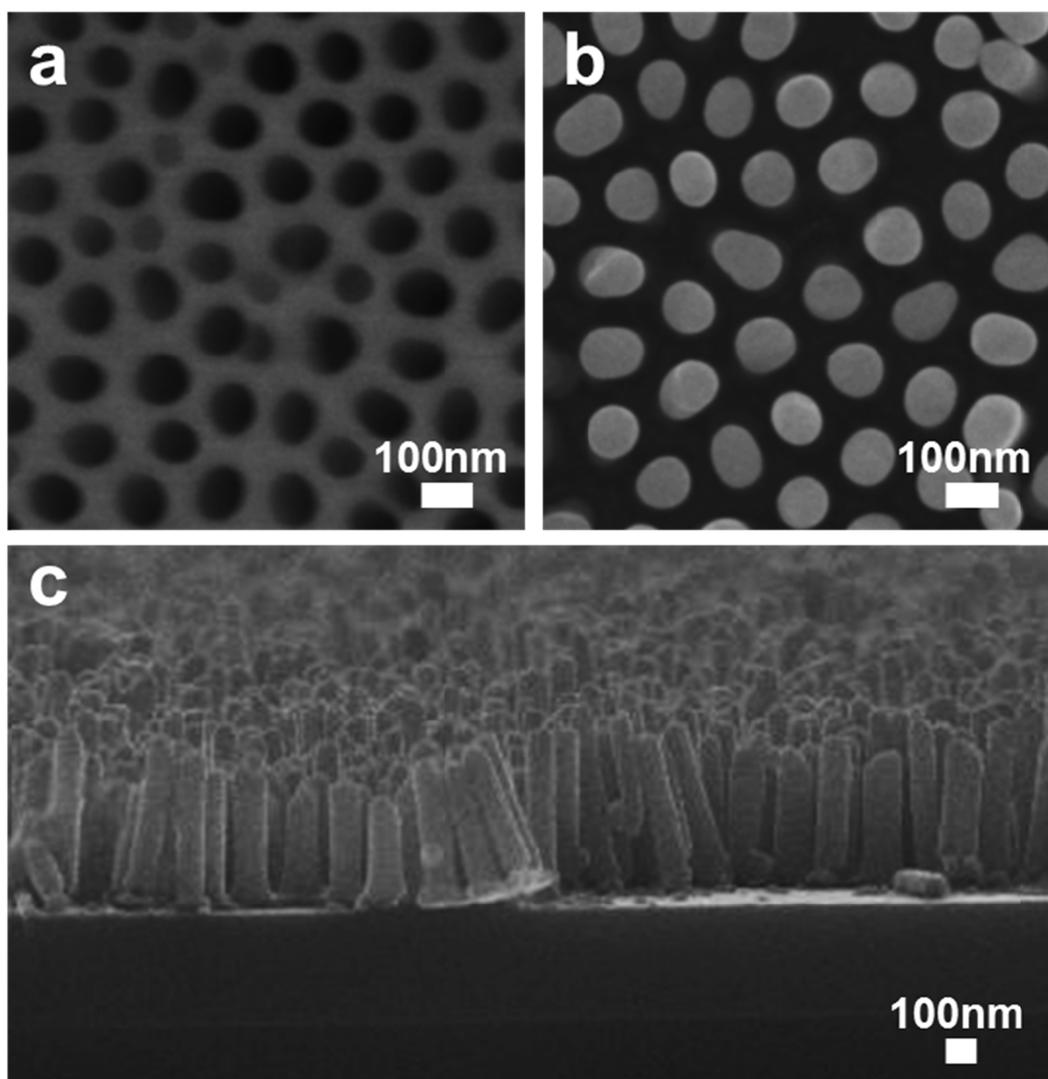
1. SEM image of thin film silver .....	2
2. SEM image of SNW .....	3
3. Stability study of silver nanograss as a SERS substrate .....	4
4. Schematic route of the preparation of patterned silver nanograss .....	5
5. CV at various concentrations of hydrogen peroxide .....	6

## 1. SEM image of thin film silver



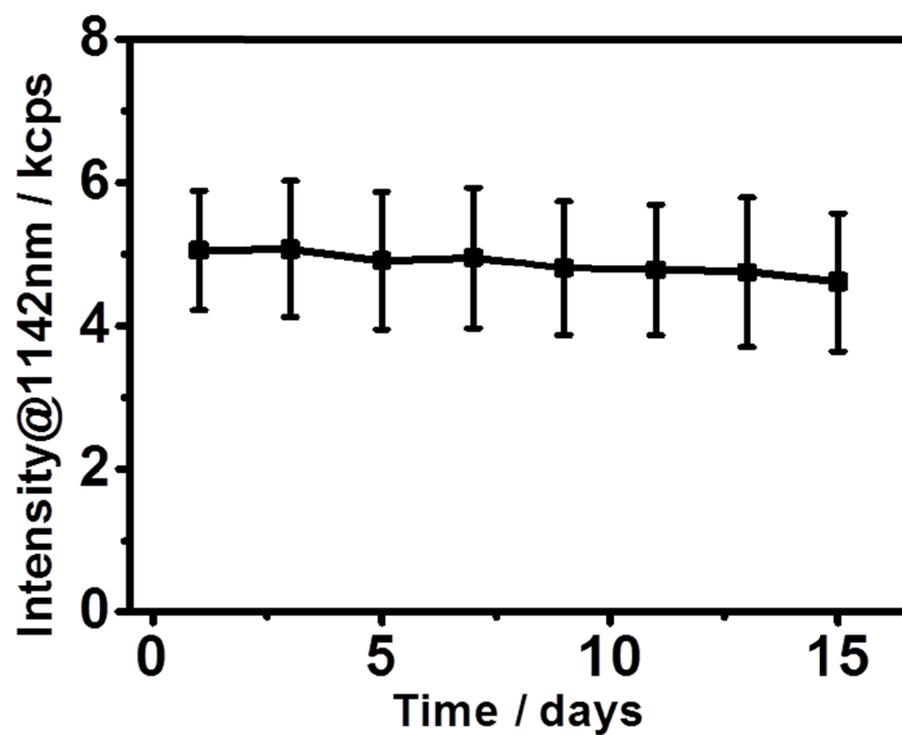
*Figure S1.* Scanning electron microscopy image of a thermally evaporated silver thin film.

## 2. SEM image of SNW



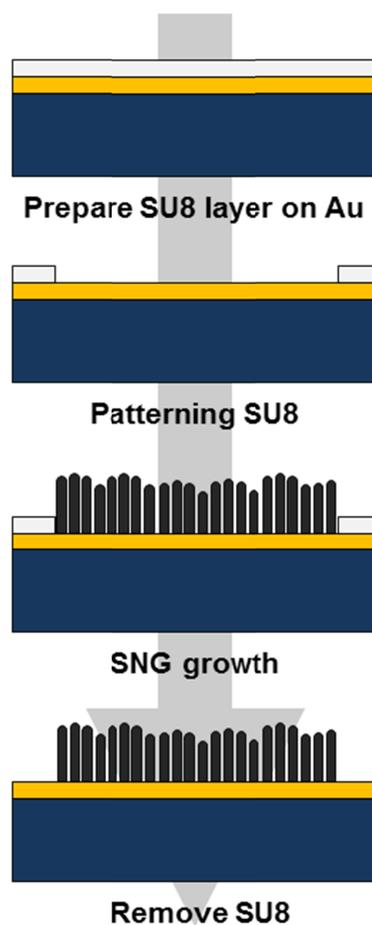
**Figure S2.** Scanning electron microscopy images of the top-view of an a) AAO template, b) silver nanowire array and c) side-view image of the silver nanowire array.

### 3. Stability study of silver nanograss as a SERS substrate



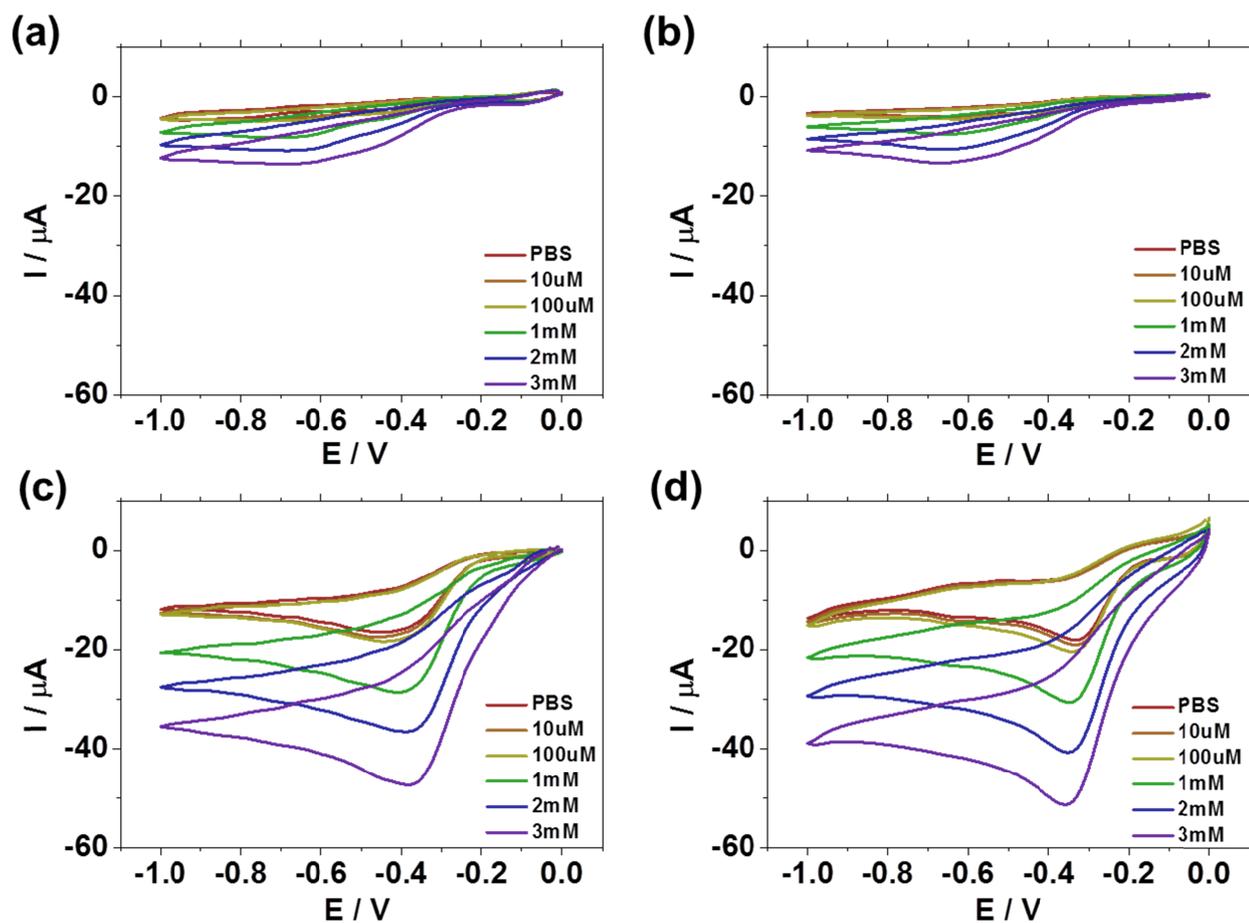
**Figure S3.** Plot of SERS intensity (1142 nm) of p-ATP ( $10^{-5}$  M) deposited silver nanograss vs time. The laser ( $\lambda = 633$  nm) on the sample was  $500 \mu\text{W}$  and the integration time was 10 s.

#### 4. Schematic route of the preparation of patterned silver nanograss



*Figure S4.* Schematic route of preparation of patterned silver nanograss.

## 5. CV at various concentrations of hydrogen peroxide



**Figure S5.** Cyclic voltammograms of a) thin film silver, b) SNG0.8, c) nanograss and d) SNW in the presence of  $\text{H}_2\text{O}_2$  at various concentrations (from the top: 0, 0.01, 0.1, 1, 2 and 3 mM) under 1x PBS solution (pH 7.0). Scan rate = 50mV/s.