## **Supporting Information**

## Graphene encapsulated hollow TiO<sub>2</sub> nanospheres: efficient synthesis

## and enhanced photocatalytic activity

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Synthesis of TiO<sub>2</sub> NSs and the blended TiO<sub>2</sub> NSs/graphene (TiO<sub>2</sub>-G): TiO<sub>2</sub> NSs was directly produced from amine-modified TS NPs following the synthesis procedure of GT NSs. 200 mg TS NSs was etched in 2M NaOH solution and TiO<sub>2</sub> NSs was obtained after washing with water. Then, TiO<sub>2</sub> NSs was re-dispersed into graphene oxide (GO) solution, containing 4 mg GO. The dispersion was chemically reduced by hydrazine in 90°C for 1h. After filtration, washing and dryness, the blended TiO<sub>2</sub> NSs/reduced graphene oxide (TiO<sub>2</sub>-G) was obtained.



Fig. S1 AFM image (a) and corresponding thickness analysis of GO (b). TEM image (c) and electron diffraction pattern (d) of GO.



Fig. S2 SEM images of GOTS NSs with different weight ratios of TS NSs to GO: 100: 0 (A); 100:0.5 (B); 100:1 (C); 100:2 (D); 100:5 (E) and 100:10 (F).



Fig. S3 Raman spectra of TiO<sub>2</sub> NSs and GT NSs-2 from 100 nm to 1800 nm.



Fig. S4 High-resolution XPS spectra of GT NSs-2.



Fig. S5 Thermogravimetric analysis (TGA) of GT NSs-2.



Fig. S6 Photocatalytic rate constants of degradation of RhB.

Time (h)	0 h	1 h	3 h	5 h	7 h
TOC (ppm)	15.14	14.58	11.97	9.63	8.06

Table S1. TOC result for the degradation of RhB using GT NSs-2 as catalyst.

Samples	TiO <sub>2</sub> NSs	TiO <sub>2</sub> -G	GT NSs-0.5	GT NSs-1	GT NSs-2	GT NSs-5
TOC (ppm)	10.03	9.12	9.77	8.37	8.06	11.94

Table S2. TOC results for the degradation of RhB using different catalysts after irradiation for 7h.



Fig. S7 Fluorescence spectrum of TiO<sub>2</sub> NSs and GT NSs-2.