Electronic Supplementary Information (ESI) for:

Ultrafine nanoporous PdFe/Fe₃O₄ catalysts with doubly enhanced activities

towards electro-oxidation of methanol and ethanol in alkaline media

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Table S1. Real chemical compositions of the np-PdFe samples obtained by dealloying the rapidly solidified Al-Pd-Fe precursor alloys in the 20 wt.% NaOH solution.

Samples	Pd (at.%)	Fe (at.%)	Al (at.%)
np-PdFe-1	31.0	60.1	8.9
np-PdFe-2	39.0	53.6	7.4
np-PdFe-3	44.5	46.3	9.2
np-PdFe-4	55.6	34.8	9.6
np-PdFe-5	62.9	28.4	8.7



Figure S1. Section-view SEM images showing the microstructure of the np-PdFe (a: np-PdFe-1; b,c: np-PdFe-2; d: np-PdFe-3; e: np-PdFe-4; f: np-PdFe-5) samples obtained by dealloying the rapidly solidified Al-Pd-Fe precursor alloys in the 20 wt.% NaOH solution.



Figure S2. EDX spectra of the np-PdFe (a: np-PdFe-1; b: np-PdFe-2; c: np-PdFe-4; d: np-PdFe-5) samples obtained by dealloying the rapidly solidified Al-Pd-Fe precursor alloys in the 20 wt.% NaOH solution.



Figure S3. a,c,e) TEM and b,d,f) HRTEM images showing the nanoporous structure of a,b) np-PdFe-2, c,d) np-PdFe-3 and e,f) np-PdFe-4 obtained by dealloying the rapidly solidified Al₇₅Pd₁₀Fe₁₅, Al₇₅Pd_{12.5}Fe_{12.5} and Al₇₅Pd₁₅Fe₁₀ alloys in the 20 wt.% NaOH solution, respectively. Insets in a,c,e) Corresponding SAED patterns. Some ligaments are highlighted by solid arrows in b,d,f).