Interaction	A_{ij} / eV	$ ho_{ij}$ / Å	C_{ij} eV Å ⁶	Pseudo cubic lattice parameter a_p /Å	MD stability	Reference
Co ³⁺ –O ²⁻	1329.82	0.3087	0	3.766	Stable	[1-3]
Co ³⁺ –O ²⁻	1007.11	0.3069	0	3.586	-	[4]
Co ³⁺ –O ²⁻	1226.31	0.3087	0	3.712	-	[5]
Co ³⁺ –O ²⁻	1371.71	0.3087	0	3.780	-	[5]
Co ³⁺ –O ²⁻	1156.61	0.3087	0	3.679	-	[5]
Co ³⁺ –O ²⁻	1006.61	0.3087	0	3.601	-	[5]
Co ³⁺ –O ²⁻	1226.31	0.3087	0	3.915	Unstable	[5]
$\mathrm{Sr}^{2+}\mathrm{-O}^{2-}$	682.172	0.3945	0			[6]
$Ba^{2+}-O^{2-}$	905.7	0.3976	0			[7]
$O^{2-} - O^{2-}$	9597.96	0.2192	32.00			[5]

Table S1. Comparison of Buckingham pair potentials for $Co^{3+}-O^{2-}$ in BSC.



Fig. S1. Configurational energy and temperature for BSCF (d = 0.5) at 500 K as a function of time after 50 ps equilibration by MD simulation.



Fig. S2. Configurational energies for (a) BSC, (b) BSF and (c) BSCF as a function of Ba/Sr disorder parameter d. Energies were averaged during 50 ps of production run after equilibration at 1200 K by MD simulations.



Fig. S3. Mean square displacements (MSDs) of ions in BSCF as a function of time at 2000 K.



Fig. S4. Trajectories of oxygen atoms (red) in MD simulations for BSCF with ordered (d=0, left panel) and disordered (d=0.5, right panel) Ba/Sr arrangement at 1200 K. Both the figures were viewed along c-axis. Note that three oxygen layers, Ba-O, Sr-O, and TM-O₂ layers, are included. (See figure 1 in main text.)

Reference

- [1] M. Cherry, M.S. Islam, C.R.A. Catlow, J. Solid State Chem., 118 (1995) 125-132.
- [2] M.S. Islam, M. Cherry, C.R.A. Catlow, J. Solid State Chem., 124 (1996) 230-237.
- [3] M.T. Buscaglia, V. Buscaglia, M. Viviani, P. Nanni, J. Am. Ceram. Soc., 84 (2001) 376-384.
- [4] S.M.Woodley, C.R.A.Catlow, P.Piszora, K.Stempin, E.Wolska, J. Sol. State Chem., 153

(2000) 310.

- [5] D.J.Binks PhD Thesis, University of Surrey (1994).
- [6] M. McCoy, R. W. Grimes, W. E. Lee, Philosophical Magazine A, 75 (1997) 833-846.
- [7] C. R. Stanek PhD Thesis, Imperial College London (2003).