

Oxidized graphene and graphite as metal-free catalysts for aqueous sulfide oxidation

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Supplementary Material

1. Tables

Table S1 - Average diameter of graphites and graphenes obtained.

Sample	λ / nm	2θ / °	d_{002} / nm	L_a / nm	L_c / nm
GT	0.1540	26.5388	0.3355	42.6	20.6
GT-80	0.1540	26.5603	0.3352	44.1	21.3
GT-140	0.1540	26.5102	0.3358	44.1	21.3
GN	0.1540	26.2870	0.3386	22.7	11
GN-80	0.1540	26.3362	0.3380	21.7	10.5
GN-140	0.1540	26.2498	0.3391	23.3	11.3

Table S2 – Raman data of graphenes and graphites.

	G position/cm ⁻¹	G' position/cm ⁻¹	G FWHM	G' FWHM	I _G /I _{G'}
GN	1576.32	2650.94	35.29	59.94	1.02
GN 80	1579.40	2656.74	37.84	98.05	1.78
GN 140	1578.95	2656.83	30.97	51.50	1.16
GT	1580.11	2664.76	34.95	73.29	1.11
GT 80	1582.96	2663.24	35.13	68.98	0.74
GT 140	1582.42	2663.97	31.16	76.64	0.25

2. Figures

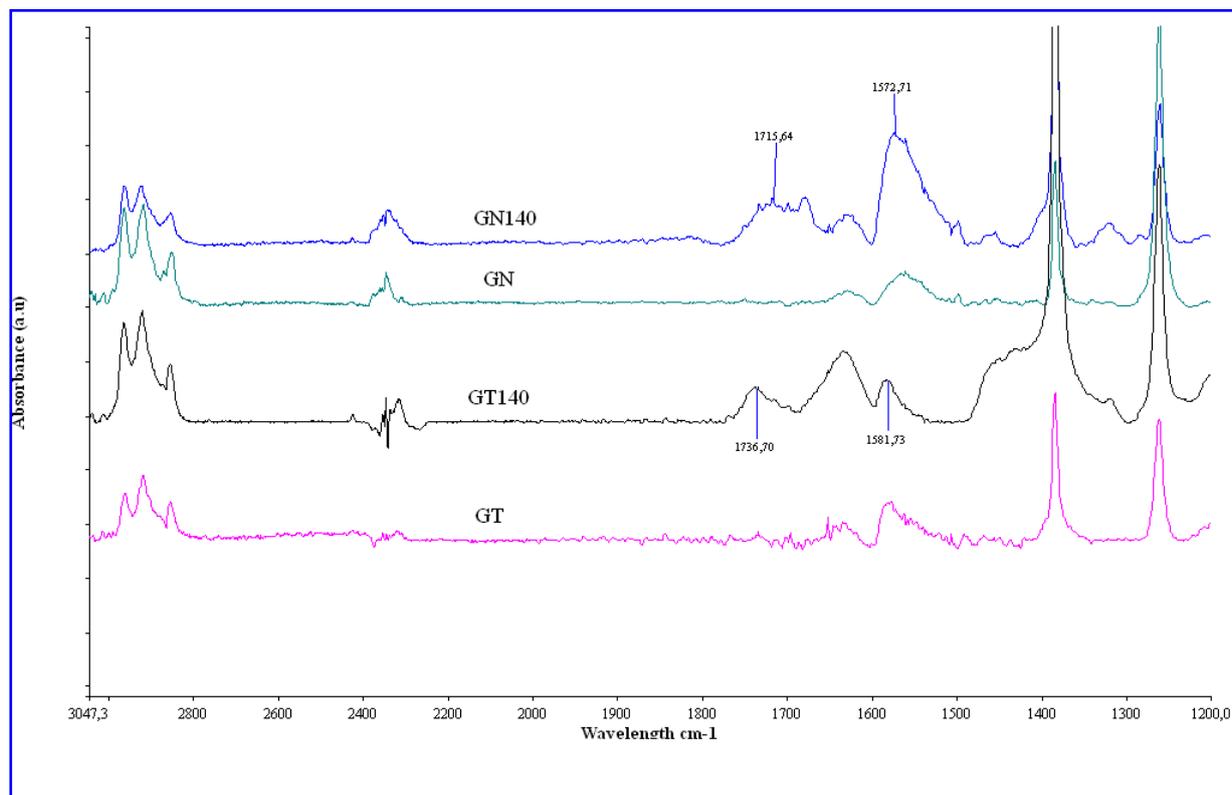


Figure S1: FTIR spectra of GN and GT in KBr before and after functionalisation with HNO₃ at 140°C.

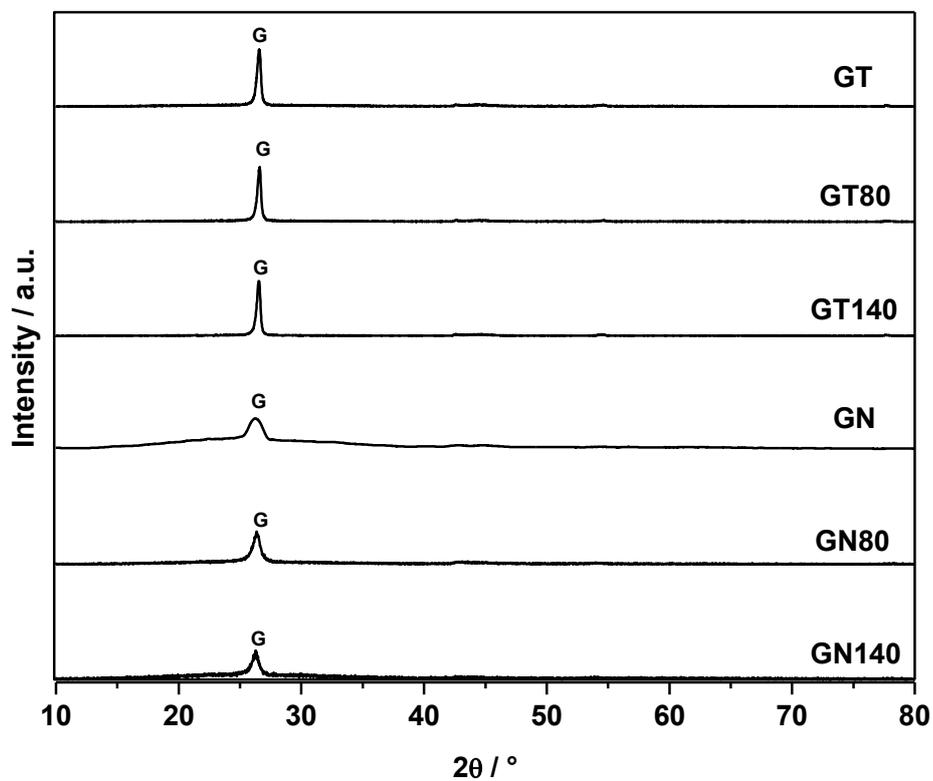


Fig. S2 - Diffractograms of graphenes and graphites functionalized by treatment with HNO₃.

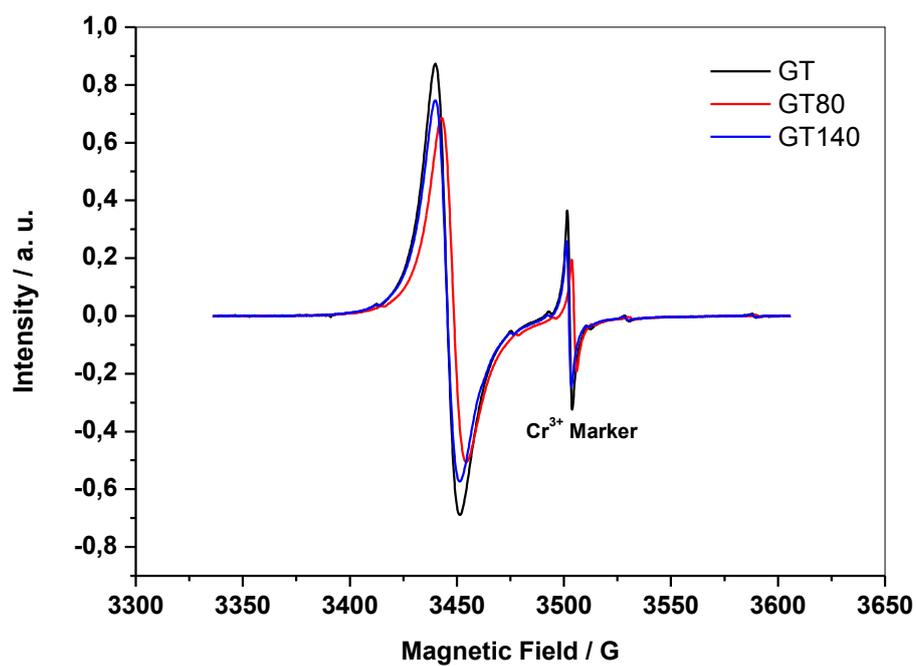


Fig. S3 - Room temperature Electron Paramagnetic Resonance spectrum of the graphite samples. The absolute line areas are normalized to the Cr³⁺ marker and the sample masses.

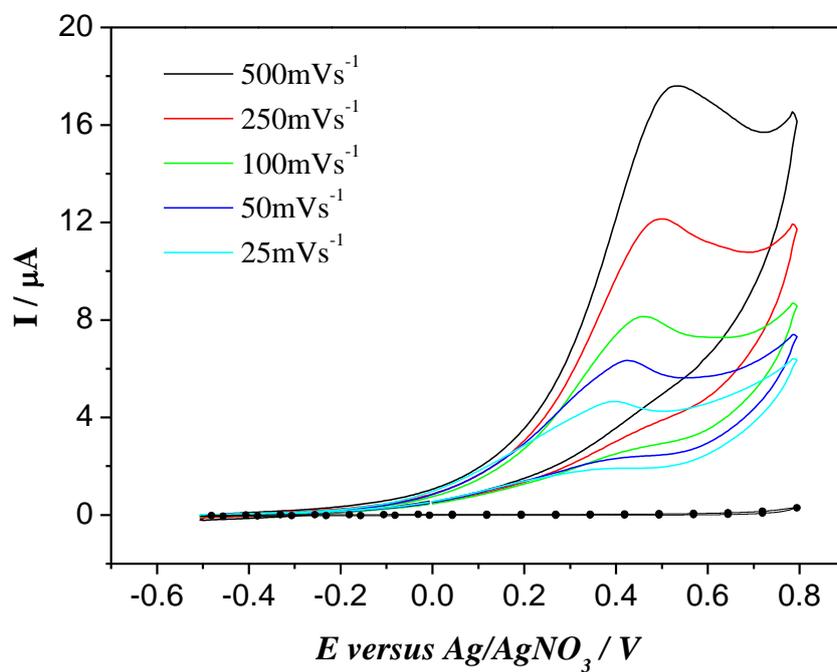


Fig. S4 - Overlaid cyclic voltammograms of GT carbon paste electrode at different scan rates in presence of 1 mM Na₂S.

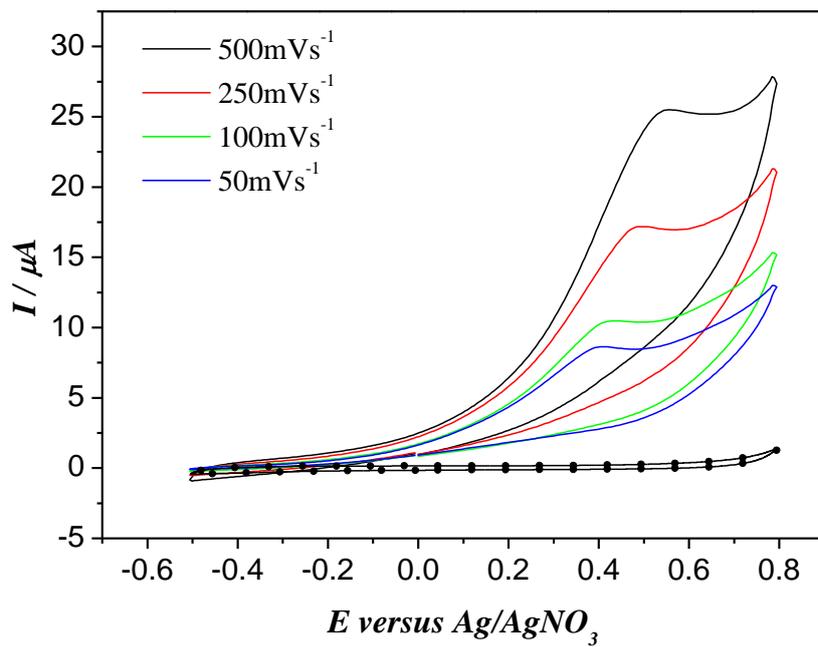


Fig. S5 - Overlaid cyclic voltammograms of GT80 carbon paste electrode at different scan rates in presence of 1 mM Na_2S .

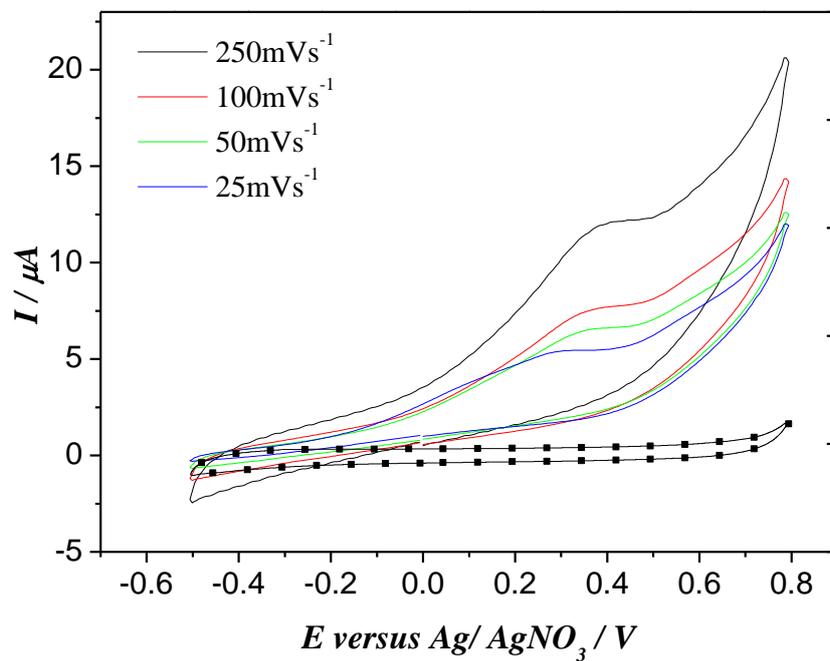


Fig. S6 - Overlaid cyclic voltammograms of GT140 carbon paste electrode at different scan rates in presence of 1 mM Na_2S .

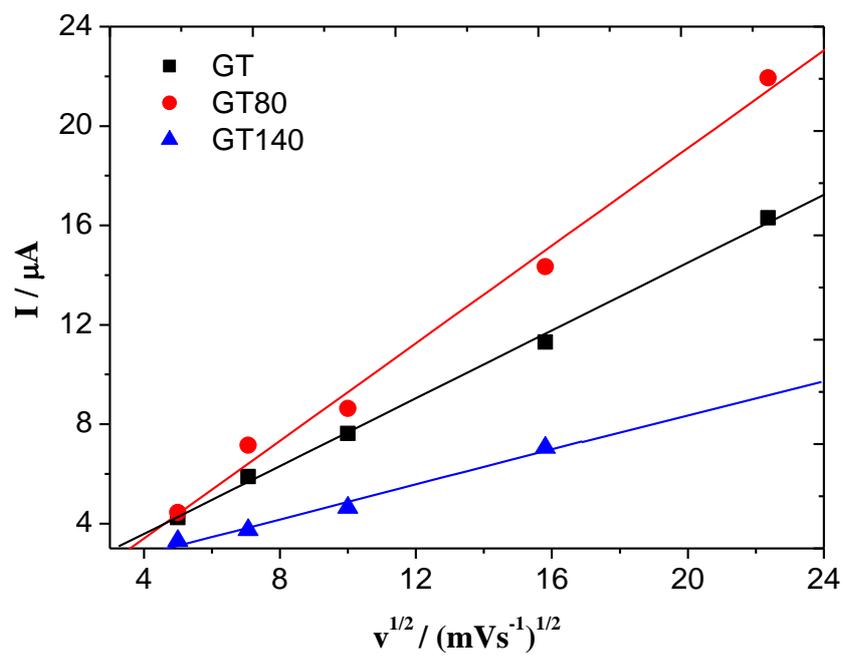


Fig. S7 - Plot of anodic peak current of sulfide oxidation vs. square root of scan rate obtained with carbon paste electrodes prepared with different graphites.