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## **Bio-inspired High Performance Electrochemical Supercapacitor based** on Conducting Polymer modified Coral-like Monolithic Carbon

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 $\label{eq:Figure S1} Figure \ S1. \ The \ illustration \ of \ three-electrode \ cell \ system.$ 



Figure S2. The X-ray photoelectron spectroscopy (XPS) curve of coral-like carbon.

Sample		The amount of aniline (µL)	Conducting polymer
C-CA1		20	PANi (Polyaniline)
C-CA2	C-CA (Carbon composited with PANi by Chemical oxidative	60	PANi
C-CA3	polymerization)	100	PANi
C-CA4		140	PANi

## Table S1. The abbreviations of hybrid materials by oxidative polymerization



Figure S3. SEM images of C-CA1 (a, b) and C-CA3 (c, d)



Figure S4. TEM image (a) and the energy dispersive X-ray (EDX) spectra (b) of C-ET3



Figure S5. N<sub>2</sub> adsorption-desorption isotherms of hybrid materials. The data are shifted by 800 cm<sup>3</sup> g<sup>-1</sup> STP relative to each other for clarity.

Sample	$S_{BET} (m^2 g^{-1})$	$V_{p} (cm^{3} g^{-1})$	D <sub>p</sub> (nm)	Content of conducting polymer (wt %) <sup>a</sup>
C-CA1	715	1.89	6.51	21
C-CA2	527	1.35	6.48	32
C-CA3	367	1.19	6.28	39
C C L L	277	0.82	6.09	42

Table S2. Structural Properties of the hybrid materials



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Figure S7. FTIR spectrum of PANi composites prepared by chemical oxidative polymerization. The data of FTIR are shifted by 50 relative to each other for clarity.



Figure S8. Cyclic voltammograms of C-CA at the scan rate of 5 mV s<sup>-1</sup> (a) and galvanostatic charge-discharge curves of the hybrid materials (b-d) at the scan rate of 0.5 A g<sup>-1</sup>.

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	2.0 A g <sup>-1</sup>	1.0 A g <sup>-1</sup>	0.5 A g <sup>-1</sup>
C-CA1	285	295	313
C-CA2	384	425	459
C-CA3	511	575	595
C-CA4	338	360	397

Table S3. Capacitance performances of C-CA



Figure S9. The Nyquist plots in the range of 10 kHz to 10 mHz for C-EA and C-ET.





Figure S11. Cycling performance at 1.0 A g<sup>-1</sup> of C-EA and C-ET.