# The room temperature formation of gold nanoparticles from the reaction of cyclohexanone and auric acid; a transition from dendritic particles to compact shapes and nanoplates

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Supporting information for this article is available on the WWW under http://www.chemeurj.org/ or from the author.

#### **Supplementary Information 1**



Figure S1: <sup>13</sup>C NMR spectrum of pure cyclohexanone standard (600 MHz, D<sub>2</sub>O)

# **Supplementary Information 2**







Figure S3: Zoomed in <sup>13</sup>C NMR spectra (600 MHz, D<sub>2</sub>O) across four key regions of  $\delta$  ppm where the top spectrum in the series represents a gold nanoparticle solution (diluted by a factor of 100) formed from the reaction of auric acid (0.28 mM) and cyclohexanone (0.48 M) in D<sub>2</sub>O at room temperature after 1 hr and the bottom spectrum in the series represents cyclohexanone alone in D<sub>2</sub>O (0.48 M). The peaks highlighted in dotted red relate were identified as 2-chlorocyclohexanone environments

#### Supplementary Information 3(b)



Figure S3: Zoomed in <sup>13</sup>C NMR spectra (600 MHz,  $D_2O$ ) across four key regions of  $\delta$  ppm where the top spectrum in the series represents a gold nanoparticle solution (diluted by a factor of 100) formed from the reaction of auric acid (0.28 mM) and cyclohexanone (0.48 M) in  $D_2O$  at room temperature after 1 hr and the bottom spectrum in the series represents cyclohexanone alone in  $D_2O$  (0.48 M). The peaks highlighted in dotted red relate were identified as 2-chlorocyclohexanone environments

### Supplementary Information 3(c)



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