

Supporting Information for
A sandwich N-doped graphene/Co₃O₄ hybrid: an efficient catalyst for
selective oxidation of olefins and alcohols

Renfeng Nie^a, Juanjuan Shi^a, Weichen Du^a, Wensheng Ning^b, Zhaoyin Hou^{*,a}, Feng-Shou Xiao^a

a Institute of Catalysis, Department of Chemistry, Zhejiang University, Hangzhou 310028, China

b College of Chemical Engineering and Materials Science, Zhejiang University of Technology, Hangzhou 310014, China

*Address correspondence to zyhou@zju.edu.cn

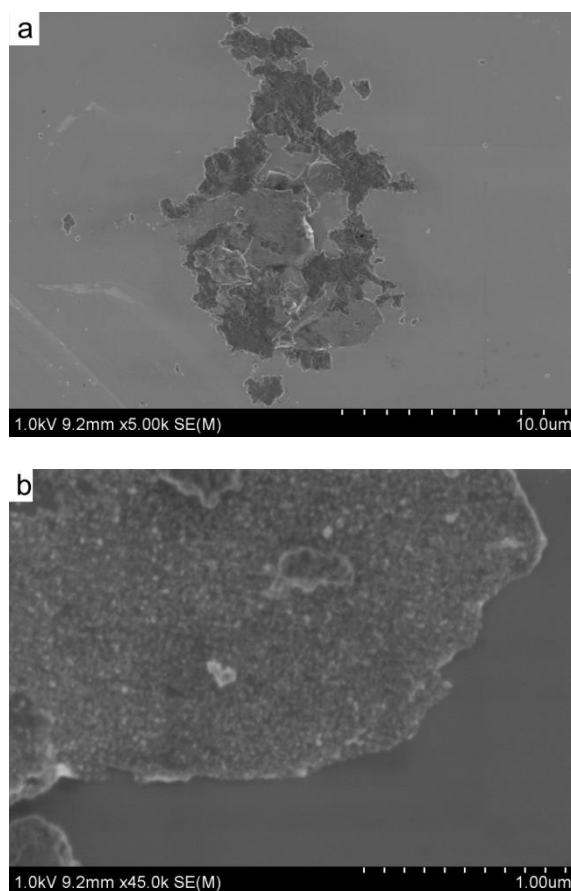


Figure S1. SEM images of Co₃O₄/RGO-N hybrid

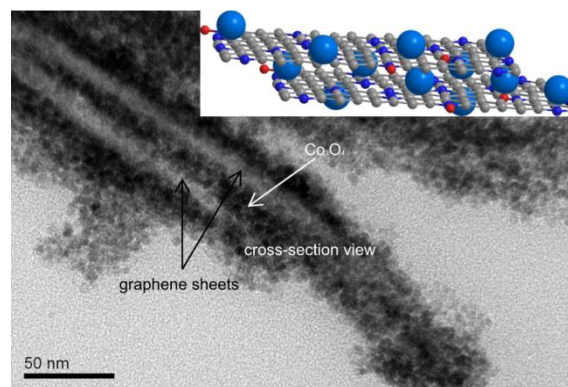


Figure S2. Structural illustration of the N-doped graphene- Co_3O_4 sandwich. TEM image of cross-section view of $\text{Co}_3\text{O}_4/\text{RGO-N}$ hybrid, the inset is the structural model.

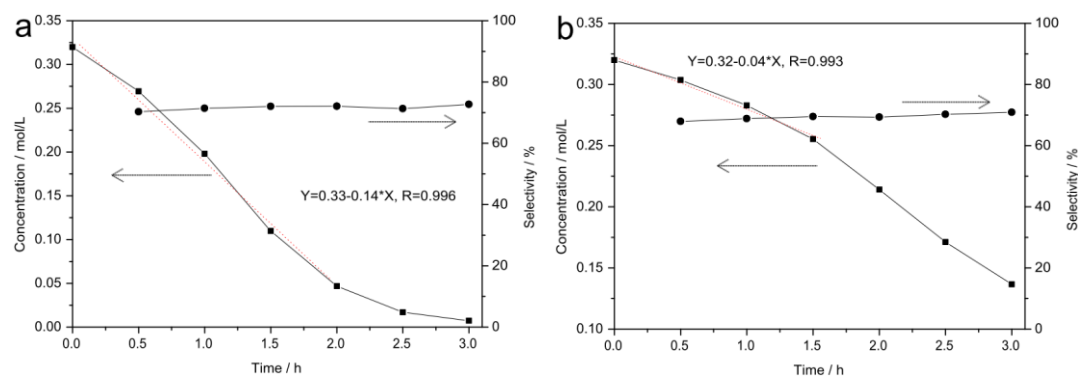


Figure S3. Time-course of epoxidation of styrene with O_2 catalyzed by (a) $\text{Co}_3\text{O}_4/\text{RGO-N}$ and (b) $\text{Co}_3\text{O}_4/\text{RGO}$. Reaction conditions: temperature ($100\text{ }^\circ\text{C}$), the amount of M (0.15 mmol), styrene (8 mmol), DMF (25 mL), flow rate of O_2 (20 mL/min).

The rate constant of epoxidation reaction could be calculated from the time-concentration courses (these data were added as Fig. S3 in the supporting information). It was concluded that the rate constant of oxidation of styrene over $\text{Co}_3\text{O}_4/\text{RGO-N}$ (0.14 mol/L/h) is higher than that over $\text{Co}_3\text{O}_4/\text{RGO}$ (0.04 mol/L/h). These data indicated that styrene is more likely to be activated (lower activation energy) over $\text{Co}_3\text{O}_4/\text{RGO-N}$ due to the high dispersion of Co_3O_4 NPs of RGO-N sheets, high abundance of Co(III) species derived from the strong interaction between Co_3O_4 NPs and RGO-N , well dispersion of $\text{Co}_3\text{O}_4/\text{RGO-N}$ in solvent.