

Supporting Information

Facile 3D Self-Assembly of Porous Iron Hydroxide and Oxide Hierarchical Nanostructures for Removing Dyes from Wastewater

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Fig. S1. Photo image of iron nails in different medium for 5 days at room temperature (about 20 °C). (A) water; (B) 0.2 M NH₄Cl; (C) 0.2 M (NH₄)₂SO₄.

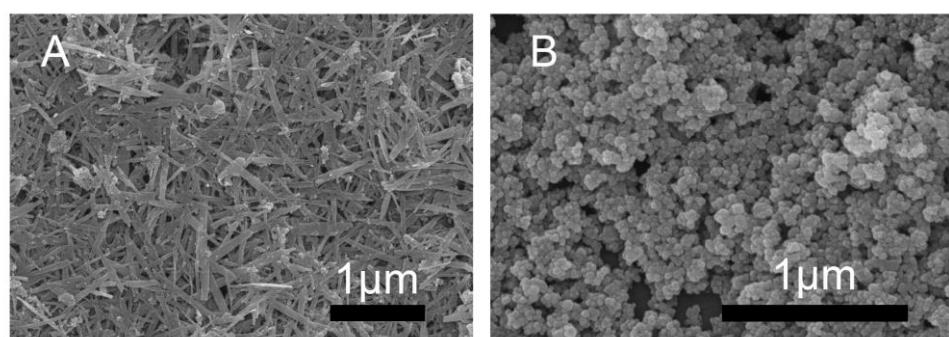


Fig.S2. SEM images of conventional α-FeOOH nanoparticles as-prepared and α-Fe₂O₃ nanoparticles.

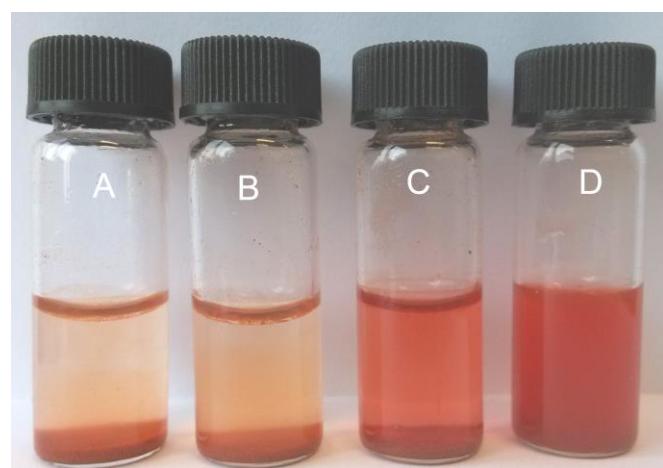


Fig. S3. Photo images after adsorption of Congo red by(A) iron hydroxide hierarchical nanostructures; (B) iron oxide hierarchical nanostructures; (C) conventional α -FeOOH and (D) α -Fe₂O₃ nanoparticles, respectively.

Table S1 BET surface area and Maximum Congo red adsorption capacities (q_m) of different adsorbents.

Adsorbents	BET surface area (m^2/g)	q_m (mg/g)	References
$\alpha\text{-FeOOH}$ Hierarchical Nanostructures	101	56.3	Present work
$\gamma\text{-Fe}_2\text{O}_3$ Hierarchical Nanostructures	114	58.2	Present work
MnO_2 Hierarchical Hollow Nanostructures		80	[10]
Mesoporous Fe_2O_3	111	53	[35]
ZnFe_2O_4 hollow nanospheres		16	[36]
ZnAl-layered double hydroxides		20	[37]
FeOOH hierarchical nanostructures	239	240	[12]
Urchin-like $\alpha\text{-FeOOH}$ hollow spheres	97	275	[38]
Ni(OH)_2 nanosheets	127	60	[39]
hierarchical spindlelike $\gamma\text{-Al}_2\text{O}_3$	149	90	[40]
Nestlike $\alpha\text{-Fe}_2\text{O}_3$ Nanostructures	152	160	[41]