Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A

Supporting information

Stable colloidal boron nitride nanosheets dispersion and its potential application in catalysis

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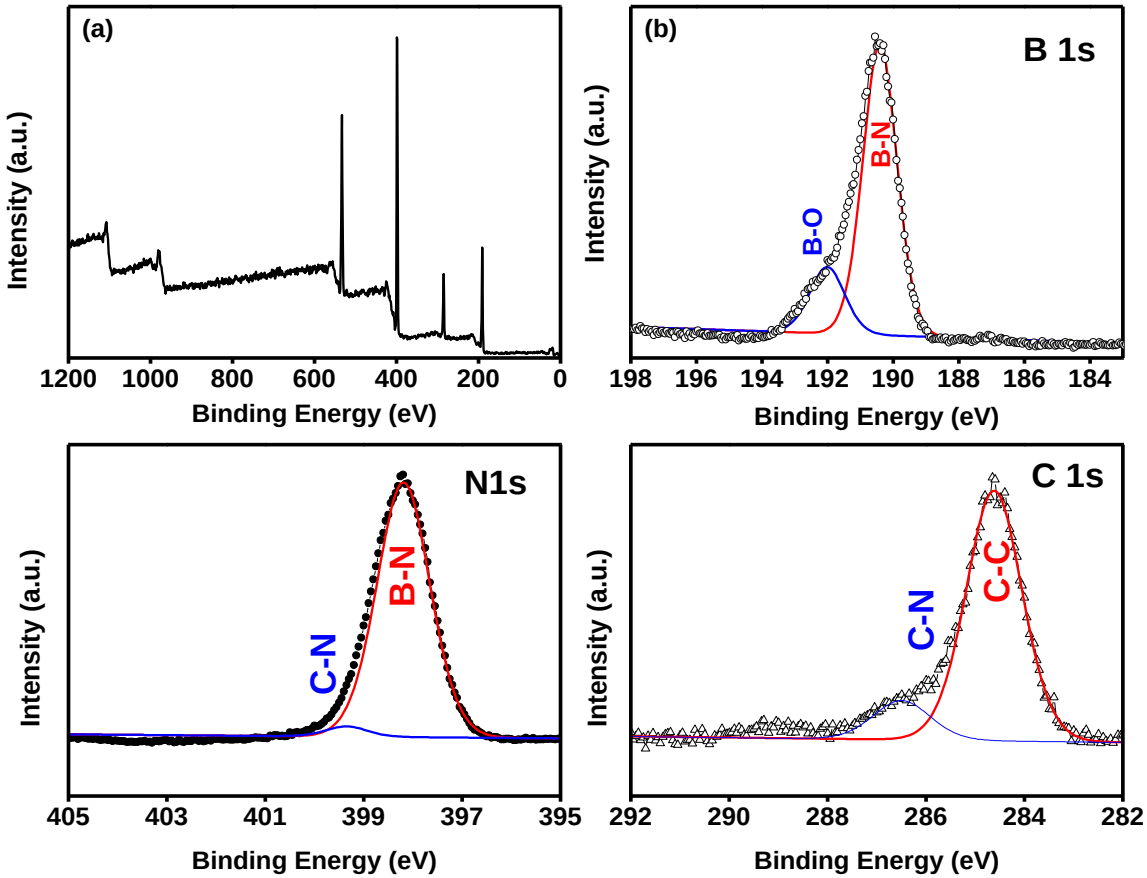

Fig. S1 XPS survey spectra of the as-obtained h-BN powder, specifying B, N, and C 1s core levels, respectively. The spectrum curves (black) are deconvoluted by Gaussian fitting (blue and red). Elemental formula and possible bonding information are given in the spectra.

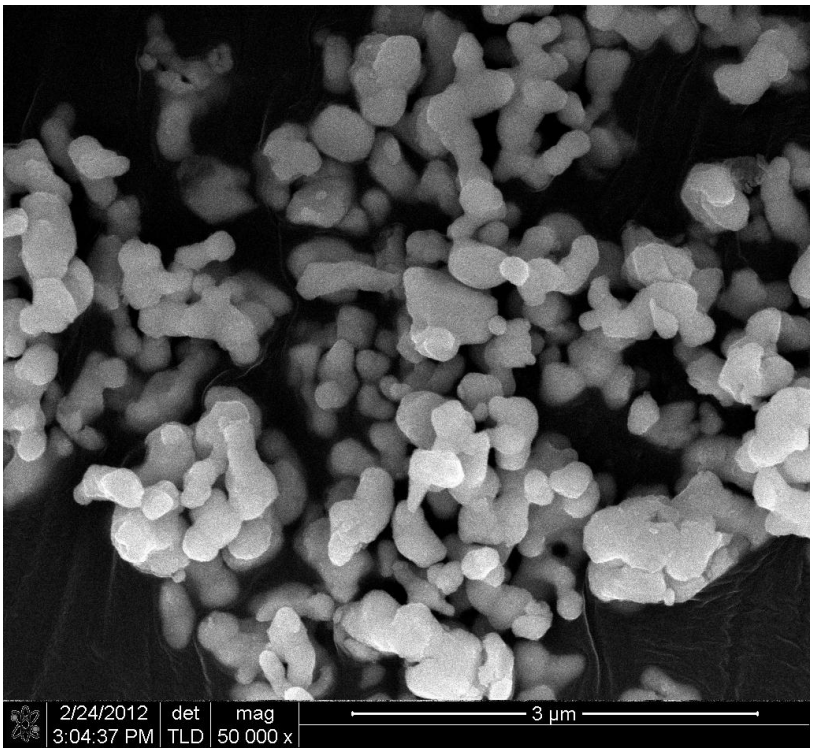

Fig. S2 SEM image of the h-BN powder particles obtained by Nag's method.

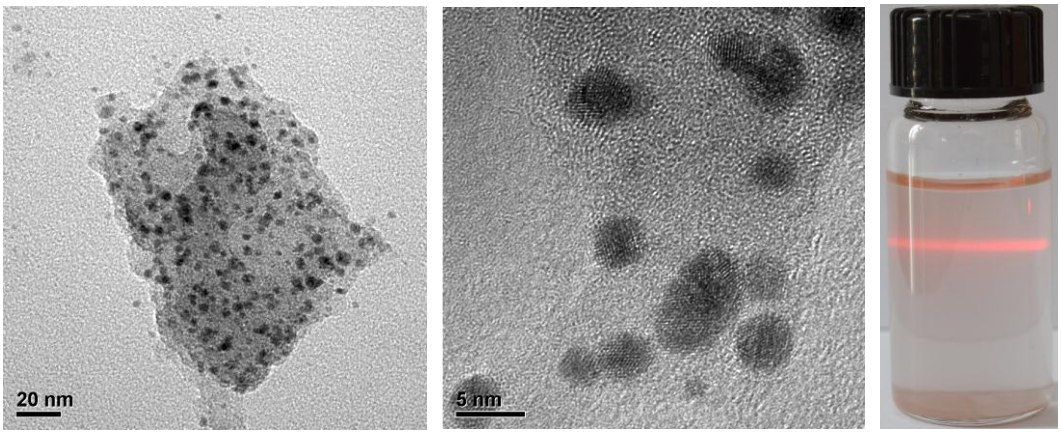

Fig. S3 TEM and HRTEM images of *h*-BN sheets/Au composites, and picture of direct of *h*-BN sheets/Au composites colloidal dispersion.

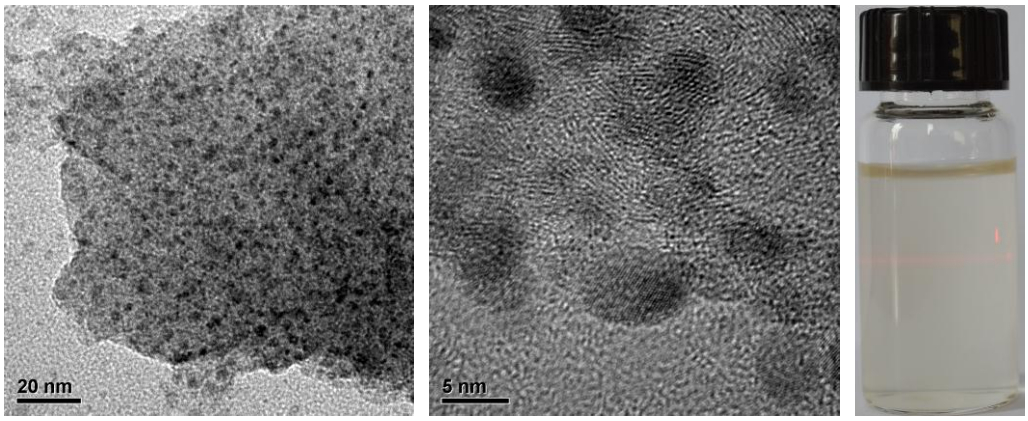

Fig. S4 TEM and HRTEM images of *h*-BN sheets/Pt composites, and picture of direct of *h*-BN sheets/Pt composites colloidal dispersion.

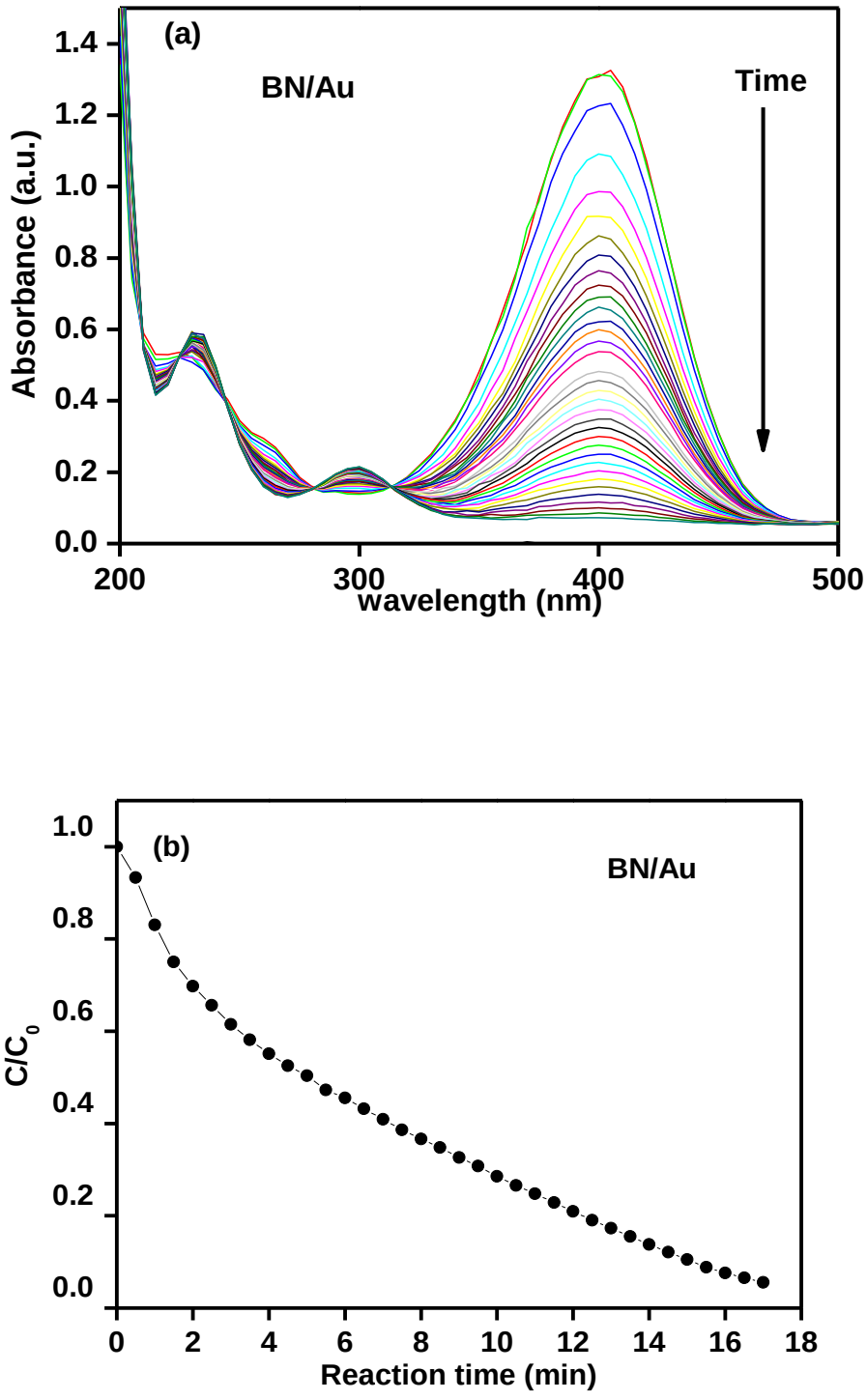

Fig. S5 (a) Evolution of UV-Vis spectra during p-nitrophenol degradation catalyzed by *h*-BN sheets/Au composites. Each curve was taken successively at 30-sec intervals. (b) Conversion of p-nitrophenol as a function of reaction time.

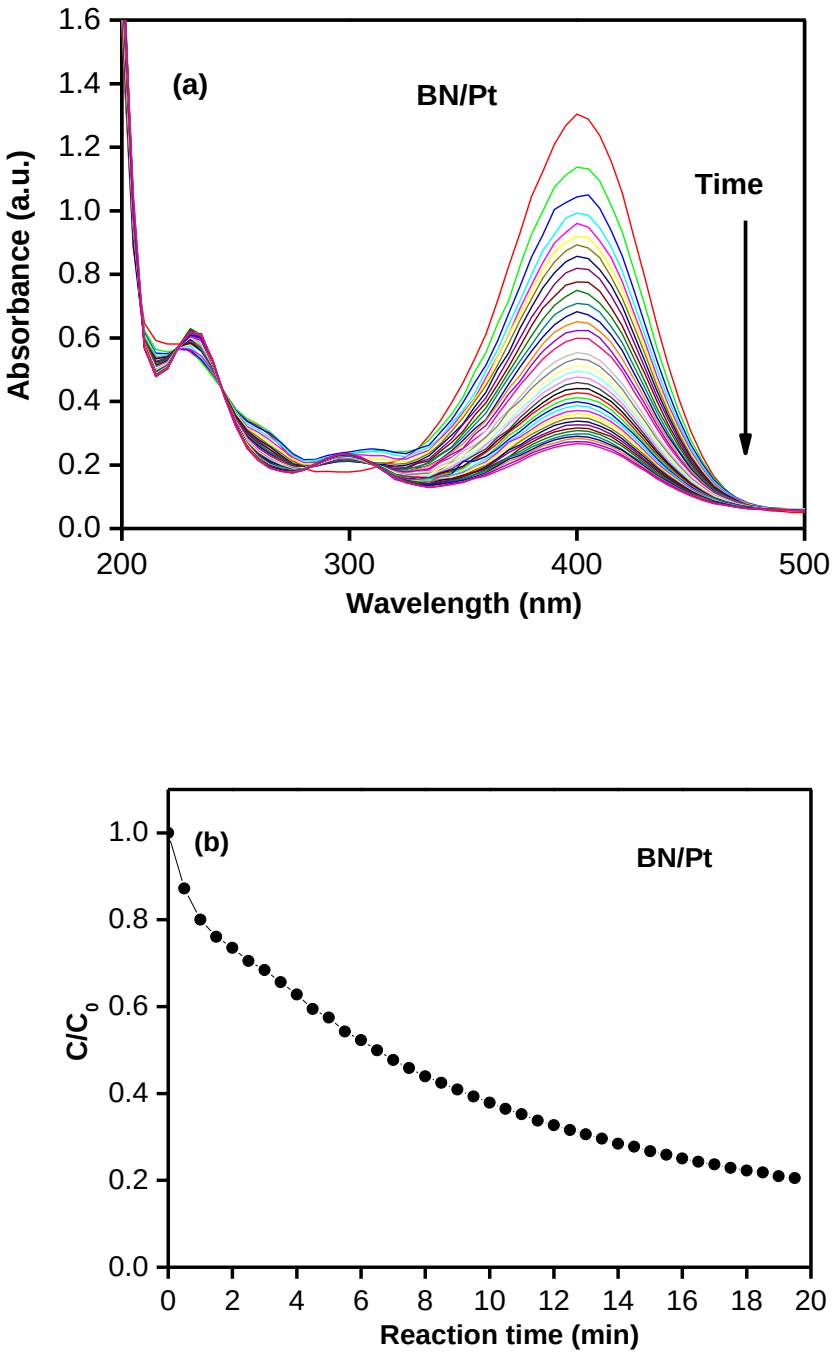

Fig. S6 (a) Evolution of UV-Vis spectra during p-nitrophenol degradation catalyzed by *h*-BN sheets/Pt composites. Each curve was taken successively at 30-sec intervals. (b) Conversion of p-nitrophenol as a function of reaction time.