

Phase transition behavior of NaCrO₂ during sodium extraction studied by synchrotron-based X-ray diffraction and absorption

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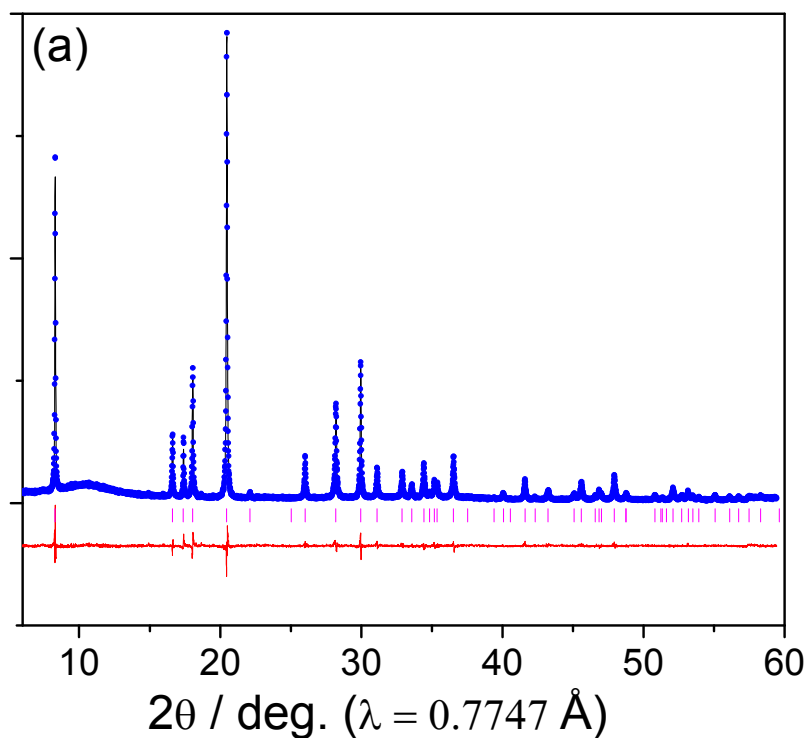


Fig. S1 Synchrotron diffraction pattern of NaCrO₂ and Rietveld refinement of its structure. The experimental (blue circles) and the calculated (black line) X-ray diffraction patterns are in good agreement with the smooth difference line (red line). The space group is R-3m. The calculated lattice parameters are: a=b=2.9768(6) Å, c= 16.9895(3) Å.

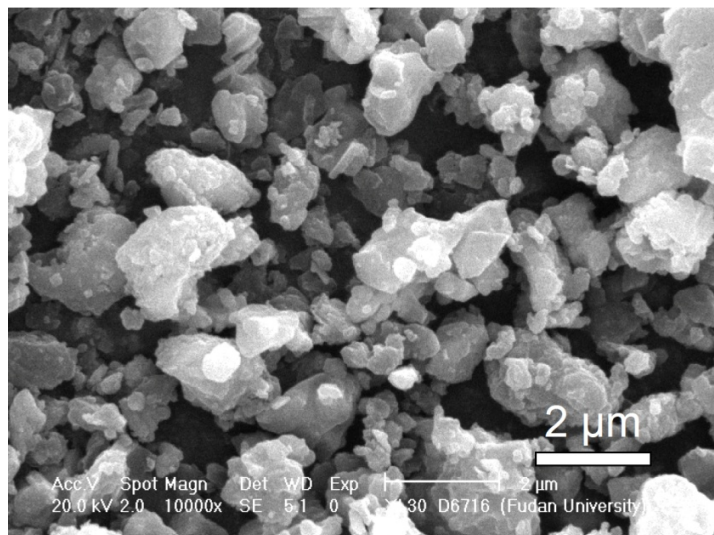


Fig. S2 SEM image of the pristine NaCrO₂ powder.

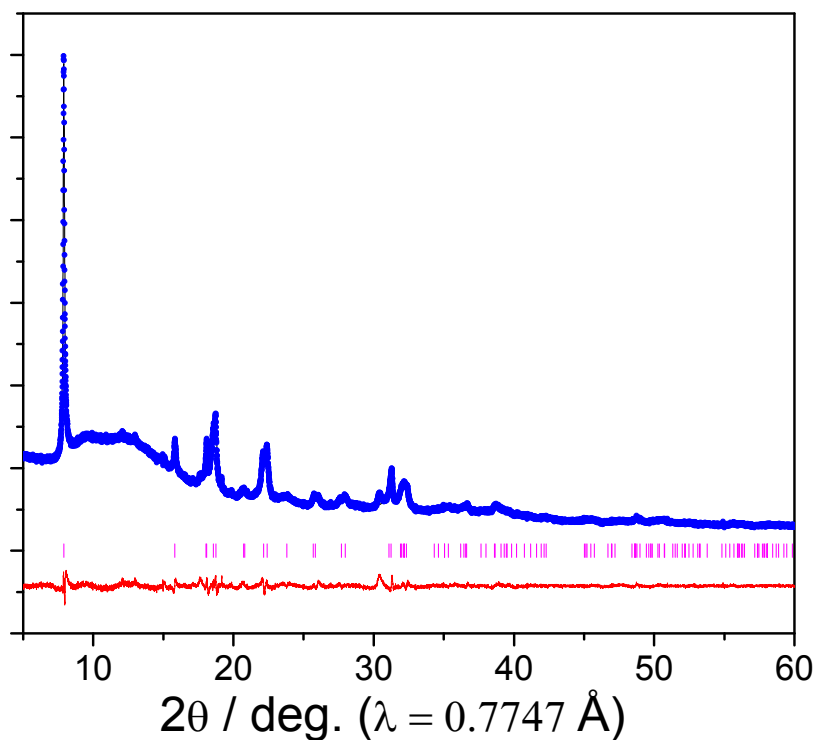


Fig. S3 *Ex-situ* synchrotron x-ray diffraction pattern of the fully charged NaCrO₂ (\sim Na_{0.5}CrO₂) and Rietveld refinement of its structure. The experimental (blue circles) and the calculated (black line) X-ray diffraction patterns are in good agreement with the smooth difference line (red line). It was indexed with a monoclinic cell (space group: C2/m) with the following parameters: $a_M (\sim \sqrt{3} a_R) = 5.0178(7)$ Å, $b_M (\sim b_R) = 2.8828(3)$, $c_M (\sim c_R/3\sin\beta) = 5.9027(5)$ Å, $\beta = 106.97^\circ$.

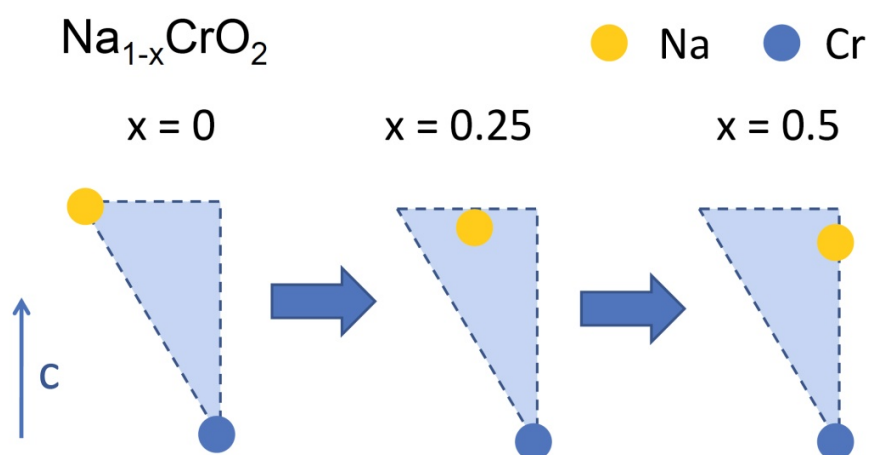


Fig. S4 Schematic for the relative position between Cr and Na ions in $\text{Na}_{1-x}\text{CrO}_2$ at different charge state ($x=0, 0.25, 0.5$). It corresponds to the Na coordination evolution from octahedral to pseudo-tetrahedral, then to trigonal-prismatic.