

SUPPORTING INFORMATION

Highly Dispersed Platinum Nanoparticles Generated in Viologen Micelles with Highly Catalytic Activity and Stability

Gui-Qi Gao, Ling Lin, Cong-Min Fan, Qing Zhu, Rui-Xia Wang, and An-Wu Xu*

Division of Nanomaterials and Chemistry, Hefei National Laboratory for Physical Sciences at Microscale, Department of Chemistry, University of Science and Technology of China, Hefei 230026, P. R. China.

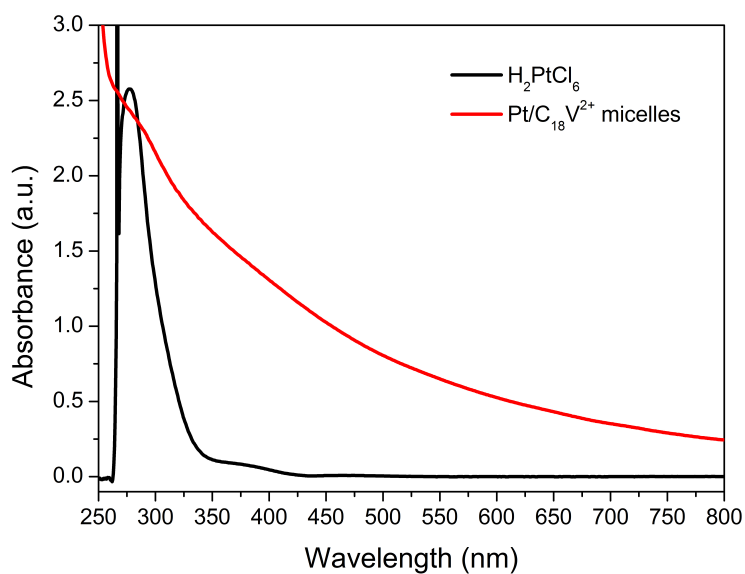


Fig. S1. UV-vis absorption spectra of H_2PtCl_6 and $\text{Pt/C}_{18}\text{V}^{2+}$ micelle solution.

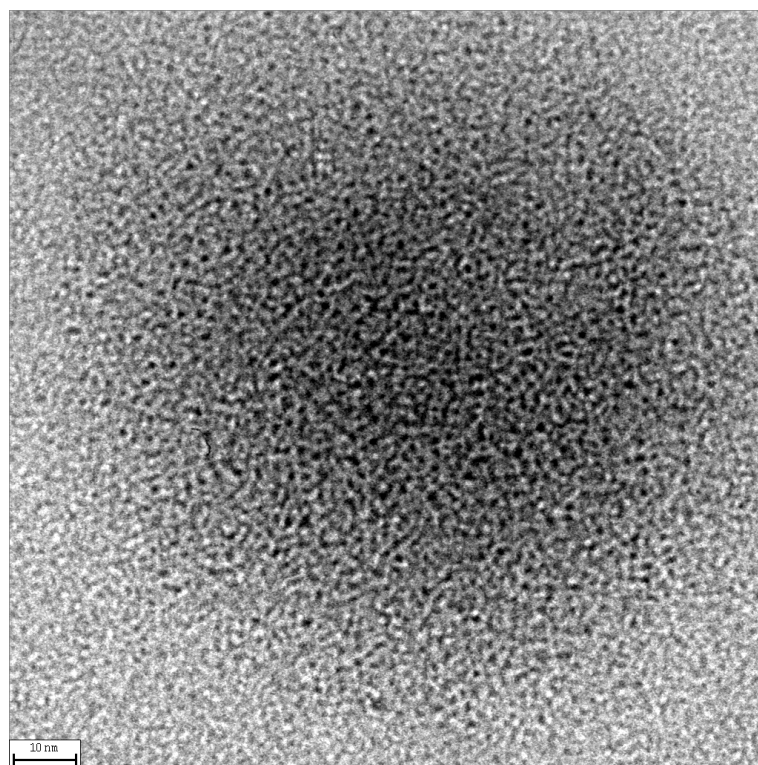


Fig. S2. HRTEM image of the as-synthesized Pt NPs/viologen micellar nanocomposites. The molar ratio of $C_{18}V^{2+}/H_2PtCl_6 = 0.60$, the volume ratio of DMF/ $H_2O = 20\%$.

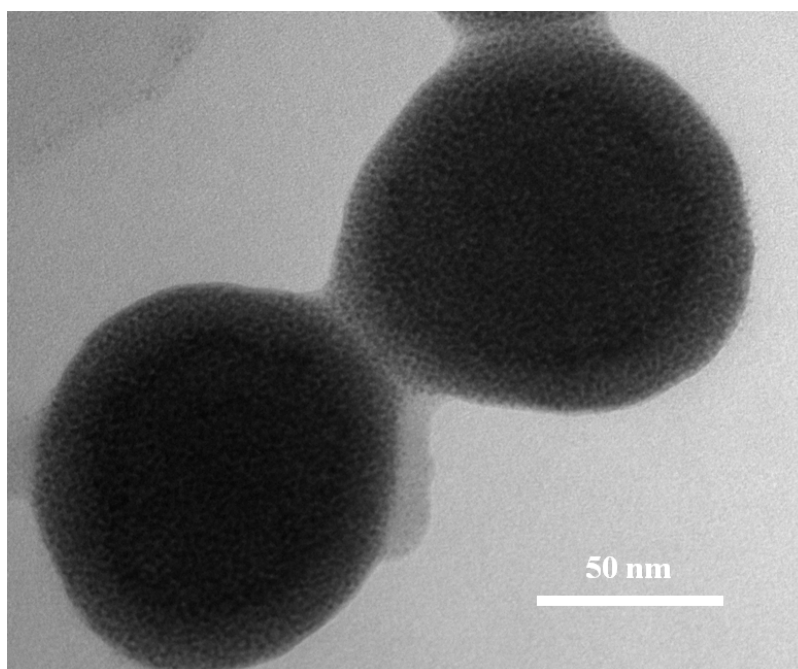


Fig. S3. HRTEM image of the as-synthesized Pt NPs/viologen micellar nanocomposites. The molar ratio of $C_{18}V^{2+}/H_2PtCl_6 = 0.30$, the volume ratio of DMF/ $H_2O = 50\%$.

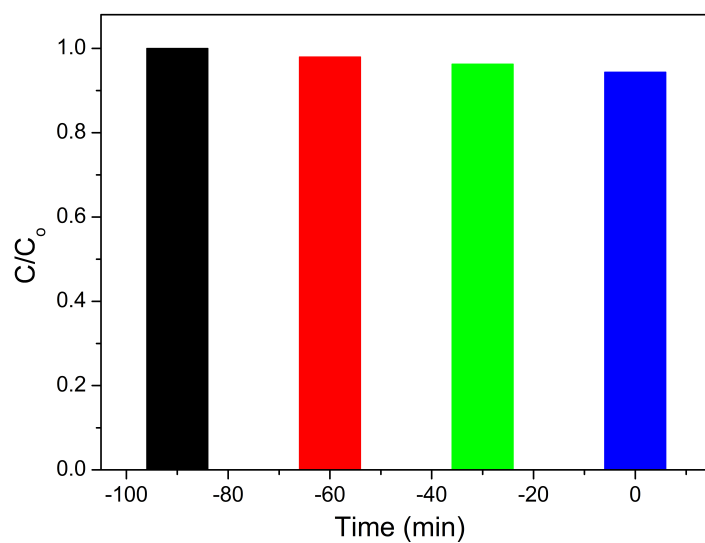


Fig. S4. 4-NP saturation adsorption on Pt/C₁₈V²⁺ micellar catalyst before adding NaBH₄.

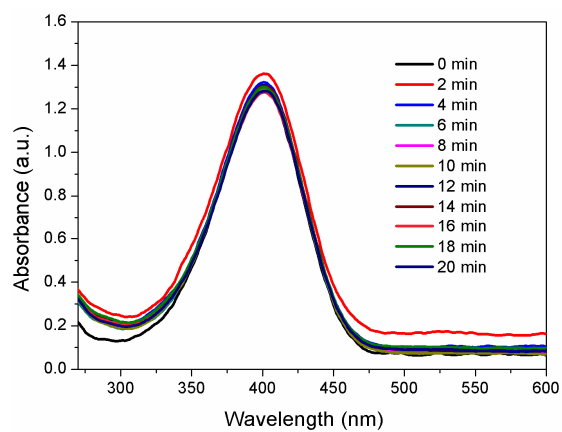


Fig. S5. Time-dependent UV-vis absorbance of 4-NP catalyzed by C₁₈V²⁺ at room temperature. The time difference between two neighbor curves is 2 min.

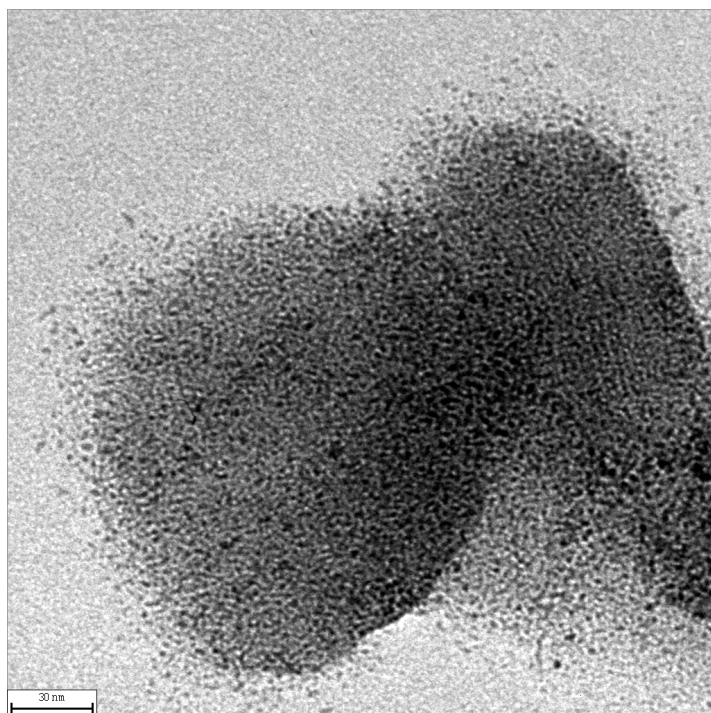


Fig. S6. TEM image of Pt/C₁₈V²⁺ micelles after five cycles of catalytic reaction.

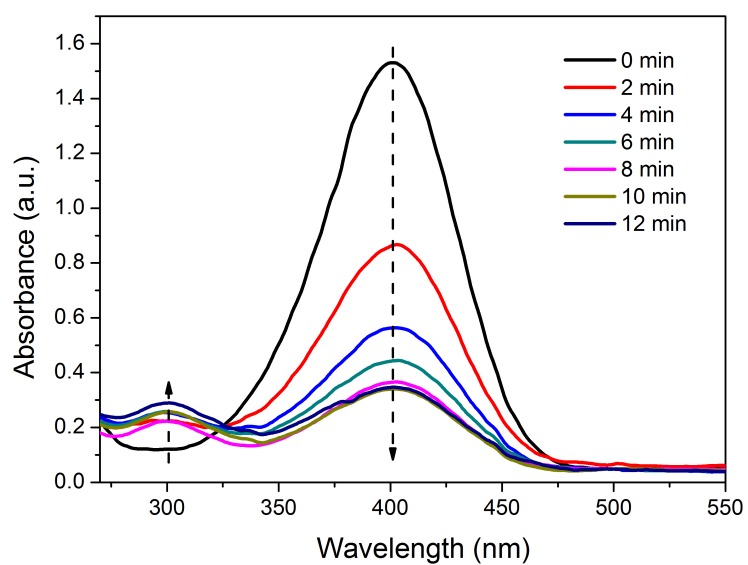


Fig. S7. Time-dependent UV-vis absorbance of 4-NP catalyzed by Pt/C₁₈V²⁺ micelles kept in ethanol for one month.