

Supporting Information  
For

**Tubular graphitic-C<sub>3</sub>N<sub>4</sub>: A Prospective material for energy storage  
and green photocatalyst**

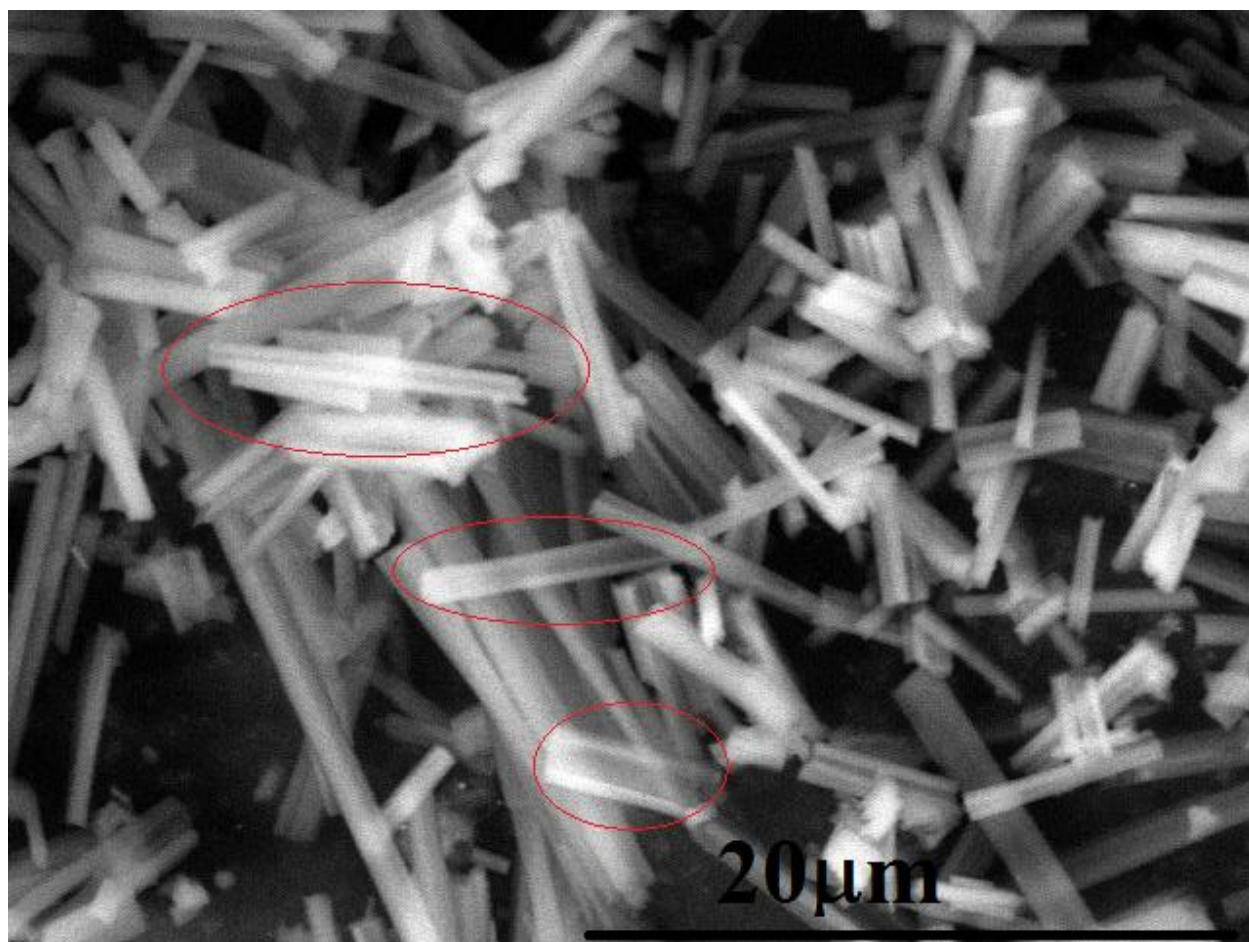


Figure S1. SEM image of tubular g-C<sub>3</sub>N<sub>4</sub>.

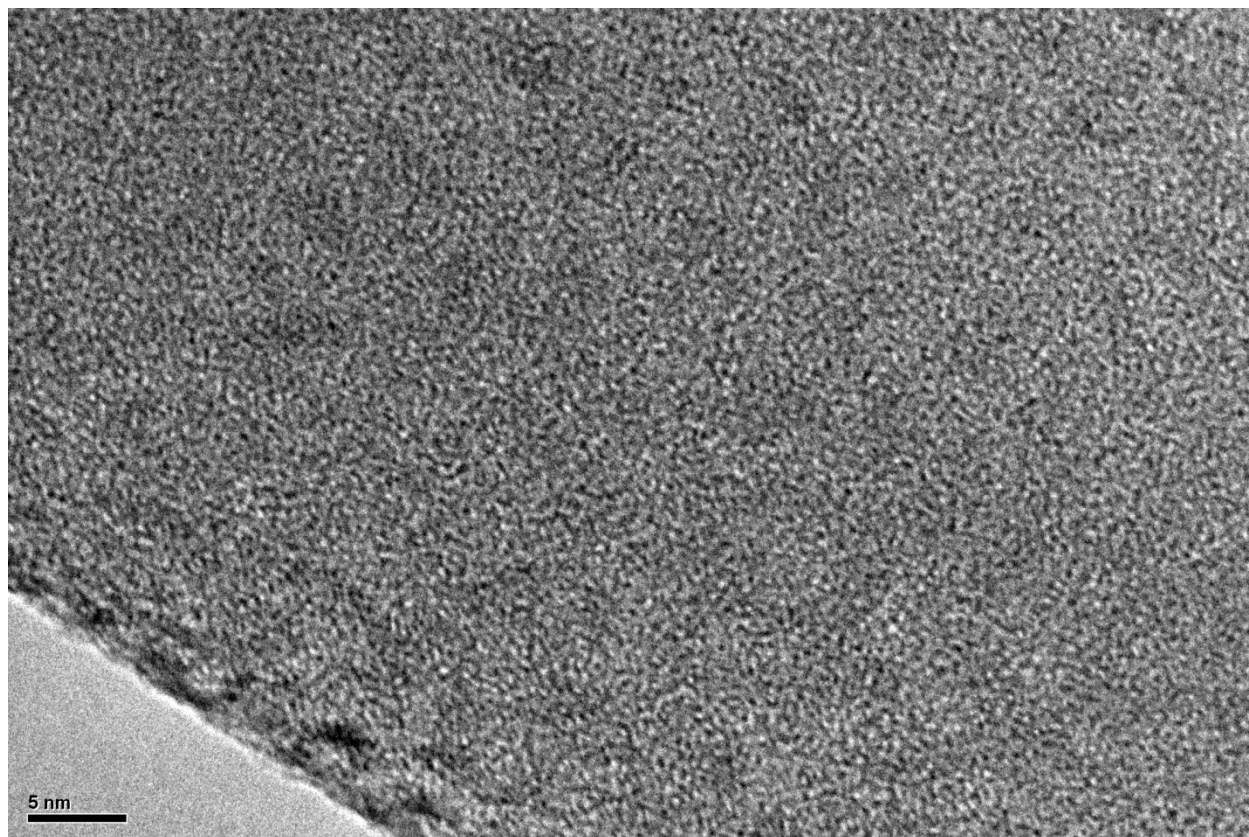


Figure S2. HRTEM image of tubular g-C<sub>3</sub>N<sub>4</sub>.

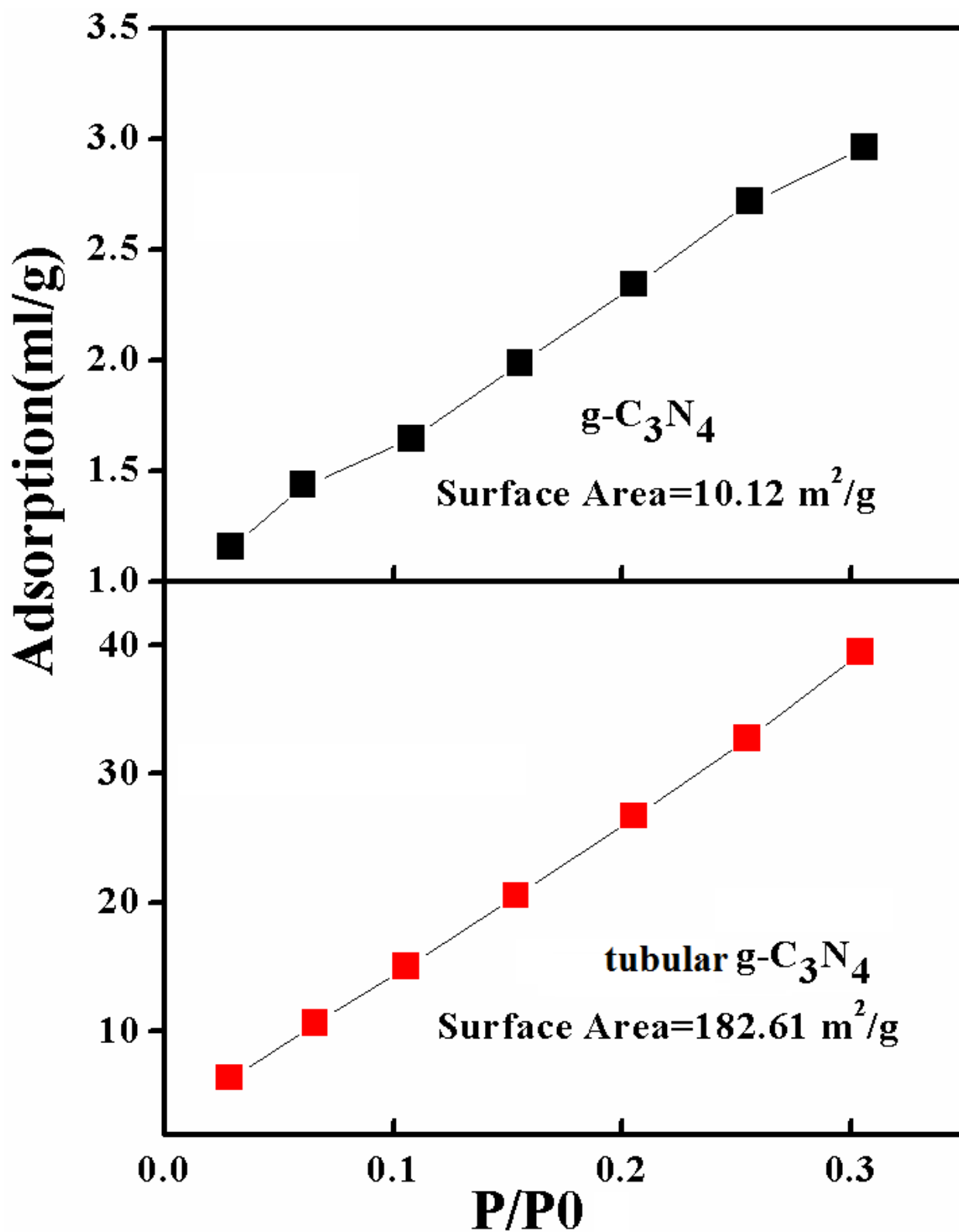


Figure S3. Nitrogen- adsorption curves of  $g-C_3N_4$  and tubular  $g-C_3N_4$  for BET surface area.

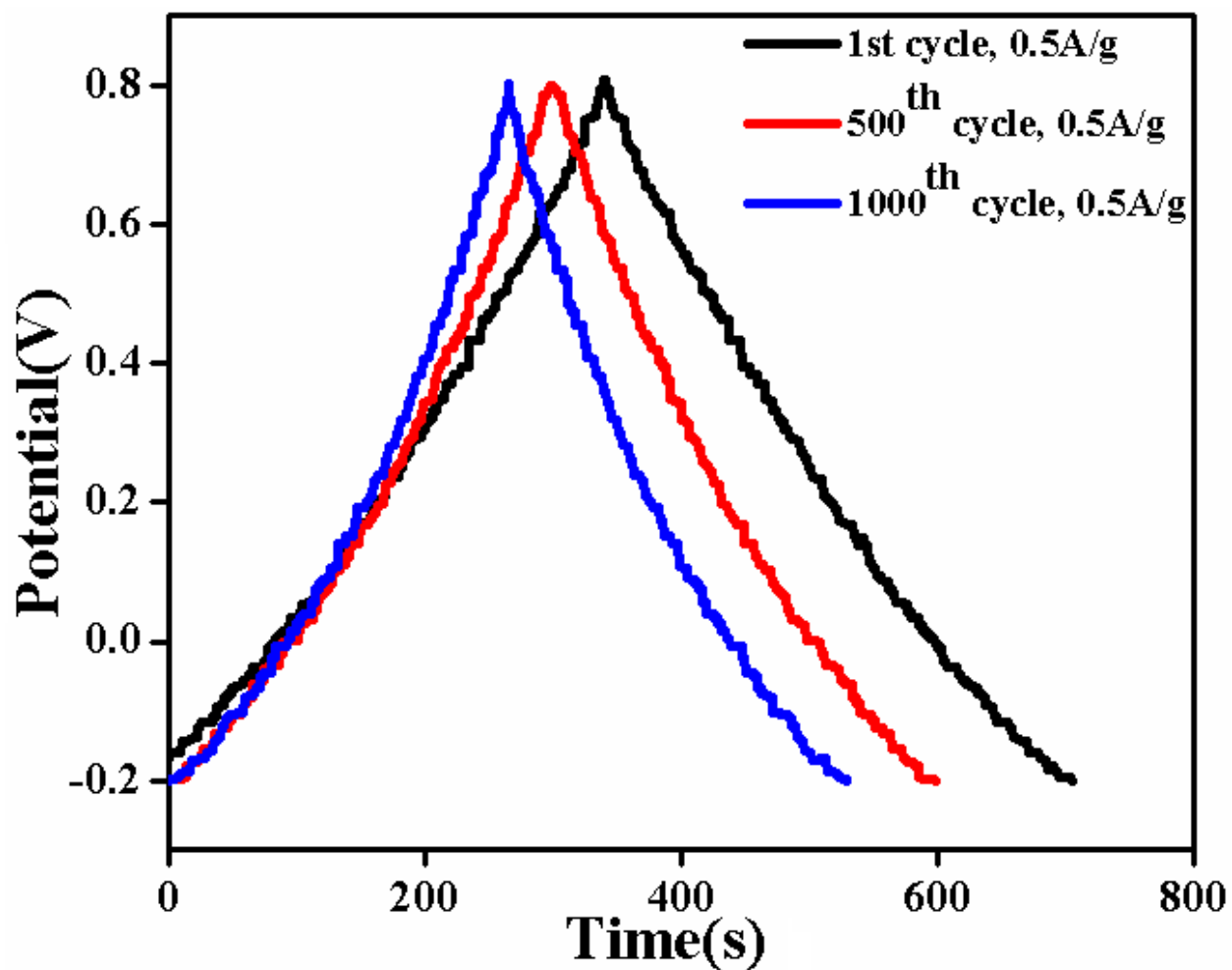


Figure S4. 1<sup>st</sup>, 500<sup>th</sup> and 1000<sup>th</sup> charging - discharging curves of g-C<sub>3</sub>N<sub>4</sub> within a potential window of -0.2 –0.8 V for current density 0.5 A/g.

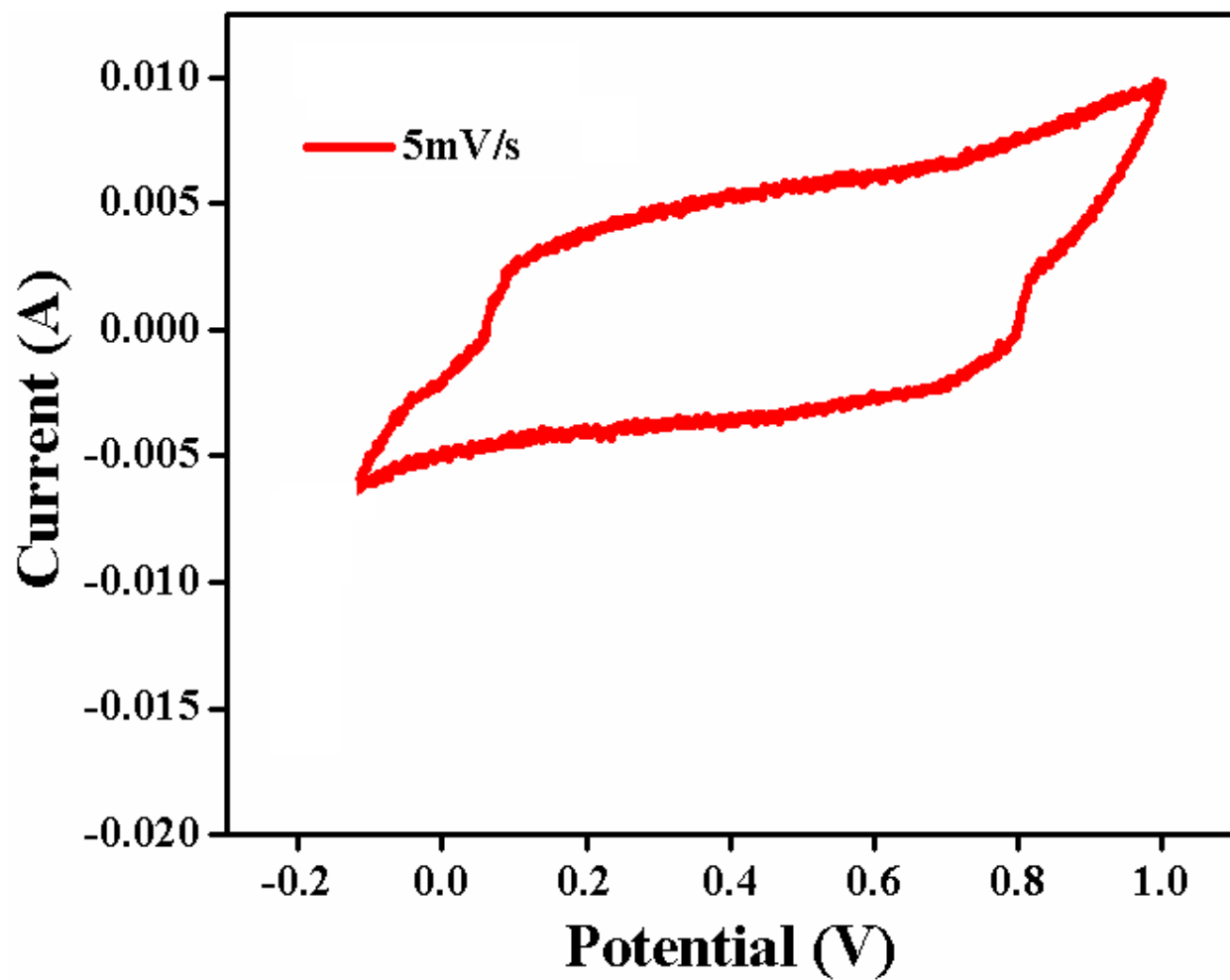


Figure S5. C-V curve of tubular g-C<sub>3</sub>N<sub>4</sub> within a potential window of -0.2 –1.0V using a three-electrode cell at a scan rate of 5mV/s.

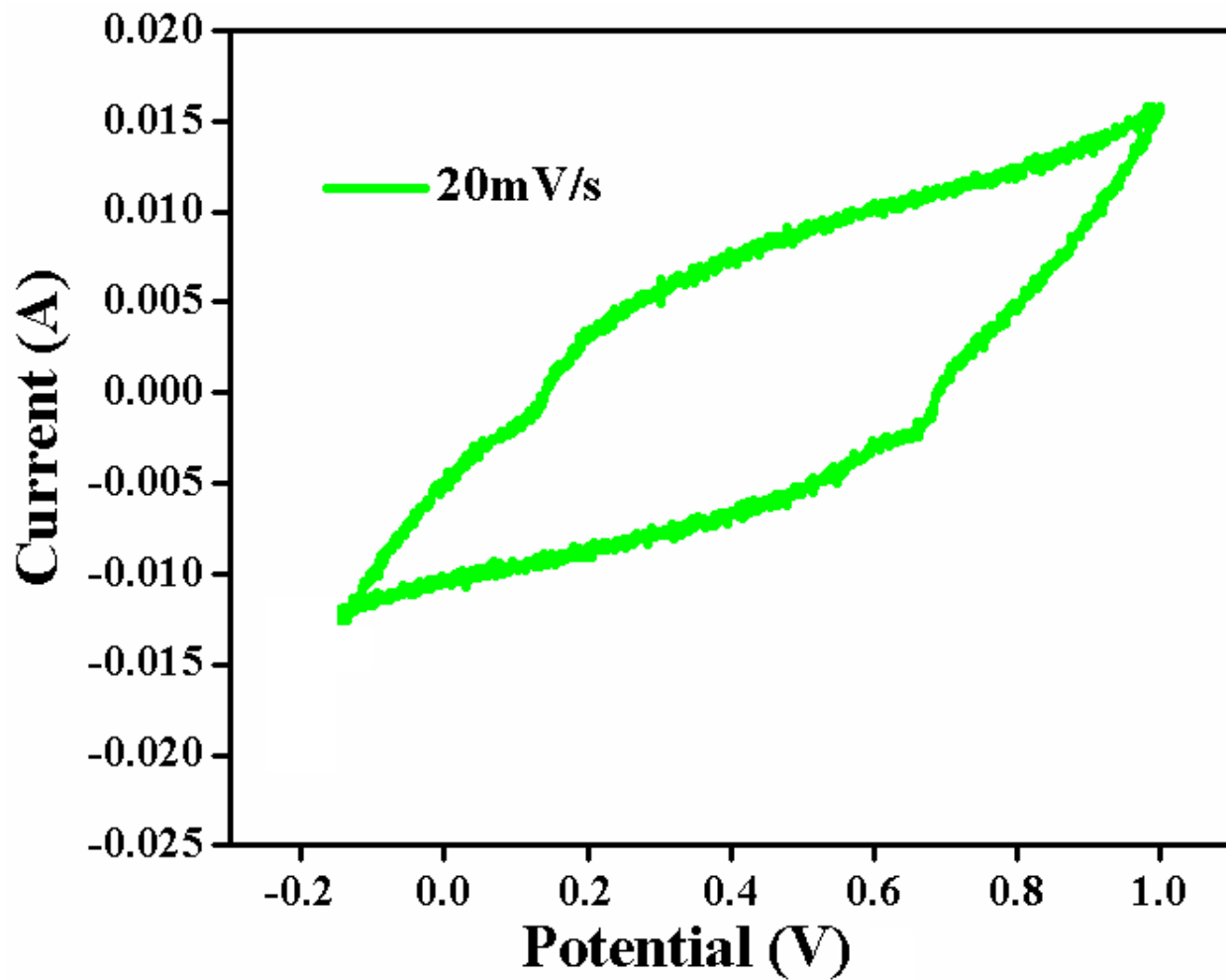


Figure S6. C-V curve of tubular g-C<sub>3</sub>N<sub>4</sub> within a potential window of -0.2 –1.0V using a three-electrode cell at a scan rate of 20mV/s.

**Table S1.** Specific capacitance comparison of the best-performing nitrogen-doped carbon materials in the literature.

Materials	Electrolyte[L <sup>-1</sup> ]	Capacitance [Fg <sup>-1</sup> ]	Current Density (Ag <sup>-1</sup> )	Reference
N-Enriched Nanocarbons	1M H <sub>2</sub> SO <sub>4</sub>	210	0.1	1
N-enriched carbon	1 M H <sub>2</sub> SO <sub>4</sub>	201	0.5	2
MWCNTs	4 M H <sub>2</sub> SO <sub>4</sub>	62	0.2	3
CNTs	6 M KOH	198	0.05	4
N-enriched carbon	1 M H <sub>2</sub> SO <sub>4</sub>	201	0.5	5
N-enriched carbon	1 M TEABF <sub>4</sub>	52	1mA /cm <sup>2</sup>	6
O-rich carbons	1 MH <sub>2</sub> SO <sub>4</sub>	198	1	7
N-carbon	5 M H <sub>2</sub> SO <sub>4</sub>	211	1	8
CNTs/ N-carbon	1 M H <sub>2</sub> SO <sub>4</sub>	100	0.2	9
N-Doped Graphene	6 M KOH	246	1	10
N-doped carbonnanocage	6 M KOH	248	1	11
N-Doped Carbon	6 M KOH	202	1	12
Porous 3D graphene	1M TEABF <sub>4</sub>	231	1	13
Acrylonitrile–propylene	1M H <sub>2</sub> SO <sub>4</sub>	340	0.2	14
Melamine	6 M KOH	280		14
Ethylene diamine–carbon tetrachloride	6 M KOH	318		14
Urea–brown coal	6 M KOH	341		14
Tubular g-C <sub>3</sub> N <sub>4</sub>	6 M KOH	233	0.2	Our work



**Table S2.** First order rate constant comparison between different reported results.

Dye	Photocatalysts	k(min <sup>-1</sup> )	Reference	Our Work
MO	Born doped g-C <sub>3</sub> N <sub>4</sub>	0.004	15	0.0067
	g-C <sub>3</sub> N <sub>4</sub> at 600 <sup>0</sup> C	0.003	15	
	g-C <sub>3</sub> N <sub>4</sub> at 580 <sup>0</sup> C	0.004	15	
	g-C <sub>3</sub> N <sub>4</sub>	0.005	16	
MB	TiO <sub>2</sub> nanotubes	0.024	17	0.021
	TiO <sub>2</sub>	0.0012	18	
	g-C <sub>3</sub> N <sub>4</sub> nanoplates	0.0016	19	
	g-C <sub>3</sub> N <sub>4</sub> nanorods	0.002	19	

## Supplementary References

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