

Supporting Information

Understanding the fast lithium storage performance of hydrogenated TiO₂ nanoparticles

Yong Yan,^{a+} Bo Hao,^{b+} Dong Wang,^{a} Ge Chen,^{b*} Eric Markweg,^c Arne Albrecht,^d and Peter Schaaf^a*

[a] Chair materials for Electronics, Institute of Materials Engineering and Institute of Micro- and Nanotechnologies MarcoNano[®], Ilmenau University of Technology, Gustav-Kirchhoff-Str. 5, 98693 Ilmenau, Germany

[b] College of Environmental & Energy Engineering, Beijing University of Technology, Pingleyuan 100, 100124, Beijing, P.R. China

[c] Chair micromechanical Systems, Institute of Micro- and Nanotechnologies MarcoNano[®], Ilmenau University of Technology, Max-Planck-Ring 12, 98693 Ilmenau, Germany

[d] Center for Micro- and Nanotechnologies MacroNano[®], Ilmenau University of Technology, Gustav-Kirchhoff-Str. 7, 98693 Ilmenau, Germany

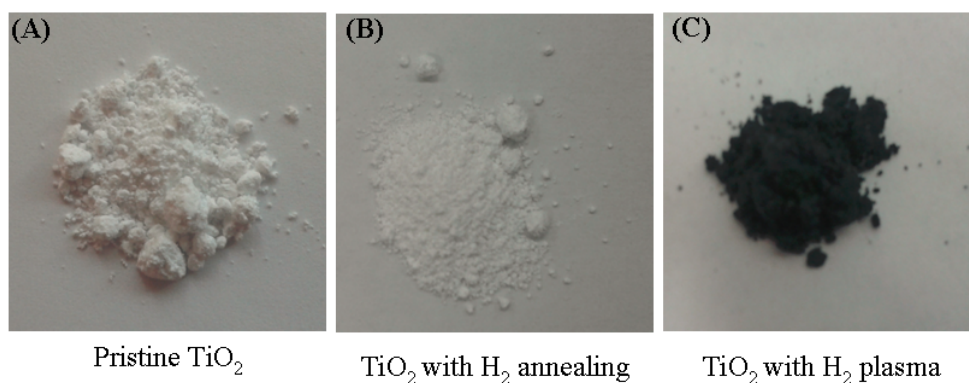


Figure S1. The photographs of TiO₂ before and after hydrogenation: (A) Pristine TiO₂. (B) TiO₂ after thermal annealing under H₂ atmosphere without plasma. (C) TiO₂ after H₂ plasma treatment.

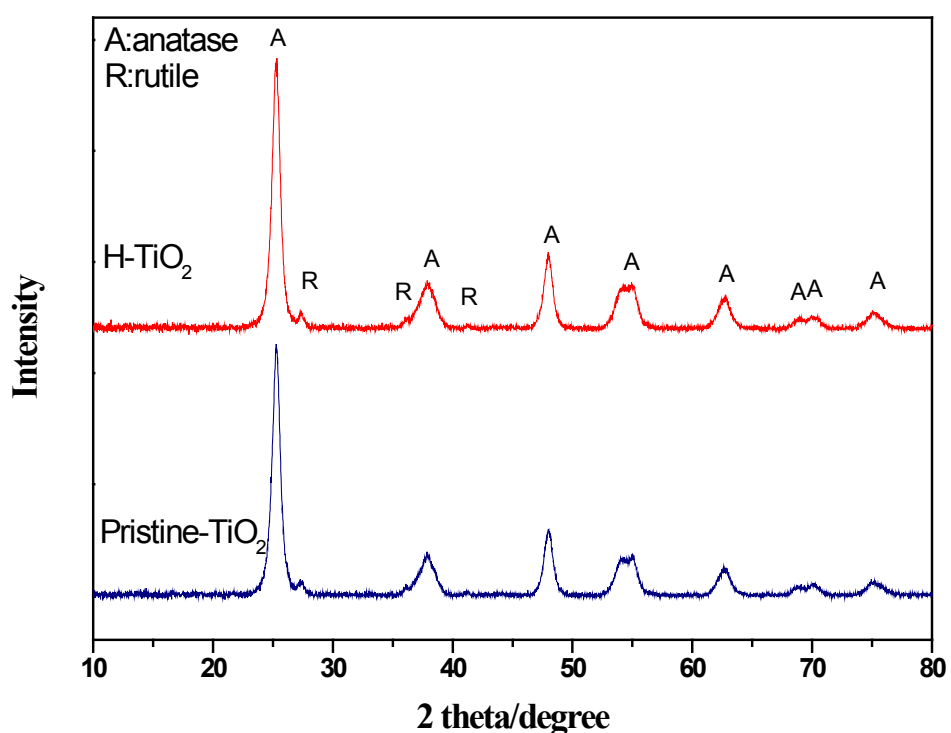


Figure S2. XRD pattern of pristine- and H-TiO₂.

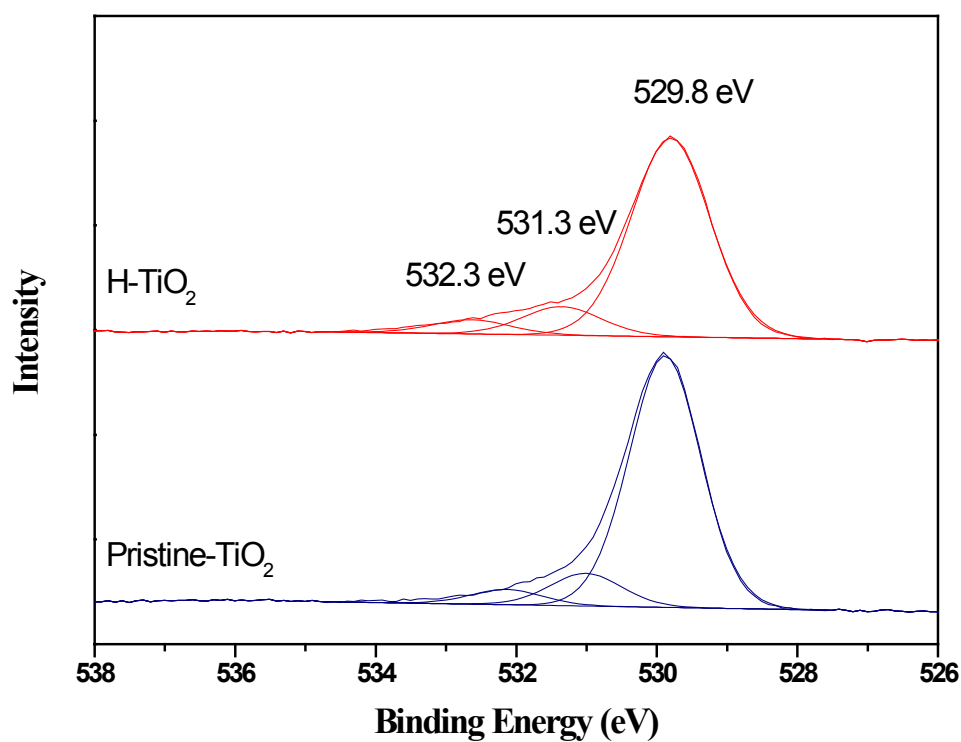


Figure S3. XPS O 1s core level spectrum of pristine- and H-TiO₂.

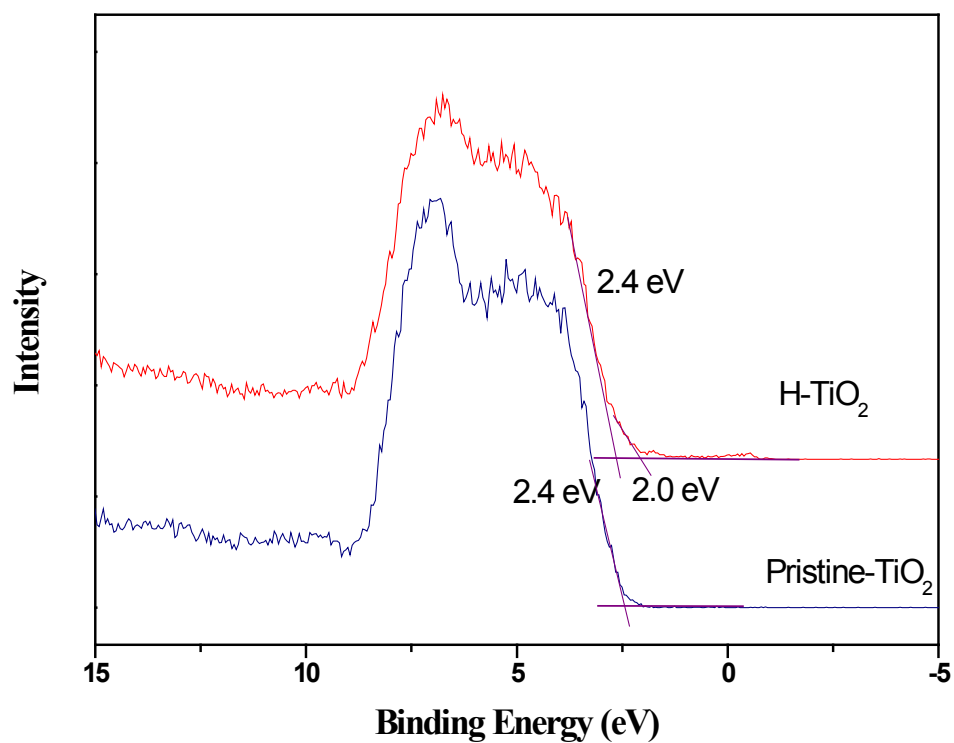


Figure S4. XPS valence band spectra of pristine- and H-TiO₂.

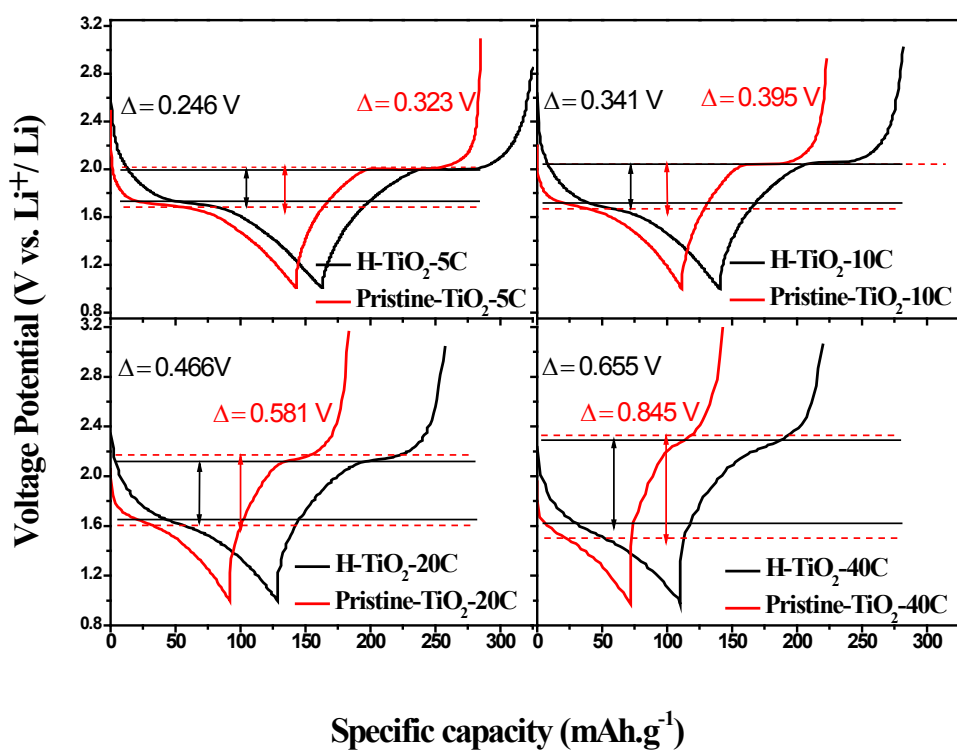


Figure S5. Polarization of ΔE versus rate plots of pristine- and H-TiO₂ electrodes.

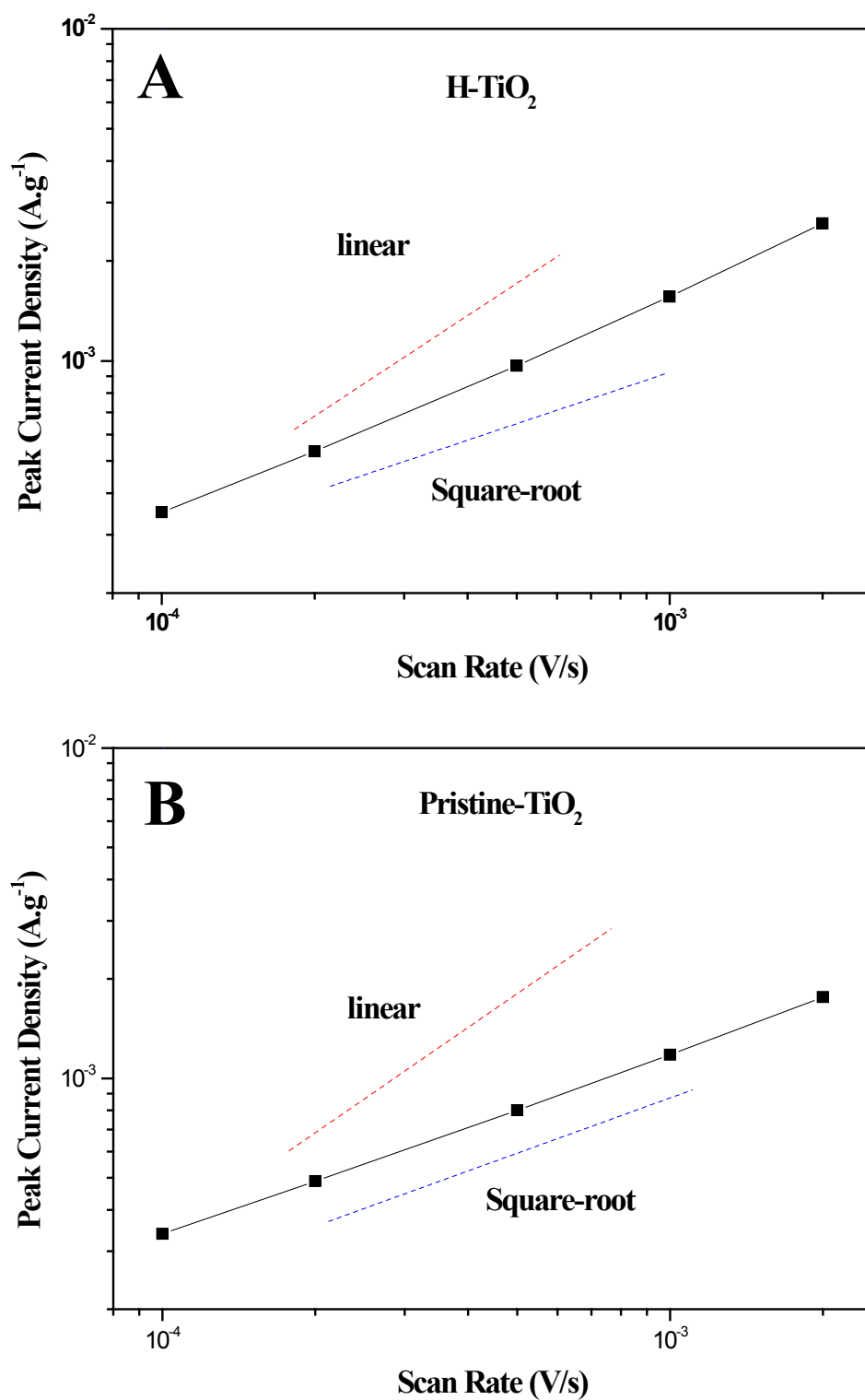


Figure S6. The peak discharge current of pristine- and H-TiO₂ electrodes measured at various scan rates. (A) H-TiO₂ electrode. (B) Pristine-TiO₂ electrode.

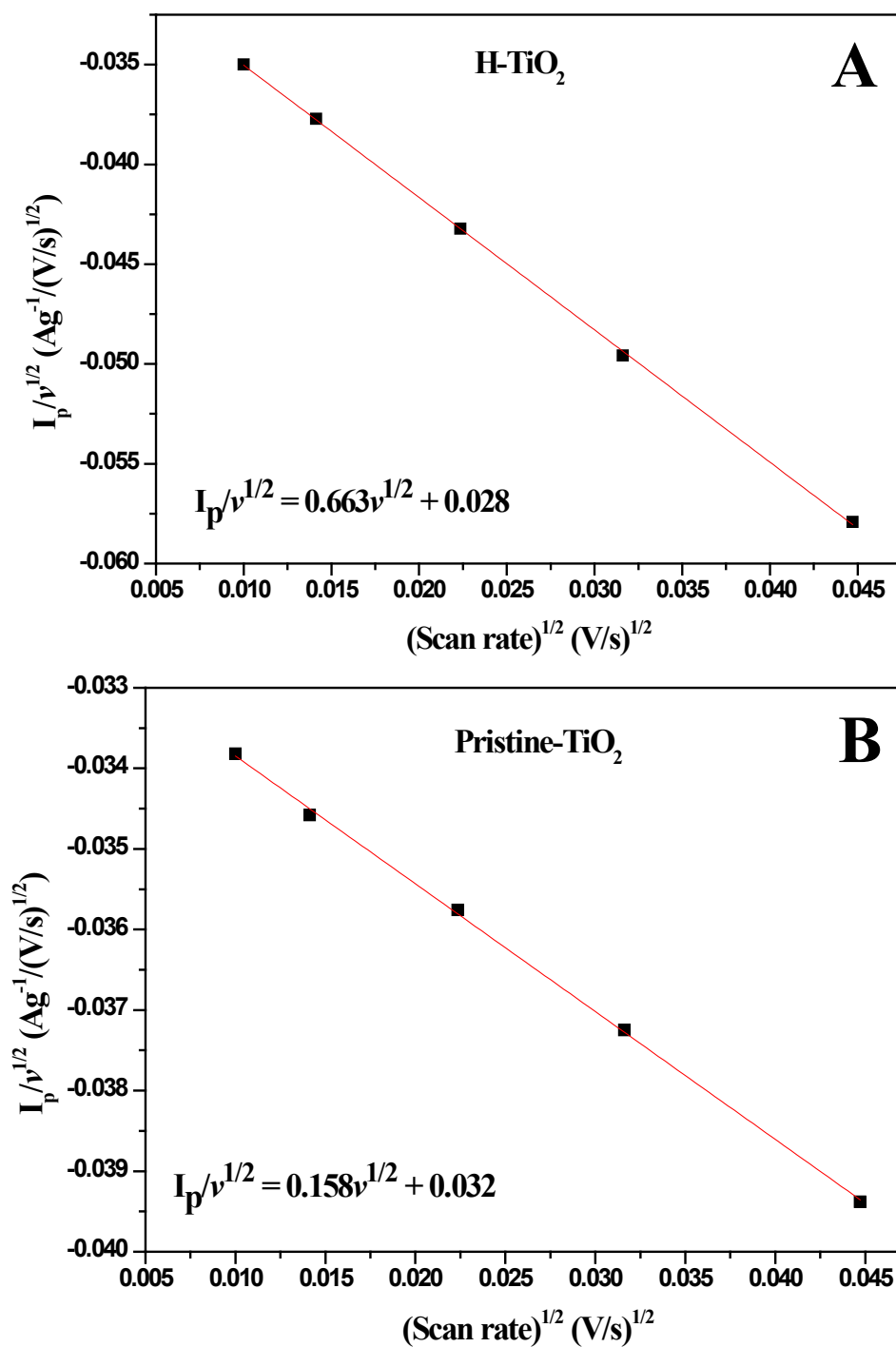


Figure S7. The calculated C_1 and C_2 for two samples using Eq. (2) that correspond to the slope and the y-axis intercept point, respectively. (A) H-TiO₂. (B) Pristine-TiO₂.