

Supporting information:

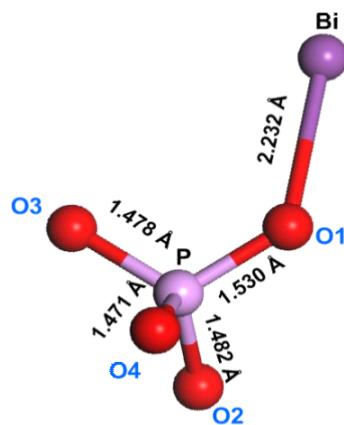


Figure S1. The ball-and-stick model of BiPO_4 coordination

As seen from **Figure S1**, the bond length between oxygen (O) atom and phosphorus (P) atom or bismuth (Bi) atom is not equal in BiPO_4 . Based on the bond length (representing bond energy) between each two atoms, O atom maybe removed in turn, from O1, O2, O3 to O4, even the PO_4^{3-} is totally detached from BiPO_4 , generating oxygen vacancies with different number and extent.

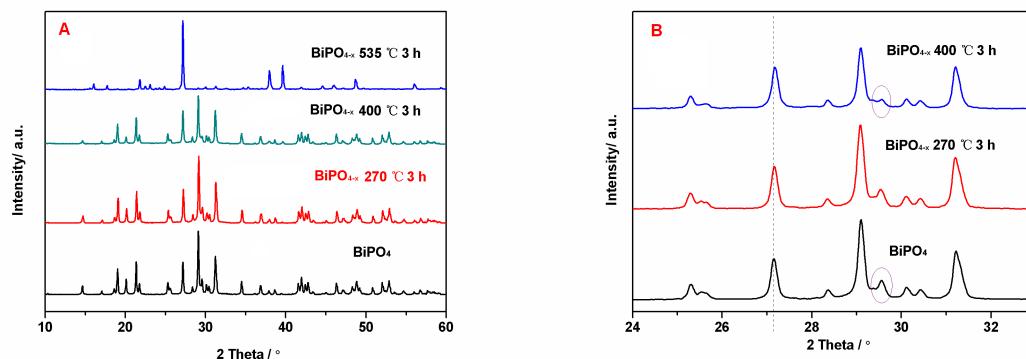


Figure S2. (A) XRD patterns of BiPO₄ and BiPO_{4-x} nanorod treated at 270, 400 and 535 °C for 3 h; (B) The enlarged XRD

patterns of BiPO₄ and BiPO_{4-x} samples at local.

From the X-ray diffraction (XRD) patterns of pure BiPO₄ and BiPO_{4-x} samples **Figure S2**, no phase transformation or any impurity is observed for BiPO₄ after H₂ reduction at 270 °C for 3 h, so there most oxygen atoms are removed from BiPO₄ surface, generating surface oxygen vacancies. However, after H₂ reduction at 400 °C for 3 h, the XRD peak of vacancy BiPO_{4-x} decrease, indicating that crystallinity decreased, and the peak at $2\theta = 27.1^\circ$ (200) moves to big angle slightly, indicating that the lattice spacing d decreases, which may be resulted from the loss of lattice oxygen and the formation of bulk oxygen vacancy, and the peak intensity. And after 535 °C 3 h H₂ reduction treated, BiPO₄ losses PO₄³⁻ totally becomes metal Bi (**Figure S2A**).

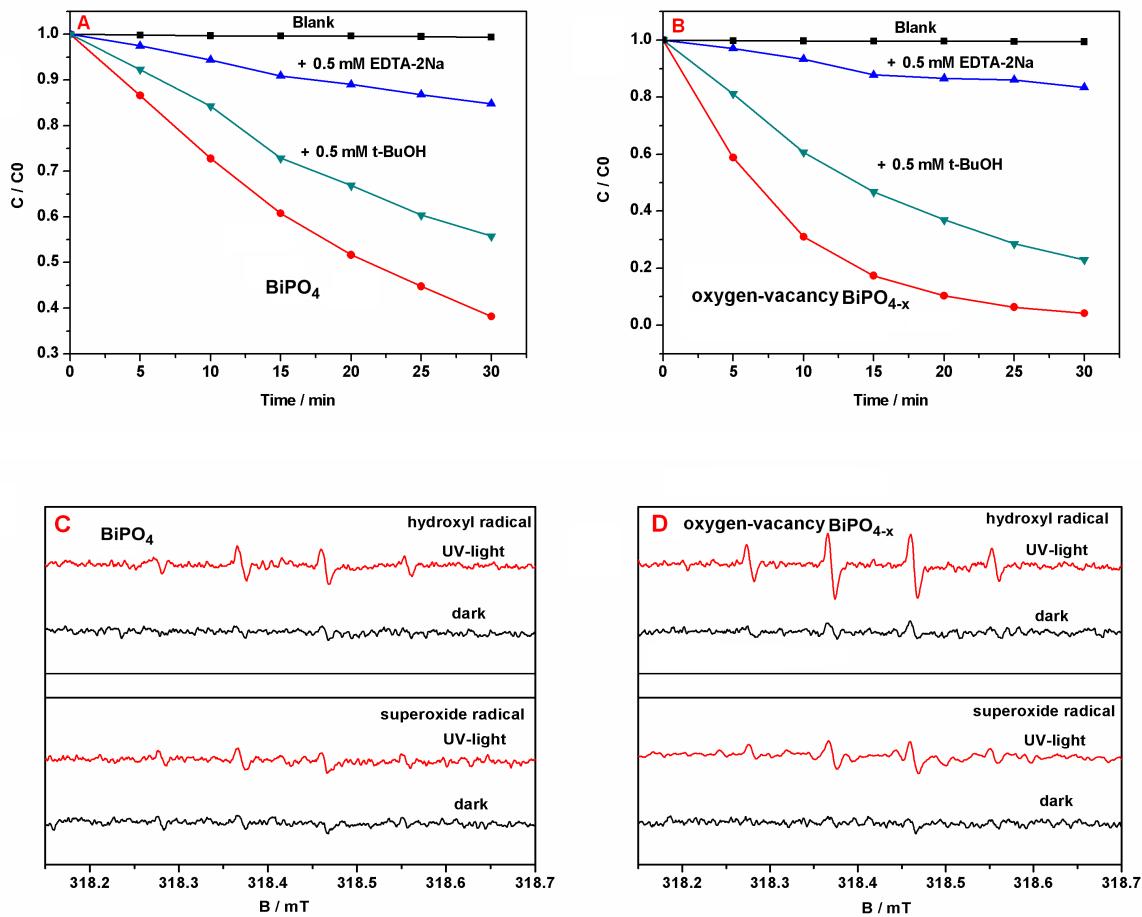


Figure S3 (A)(B) The plots of photoinduced carriers trapping in the system of photodegradation on MB over BiPO_4 and oxygen–vacancy BiPO_{4-x} , under UV light ($\lambda = 254 \text{ nm}$), respectively; **(C)(D)** DMPO spin–trapping ESR spectra of BiPO_4 and oxygen–vacancy BiPO_{4-x} photocatalysts.

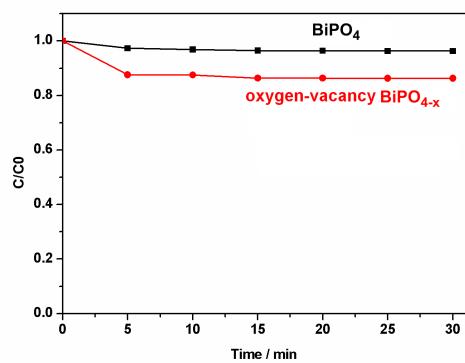


Figure S4. The adsorption of BiPO₄ and oxygen-vacancy BiPO_{4-x} on the degradation of MB solution, in dark.

As can be seen from **Figure S4**, compared pure BiPO₄, the adsorption capacity of oxygen-vacancy BiPO_{4-x} is slightly improved from about 3% for pure BiPO₄ to about 11% in dark. And after 10 min, the adsorption equilibrium achieved for both of them

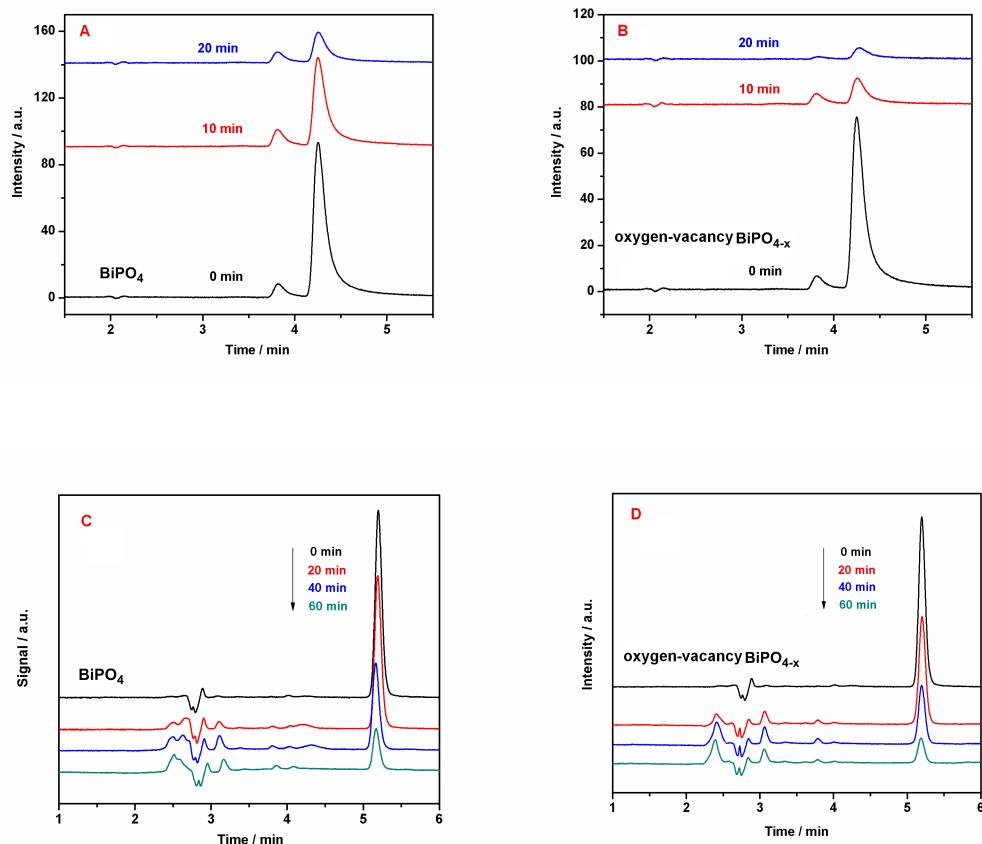


Figure S5. HPLC of intermediates at different irradiation intervals on MB (**A**)**(B)** and phenol (**C**)**(D)** by BiPO₄ and oxygen-vacancy BiPO_{4-x}, respectively.

The intermediates of MB and phenol over BiPO₄ and oxygen-vacancy BiPO_{4-x} were determined by HPLC (Lumtech) analysis with a UV detector at 663 nm and 270 nm, respectively.

To demonstrate the change and reveal some details of the reaction process of pure BiPO₄ and oxygen-vacancy BiPO_{4-x}, photodegradation intermediates distribution on MB and phenol at intervals during the photocatalytic process were monitored by HPLC, and the results are displayed in **Figure S5**. From **Figure S5A and B**, it can be found that the degradation intermediates on MB by oxygen-vacancy BiPO_{4-x} hardly any changes but the extent and rate greatly increased, compared with that of pure BiPO₄. In addition, **Figure S5C**

and D shows that the degradation intermediates on phenol over two samples only little change in concentration, and the intermediates distribution is almost no different. Thses indicate that the photodegradation on MB or phenol by oxygen-vacancy BiPO_{4-x} are both enhanced, compared with pure BiPO_4 , and photodegradation intermediates are not changed.

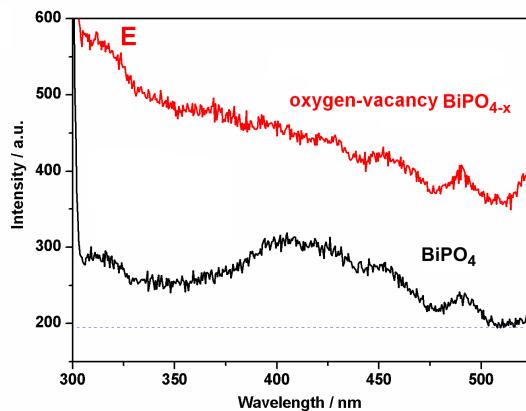


Figure S6. The PL (270 nm excitation) spectra of BiPO₄ and oxygen–vacancy BiPO_{4-x}.

The room temperature photoluminescence (PL) spectra of BiPO₄ and oxygen-vacancy BiPO_{4-x} samples were investigated utilizing the PerkinElmer LS55 spectrophotometer equipped with xenon (Xe) lamp with an excitation wavelength of 277 nm.

Photoluminescence spectra (PL) can reveal surface structure characteristics of nanometer semiconductor materials, such as surface oxygen vacancies and defects etc., and surface oxygen vacancy is positively related with the PL signal.¹⁻³ The room temperature photoluminescence (PL) spectra of BiPO₄ and oxygen-vacancy BiPO_{4-x} at wavelength 270 nm excitation were investigated, as shown in Fig.S6. Compared with pure BiPO₄, the PL peak of oxygen–vacancy BiPO_{4-x} is obviously enhanced in the range of 300–520 nm, which is induced by oxygen vacancies. The larger the concentration of surface oxygen vacancy, the stronger the PL signal.^{2,3} Therefore, more surface oxygen vacancies are produced on oxygen-vacancy BiPO_{4-x} nanorod.

References

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- (2) Jing, L. Q.; Yuan, F. L.; Hou, H. G.; Xin, B. F.; Cai, W. M.; Fu, H. G. *Sci. in China Ser. B Chem.* **2005**, *48*, 25.
- (3) Jing, L. Q.; Qu, Y. C.; Wang, B. Q.; Li, S. D.; Jiang, B. J.; Yang, L. B.; Fu, W.; Fu, H. G.; Sun, J. Z. *Sol. Energy Mater. Sol. Cells* **2006**, *90*, 1773.