

SUPPORTING INFORMATION

Black mesoporous anatase TiO₂ nanoleaves: A high capacity and high rate anode for aqueous Al-ion batteries

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Summary: 20 Pages; 2 Tables; 26 Figures;

Table S1 XRD Rietveld refinement results and conductivity of the samples

Samples	Composition	a / nm	c / nm	Rp/%	Rwp/%	Rb/%	Conductivity/ Scm ⁻¹
commercial white anatase TiO ₂	TiO ₂	0.38007(3)	0.94898(4)	8.09	6.01	4.12	9.7×10 ⁻⁷
black anatase TiO ₂ nanoleaves	TiO _{1.976} (NH) _{0.024}	0.38124(2)	0.95099(1)	8.23	5.45	4.04	3.0×10 ⁻¹
commercial white anatase TiO ₂ after SPP treatment with NH ₄ Cl	TiO _{1.983} (NH) _{0.017}	0.38092(4)	0.95075(7)	8.75	5.89	4.07	9.9×10 ⁻²
white anatase TiO ₂ ¹⁸	TiO ₂	0.3887(7)	0.94894(7)	-	-	-	2.2×10 ⁻⁷
black anatase TiO ₂ ¹⁸	TiO _{1.974} (NH) _{0.026}	0.38038(3)	0.95148(7)	-	-	-	8.2×10 ⁻²

Table S2 Impedance parameters of the samples

Samples	Cycles	R_e / Ω	R_{sei} / Ω	R_{ct} / Ω	Z_w / Ω
Commercial white TiO ₂ anatase	no cycling	1.2	-	68.7	203
	the 1 st cycle	1.8	2.3	103.1	70.6
	the 10th cycle	2.6	2.7	72.0	47.5
	the 30 th cycle	3.9	3.2	79.2	55.5
	the 50 th cycle	4.2	3.6	85.3	60.3
Black anatase TiO ₂ nanoleaves	no cycling	1.1	-	24.7	65.2
	the 1 st cycle	1.7	4.8	30.8	20.4
	the 10th cycle	2.4	5.2	16.7	18.7
	the 30 th cycle	3.7	5.4	17.8	19.6
	the 50 th cycle	4.2	5.6	18.5	20.3

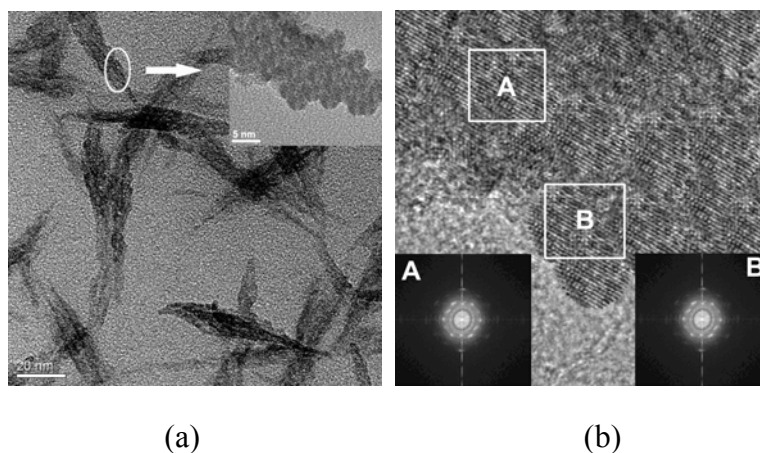


Fig.S1 (a) High magnification TEM image (the inset is the enlargement of the frame), and (b) the FFT patterns originated from the marked areas in the HRTEM image of the as-prepared black anatase TiO₂ nanoleaves.

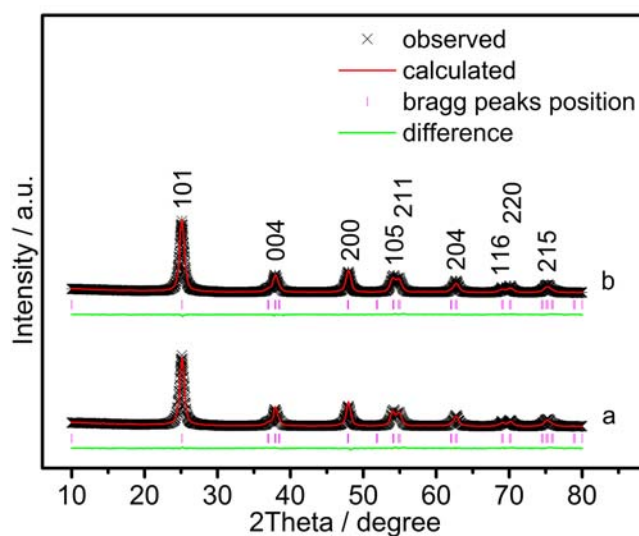


Fig.S2 XRD Rietveld refinement for the diffraction patterns of (a) the commercial white anatase TiO₂ and (b) the commercial white anatase TiO₂ after treated by SPP with NH₄Cl.

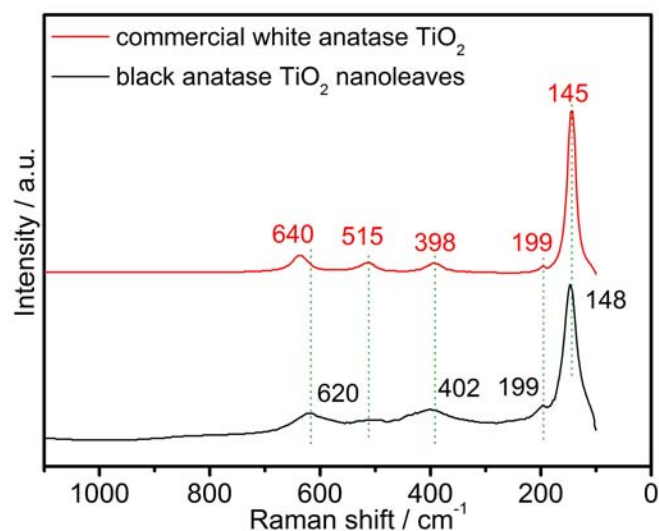


Fig.S3 Raman spectra of the commercial white anatase TiO_2 and black anatase TiO_2 nanoleaves.

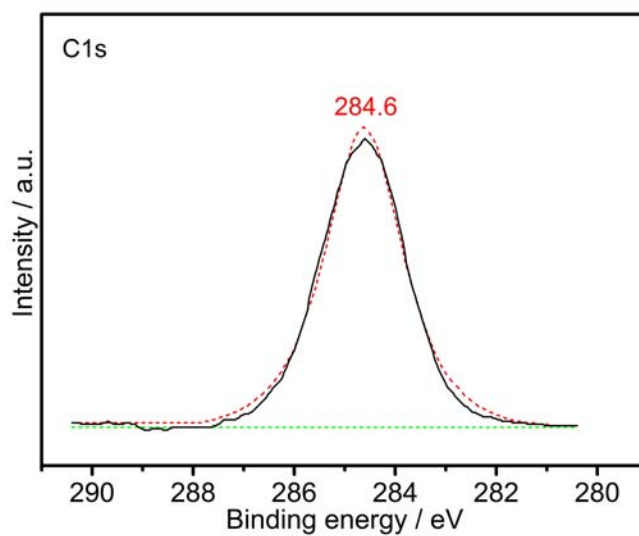


Fig.S4 C1s high-resolution XPS spectrum for the black anatase TiO_2 nanoleaves.

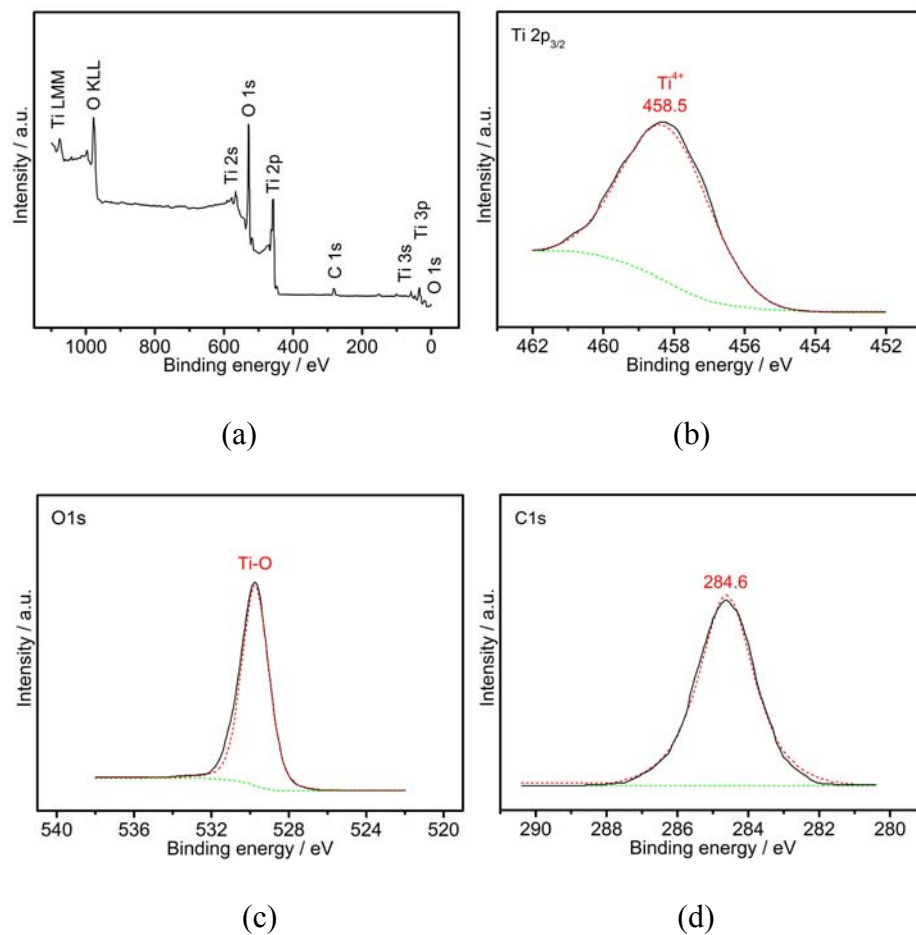


Fig.S5 XPS spectra for the commercial white anatase TiO₂: (a) survey spectrum; (b) Ti 2p_{3/2}; (c) O 1s and (d) C 1s high-resolution spectra.

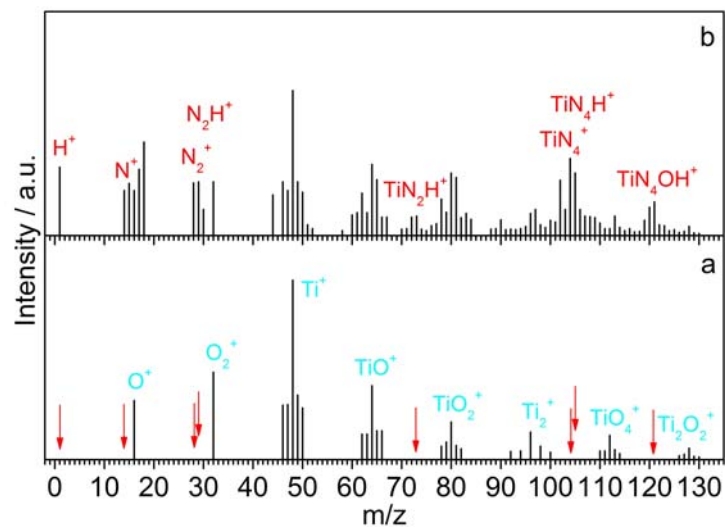


Fig.S6 ToF-SIMS spectra of (a) the commercial white anatase TiO₂ and (b) the commercial white anatase TiO₂ after treated by SPP with NH₄Cl.



(a)



(b)

Fig.S7 Photographs of (a) the commercial white anatase TiO₂ nanoleaves and (b) the commercial white anatase TiO₂ nanoleaves after treated by SPP with NH₄Cl.

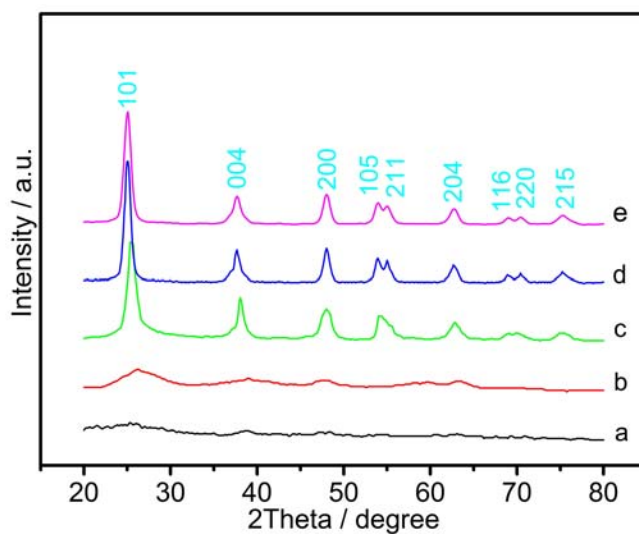


Fig.S8. XRD patterns of the black anatase TiO₂ nanoleaves prepared during SPP: (a) 0.5 min; (b) 1 min; (c) 2 min; (d) 5 min; (e) 10 min.

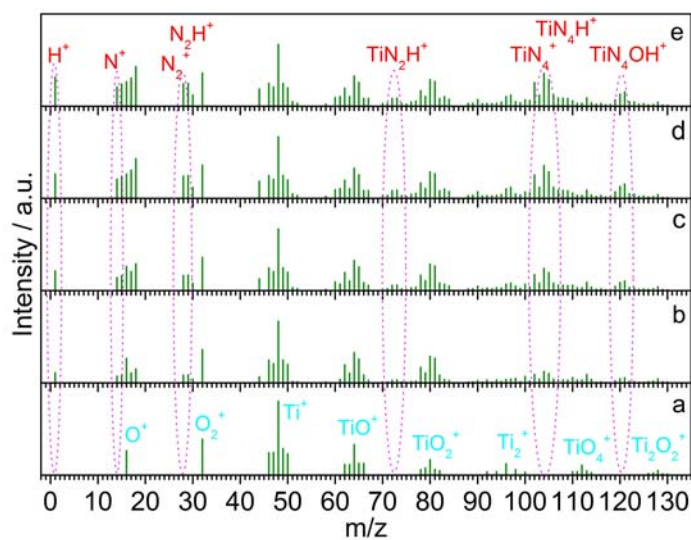


Fig.S9. ToF-SIMS spectra of the black anatase TiO₂ nanoleaves prepared during SPP: (a) 0.5 min; (b) 1 min; (c) 2 min; (d) 5 min; (e) 10 min.

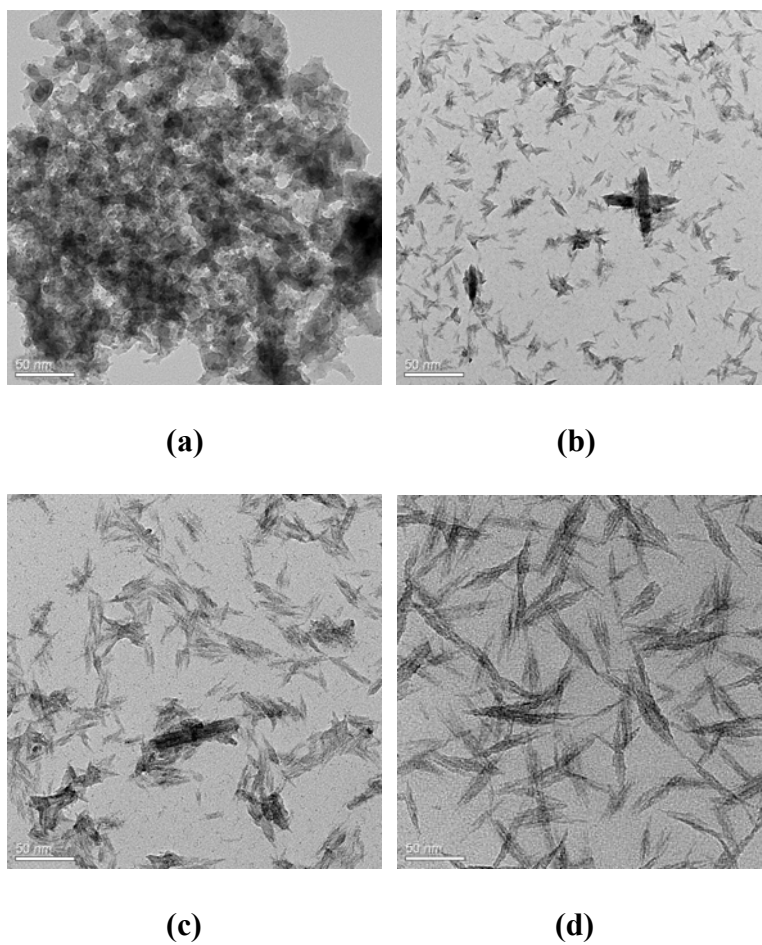


Fig.S10 TEM images of the black anatase TiO_2 nanoleaves obtained with different concentration of Brij-30: (a) 0 M, (b) 0.01 M, (c) 0.02M, (d) 0.05 M.

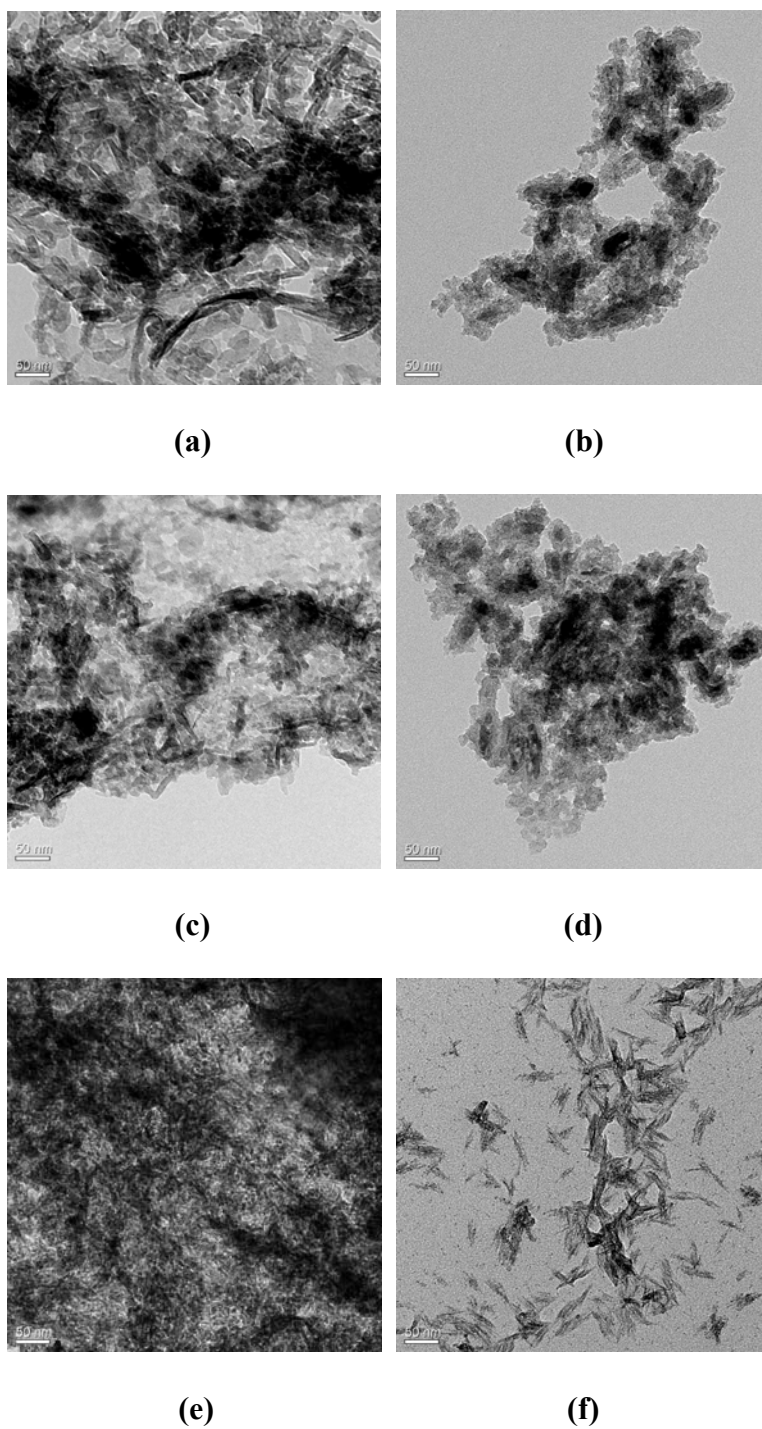


Fig.S11. TEM images of the black anatase TiO₂ nanoleaves prepared by SPP with (a) P123; (b) ethylenediamine; (c) SDBS; (d) PVA; (e) PVP and (f) either of (a)–(e) after additional of Brij-30.

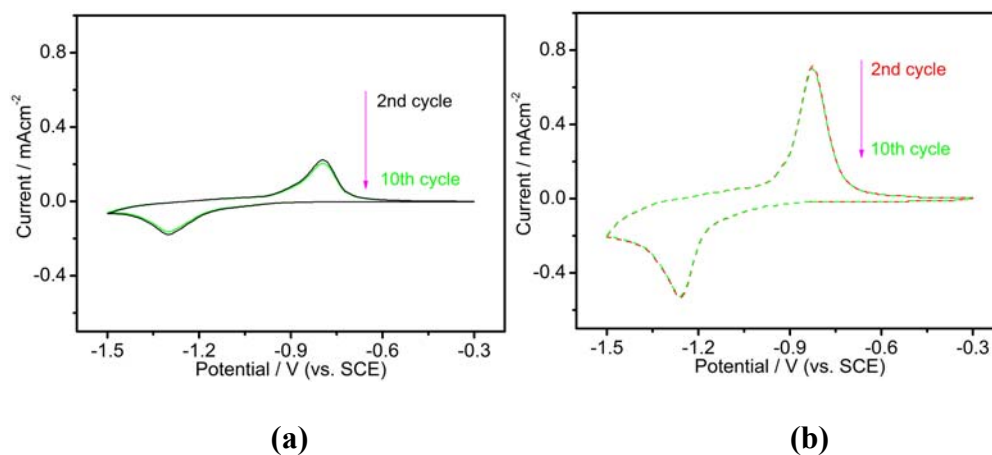


Fig.S12 Cyclic voltammograms of (a) commercial white anatase TiO_2 and (b) black anatase TiO_2 nanoleaves electrodes at a scanning rate of 1mVs^{-1} from the second to the tenth cycle.

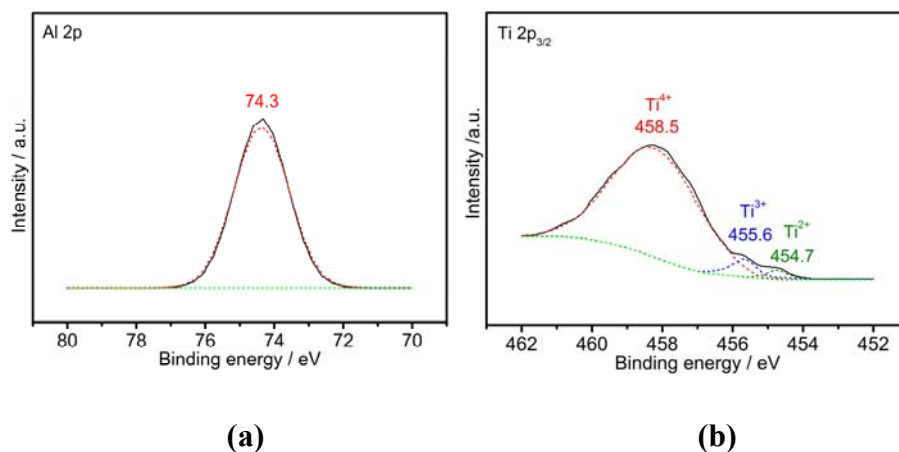


Fig.S13 XPS spectra of the commercial white anatase TiO_2 after being 1st discharged at 0.05Ag^{-1} between -1.4 and -0.4 V (vs. SCE) with argon-ion sputtering for 3 min: (a) Al 2p and (b) Ti 2p_{3/2}.

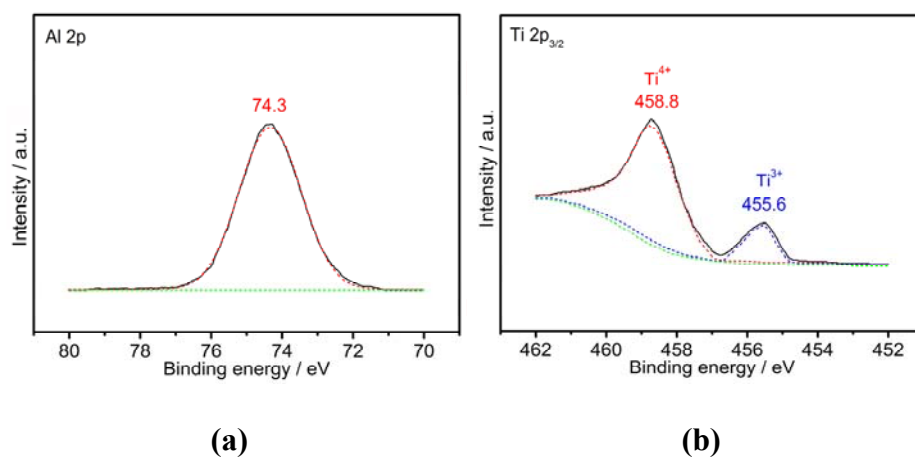


Fig.S14 XPS spectra of the black anatase TiO₂ nanoleaves after being 1st discharged at 0.05 Ag⁻¹ between -1.4 and - 0.4 V (vs. SCE) with argon-ion sputtering for 3 min: (a) Al 2p and (b) Ti 2p_{3/2}.

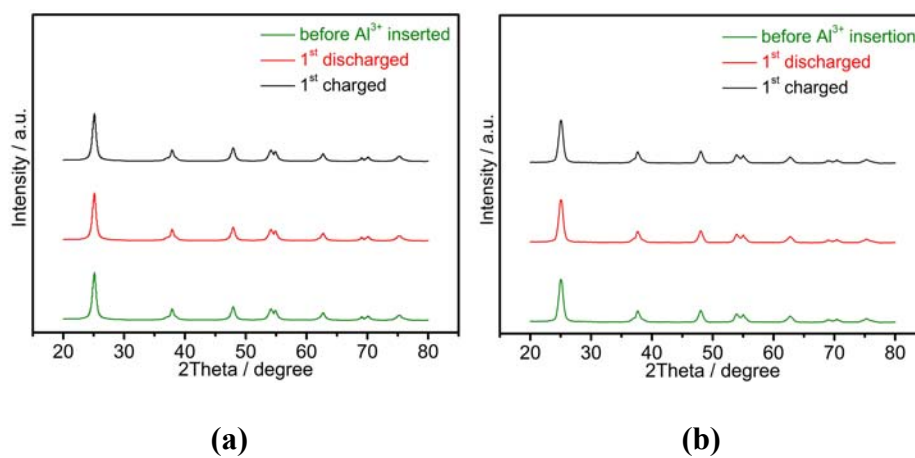


Fig.S15 XRD patterns of the (a) commercial white anatase TiO₂; (b) black anatase TiO₂ nanoleaves before Al³⁺ insertion and after being 1st discharged and charged at 0.05 Ag⁻¹ between -1.4 and - 0.4 V (vs. SCE).

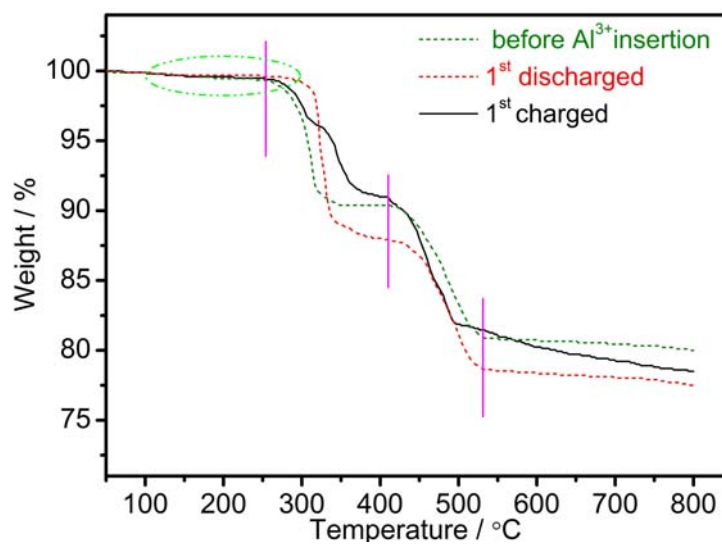


Fig.S16 The typical curves of black anatase TiO₂ nanoleaves before Al³⁺ insertion and after being 1st discharged and charged at 0.05 Ag⁻¹ between -1.4 and - 0.4 V (vs. SCE).

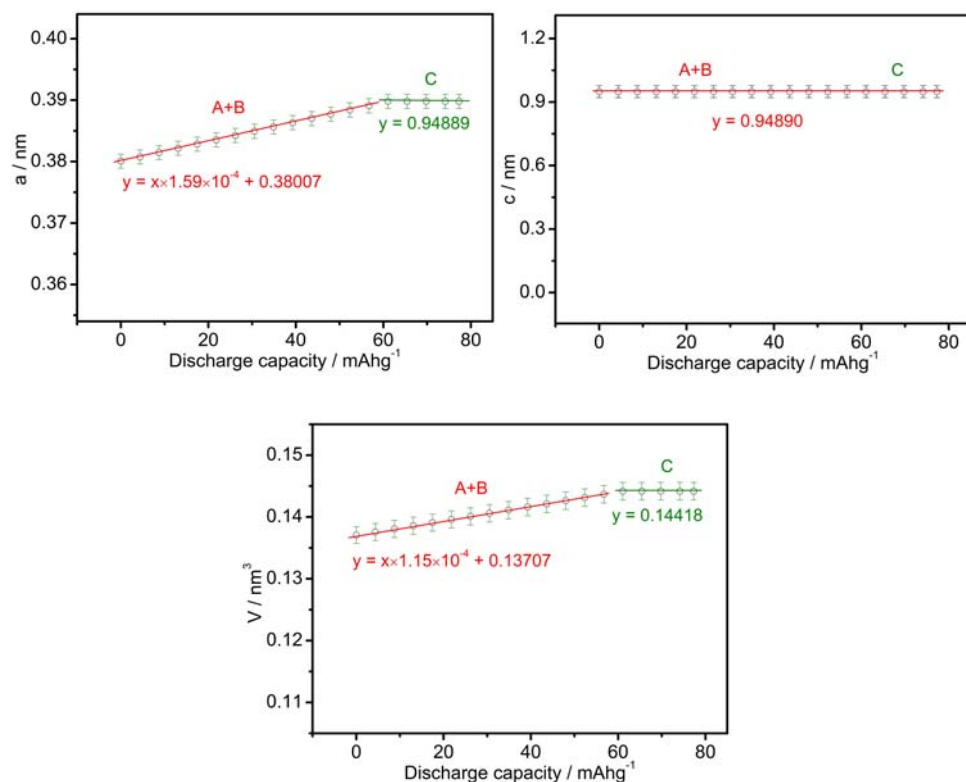


Fig.S17 The changes of crystal parameters (a , c and lattice volume V) of commercial white anatase TiO₂ obtained from the in-situ XRD in A, B and C regions labeled in Fig.8 during the 1st discharge at 0.05 Ag⁻¹ between -1.4 and -0.4 V (vs. SCE).

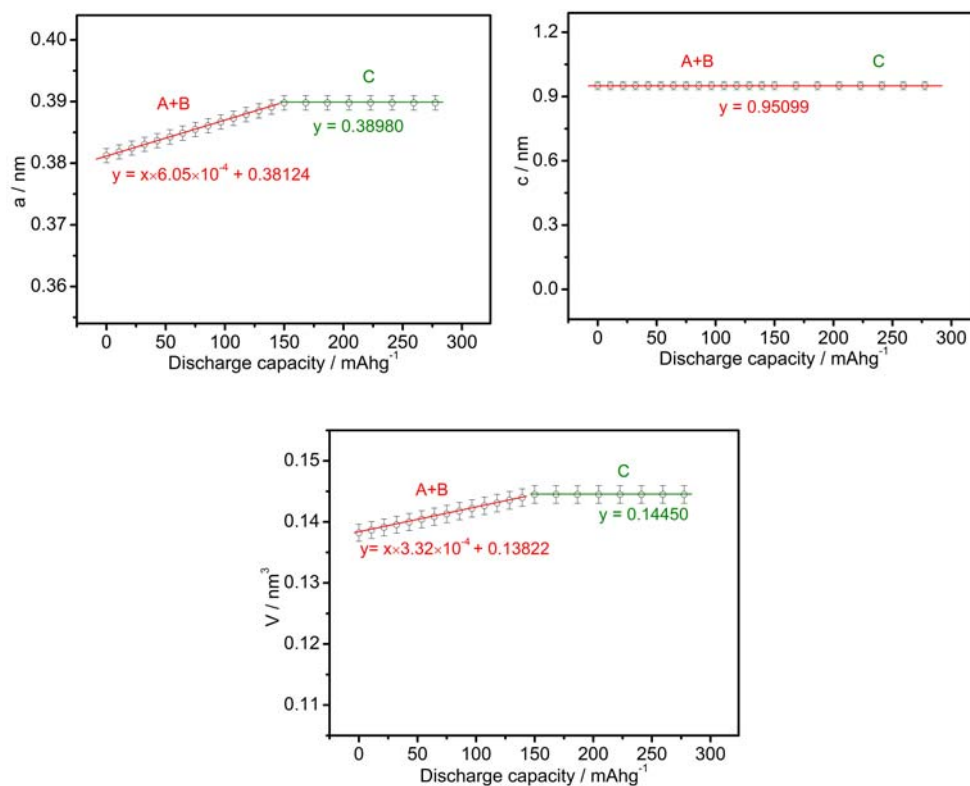


Fig.S18 The changes of crystal parameters (*a*, *c* and lattice volume *V*) of black anatase TiO₂ nanoleaves obtained from the in-situ XRD in A, B and C regions labeled in Fig.8 during the 1st discharge at 0.05 Ag⁻¹ between -1.4 and - 0.4 V (vs. SCE).

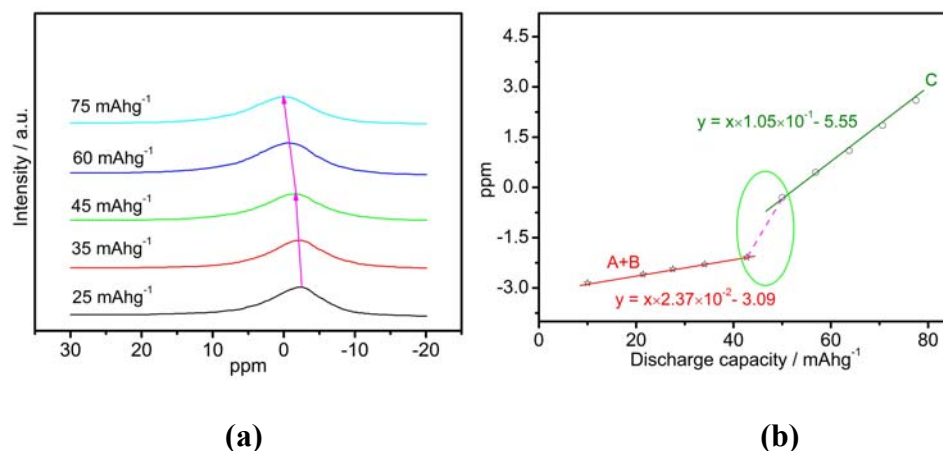


Fig.S19 (a) The typical *ex situ* ^{27}Al MAS NMR curves and **(b)** the variation of shifts of commercial white anatase TiO_2 in A, B and C regions labeled in Fig.8 during the 1st discharge at 0.05 Ag^{-1} between -1.4 and - 0.4 V (vs. SCE).

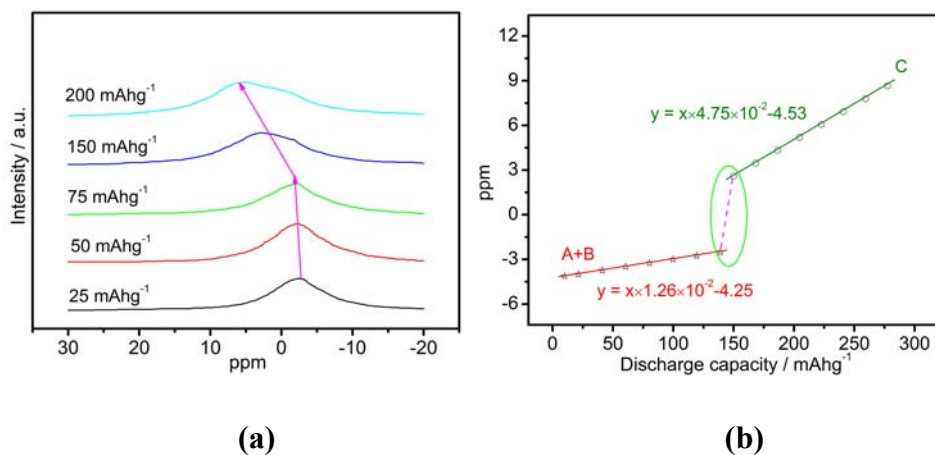


Fig.S20 (a) The typical *in-situ* ^{27}Al MAS NMR curves and **(b)** the variation of shifts of black anatase TiO_2 nanoleaves in A, B and C regions labeled in Fig.8 during the 1st discharge at 0.05 Ag^{-1} between -1.4 and - 0.4 V (vs. SCE).

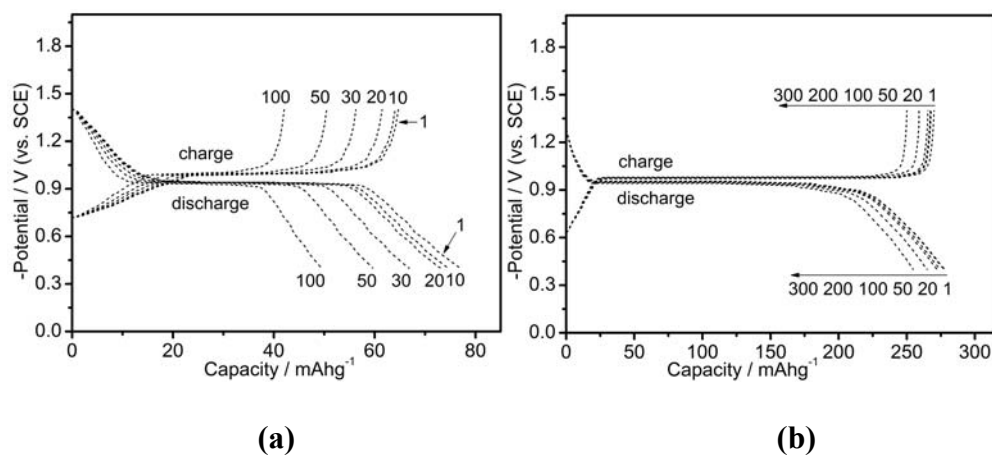


Fig.S21 Charge-discharge curves of (a) commercial white anatase TiO_2 and (b) black anatase TiO_2 nanoleaves electrodes under a current rate of 0.05 Ag^{-1} at different cycles.

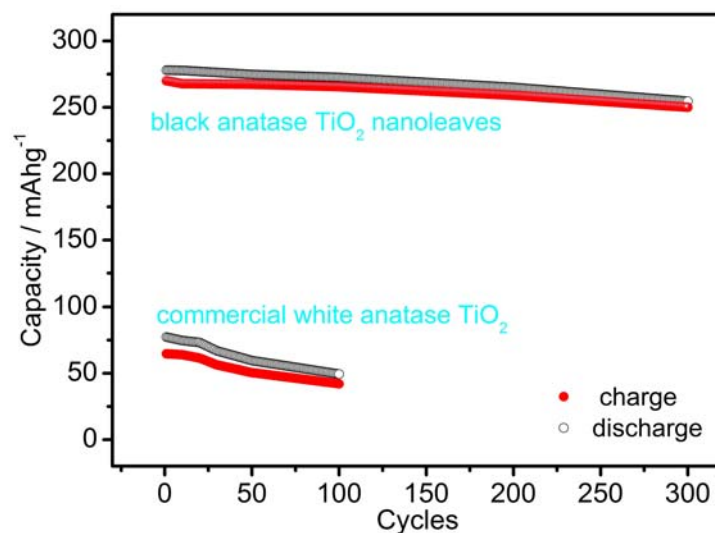


Fig.S22 Cycle performance of commercial white anatase TiO_2 and black anatase TiO_2 nanoleaves electrodes at a current rate of 0.05 Ag^{-1} .

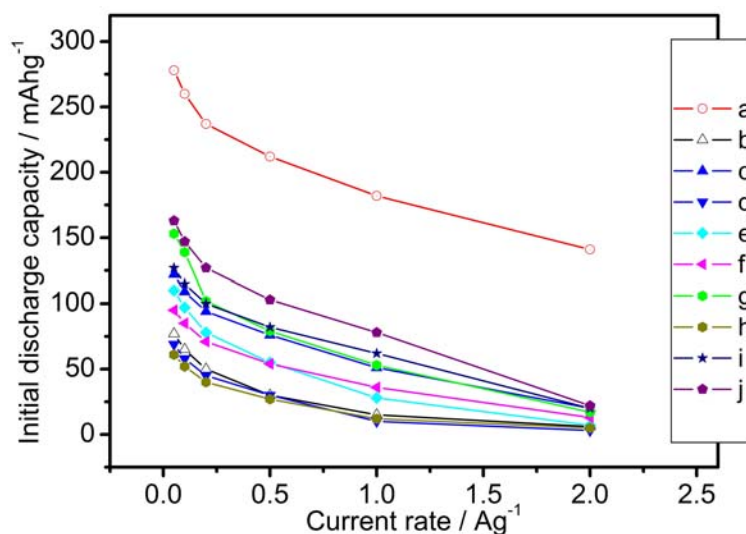


Fig.S23 Rate performance comparison of the samples (a) black anatase TiO₂ nanoleaves in this work; (b) commercial white anatase TiO₂; (c) black anatase TiO₂ prepared from (b) after SPP treatment with NH₄Cl; (d) commercial TiO₂-P25; (e) commercial anatase TiO₂-UV100; (f) commercial white anatase TiO₂ nanotube; (g) black anatase TiO₂ nanotubes prepared from (f) after SPP treatment with NH₄Cl; (h) amorphous white anatase TiO₂ obtained in our work after SPP for 0.5 min; (i) black anatase TiO₂ obtained in our work after SPP for 1 min; (j) black anatase TiO₂ obtained in our work after SPP for 2 min.

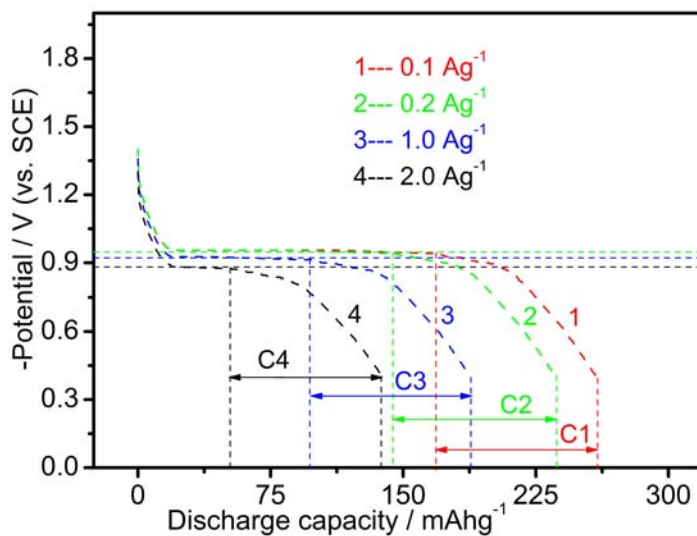


Fig.S24 The 1st insertion curves at various rates of the TiO_2 nanoleaves.

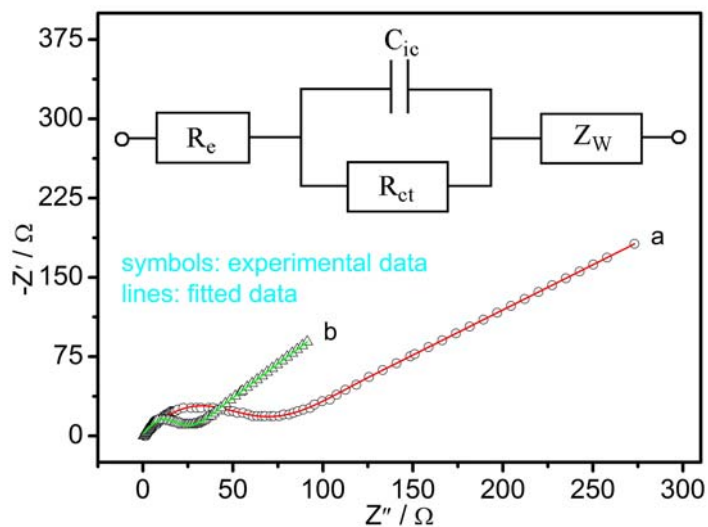


Fig.S25 Nyquist plots with fitted lines of (a) commercial white anatase TiO_2 and (b) black anatase TiO_2 nanoleaves. The inset is the corresponding equivalent circuits.

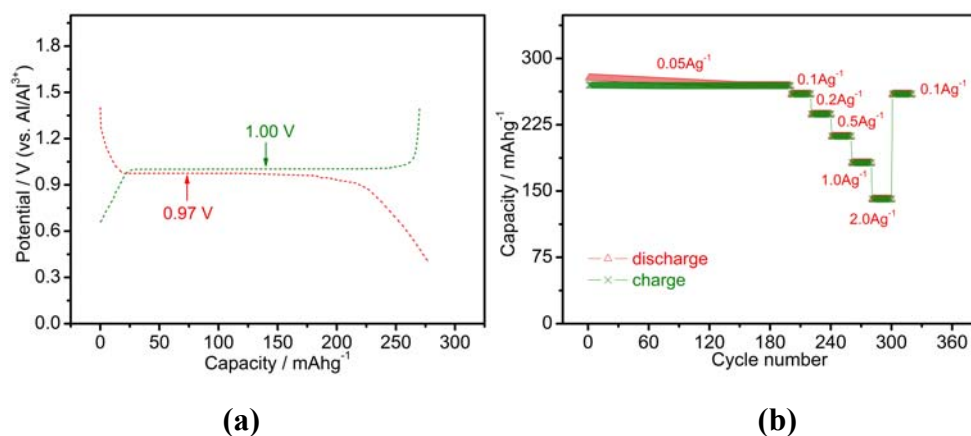


Fig.S26 (a) The initial charge/discharge curves and (b) the variation of charge/discharge capacities with cycle numbers of TiO_2/Al two-electrode cell at different current rates between 0.4 and 1.4 V (vs. Al/Al^{3+})