Supporting Information

Highly Sensitive and Selective Fluorescence Turn-on Detection of Lead Ion in Water Using Fluorene-Based Compound and Polymer

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Scheme S1. Synthesis of compound 8 from 7a

Compound 8. Compound **7a** (0.025 mg, 0.05 mmol) was dissolved in 3 mL of methanol under argon atmosphere and 1.00 g of SOCl₂ was added to the solution dropwise. The solution was stirred overnight and monitored by the TLC (hexane/acetone 3:1). The mixture was neutralized by 10% sodium bicarbonate solution and extracted three times with dichloromethane. The solvent was evaporated under the reduced pressure and purified by silica gel column chromatography using hexane and acetone (3:1) as eluting solvent. Compound **8** was obtained as yellow viscous oil after removing the solvent. ¹H NMR (300 MHz, CDCl₃,): δ 8.01 (1H, d), 7.78 (3H, m), 7.64 (1H, s), 7.55 (2H, d), 7.35 (3H, m), 7.15 (1H, m), 6.79 (2H, d), 4.19 (2H, s), 4.16 (2H, s), 3.75 (18H, m). ESI-MS calculated for [MH⁺]: 546.24, found 546.22. IR (NaCl plate, cm⁻¹): 3012, 2951, 2900,1754, 1595, 1600, 1519, 1219.



Figure S1. IR spectrum (NaCl plate) of compound 1.



Figure S2. ¹H NMR spectrum (300 MHz, CDCl₃) of compound **1**.



Figure S3. ¹³C NMR spectrum (75 MHz, CDCl₃) of compound 1.



Figure S4. IR spectrum (NaCl plate) of compound 2.



Figure S5. ¹H NMR spectrum (300 MHz, CDCl₃) of compound **2**.



Figure S6. ¹³C NMR spectrum (75 MHz, CDCl₃) of compound 2.







Figure S8. IR spectrum (KBr Pellet) of compound 3.



Figure S9. ¹H NMR spectrum (300 MHz, CDCl₃) of compound **3**.



Figure S10. ¹³C NMR spectrum (75 MHz, CDCl₃) of compound 3.



Figure S11. Mass spectrum of compound 3.



Figure S12. IR spectrum (NaCl plate) of compound 4.



Figure S13. ¹H NMR spectrum (300 MHz, CDCl₃) of compound 4.



Figure S14. ¹³C NMR spectrum (75 MHz, CDCl₃) of compound 4.



Figure S15. Mass spectrum of compound 4.



Figure S16. ¹H NMR spectrum (300 MHz, CDCl₃) of 2,7-dibromofluorene.



Figure S17. ¹³C NMR spectrum (75 MHz, CDCl₃) of 2,7-dibromofluorene.



Figure S18. IR spectrum (KBr Pellet) of compound 5a.



Figure S19. ¹H NMR spectrum (300 MHz, CDCl₃) of compound 5a.



Figure S20. ¹³C NMR spectrum (75 MHz, CDCl₃) of compound 5a.







Figure S22. IR spectrum (KBr Pellet) of compound 5b.



Figure S23. ¹H NMR spectrum (300 MHz, CDCl₃) of compound 5b.



Figure S24. ¹³C NMR spectrum (75 MHz, CDCl₃) of compound 5b.



Figure S25. Mass spectrum of compound 5b.



Figure S26. IR spectrum (NaCl Plate) of compound 6a.



Figure S27. ¹H NMR spectrum (300 MHz, CDCl₃) of compound **6a**.



Figure S28.Mass spectrum of compound 6a.



Figure S29. IR spectrum (NaCl Plate) of compound 6b.



Figure S30. ¹H NMR spectrum (300 MHz, CDCl₃) of compound 6b.



Figure S31. ¹³C NMR spectrum (75 MHz, CDCl₃) of compound 6b.



Figure S32 .Mass spectrum of compound 6b.



Figure S33.¹H NMR spectrum (300 MHz, D₂O) of compound 7a.



Figure S34. IR spectrum (KBr pellet) of compound 7a.



Figure S35. Mass spectrum of compound 7a.



Figure S36. Normalized absorption and emission spectra of compound 7a in water.



Figure S37. ¹HNMR spectrum (300 MHz, CDCl₃) of compound 8.



Figure S38. Mass spectrum of compound 8.



Figure S39. ¹H NMR spectrum (300 MHz, CDCl₃) of polymer P1.



Figure S40. Normalized absorption and emission spectra of polymer P1 in DMF and compound 6a in CHCl₃.



Figure S41. IR spectrum of polymer P1 (NaCl plate).



Figure S42. IR spectrum of polymer P2 (KBr pellet).





Figure S43. IR spectrum of polymer P3 (KBr pellet).



Figure S44. Normalized absorption and emission spectra of (a) polymer **P2** inwater: THF (1:1) and (b) polymer **P3** in DMF.



Figure S45. Photograph of $7a-Pb^{2+}$ complex deposited at the bottom of volumetric flask.



Figure S46. Photograph of **P2**-Pb $^{2+}$ deposited at the bottom of vials.



Figure S47. TGA of polymer P1.



Figure S48. DSC trace of polymer P1



Figure S49. Changes of absorption spectra of **7a** in water $(5 \times 10^{-5} \text{ M})$ upon addition of Pb²⁺ (0 to $11 \times 10^{-6} \text{ M}$).



Figure S50. Job's plot of compound **7a** and Pb^{2+} . The total concentration of compound **7a** and Pb^{2+} was kept at 6.0 x 10⁻⁴M. Excitation at 400 nm and emission intensity at 489 nm.



Figure S51. Changes of fluorescence spectra of polymer **P3** (2.8×10^{-4} M) in DMF upon addition of Pb²⁺ (0 to 584 ×10⁻⁶ M) (excitation at 400 nm).



Figure S52. Changes of fluorescence spectra of polymer **P3** (2.8×10^{-4} M) in pure water/THF (1:1) with 2.5 mg of LiOH upon addition of Pb²⁺ (0 to 904 × 10⁻⁶ M) (excitation at 400 nm).