Supporting information



Fig.S1 XRD patterns of GO and Graphene.



Fig.S2 XPS survey spectra of GO, BT and BTG-0.75 wt%.

	XPS			ICP			
Sample	Bi	Ti		Bi	Ti		
	(at %)	(at %)	n (11/B1)	(wt %)	(wt %)	n (11/B1)	
BT	3.78	17.82	4.71	29.45	31.66	4.69	
BTG-0.75 wt%	2.48	10.10	4.07	27.44	31.14	4.95	

Table S1 The element content from the XPS and ICP analysis.

Table S2 The elemental analysis of C, H, O in the sample BT and BTG-0.75 wt%.

sample	C (wt%)	O (wt%)	H (wt%)
BT	8.55	6.275	2.091
BTG-0.75 wt%	10.5	5.269	2.342



Fig.S3 SEM of GO (a), GO after solvothermal treatment (b), BT (c) and (d).



Fig.S4 Photograph of the BT and BTG-X



Fig.S5 EDS analysis of different particles in BTG-0.75 wt%.

Table S3 The elemental	composition	from	EDS	result.
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$DTC = 0.75 \cdots 10^{7}$	Elements (at%)					
B1G-0.75 wt%	Bi	Ti	С	C Br	0	Ti:Bi:Br
Particle 1	0.99	6.26	88.68	0.65	3.42	6.32/1/0.66
Particle 2	0.68	4.21	91.68	0.49	2.95	6.19/1/0.72
Particle 3	0.98	6.59	88.96	0.7	2.76	6.72/1/0.71



Fig.S6 Time profile of RhB absorbance spectrum observed during photodegradation with BTG-0.75 wt% under visible light irridiation.



Fig. S7 Variation of RhB concentration against irradiation time using control sample under visible light irradiation ($\lambda \ge 400$ nm).



Fig.S8 Reaction process on 25 mg BTG-0.75 wt% with different pollutant (a) MO, (b) SA, (c) phenol under visible light irradiation.