

Supporting Information for

Versatile Superhydrophobic and Photocatalytic Films Generated from $\text{TiO}_2\text{-SiO}_2\text{@PDMS}$ and their Applications on Fabrics

Zheng-Yan Deng, Wei Wang, Li-Hua Mao, Cai-Feng Wang, Su Chen*

State Key Laboratory of Material-Oriented Chemical Engineering and College of
Chemistry and Chemical Engineering, Nanjing University of Technology, Nanjing
210009, P. R. China

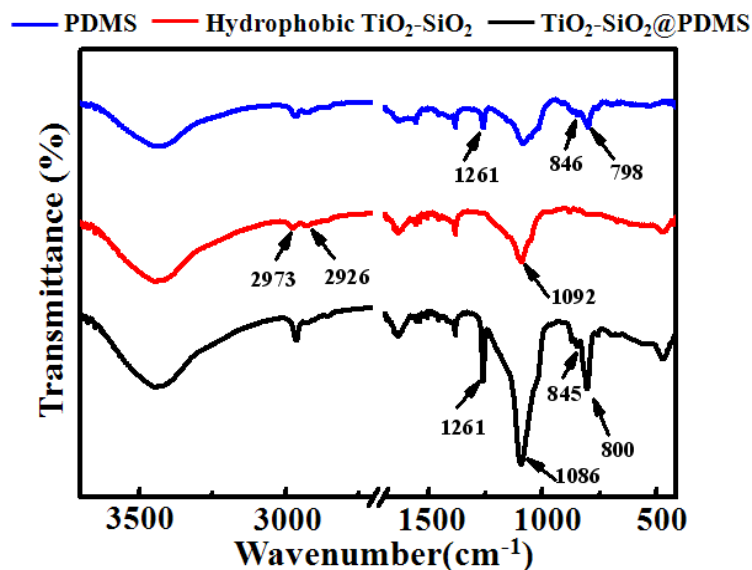


Figure S1. FT-IR spectra of polymer PDMS (blue), hydrophobic $\text{TiO}_2\text{-SiO}_2$ (red), and $\text{TiO}_2\text{-SiO}_2\text{@PDMS}$ hybrid powder (black).

The FTIR spectra of hydrophobic $\text{TiO}_2\text{-SiO}_2$ hybrid gels should have been assigned to the hydrolysis of HMDS monomers to replace the Si-OH or Ti-OH groups with $-\text{Si}(\text{CH}_3)_3$ and $\text{Ti-Si}(\text{CH}_3)_3$ groups, which are attributed to the hydrophobicity

of hybrid gels. The absorption bands observed at 2973 and 2926 cm^{-1} are due to C–H stretching. For PDMS-containing hybrid powder, the additional absorption bands at about 1261 cm^{-1} (Si-CH₃ stretching) and 845 (CH₂ rocking) conform the existence of -CH₃ groups on the surface of modified hybrid gels. It is worth noting that the peak intensity of at 1086 cm^{-1} markedly increases and becomes higher than that hybrid powder gels and pure PDMS, which results in the introduction of Si-O-Si stretching vibrations.

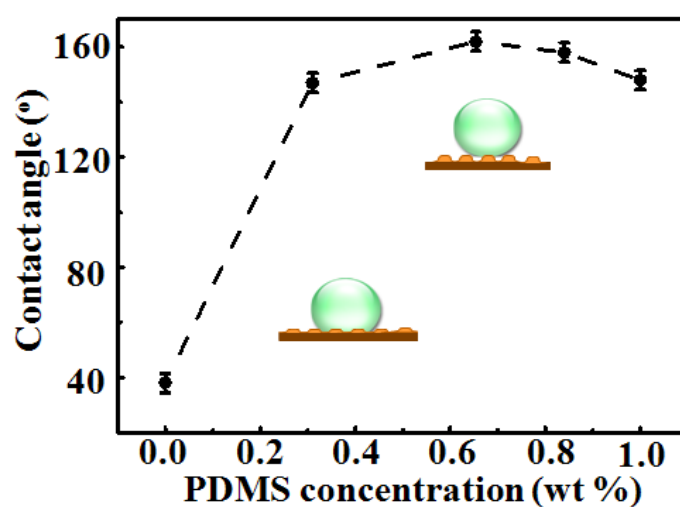


Figure S2. Effect of PDMS concentration on water contact angle (WCA) values

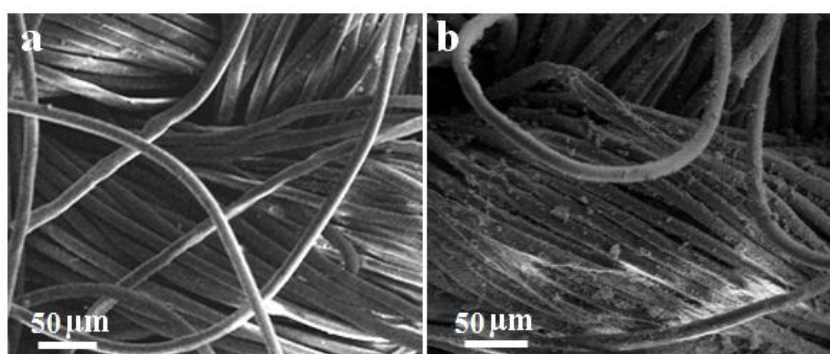


Figure S3. SEM images of the polyester/cotton fabrics. (a) The pristine fabrics, scale bar: 50 μm . (b) Treated fabrics, scale bar: 50 μm