Supporting Information

A direct novel synthesis of highly uniform dispersed ruthenium nanoparticles

over P6mm ordered mesoporous carbon by host/guest complexes

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Figure S1. Wide angle XRD pattern of the Ru@MCA2 material. *The diffraction peaks are indexed to pure Ru (hexagonal structure [1], red lines) and RuO*₂ (tetragonal structure [2], green lines).

The wide angle XRD analysis of the composite material prepared from host-guest complex and RuCl₃ (Ru@MCA-2) is given in the Figure S1. Experiments were conducted on a Bruker D8 Advance diffractometer in Bragg-Brentano geometry equipped with a copper anode (λ =1.5418 Å) and a 1D PSD Lynxeye detector. The scattering intensity was measured over an angular range of 15°< 20□< 80° with a step-size of Δ (20)= 0.02°.

The XRD pattern shows a broad peak at $2\theta = 23^{\circ}$ and a less intense peak at $2\theta = 44^{\circ}$ attributed to the (002) and (100) diffraction peaks in carbon materials, with limited domain sizes of stacked crystalline graphite. No reflection peaks of ruthenium in the form of metallic or oxide phase

emerge from the XRD pattern, indicating that the ruthenium phase forms very small particles within the carbon matrix, which cannot be detected by wide angle XRD analysis [1,2].

- R. W. G. Wyckoff, In *Crystal Structures*. Second Edition, Interscience Publishers, Vol. 1, pp. 7–83, New York, 1963.
- [2] J. Haines, J. M. Léger, O. Schulte and S. Hull, Acta Crystallogr. Sect. B, 1997, 53, 880.



Figure S2. Raman spectra of the MCA-3 and MCA-4 synthesized materials in the spectral region 800-2000 cm⁻¹.

Carbon	S _{BET} (m ² g ⁻¹)	V _{tot} (cm ³ .g ⁻¹)	V _{micro} (cm ³ .g ⁻¹)	% micro ^a	Average pore diameter (nm)	Titration results (μmol g ⁻¹) ^b
AC-WV	1690	1.32	0.54	41	3.1	39
AC-NorA	1207	0.71	0.42	60	2.4	539

Table S1. Textural characteristics and Boehm's titrations of the control carbon supports

^a Relative percentage of microporosity defined as the ratio of the micropore volume to the total pore volume.

^b The Boehm's titration method was used to determine the number of acidic and basic groups [1]. In a typical experiment, 1.0 g of carbon sample was equilibrated for 24 h with 50 mL of NaOH solution (0.05 mol L^{-1}). Then, 5 mL of each filtrate was titrated with HCl (0.05 mol L^{-1}). The numbers of acidic sites were calculated from the assumption that NaOH neutralizes carboxylic, phenolic, and lactonic groups,. Basic surface sites were determined similarly by using HCl solution (0.05 mol L^{-1}) as reactant.

[1] H.P. Boehm, Carbon 32 (1994) 759.