

Electronic Supplementary Information

Synthesis of corundum type In_2O_3 porous spheres and their photocatalytic properties

Binglin Tao^a, Ying Zhang^{*a}, Dezhi Han^b, Yanpeng Li^a and Zifeng Yan^{*a}

^a State Key Laboratory for Heavy Oil Processing, Key Laboratory of Catalysis, China University of Petroleum, Qingdao 266580, China.

^b Key Laboratory of Biofuels, Qingdao Institute of Bioenergy and Bioprocess Technology, Chinese Academy of Sciences, Qingdao 266101, China

1. Experimental details for recovering the catalysts

All the suspensions, including the samples for UV-Vis spectra analysis and the residue in reactor were transferred to centrifuge tubes and washed with ethanol and water for 3 times, respectively. In order to fully recover the spent catalysts, the yellow precipitate was centrifuged at 12000 rpm for 20min to remove the RhB molecules. In this way, the catalysts suffered little loss. Then the catalyst was re-dispersed in 100 mL fresh RhB solution via sonication.

2. Supporting figures

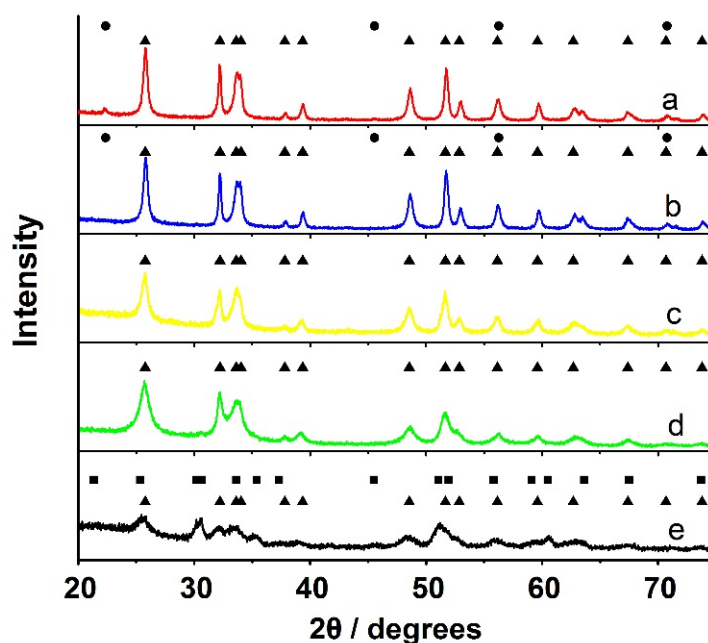


Figure S1. XRD patterns of the obtained samples at different water-ethylene glycol volume ratios: (a) 60:0, (b) 50:10, (c) 40:20, (d) 30:30, (e) 20:40, respectively. All the samples were synthesized at the temperature of 200 °C and reaction time of 16 h. • $\text{In}(\text{OH})_3$ ▲ InOOH ■ $\text{c-In}_2\text{O}_3$

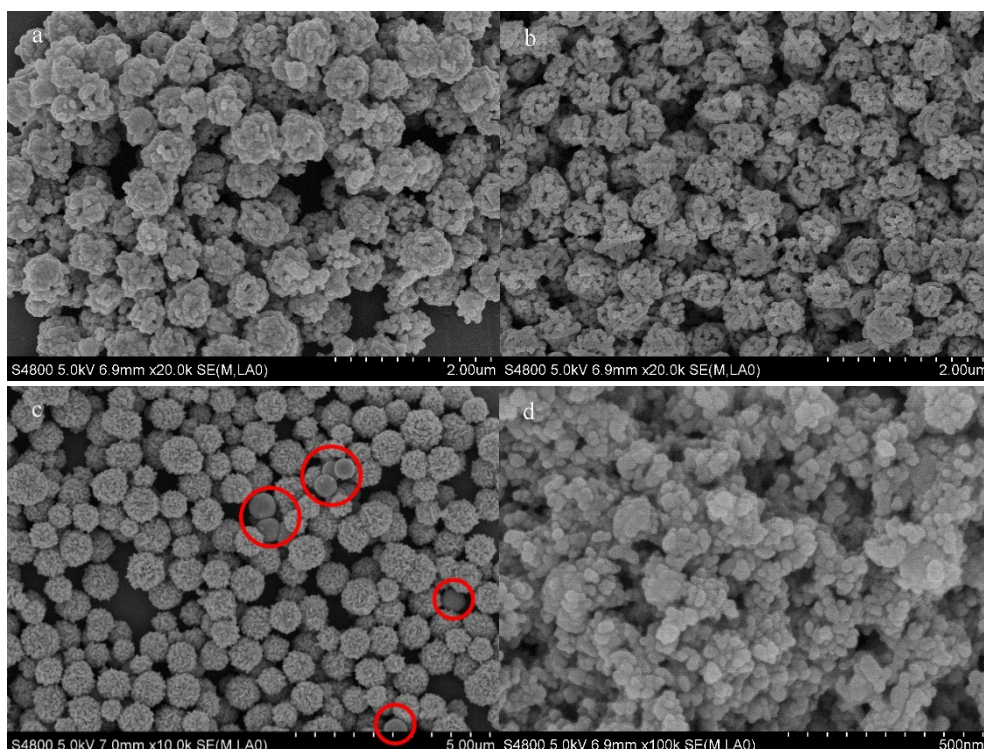


Figure S2. SEM images of the products at different water-ethylene glycol volume ratios: (a) 60:0, (b) 50:10, (c) 40:20, (d) 20:40, respectively. All the samples were synthesized at the temperature of 200 °C and reaction time of 16 h.

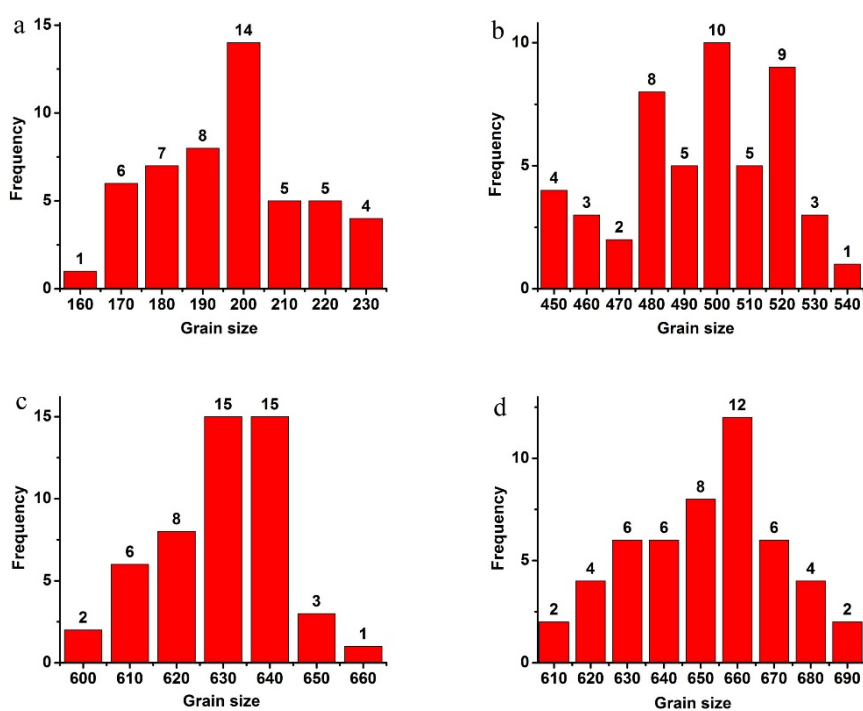


Figure S3. The size distribution histograms for InOOH spheres at different reaction intervals: (a) 2 h, (b) 6 h, (c) 10 h, (d) 16 h, respectively.

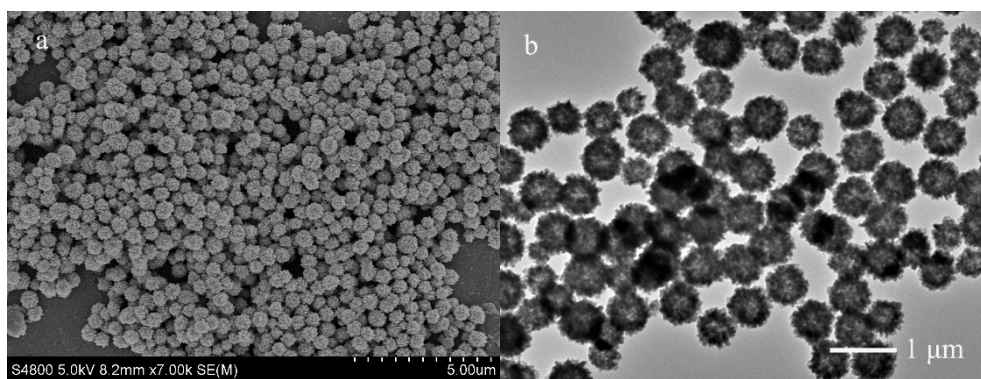


Figure S4. Low magnification SEM (a) and TEM (b) images of h-In₂O₃ porous spheres after calcination at 400 °C. The sample was synthesized at a water-ethylene glycol volume ratio of 30:30, temperature of 200 °C and reaction time of 16 h.

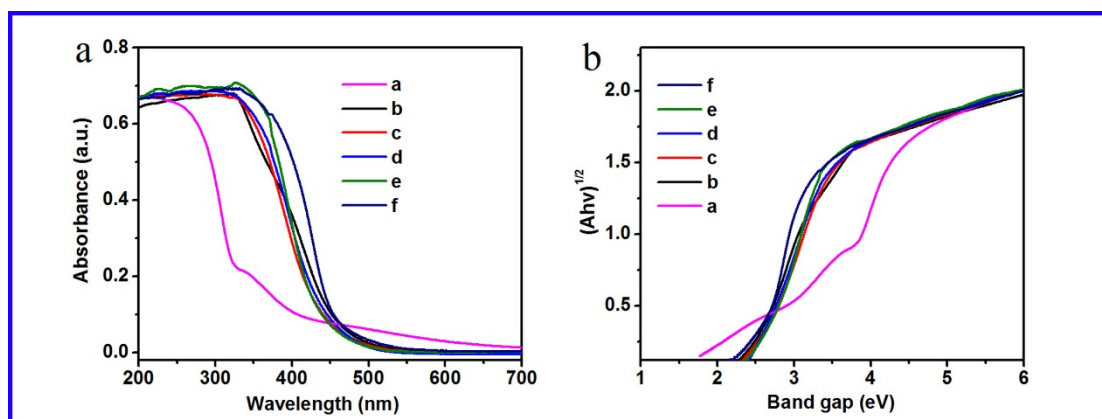


Figure S5. UV-Vis diffuse reflectance spectra and plots of $(Ah\nu)^{1/2}$ versus $h\nu$ for the band gaps of different samples: (a) InOOH porous spheres; (b) h-In₂O₃ prepared at 140 °C; (c) h-In₂O₃ prepared at 160 °C; (d) h-In₂O₃ prepared at 180 °C; (e) h-In₂O₃ porous spheres; (f) commercial c-In₂O₃. The samples were synthesized at water-ethylene glycol volume ratio of 30:30 and reaction time of 16 h.

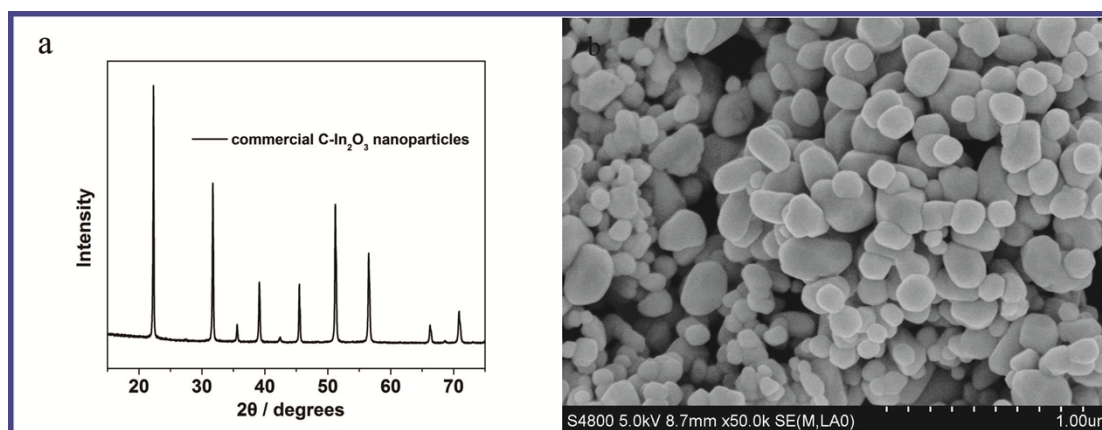


Figure S6. (a) XRD patterns and (b) SEM image of the commercial c-In₂O₃ nanoparticles