

Electronic Supplementary Information (ESI)

Magnetic Iron Oxide Microspheres Coated by Hierarchical Structural Silica: Highly Stable Composite System and Ideal Catalyst Supports

Changzi Jin^{ab}, Yanjie Wang^a, Haisheng Wei^a, Hailian Tang^a, Xin Liu^{ab}, Ting Lu^a, Junhu Wang^{ab*}*

^aLaboratory of Catalysts and New Materials for Aerospace, ^bMössbauer Effect Data Center, Dalian
Institute of Chemical Physics, Chinese Academy of Sciences, 457 Zhongshan Road, Dalian, 116023,
China

*To whom correspondence should be addressed.

Email: wangjh@dicp.ac.cn; jinchangzi@dicp.ac.cn.

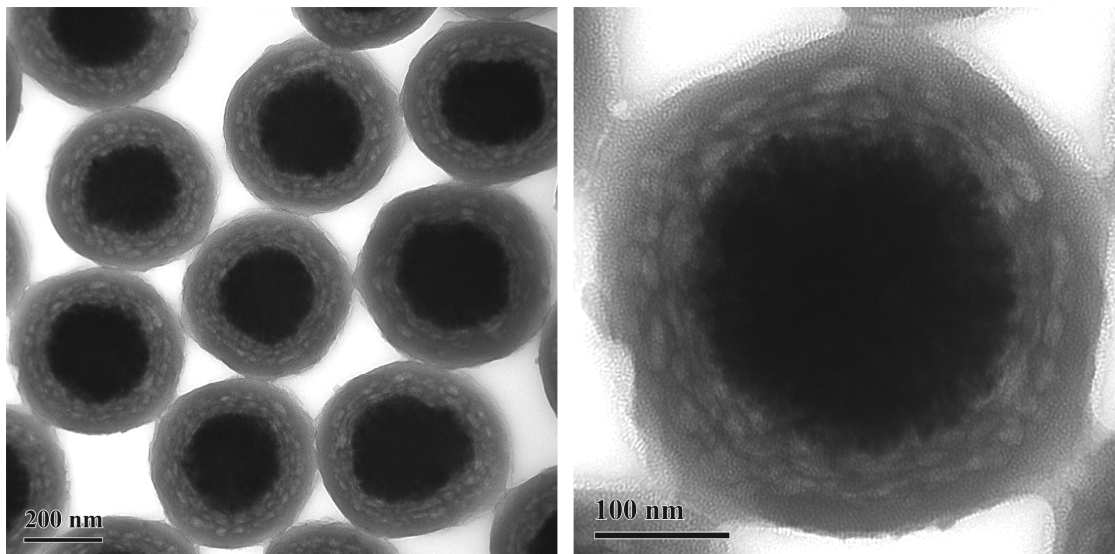


Figure S1 TEM images of MIO@H-SiO₂, whose synthesis adopt 24 h of pseudomorphic transformation.

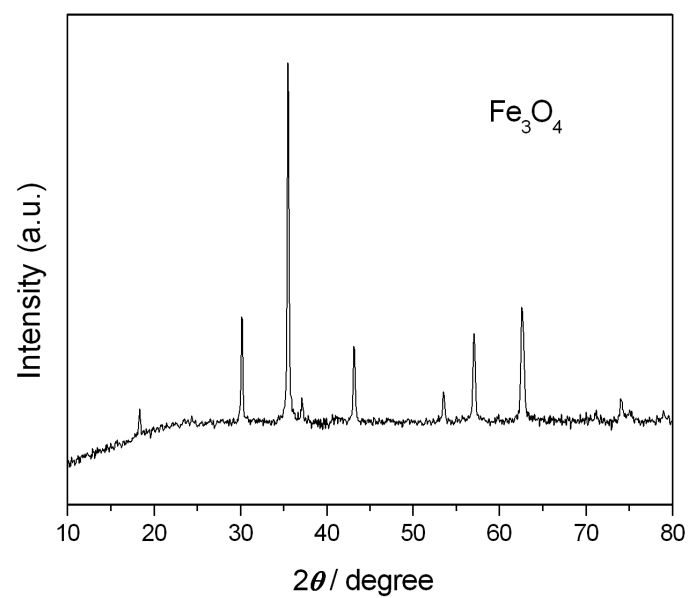


Figure S2 XRD pattern of MIO cores.

Table S1 Mössbauer hyperfine parameters of MIO core, MIO@nSiO₂/H-SiO₂ and Pt / MIO@nSiO₂/H-SiO₂

Sample	IS (mm/s)	QS (mm/s)	2 Γ (mm/s)	B (T)	A (%)	Component	Phase
MIO core	0.63	0.02	0.50	45	43	Fe ^{2.5+}	Fe ₃ O ₄
	0.31	-0.02	0.43	49	57	Fe ³⁺	
MIO@nSiO ₂ /H-SiO ₂	0.21	0	0.57	48	68	Fe ³⁺	γ -Fe ₂ O ₃
	0.19	-0.03	0.96	44	32	Fe ³⁺	
	0.61	-0.01	0.58	45	55	Fe ^{2.5+}	
Pt / MIO@nSiO ₂ /H-SiO ₂	0.30	0	0.38	48	37	Fe ³⁺	Fe ₃ O ₄ /Fe
	0	0	0.58	33	8	Fe ⁰	

IS--isomer shift relative to α -Fe; QS--quadrupole splitting; 2 Γ --line width; B--magnetic field; A--relative subspectral area

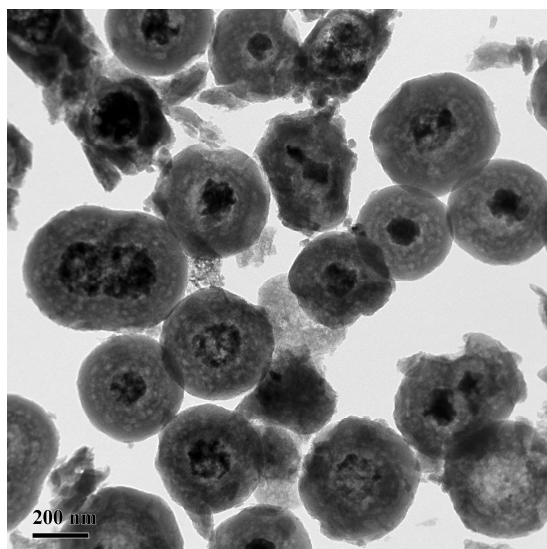


Figure S3 TEM image of MIO@H-SiO₂ immersed in 1 M HCl for 24 h

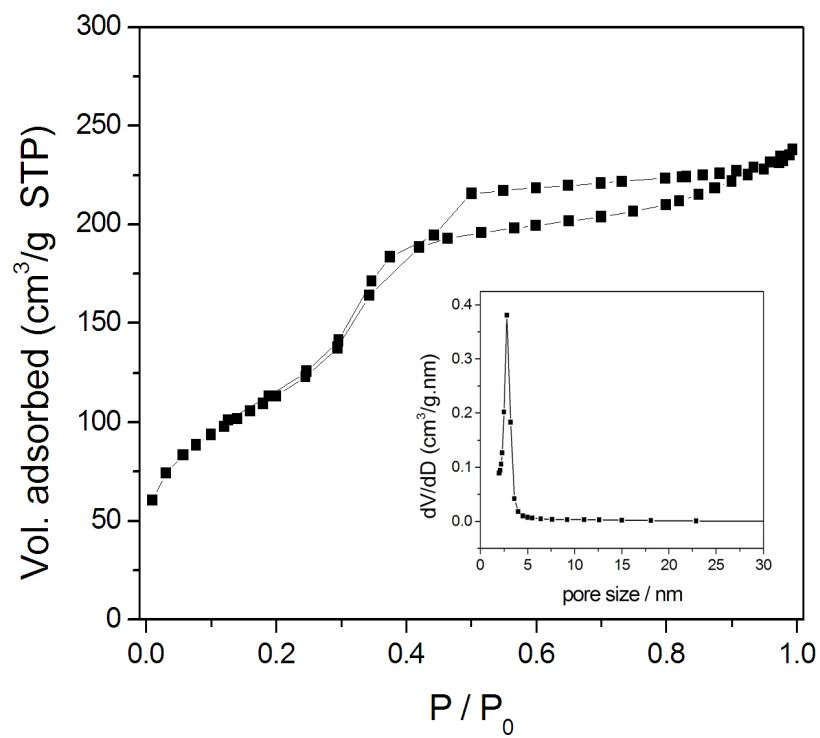


Figure S4 N₂ physical adsorption-desorption isotherm and pore size distribution (inset) of HCl treated MIO@nSiO₂/H-SiO₂

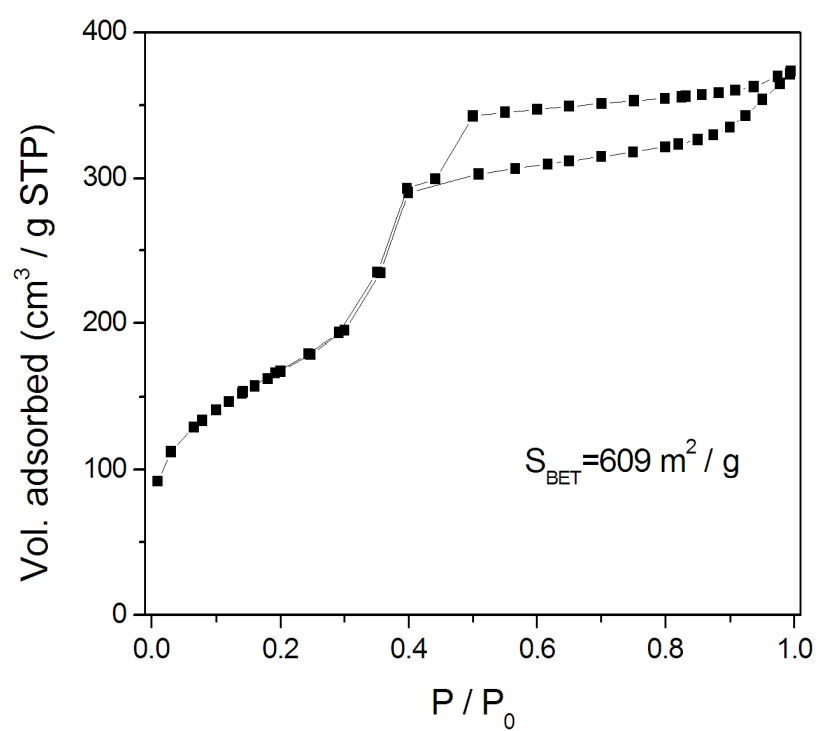


Figure S5 N₂ physical adsorption-desorption isotherm of MIO@H-SiO₂

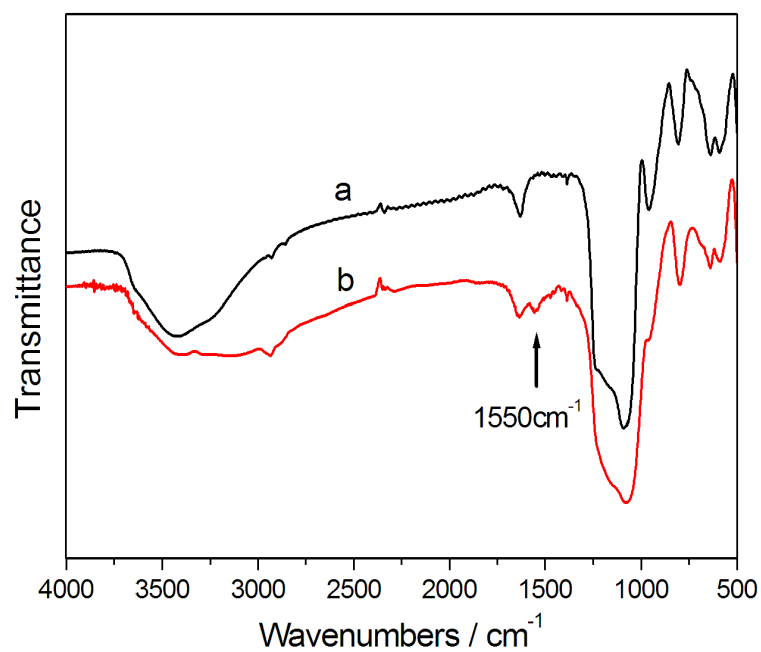


Figure S6 FT-IR spectra of (a) MIO@nSiO₂/H-SiO₂ and (b) APTS-MIO@nSiO₂/H-SiO₂. The band at 1550cm⁻¹ is assigned to the N-H bend vibration, which indicates the existence of -NH₂ group in APTS modified sample.

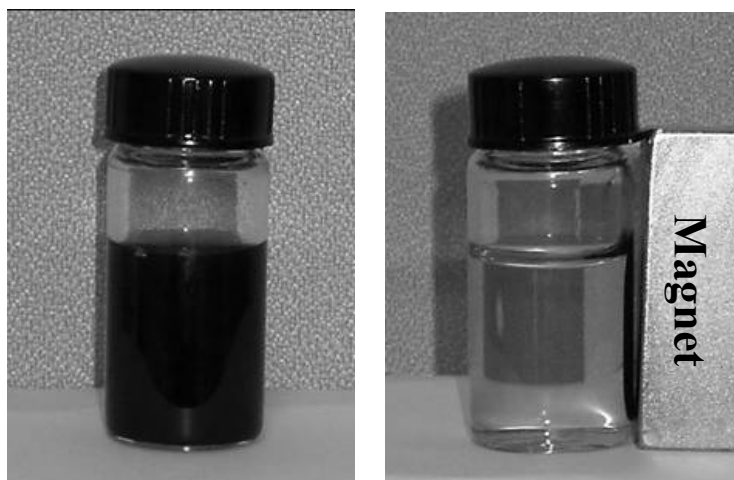


Figure S7 The photos described the convenient separation of the catalysts by using external magnet.