

Supporting Information for

Graphene Enhanced Carbon-Coated Tin Dioxide Nanoparticles for Lithium-ion Secondary Battery

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As shown in Fig. S1(a), no obviously strong peaks can be noticed in the XRD pattern of SnO₂/C-50 sample. However, just as reported in the paper, there are four sharp peaks in the sample of SnO₂/C/GN-1.5, suggesting its high crystallinity due to the introduction of GO.

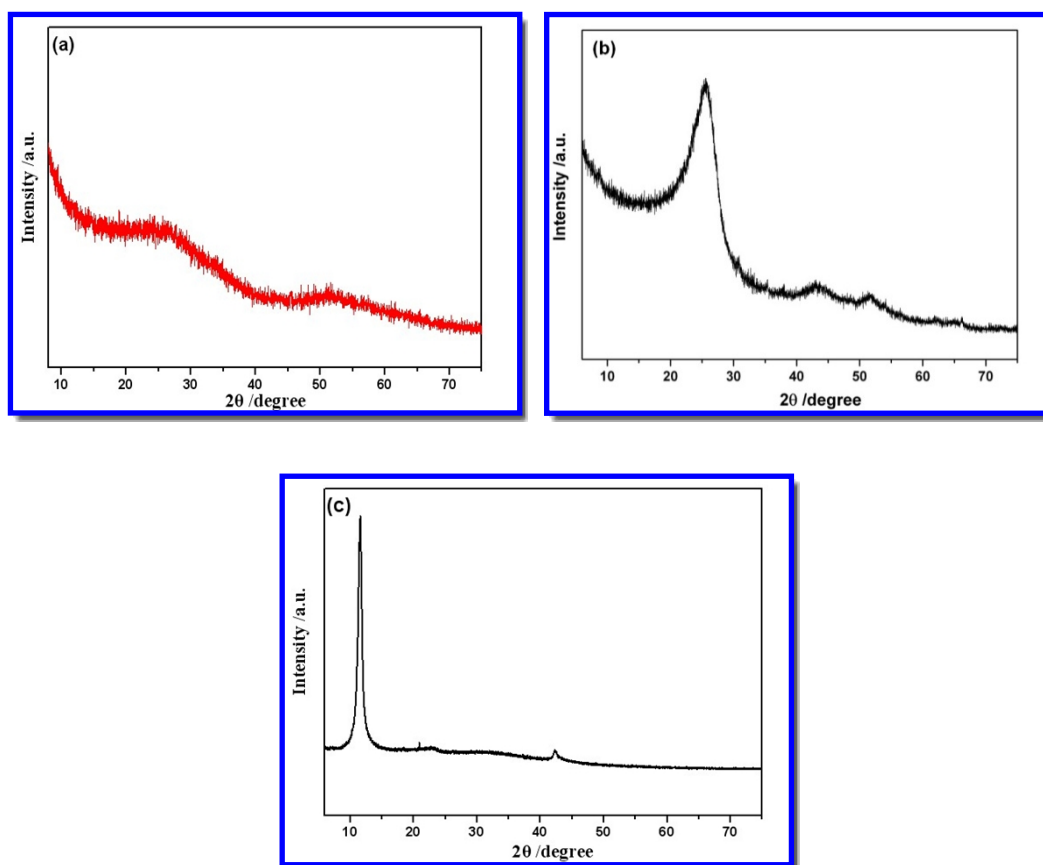


Fig. S1. XRD patterns of (a) SnO₂/C-50, (b) C/GN and (c) graphite oxide

The FT-IR spectrum of SnO₂ is revealed in Fig. S2(a), a strong peak at 600 cm⁻¹ can be ascribed to Sn-O.

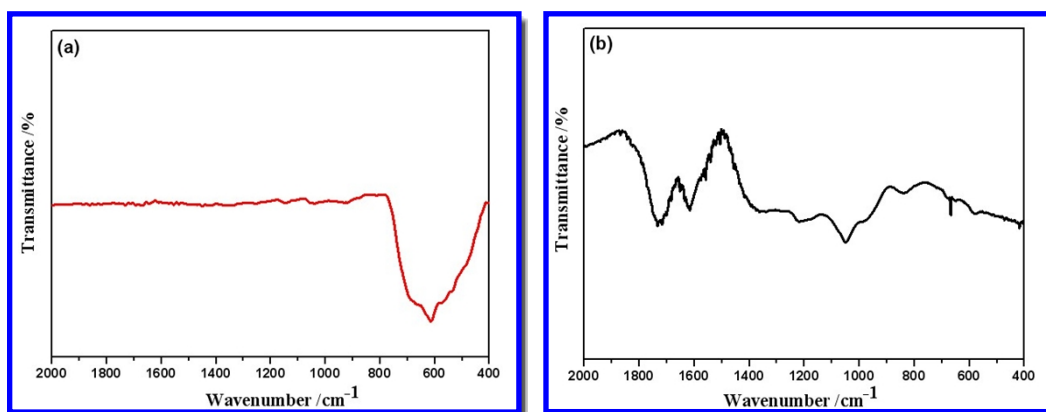


Fig. S2. FT-IR spectra of (a) SnO₂ and (b) graphite oxide

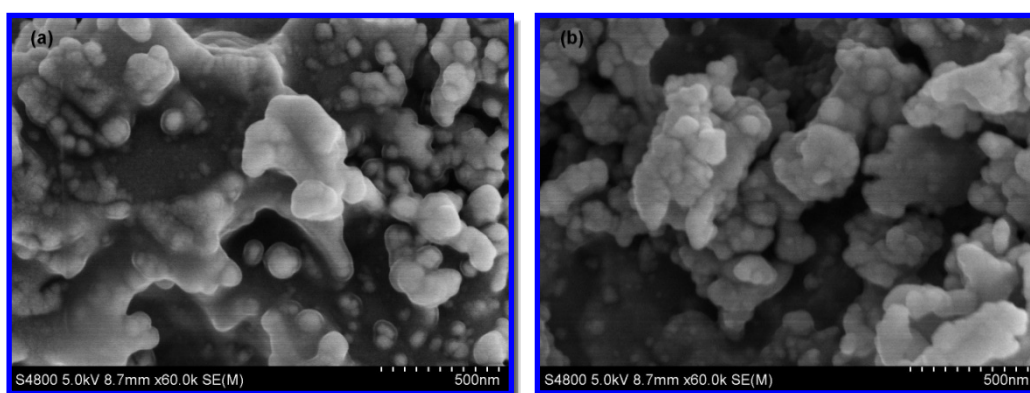


Fig. S3. SEM images of (a) SnO₂/C-50 and (b) SnO₂/C/GN-1.5

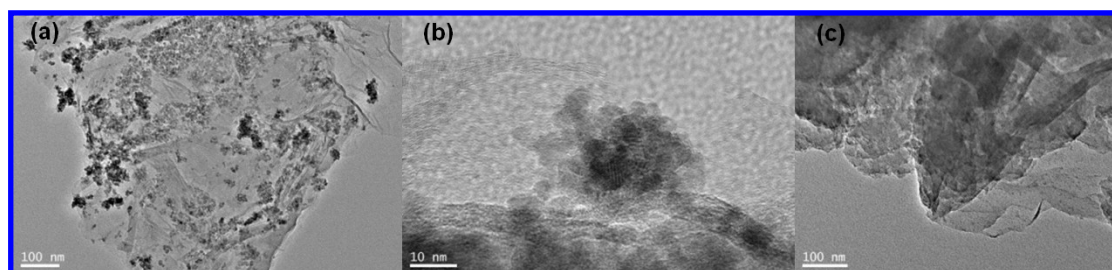


Fig. S4. TEM images of (a) SnO₂/GN, (c) C/GN and HRTEM image of (b) SnO₂/GN

The rate capability of SnO₂/C/GN-1.5 is evaluated by charging/discharging at various current densities from 100 to 800 mA g⁻¹ (Fig. S5(a)). It is obvious that the sample shows excellent cyclic capacity retention at each current density except 800 mA g⁻¹. Remarkably, at a current density of 400 mA g⁻¹, SnO₂/C/GN-1.5 delivers a high capacity of 560 mAh g⁻¹. Importantly, after the high-current-density measurements, the capacity of the SnO₂/C/GN-1.5 at 100 mA g⁻¹ can recover to the initial value, indicating its high reversibility.

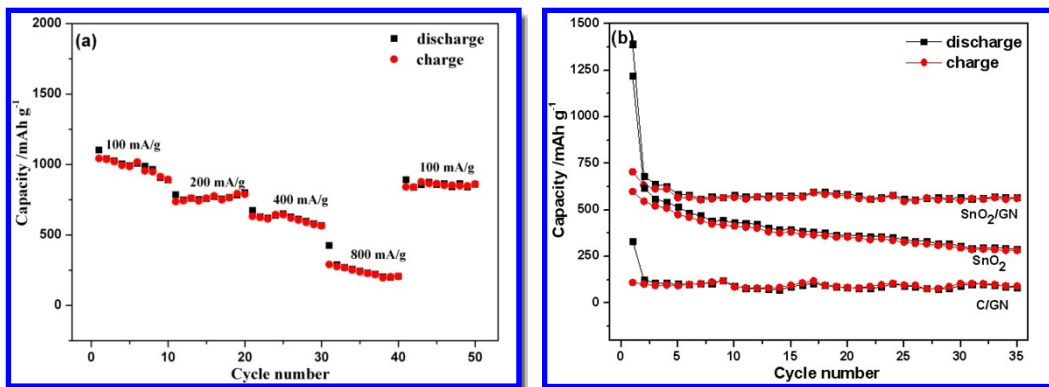


Fig. S5. (a) Rate performance of SnO₂/C/GN-1.5 and (b) the cyclic performance of SnO₂/GN, SnO₂ and C/GN