

Supporting information

Hydrogen-Treated Hematite Nanostructures with Low Onset Potential for Highly Efficient Solar Water Oxidation

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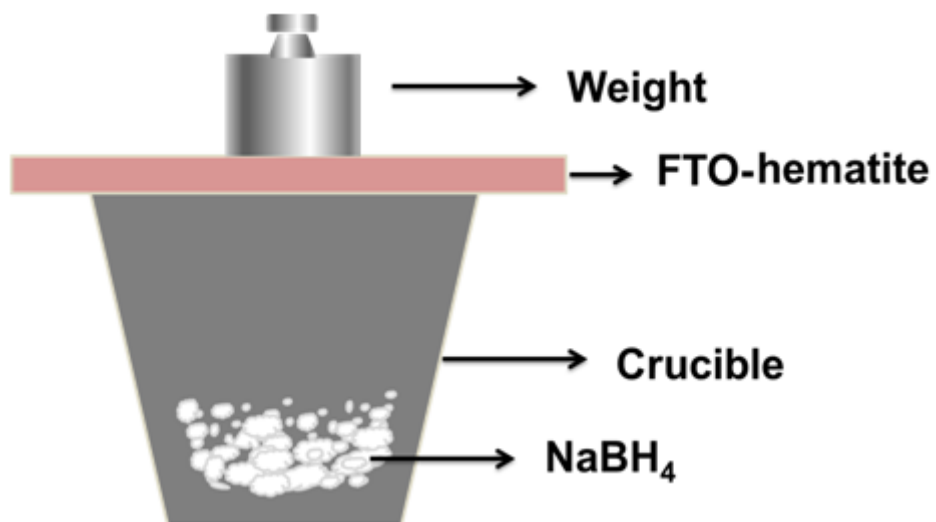


Figure S1: Illustration of the experimental setup for H₂ treatment.

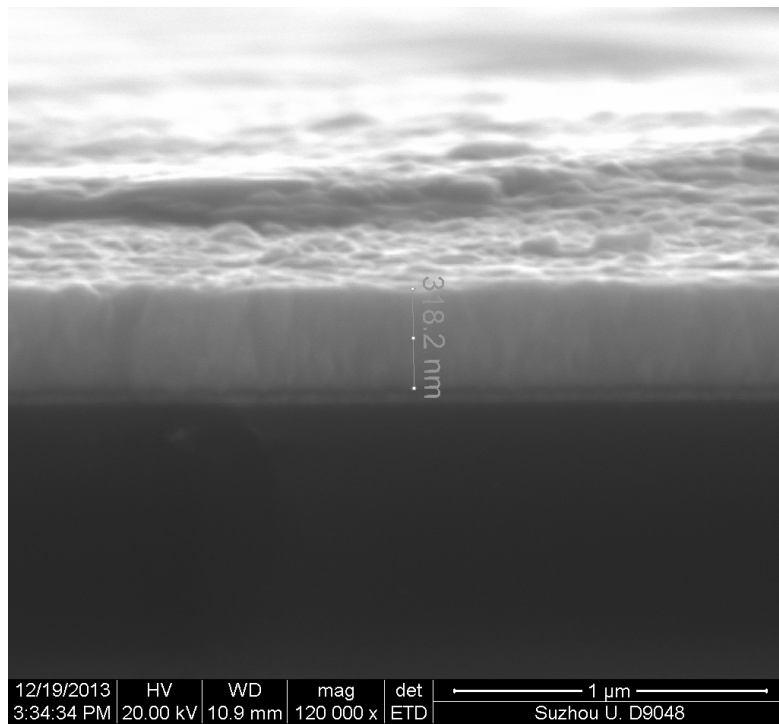


Figure S2: SEM image of the cross section of H₂-treated (8 mmol NaBH₄) hematite nanostructures.

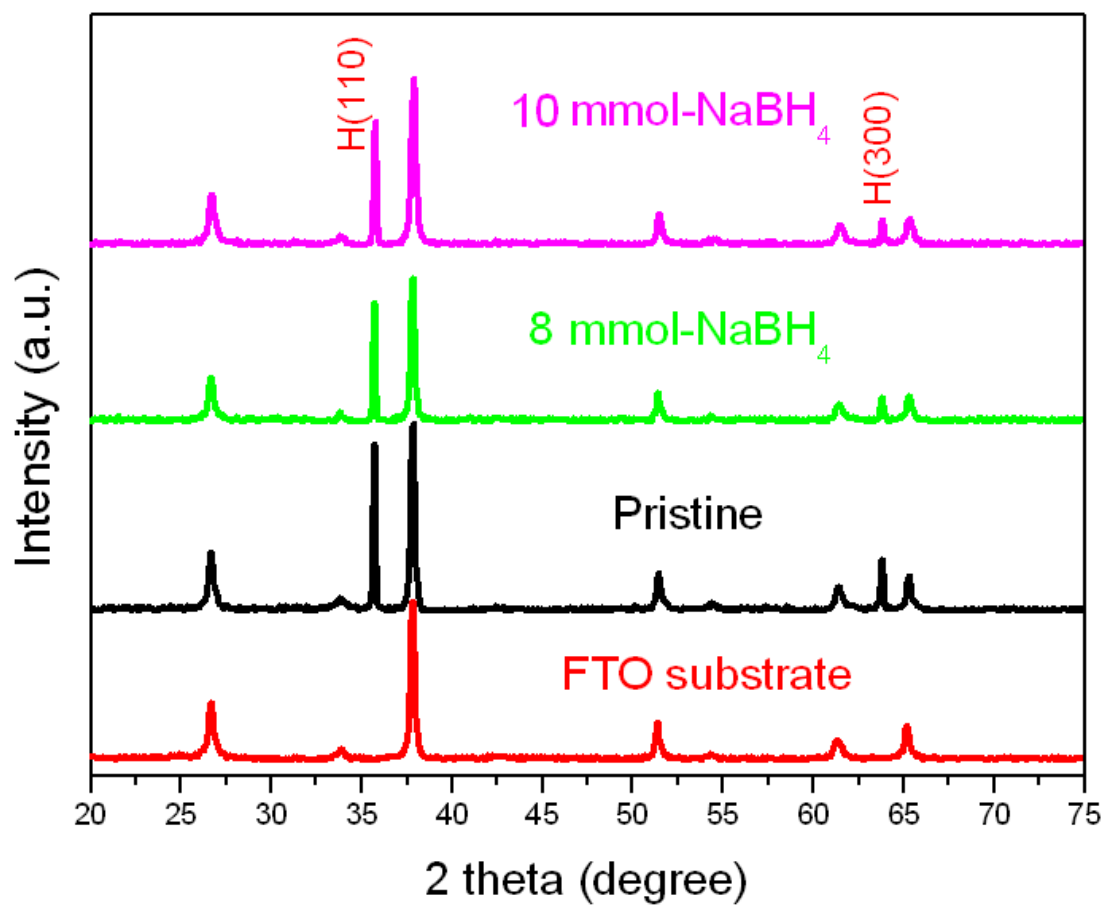


Figure S3: XRD spectra of hematite nanostructures before and after H₂ treatment.

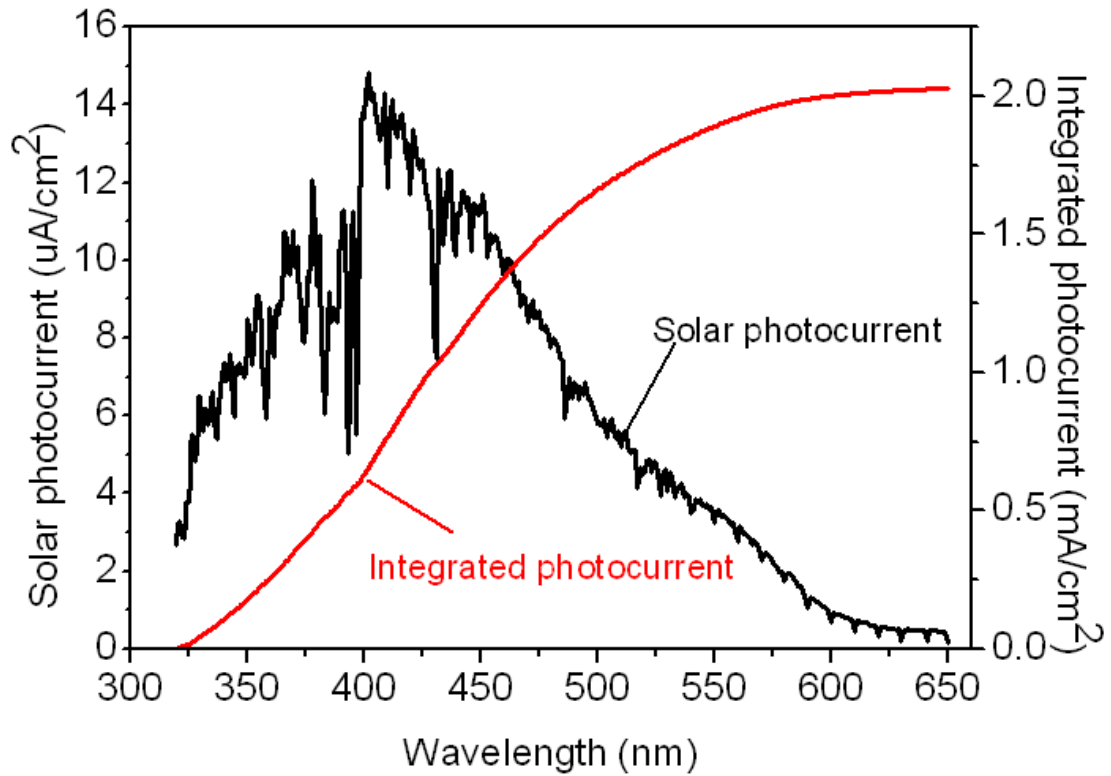


Figure S4: The integrated photocurrent of H₂-treated (8 mmol NaBH₄) hematite based on the IPCE data (320 nm to 650nm) at 1.23 V vs. RHE. The photocurrent density was calculated by integrating the IPCE spectra with a standard AM 1.5G solar spectrum (ASTMG-173-03), using the following equation (Ref: Nano Lett., 2011, 11, 3503):

$$I = \int_{320}^{650} \frac{1}{1240} \lambda \text{IPCE}(\lambda) E(\lambda) d(\lambda)$$

where $E(\lambda)$ is the solar spectral irradiance at a specific wavelength (λ) and $\text{IPCE}(\lambda)$ is the obtained IPCE profile as a function of wavelengths (λ) at 1.23 V vs. RHE. The calculated photocurrents are 2.03 mA cm⁻² at 1.23 V vs. RHE.

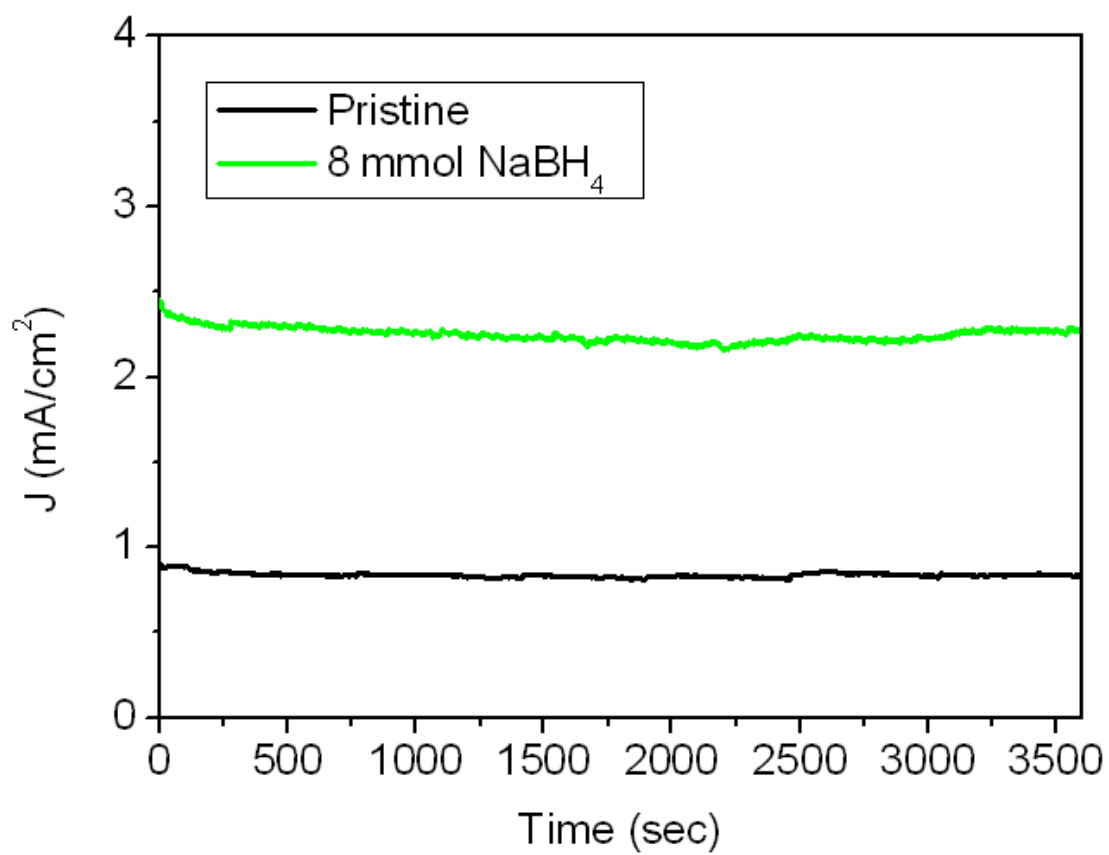


Figure S5: Photochemical stability curves for pristine and H_2 -treated (8 mmol NaBH_4) hematite electrodes collected at 1.23 V vs. RHE.

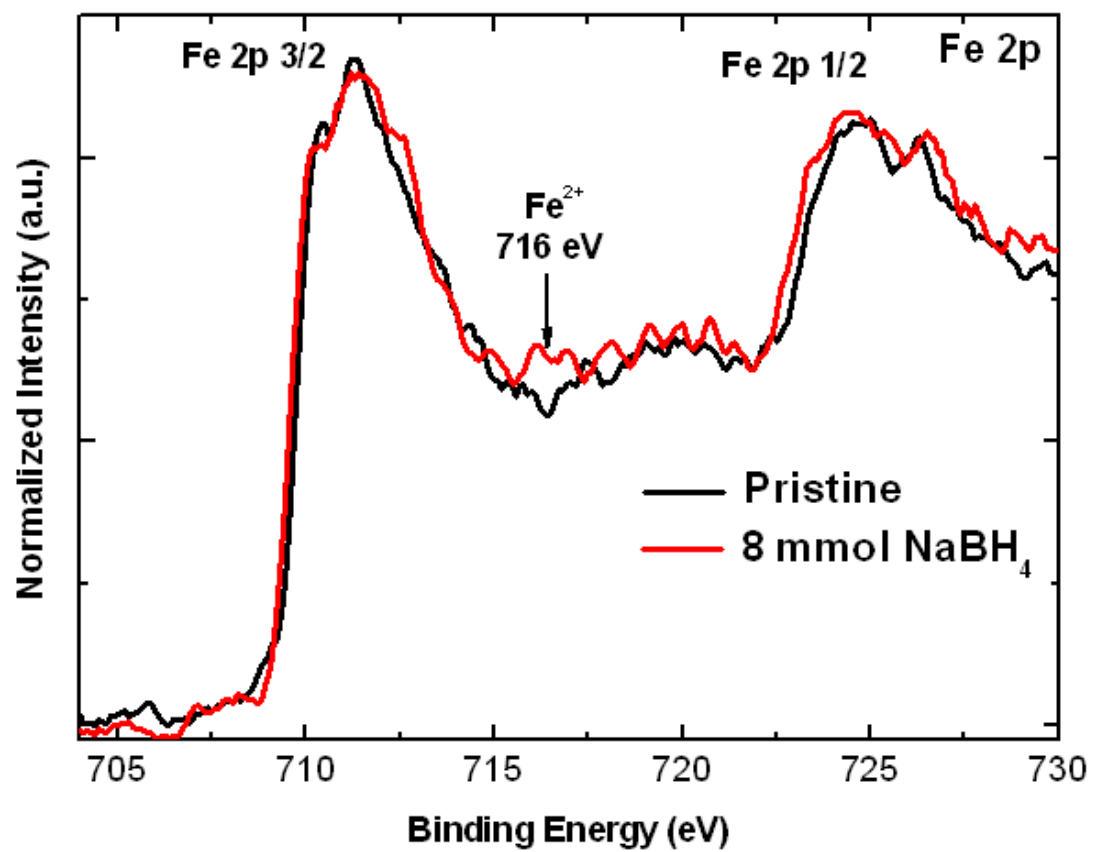


Figure S6: Fe 2p XPS spectra of pristine and H₂-treated (8 mmol NaBH₄) hematite nanostructures.