

## ESI Guide

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# Electronic Supplementary Information for Bioinspired wet-assembly fibers: from nanofragments to microhumps on string in mist

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## Supporting Information

This file contains:

Mechanism and optimizing of fragment-on-string fibers

Supplementary Figures S1-S3

Supplementary Analysis S1-S5

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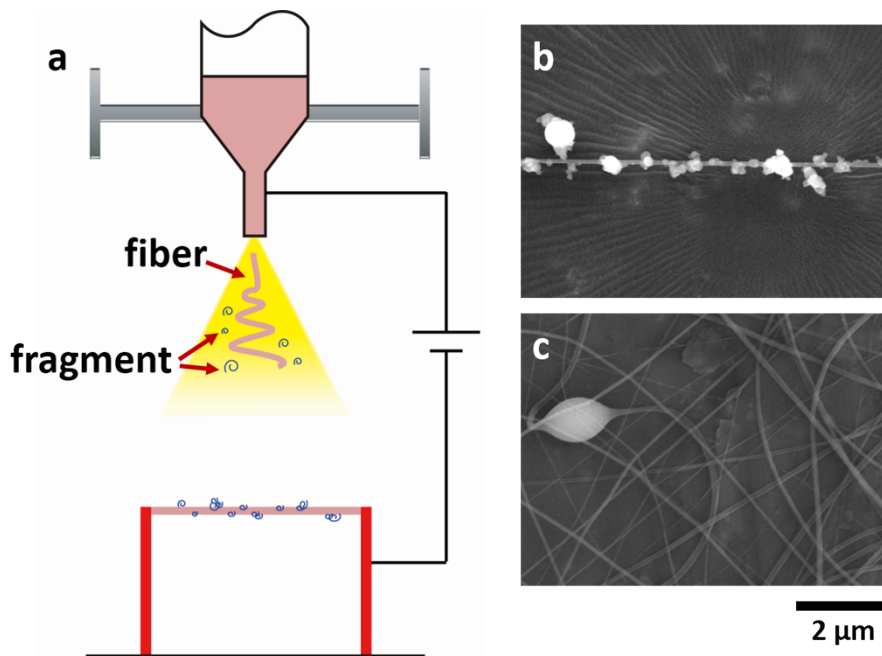
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## Mechanism and optimizing of fragment-on-string fibers

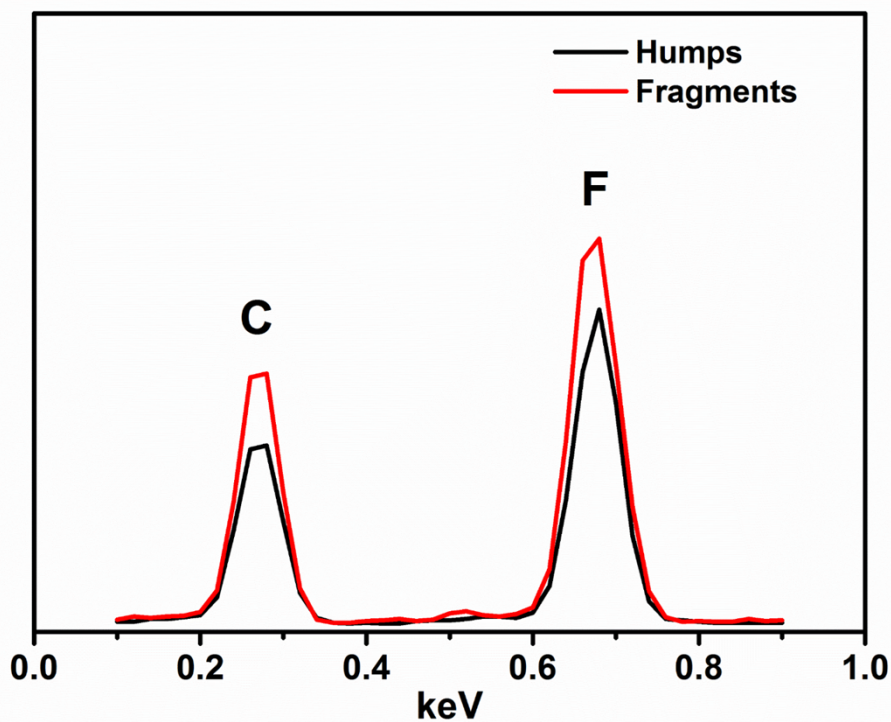
After collected by a shelf connected with the cathode (**Figure S1a**), the micro PVDF fiber can form a unique fragment-on-string structure, different from those commonly collected on the foil (**Figure S1b-c**), which may be owing to the remaining DMF on the foil, dissolving the fragments caught by the fiber. Those nanofragments may be produced during the electrospinning process and then caught by the hanging PVDF fibers which act as the fibrous filtration<sup>1-4</sup> because the fragments and humps are both composed of PVDF (**Figure S2**) and their spreading density on the fiber ( $V_p/V_f$ ,  $V_p$ ,  $V_f$  represent the total volume of the fragments and the fiber holding them) is almost linear dependent on the delay time of the shelf in the electrospinning chamber after the electrospinning ends (**Figure S3a**). Moreover, the density also has an obvious S-shaped relationship with the concentration of PVDF (**Figure S3b**), as the PVDF can completely dissolve in DMF (N,N-dimethylformamide) in low concentration, leaving no fragments electrospun and caught by the fiber while when the solution is too dense, the undissolved PVDF is still in large pieces, which cannot be electrospun either, which make the value of  $V_p/V_f$  increased slow in extremely low and high concentration, but sharp in the middle part. Thus, we chose PVDF concentration of 11 wt% and delay time of 12 h as the optimal results.

**Figure S1:**



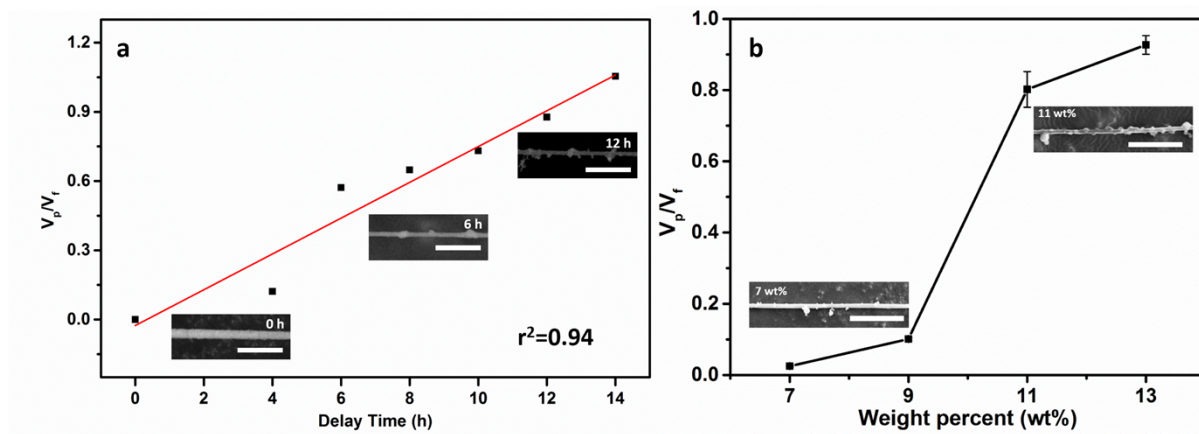
**Figure S1.** a, Illustration of the improved electrospun facilities. The spinnable PVDF solution is fed through a movable needle via a soft tube. When a high voltage is applied, an electro hydrodynamic approach is carried out. The electrospun fiber with nanofragments is formed via static delay time. The fibers on the shelf can catch the nanofragments (b) while those on the foil still remain smooth (c), due to the slow evaporating of DMF on the foil.

**Figure S2:**



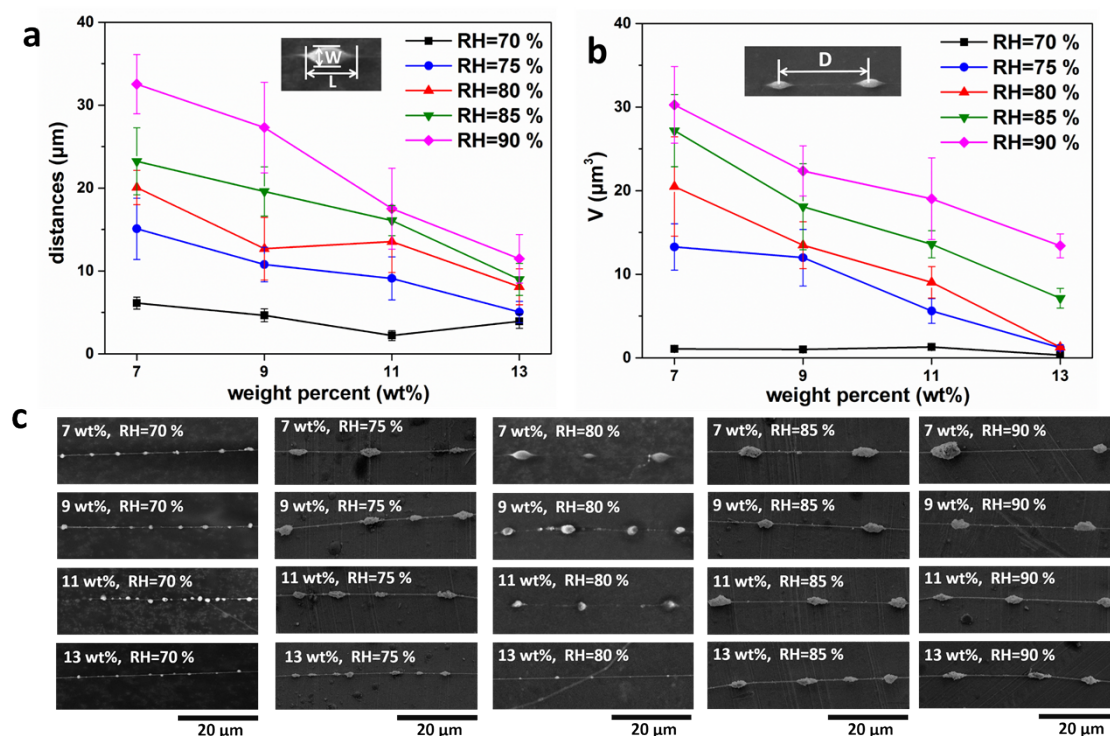
**Figure S2. EDS results of nanofragments (red) and microhumps (black).** The elements were identified as carbon (C) and fluorine (F), which can also originate from the PVDF structure. A semi-quantitative relationship of the chemical elements can be calculated based on their peak sizes in the spectrum. The results of humps and nanofragments we observed were  $\sim 0.57$  and  $\sim 0.58$ , respectively, which were quite near the theoretical one excluding errors and impurity of PVDF ( $\sim 0.63$ ).

**Figure S3;**



**Figure S3.** The relationship between density of fragments on the fiber with delay time and PVDF concentration. The density has a linear growth with the delay time continuing (a) while an S-shaped relationship with the PVDF concentration. Scale bar: 2  $\mu\text{m}$ .

**Figure S4**



**Figure S4.** The average distance between the humps and hump volumes in different PVDF concentration and RH. a, b) Relationship of hump distances and hump volumes of different PVDF concentration at different RH. With the increase of PVDF concentration, both distances (a) and hump volumes decrease. c) SEM images of the self-assembled microhumps of different PVDF concentration (7 wt%, 9 wt%, 11 wt% and 13 wt%) when RH=70 %, 75 %, 80 %, 85 % and 90 % (wet-assembly time=30 s).

## References

- 1 H.-S. Park and Y. Park, *Korean J. Chem. Eng.*, 2005, **22**, 165-172.
- 2 R. Sahay, P. S. Kumar, R. Sridhar, J. Sundaramurthy, J. Venugopal, S. G. Mhaisalkar and S. Ramakrishna, *J. Mater. Chem.*, 2012, **22**, 12953-12971.
- 3 J. Schaeffer and L. Olson, *Filtr. Sep.*, 1998, **35**, 124-129.
- 4 K. M. Yun, C. J. Hogan Jr, Y. Matsubayashi, M. Kawabe, F. Iskandar and K. Okuyama, *Chem. Eng. Sci.*, 2007, **62**, 4751-4759.

## Analysis S1:

If we note Re and Pe as Reynolds number<sup>1</sup> and Péclet number<sup>2</sup> of the stream, respectively, then we have:

$$\text{Re} = \frac{\rho v_0 d_f}{\mu} \quad (\text{I-1})$$

and

$$\text{Pe} = \frac{v_0 d_f}{D} \quad (\text{I-2})$$

where  $\rho$ ,  $v_0$ ,  $\mu$  are the mean density, mean velocity and intrinsic viscosity of the stream, respectively;  $d_f$ ,  $D$  are the diameter of the fiber and the diffusion coefficient which can be given by Stokes-Einstein equation<sup>3,4</sup>:

$$D = \frac{k_B T C_u}{3\pi\mu d_p} \quad (\text{I-3})$$

where  $k_B$ ,  $T$ ,  $d_p$  are the Boltzmann constant, absolute temperature and diameter of the particles, respectively, and  $C_u$  is Cunningham slip correction factor<sup>5</sup> due to the small diameter of the particles, which, under normal pressure and temperature, can be written as:

$$C_u = 1 + \frac{2\lambda}{d_p} [1.257 + 0.4 \exp(-1.1 \frac{d_p}{2\lambda})] \quad (\text{I-4})$$

where  $\lambda$  is the mean free path of gas molecules ( $\lambda=0.066 \mu\text{m}$ ).

In our experiment,  $T=293 \text{ K}$ ,  $v_0=0.1 \text{ m/s}$ , if we assume  $d_f \approx 200 \text{ nm}$ ,  $d_p \approx 1 \mu\text{m}$  (the diameter of water drops produced by our ultrasonic nebulizer are in the range of  $0.5 \sim 5 \mu\text{m}$ ) and the steam is saturated steam, whose intrinsic viscosity is  $9.727 \times 10^{-6} \text{ Pa}\cdot\text{s}^{-1}$ , then we can easily get  $C_u \approx 1.166$  and  $D \approx 5.174 \times 10^{-11} \text{ m}^2\cdot\text{s}^{-1}$ . According to Eq. (I-1) and (I-2), we will have  $\text{Pe} \approx 400 \gg 1$  and  $\text{Re} \approx 4 \times 10^{-11} < 1$ .

## References

1. Reynolds, O. An experimental investigation of the circumstances which determine whether the motion of water shall be direct or sinuous, and of the law of resistance in parallel channels. *Proc. R. Soc. London* **35**, 84-99 (1883).
2. Squires, T. M. & Quake, S. R. Microfluidics: Fluid physics at the nanoliter scale. *Rev. Mod. Phys.* **77**, 977-1026 (2005).
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Analysis S2:

In the range of  $Pe \gg 1$  and  $Re < 1$ ,  $\eta_f$  can be calculated through Langmuir's equation<sup>1</sup>:

$$\eta_f = \frac{1}{2La} \left( 2x \ln x - x + \frac{1}{x} \right)$$

$$La = 2 - \ln Re$$

$$x = 1 + 1.308 \left( \frac{La}{Pe} \right)^{\frac{1}{3}} \quad (II-1)$$

which can be simplified into

$$\eta_f = 1.75 \frac{1}{La^{\frac{1}{3}}} Pe^{\frac{2}{3}} = 1.75 \left[ 2 - \ln \left( \frac{\rho v_0 d_f}{\mu} \right) \right]^{\frac{1}{3}} \left( \frac{v_0 d_f}{D} \right)^{\frac{2}{3}} \quad (II-2)$$

As  $\rho$ ,  $v_0$ ,  $\mu$  and  $D$  are all gas constant which are irrelevant to  $d_f$ , Eq. (II-2) can be seen as a function of  $d_f$ :

(II-3)

$$\eta_f = f(d_f) = C(-3.18 - \ln d_f)^{-1/3} d_f^{-2/3}$$

As the order of magnitude of  $d_f$  is about  $10^{-7}$  (a few hundred nanometers), we can get

$$\eta_f \approx C_0 \cdot d_f^{-\frac{2}{3}} > 0 \quad (II-4)$$

References:

1. Langmuir, I. & Suits, C. G. *The Collected Works of Irving Langmuir: Atmospheric phenomena*. (Published with the editorial assistance of the General Electric Company by Pergamon Press, 1961).



Analysis S3:

According to Young's law, we can first define the coalescence-induced difference of the surface energy ( $\Delta E_c$ ) as1:

$$\Delta E_c = \gamma_{LV} \Delta A_{LV} - \gamma_{LV} \Delta A_{SL} \cos \theta \quad (\text{III-1})$$

where  $\gamma_{LV}$ ,  $\Delta A_{LV}$ ,  $\Delta A_{SL}$ ,  $\theta$  is the surface energy of liquid-vapor interface, the changes of liquid-vapor and solid-liquid surface areas, and the Young's contact angle, respectively. We assume that the larger drop (with the radius of  $R_0$ ) were made from two smaller ones of the same size (both with the radius of  $r_0$ ), due to the mass conservation principle,  $R_0$  and  $r_0$  have a relationship :

$$2 \times \frac{4}{3} \pi r_0^3 = \frac{4}{3} \pi R_0^3$$

$$R_0 = 2^{\frac{1}{3}} r_0 \quad (\text{III-2})$$

Then Eq. (III-1) will be written as:

$$\Delta E_c = \gamma_{LV} (4\pi R_0^2 - 2 \times 4\pi r_0^2) - \gamma_{LV} \cos \theta \frac{\pi d_f^2}{4} (4r_0 - 2R_0)$$

$$= -\pi \gamma_{LV} (1.65r_0^2 + 0.37r_0 d_f^2 \cos \theta)$$

$$\approx -1.65\pi \gamma_{LV} r_0^2 < 0 \quad (\text{as } r_0 \gg d_f) \quad (\text{III-3})$$

References:

1. Boreyko, J. B. & Chen, C.-H. Self-Propelled Dropwise Condensate on Superhydrophobic Surfaces. *Phys. Rev. Lett.* **103**, 184501 (2009)

Analysis S4:

The capillary adhesion for a larger hanging-drop on a rough-curve fiber ( $f_c$ ) can be described as<sup>1,2</sup>:

$$f_c = r \cdot 4\pi b \gamma \sin \alpha \quad (\text{V-1})$$

where  $r$  is the roughness,  $\alpha$  is off-axis angle ( $\sin \alpha = 1 - h/R$ ,  $h$  is the height of drop above the fiber,  $R$  is the radius of drop),  $\gamma$  is the surface tension of water,  $b$  can be calculated by averaging the number of drop covering the beads and the spacing of beads, which is:

$$b \sim \frac{k b_r + n b_a}{k + n} \quad (\text{V-2})$$

where  $k$  and  $n$  were the number of ‘knots’ and of spacing between ‘knots’ covered by drop, respectively; and  $b_r$ ,  $b_a$  are the maximum and minimum radii of silk with periodic knots, respectively.

In our experiment, as the largest drop hung on one segment of fiber between two beads, we have  $k=2$ ,  $n=1$ ,  $b_r=R_{\text{beads}} \approx 2.16 \mu\text{m}$  (Fig. 3b) and  $b_a=R_{\text{fiber}} \approx 0.30 \mu\text{m}$  (Fig. 3b and Fig. S2b). According to Eq. (V-2),  $b$  is  $1.54 \mu\text{m}$ .

As the surface tension ( $f_c$ ) should be balanced by the gravity, we then have:

$$\begin{aligned} r \cdot 4\pi b \gamma \sin \alpha &= 4\pi \rho_w g R^3 / 3 \\ r &= \rho_w g R^3 / 3b \gamma \sin \alpha \end{aligned} \quad (\text{V-3})$$

Given  $\sin \alpha \approx 1$  (as  $h$  can be ignored relative to  $R$ ),  $r$  can be estimated roughly to  $\sim 7.98$ , larger than that of the natural spider silk<sup>1</sup>.

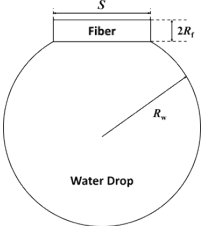
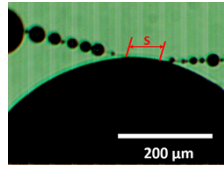
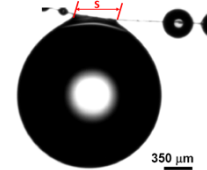
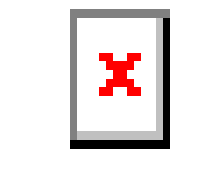
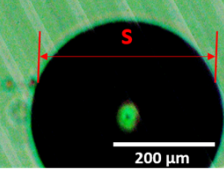
References:

1. Huang, Z., Chen, Y., Zheng, Y. & Jiang, L. Capillary adhesion of wetted cribellate spider capture silks for larger pearly hanging-drops. *Soft Matter* **7**, 9468-9473 (2011).
2. Wenzel, R. N. Resistance of solid surfaces to wetting by water. *Ind. Eng. Chem.* **28**, 988-994 (1936)

Analysis S5:

BWMF may collect a larger drop, which the volume ( $V_{\text{drop}}$ ) of drop can be estimated as following **table S1**, and some parameters can be defined:  $L$  is the wet-contact length of the fiber;  $R_f$ ,  $R_w$  is the radius of the fiber and the drop, respectively. Thus  $V_{\text{fiber}} = \pi L R_f^2$ , and  $V_{\text{drop}} = 4\pi R_w^3/3$ , where  $V_{\text{fiber}}$  is volume of fiber in wet-contact.

**Table S1** Estimation of the collected drop volume and compare of volumes

	Illustration	BWMF	Spider silk <sup>1</sup>	Dip-coating fiber <sup>2</sup>	Smooth PVDF fiber
					
$S$ ( $\mu\text{m}$ )		80	519	1130	338
$R_f$ ( $\mu\text{m}$ )		0.30	6.5	15	0.30
$R_w$ ( $\mu\text{m}$ )		650	950	1400	185
$V_{\text{drop}}$ ( $\mu\text{m}^3$ )	$4\pi R_w^3/3$	$1.15 \times 10^9$	$3.59 \times 10^9$	$1.14 \times 10^{10}$	$2.65 \times 10^7$
$V_{\text{fiber}}$ ( $\mu\text{m}^3$ )	$\pi S R_f^2$	22.6	$6.89 \times 10^4$	$7.99 \times 10^5$	95.6
$V_{\text{drop}}/V_{\text{fiber}}$		$5.09 \times 10^7$	$5.21 \times 10^4$	$1.43 \times 10^4$	$2.77 \times 10^5$

Note:  $S$  is wet contact length between fiber and drop.

References:

1. Huang, Z., Chen, Y., Zheng, Y. & Jiang, L. Capillary adhesion of wetted cribellate spider capture silks for larger pearly hanging-drops. *Soft Matter* **7**, 9468-9473 (2011).
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