Photoactive Fe₂O₃/Cu₂O Heterostructured

Nanocrystals

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Supporting Information



Figure S1. ¹H NMR spectrum of Fe(oleate)₃ precursor acquired in CDCl₃



Figure S2. IR Spectrum of Fe(oleate)₃ precursor



Figure S3. TGA scans of $Fe(oleate)_3$ and Cu(I) acetate under N_2 showing decomposition temperatures of the organometallic precursors



Figure S4. a) Representative TEM image and particle size distribution of isolated Cu_2O nanocrystals **b)** Representative TEM image and particle size distribution of isolated γ -Fe₂O₃ nanocrystals



Figure S5. Particle size distribution of the γ - Fe₂O₃ and Cu₂O domains in as-synthesized hetero-nanocrystals in dimer and oligomer morphologies.



Figure S6. a) Representative TEM image of HNCs showing dimer and trimer morphologies. b) TEM image of a physical mixture of γ -Fe₂O₃ and Cu₂O showing the absence of any ordering into hetero-architectures



Figure S7. a) & b) Low resolution TEM images of Fe_2O_3/Cu_2O HNCs after size selective precipitation removing the majority of trimers and higher oligomers. **c)** Particle size distribution following size-selective precipitation treatment of the Fe_2O_3/Cu_2O nanocrystals.



Figure S8. XPS survey spectra of a) γ -Fe₂O₃ nanocrystals b) Cu₂O particles and c) γ -Fe₂O₃/Cu₂O HNCs



Figure S9. Optical absorption spectra of **a**) pure γ -Fe₂O₃ **b**) pure Cu₂O and **c**) γ -Fe₂O₃/Cu₂O HNCs manipulated using the Tauc relation (see reference 58 in the manuscript) to determine their optical bandgaps



Figure S10. Infrared transmission spectra of as γ -Fe₂O₃/Cu₂O HNCs before and after ligand exchange with NOBF₄ showing removal of the oleate capping ligands