

Carboxylate- Intercalated Layered Double Hydroxides for H₂ Sorption

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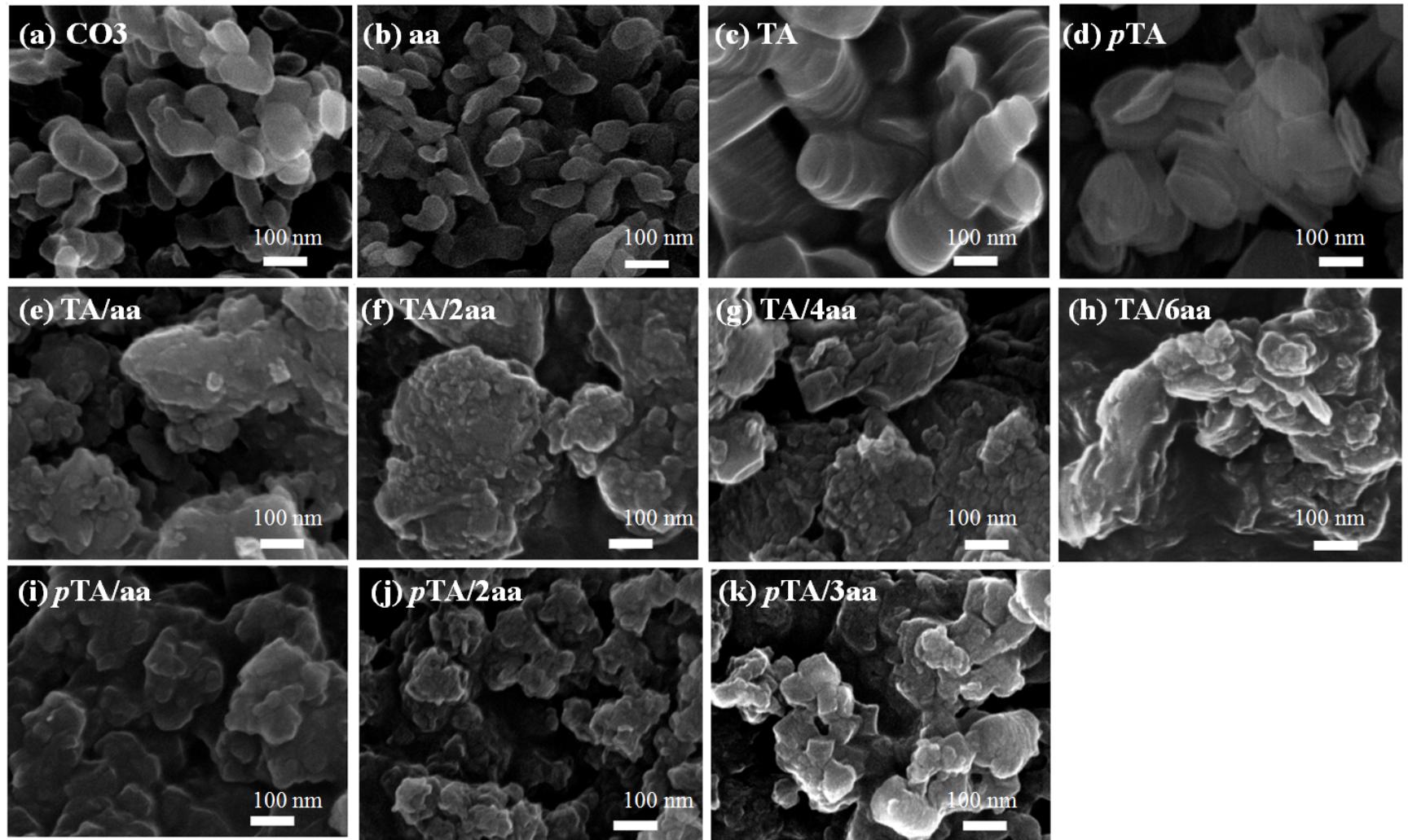


Figure S1. SEM micrographs of LDHs with different anions (a) CO₃, (b) aa, (c) TA, (d) pTA, (e) TA/aa, (f) TA/2aa, (g) TA/4aa, (h) TA/6aa, (i) pTA/aa, (j) pTA/2aa, and (k) pTA/3aa.

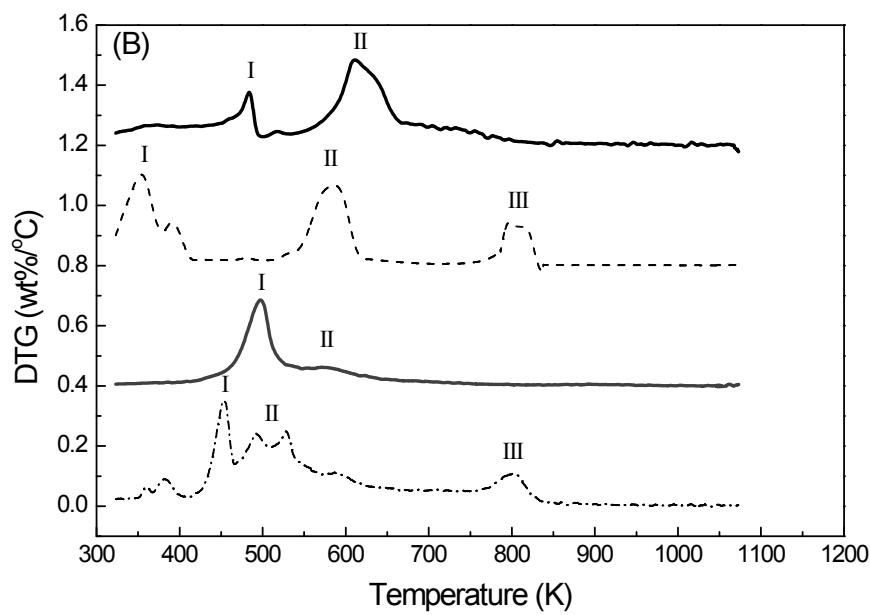
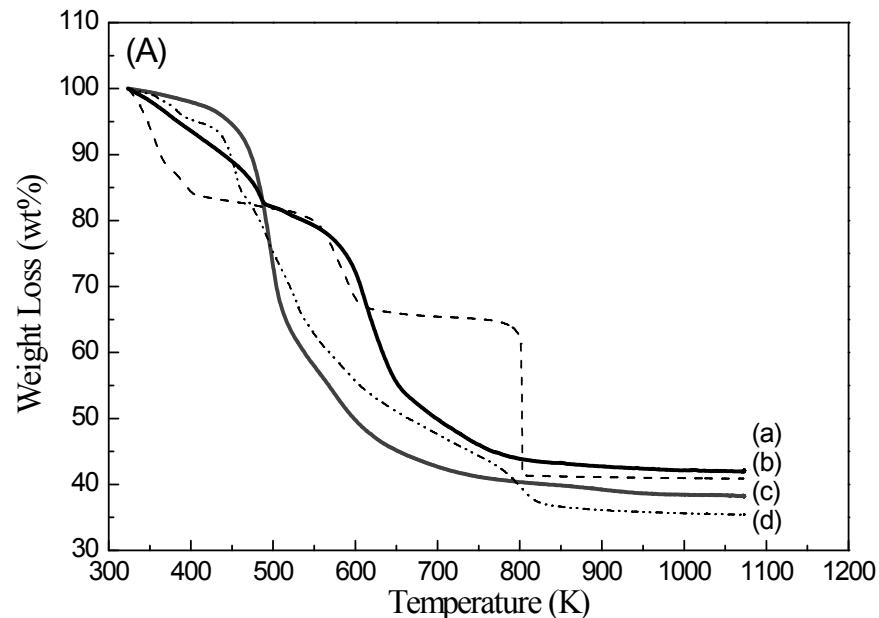


Figure S2. (A) TGA and (B) DTG profiles of LiAl- LDHs with different interlayer anions (a) aa⁻
 (b) TA²⁻ (c) CO₃²⁻, and (d) pTA⁻.

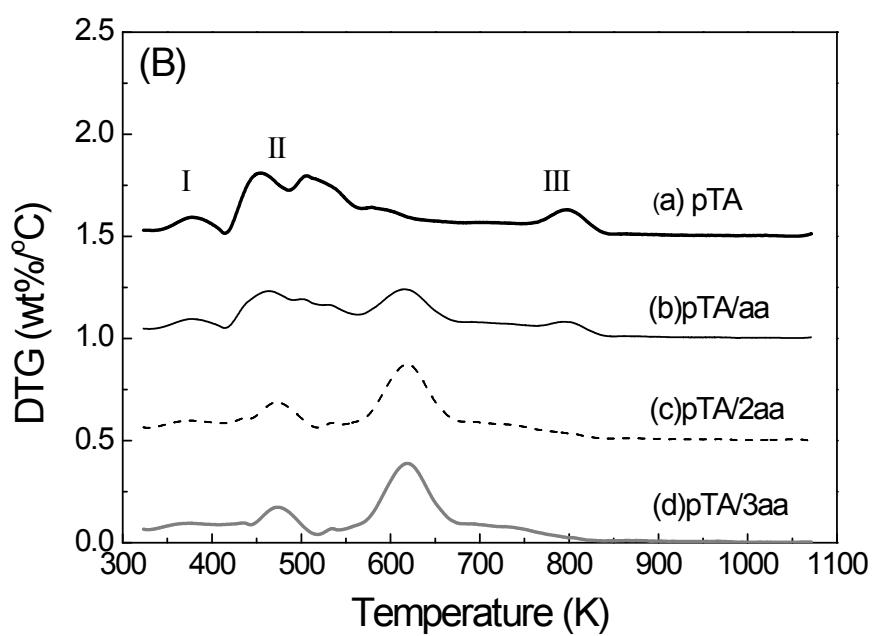
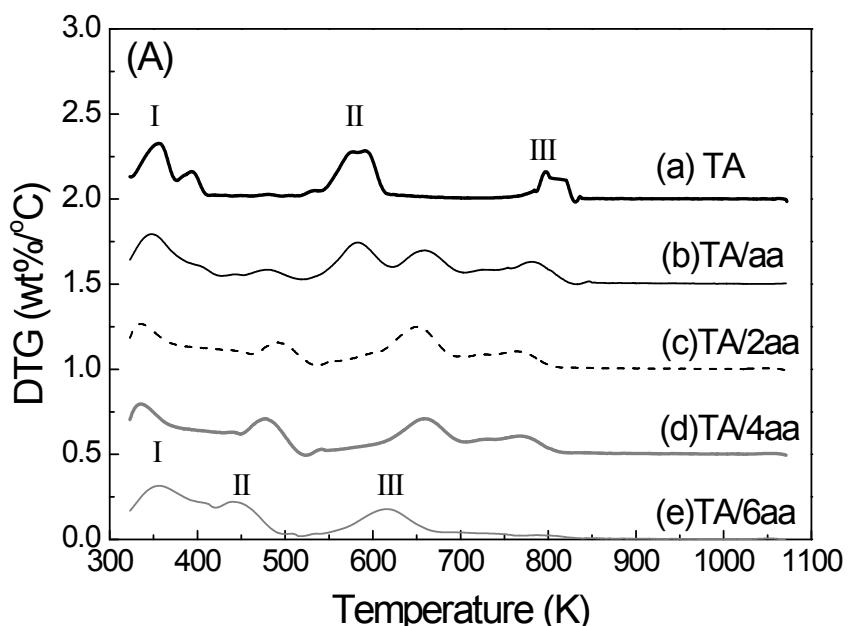


Figure S3. DTG profiles of LiAl- LDHs with mixed acetate and arylate anions (A) TA/naa and (B) pTA/naa.

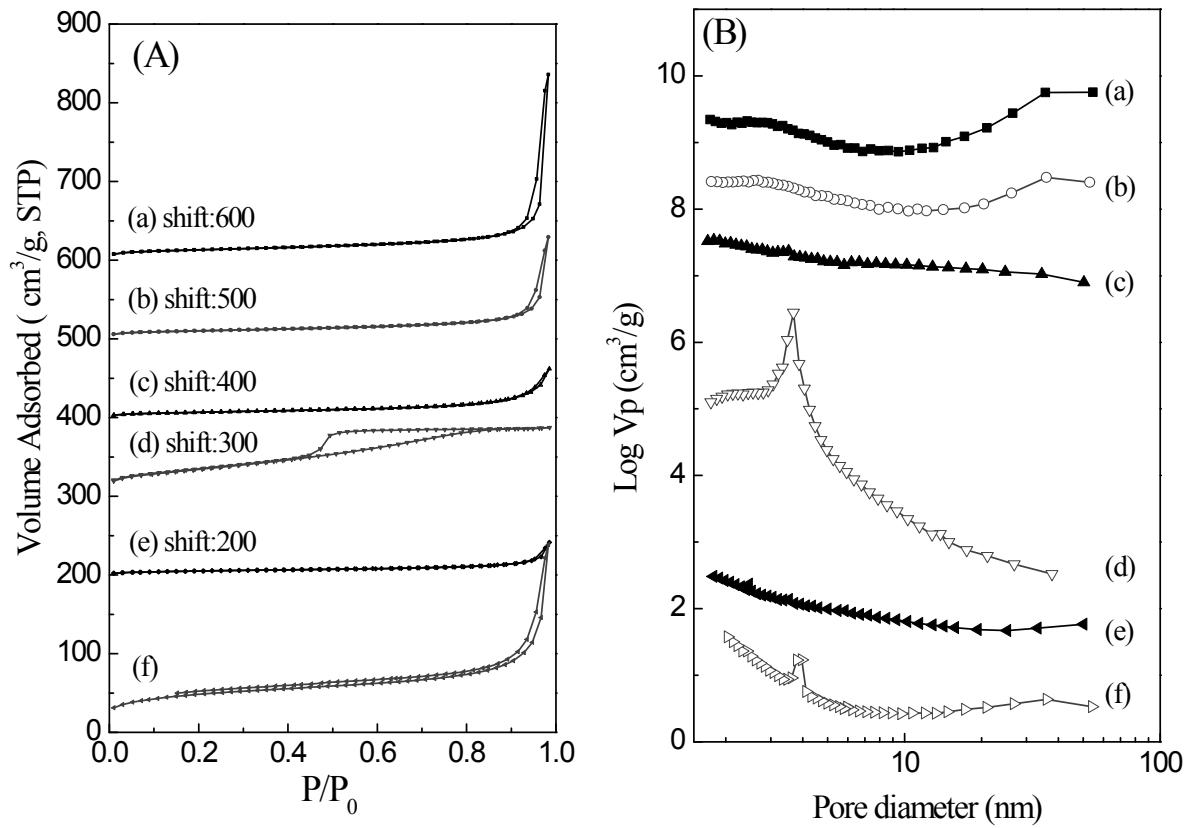


Figure S4. (A) N₂ adsorption-desorption isotherms and (B) BJH desorption pore size distribution of LiAl-LDHs with different interlayer anions:
 (a) CO₃, (b) aa, (c) TA, (d) TA/4aa, (e) *p*TA, and (f) *p*TA/3aa.

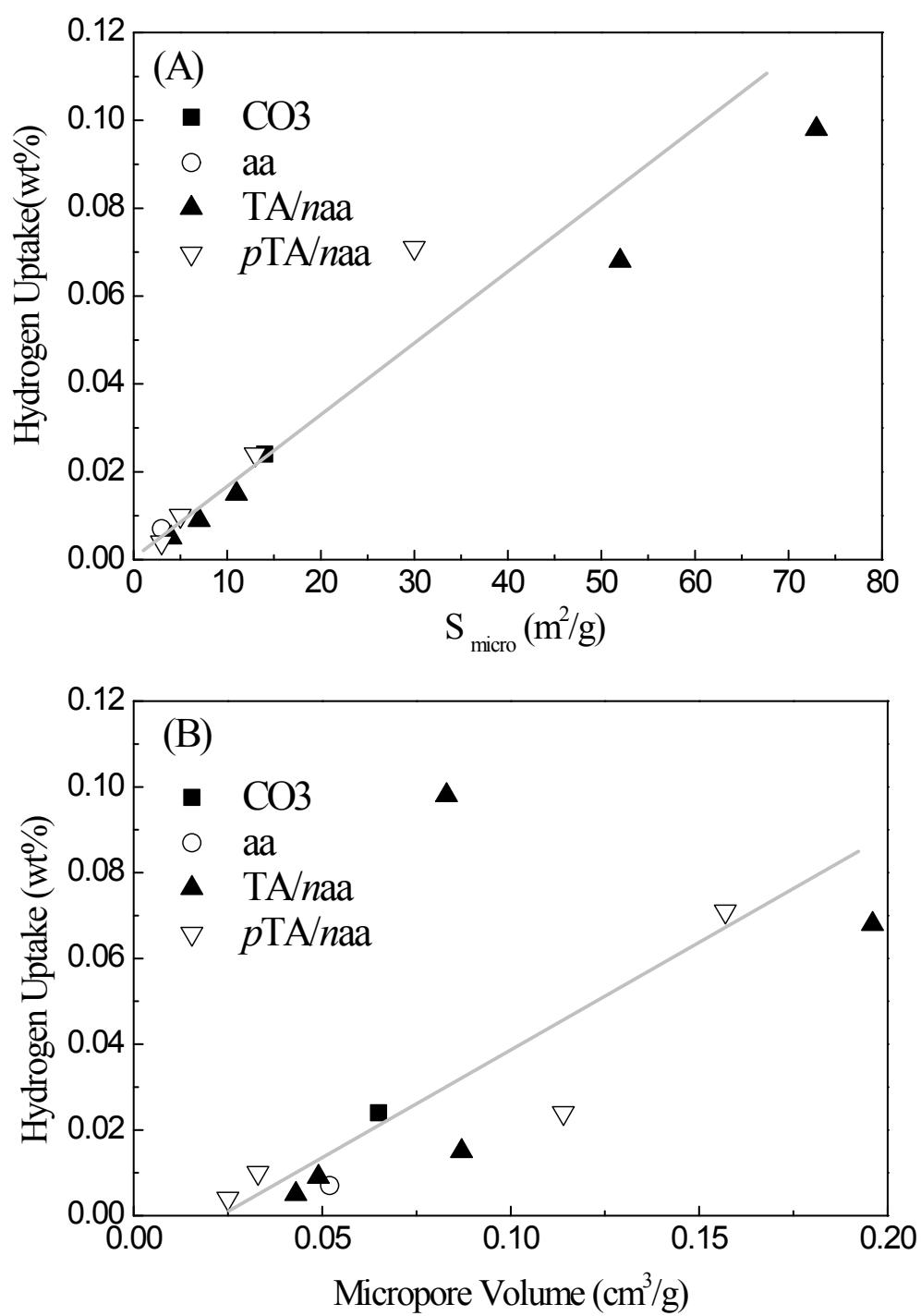


Figure S5. Relationship between the hydrogen uptake at 298K /100 atm and (A) microporous surface area, (B) micropore volume of LDHs with different interlayer anions.