# **Electronic Supplementary Information (ESI)**

## Gas uptake, molecular sensing and organocatalytic

# performances of a multifunctional carbazole-based conjugated

# microporous polymer

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Section 1. TGA Profile



Fig. S1 TGA curve of MFCMP-1 under nitrogen.

Section 2. FT IR Spectra



Fig. S2 IR spectra of MFCMP-1 (red line) and its monomer (black line).

Section 3. Powder X-Ray Diffraction



Fig. S3 PXRD curve of MFCMP-1.

Section 4. SEM, DLS and TEM



Fig. S4 (a) SEM image, (b) DLS profile and (c) TEM image of MFCMP-1.

Section 5. Uptake of CO<sub>2</sub> at high pressure



Fig. S5 Adsorption isotherm of MFCMP-1 for  $CO_2$  at 298 K and 10 bar.

#### Section 6. Adsorption Selectivity



Fig. S6  $CO_2/N_2$  and  $CO_2/CH_4$  initial slope selectivity studies for MFCMP-1.

#### Section 7. Stability of Water Vapor



**Fig. S7** (a) IR spectra of MFCMP-1 samples fresh (red line) and after reaction 5h in boiling water (black line) (b) Nitrogen adsorption/desorption isotherms ( $\bullet$ : adsorption, o: desorption) of MFCMP-1 samples after reaction 5h in boiling water, the surface area is 768 m<sup>2</sup> g<sup>-1</sup>.

## Section 8. FT IR Spectra



**Fig. S8** IR spectra of MFCMP-1 samples fresh (red line) and after six-time catalytic reaction (black line). It is evident that the structure of MFCMP-1 was maintained after catalytic reactions.

## Section 9. Catalytic Data

**Table S1.** The Knoevenagel condensation of benzaldehyde with malononitrile catalyzed by MFCMP-1 in varying solvent <sup>a)</sup>

NC

	$\rightarrow$ $^{O}_{+}$ $\stackrel{CN}{_{Sa}}$	CMP-1	$\sim$ CN + H	I <sub>2</sub> O
1a $2a$				
Entry	Solvent	Amount (mL)	Time (h)	<b>Yield</b> (%) <sup>b</sup>
1	Dioxane	0.8	4	8
2	THF	0.8	4	7
3	Toluene	0.8	4	5
4	EtOAc	0.8	4	4
5	DMF	0.8	4	5
6	H <sub>2</sub> O	0.8	4	48
7	Methanol	0.8	4	41
8	Ethanol	0.8	4	65
9	Ethanol/H <sub>2</sub> O	0.4 /0.4	4	76
10	MeOH/H <sub>2</sub> O	0.4 /0.4	4	79
11	DMF/H <sub>2</sub> O	0.4 /0.4	4	91
12	Dioxane/H <sub>2</sub> O	0.4 /0.4	4	99
13	Dioxane/H <sub>2</sub> O	0.4 /0.4	2	91
14	Dioxane/H <sub>2</sub> O	0.4 /0.4	1	82
15	Dioxane/H <sub>2</sub> O	0.4 /0.4	30min	68
16	Dioxane/H <sub>2</sub> O	0.4 /0.4	10min	32
<b>17</b> <sup>c)</sup>	Dioxane/H <sub>2</sub> O	0.4 /0.4	2	98
<b>18</b> <sup>d)</sup>	Dioxane/H <sub>2</sub> O	0.4 /0.4	10min	22
<b>19</b> <sup>d)</sup>	Dioxane/H <sub>2</sub> O	0.4 /0.4	30min	30
<b>20</b> <sup>d)</sup>	Dioxane/H <sub>2</sub> O	0.4 /0.4	1	39
<b>21</b> <sup>d)</sup>	Dioxane/H <sub>2</sub> O	0.4 /0.4	2	59
<b>22</b> <sup>d)</sup>	Dioxane/H <sub>2</sub> O	0.4 /0.4	4	76
<b>23</b> <sup>d)</sup>	Dioxane/H <sub>2</sub> O	0.4 /0.4	7	95

<sup>a)</sup> Rection conditions: benzaldehyde (0.2 mmol), malononitrile (0.2 mmol), MFCMP-1 (1.0 mol % of the substrate), 25 °C; <sup>b)</sup> Determined by GC using undecane as internal standard; <sup>c)</sup> malononitrile (0.3 mmol); <sup>d)</sup> MFCMP-1 0.5 mol % of the substrate.

## Section 10. Sorption Isotherms



**Fig. S9** Nitrogen adsorption ( $\bullet$ ) and desorption ( $\circ$ ) isotherm profiles of MFCMP-1 prepared in CH<sub>3</sub>CN/CHCl<sub>3</sub> (a) and CH<sub>3</sub>CN (b).

#### **Section 11. The Characterization of Products**



**2a:** White solid; m.p. 83-84 °C (lit.,<sup>2</sup> 84 °C); <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$  7.54 (t, J = 9.0 Hz, J = 6.0 Hz, 2H), 7.64 (t, J = 6.0 Hz, 1H), 7.78 (s, 1H), 7.92 (d, J = 9.0 Hz, 2H) ppm. GC-MS retention time 5.442 min., m/z (EI) 154 (M+, 81), 127 (98), 103 (100).

<sup>1</sup>H-NMR spectrum of 2a







#### GC-MS spectra of 2a



**2b:** Pale solid; m.p. 164-166 °C (lit.,<sup>3</sup> 167-168 °C); <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$  7.88 (s, 1H), 8.08 (d, *J* = 9.0 Hz, 2H), 8.39 (d, *J* = 9.0 Hz, 2H) ppm. GC-MS retention time 5.785 min., m/z (EI) 199 (M+, 13), 153 (16), 141 (23), 126 (35), 114 (28), 100 (29), 75 (49), 63 (29).

<sup>1</sup>H-NMR spectrum of 2b





#### GC-MS spectra of 2b



**2c:** White solid; m.p. 135-136 °C (lit.,<sup>4</sup> 136-138 °C); <sup>1</sup>H NMR

 $(300 \text{ MHz}, \text{ CDCl}_3) \delta 7.81-7.88 \text{ (m, 5H)}, 8.35 \text{ (d, } J = 9.0 \text{ Hz}, 1\text{H}), 8.45 \text{ (s, 1H) ppm.}$ GC-MS retention time 5.309 min., m/z (EI) 199 (M+, 16), 144 (18), 126 (39), 114 (47), 92 (100), 75 (53), 51 (50).

<sup>1</sup>H-NMR spectrum of 2c









**2d**: White solid; m.p. 155-157 °C (lit.,<sup>5</sup> 159-161 °C); <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$  7.69 (d, J = 9.0 Hz, 2H), 7.72 (s, 1H), 7.78 (d, J = 9.0 Hz, 2H) ppm. GC-MS retention time 5.349 min., m/z (EI) 234 (M+, 32), 153 (100), 126 (45), 99 (36), 75 (63), 63 (38), 50 (69).







GC-MS spectra of 2d



Br **2e**: White solid; m.p. 90-92 °C (lit.,<sup>5</sup> 92-93 °C); <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$  7.48 (m, 2H), 7.75 (d, J = 6.0 Hz, 1H), 8.12 (d, J = 6.0 Hz, 1H), 8.22 (s, 1H) ppm. GC-MS retention time 4.986 min., m/z (EI) 234 (M+, 21), 153 (100), 126 (38), 99 (24), 75 (37), 63 (22), 50 (30).

<sup>1</sup>H-NMR spectrum of **2e** 









**2f**: Yellow solid; m.p. 114-115 °C (lit.,<sup>2</sup> 116 °C); <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$  3.91 (s, 3H), 7.02 (d, *J* = 9.0 Hz, 2H), 7.65 (s, 1H), 7.92 (d, *J* = 9.0 Hz, 2H) ppm. GC-MS retention time 5.498 min., m/z (EI) 184 (M+, 66), 141 (35), 133 (26), 114 (100), 88 (38), 64 (28).

<sup>1</sup>H-NMR spectrum of **2f** 





GC-MS spectra of 2f



**2g**: Pale yellow solid; m.p. 79-80 °C (lit.,<sup>6</sup> 79-80 °C); <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$  3.93 (s, 3H), 6.99 (d, J = 9.0 Hz, 1H), 7.08 (t, J = 9.0 Hz, 1H), 7.58 (t, J = 9.0 Hz, 1H), 8.19 (d, J = 9.0 Hz, 1H), 8.31 (s, 1H) ppm. GC-MS retention time 5.208 min., m/z (EI) 184 (M+, 36), 119 (79), 114 (87), 91 (100), 88 (39), 78 (56), 63 (47), 51 (40).

<sup>1</sup>H-NMR spectrum of **2g** 









**2h**: White solid; m.p. 130-131 °C (lit.,<sup>3</sup> 132-134 °C); <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$  2.46 (s, 3H), 7.34 (d, J = 9.0 Hz, 2H), 7.72 (s, 1H), 7.82 (d, J = 9.0 Hz, 2H) ppm. GC-MS retention time 4.672 min., m/z (EI) 168 (M+, 59), 141 (63), 114 (51), 63 (75), 51 (63).

<sup>1</sup>H-NMR spectrum of **2h** 







**2i**: Bright orange solid; m.p. 164-165 °C (lit.,<sup>8</sup> 168-169 °C); <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>)  $\delta$  7.59–7.71 (m, 3H), 7.96 (d, J = 6.0 Hz, 2H), 8.12 (d, 6.0 Hz, 1H), 8.28 (d, 6.0 Hz, 1H), 8.66 (s, 1H) ppm. GC-MS retention time 6.338 min., m/z (EI) 204 (M+, 40), 177 (54), 150 (17), 126 (13), 75 (52), 63 (38), 51 (32).

<sup>1</sup>H-NMR spectrum of **2i** 



GC-MS spectra of 2i







**2j**: Light pink solid; m.p. 70-71 °C (lit.,<sup>9</sup> 72-73 °C); <sup>1</sup>H NMR

 $(300 \text{ MHz}, \text{CDCl}_3) \delta 6.72 \text{ (dd, 1H)}, 7.36 \text{ (d, } J = 3.0 \text{ Hz}, 1\text{H}), 7.52 \text{ (s, 1H)}, 7.81 \text{ (d, } 3.0 \text{ Hz}, 1\text{H}) \text{ ppm. GC-MS retention time } 3.287 \text{ min., m/z} \text{ (EI) } 144 \text{ (M+, 100)}, 115 \text{ (59)}, 89 \text{ (68)}, 62 \text{ (53)}.$ 

<sup>1</sup>H-NMR spectrum of **2**j





GC-MS spectra of 2j



2k: Bright yellow solid; m.p. 299-301 °C; <sup>1</sup>H

NMR (300 MHz, CDCl<sub>3</sub>) δ 7.86 (s, 2H), 7.96 (s, 2H) ppm. GC-MS retention time 12.854 min., m/z (EI) 237(M+, 15), 209 (6), 185 (6), 121(4), 94 (5).

<sup>1</sup>H-NMR spectrum of **2k** 





GC-MS spectra of 2k



**2l**: Light yellow oil; <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>) δ 1.68 (t, 2H), 1.83 (t, 4H), 2.66 (t, J = 6.0 Hz, 4H) ppm. GC-MS retention time 3.554 min., m/z (EI) 146 (M+, 5.01), 92 (13), 81 (13), 68 (19), 55 (100).

<sup>1</sup>H-NMR spectrum of 21





GC-MS spectra of 21



#### **Section 12. References**

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