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Supporting Information

**Controlled Mechanical Cleavage of Bulk Niobium Diselenide
to Nanoscaled Sheet, Rod, and Particles Nanostructures for
Pt-free Dye-sensitized Solar Cells**

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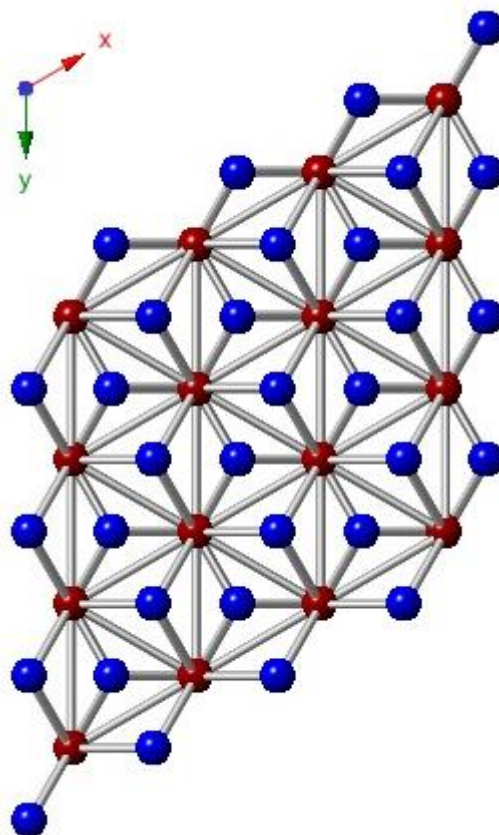


Figure S1. Crystal structure model of NbSe₂: The (001) section of the crystal structure model of the NbSe₂ supercell; blue and red balls represent Se and Nb atoms, respectively. Six equivalent Nb–Se distances are found within the trigonal prism around Nb. The calculated lattice constants ($a = b = 3.0 \text{ \AA}$; $c = 12.547 \text{ \AA}$) are consistent with the standard data recorded in CPDS 01-089-4313.

S1 Scale-up:

This method can be scaled-up readily to produce large quantities (liters) of dispersions. At an initial concentration of 0.5 wt% and a grinding time 10 h we produced 250 mL of the dispersion in NMP (Figure S2) with a comparable concentration (0.3 wt%) and very high stability for a very long time period (> 6 month). This method has the advantage of allowing large volumes of dispersions to be prepared at once.



Figure S2. High yield of nanomaterials dispersion: Photograph of a high-quality dispersion of NbSe₂ in NMP (250 mL, 0.3 wt%).

S2 Sedimentation Tests:

We performed sedimentation tests by storing NbSe₂ dispersions (prepared with and without grinding) in NMP for more than six months. The material prepared without grinding precipitated completely after several minutes; in contrast, the dispersions of NbSe₂ nanosheets and nanorods were highly stable, without any precipitation. Thus, the grinding process increased the compatibility between the NbSe₂ nanomaterials and the organic solvent, increased the repulsive forces among the particles, and prevented the materials from undergoing serious aggregation.

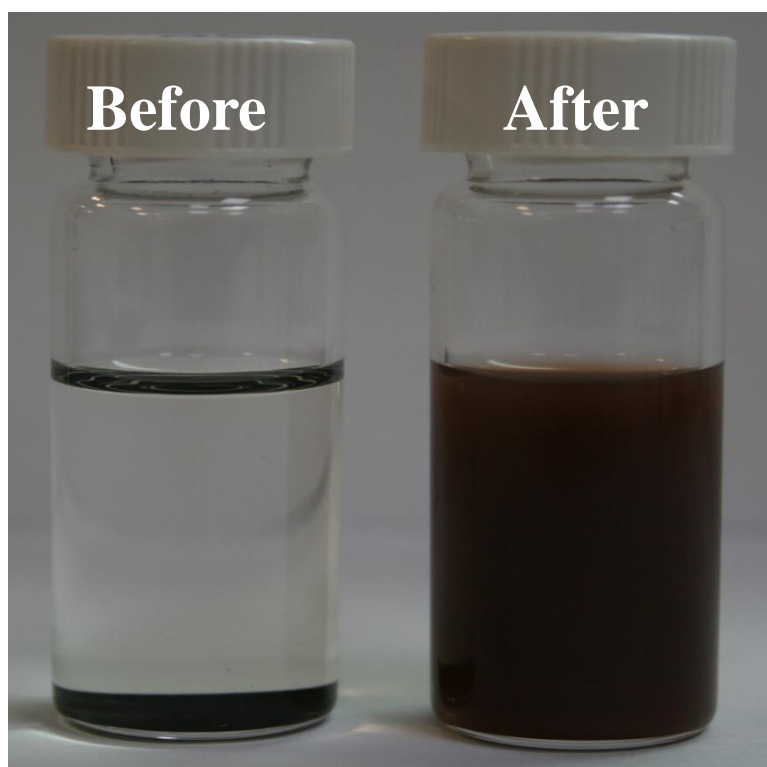


Figure S3. Sedimentation test: NbSe₂ dispersions (3 mL) before and after grinding for 10 h in NMP (starting concentration: 0.5 wt %). The dispersion prepared without grinding resulted in complete precipitation at the bottom of the vial within several minutes.

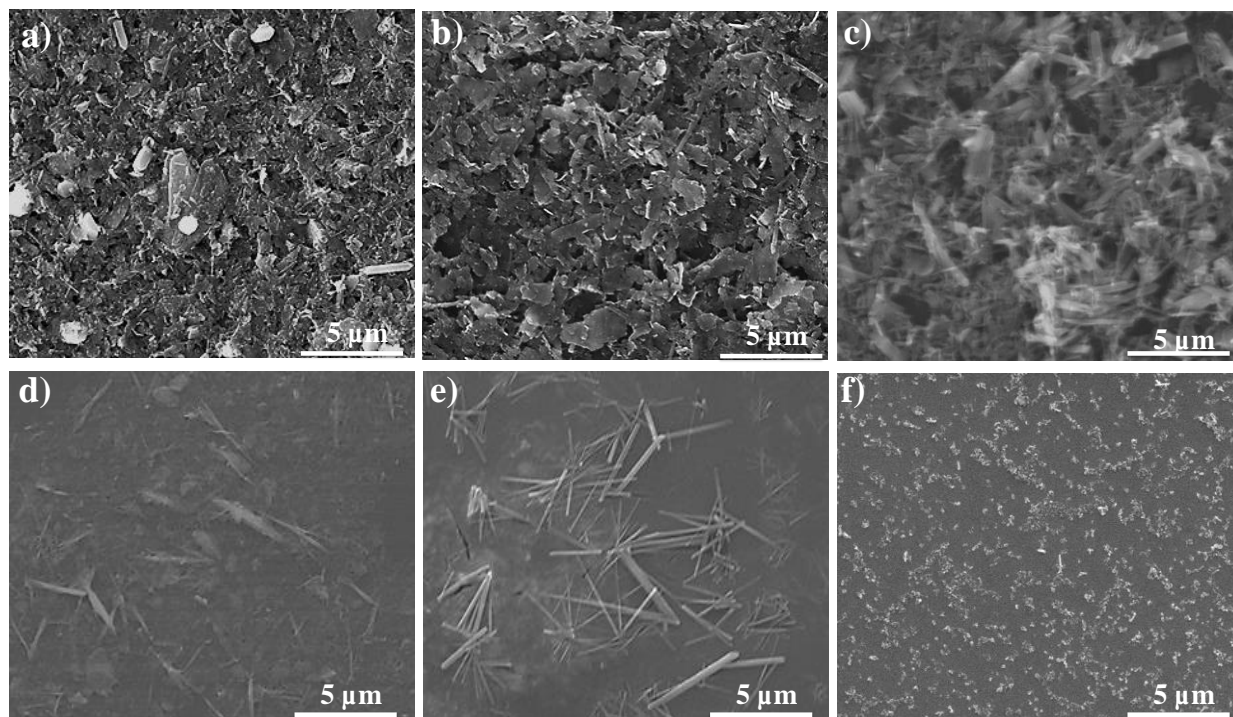


Figure S4. Morphologies of NbSe₂ materials during the conversion process: SEM images of the samples obtained at different time intervals during the conversion process from a 3D bulk NbSe₂ material into 2D nanosheets and then to 1D nanorods under the effect of mechanical forces: (a) pristine NbSe₂; after (b) 2, (c) 4, (d) 6, (e) 8, and (f) 10 h. Each dispersion was diluted tenfold with IPA and then drop-cast onto Si/SiO₂ and annealed (70 °C, 10 min).

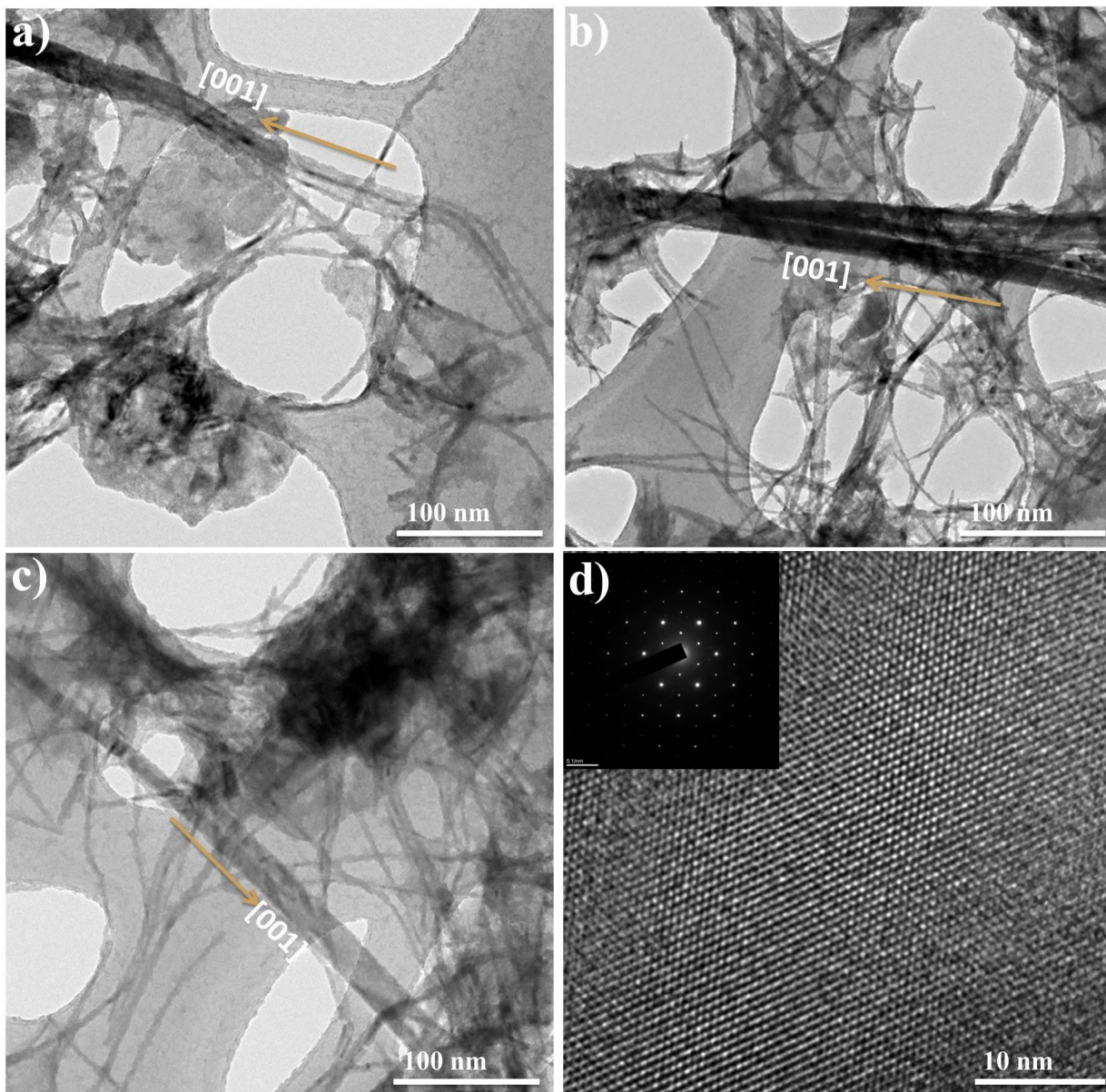


Figure S5. TEM analysis of fractured NbSe₂ nanostructures: (a, b, and c) TEM images for different fractured nanorods (the direction of fracturing showed with arrow) and (d) HRTEM image of flattened area of (c): The inset is the corresponding SAED of (d).

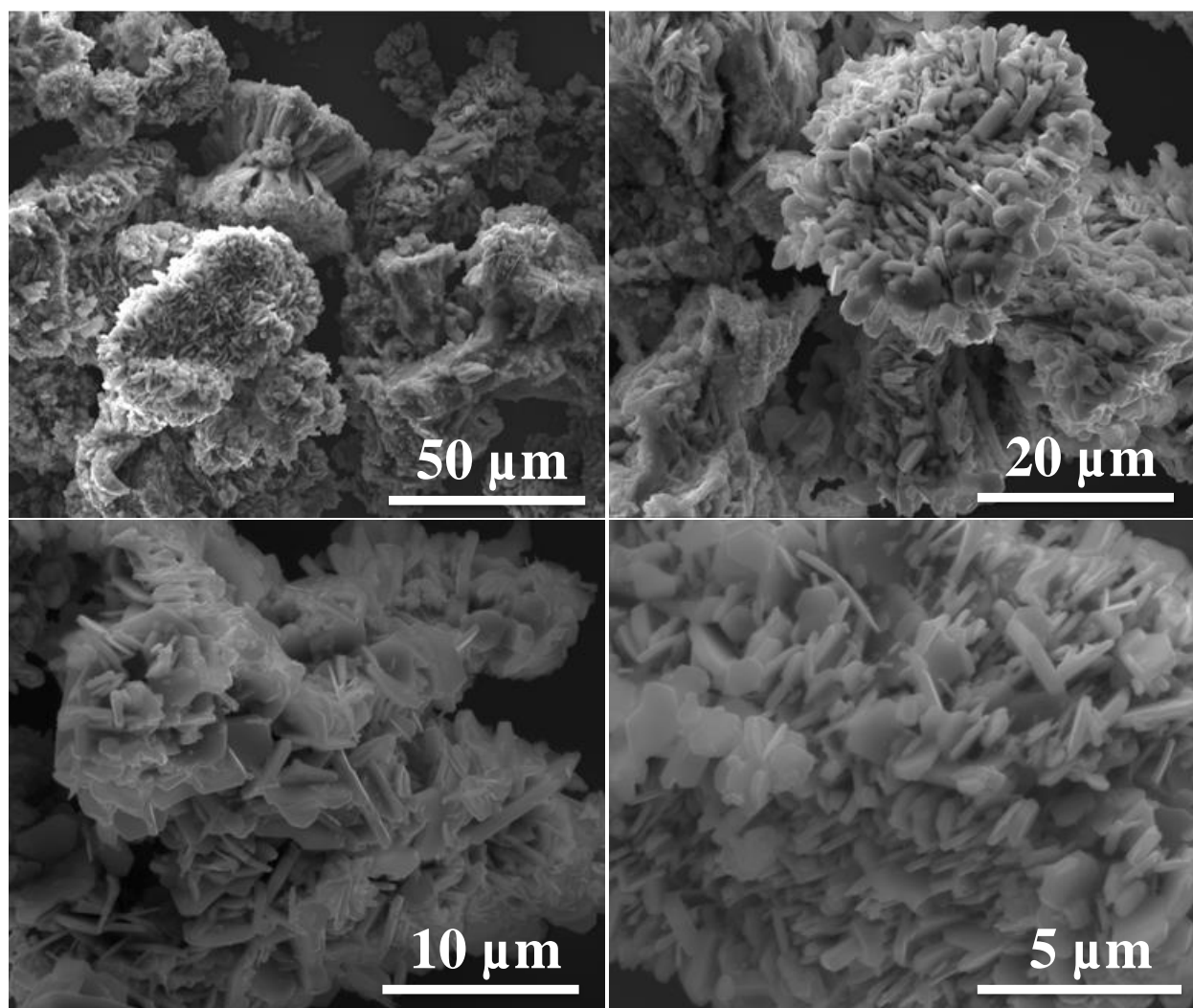


Figure S6. Morphology of pristine NbSe₂ materials: SEM images at different magnifications of the 3D network in the pristine NbSe₂ material, which consisted of a highly dense, layer-by-layer arrangements of single sheets with large lateral dimensions (>1 μm) and thicknesses of greater than 0.5 μm.

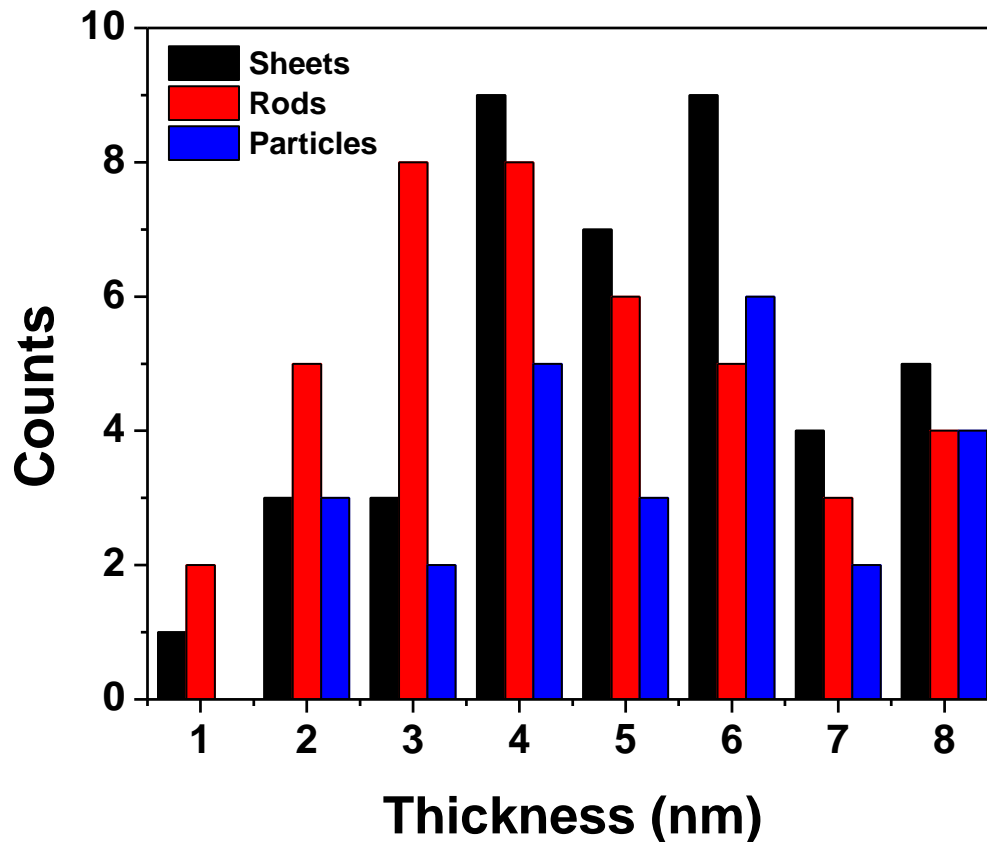


Figure S7. Histogram of NbSe₂ nanostructures: The thickness of the nanostructure from the height profile AFM images of 42 different nanosheets, 40 different nanorods, and 30 different nanoparticles. The average thicknesses approximately are 10, 5, and 3 nm for nanosheets, nanorods, and nanoparticles respectively.

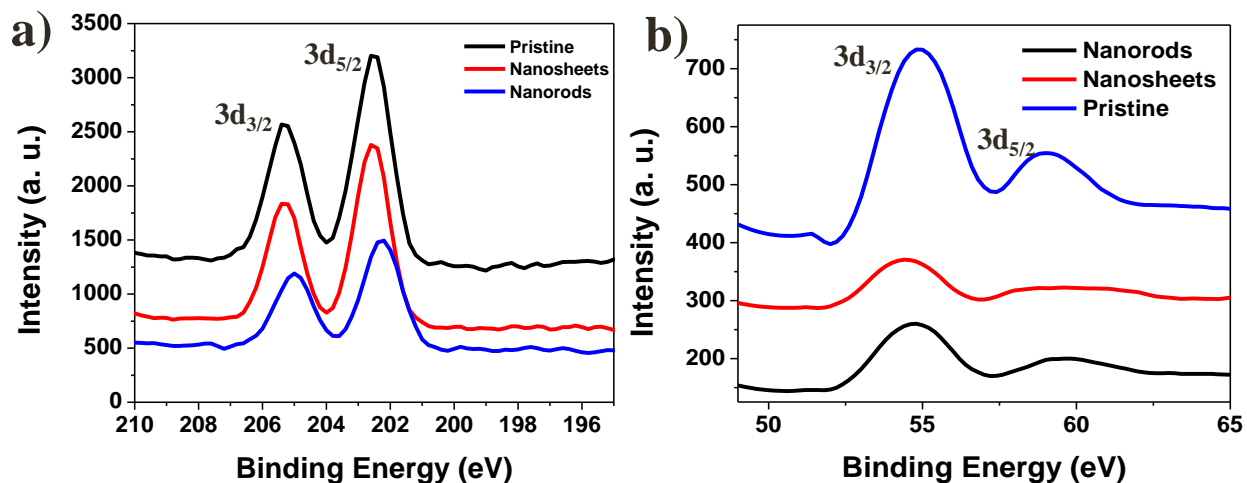


Figure S8. XPS analyses of NbSe₂ nanorods: XPS data of the binding energies of (a) Nb 3d and (b) Se 3d orbitals in NbSe₂ nanorod samples, prepared by placing drops of dispersions (diluted tenfold with IPA) onto Si/SiO₂ wafers and drying under an ambient atmosphere at 70 °C for 10 min. The peaks at 202.3 and 205.2 eV can be assigned to Nb 3d_{5/2} and Nb 3d_{3/2} orbitals, respectively; the peaks at 58 and 55.3 eV can be assigned to Se 3d_{5/2} and Se 3d_{3/2} orbitals, respectively. In addition, stoichiometry of niobium diselenide in the nanorods was confirmed from a Nb-to-Se atomic ratio of 1:2.

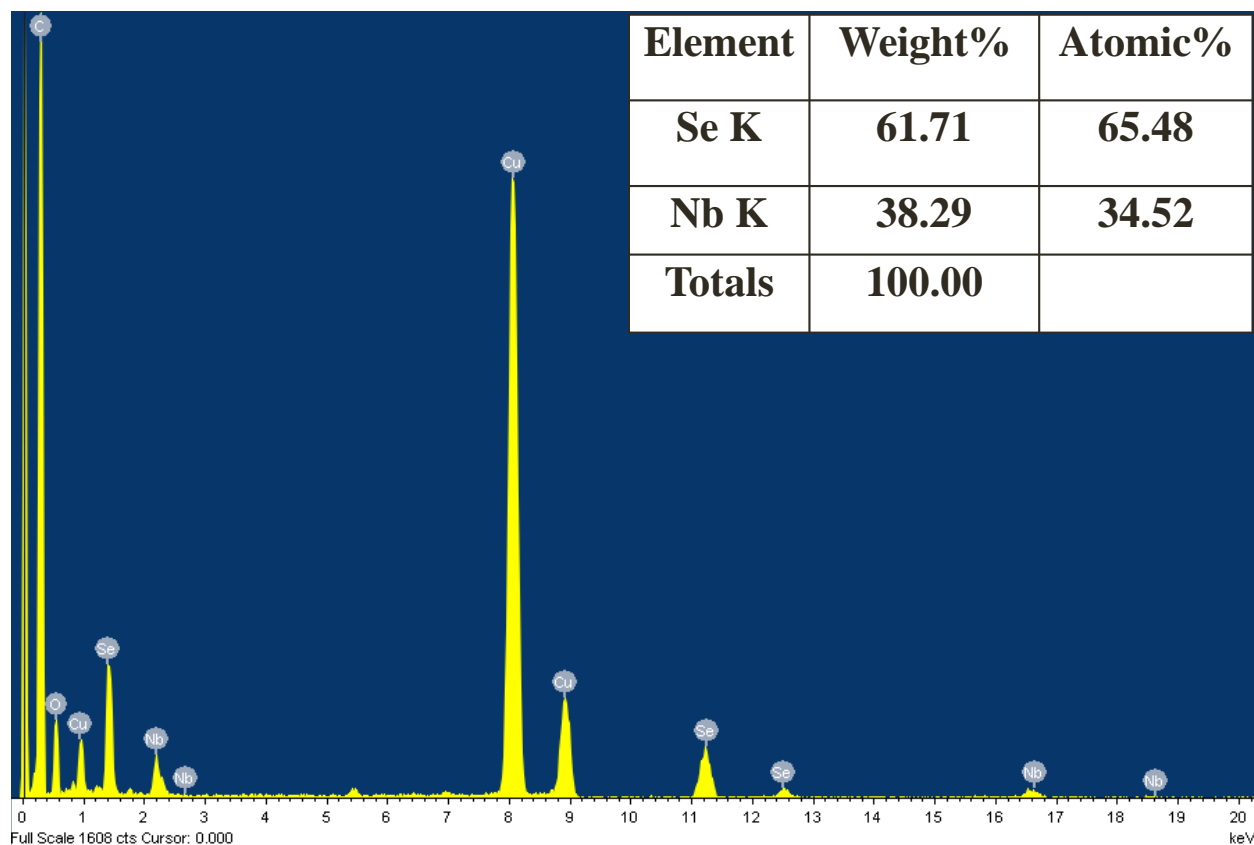


Figure S9. Compositional analysis of ultrathin NbSe₂ nanorods: Energy-dispersive X-ray analysis (EDS) of NbSe₂ nanorods revealing the atomic ratio of Nb-to-Se to be 1:1.9 (see inset Table), in agreement with the stoichiometric ratio of NbSe₂. Signals for copper atoms arose from the holey carbon-coated copper grid.

S3 Optical and Electrical Properties of thin films:

Both the transmittance (T) and sheet resistance (R_s) data of transparent conducting films are linked together and determined by the response of electrons to either static (voltage) or dynamic (light) electric fields. The relationship between them for a thin conducting film can be given by

$$T = \left[1 + \frac{Z_0}{2R_s} \frac{\sigma_{OP}}{\sigma_{DC}} \right]^{-2} \dots\dots\dots S1$$

Where Z_0 is the impedance of free space and has the value 377Ω . This relationship is controlled by the conductivity ratio, σ_{DC}/σ_{OP} , which can then be used as a Figure of Merit for transparent conducting films. It is well established that thick films give low values of T and high values of R_s , while thin films give high values of T and low values of R_s . The minimum industry standard for transparent conductors requires a value of σ_{DC}/σ_{OP} of 35 (giving $T = 90\%$ for $R_s = 100 \Omega/\text{sq}$). For most current-driven applications, however, more stringent conditions ($R_s < 10 @ T > 85\%$) are necessary.²⁻⁴ For films of NbSe₂ nanosheets and rods (Figure S11), the values of σ_{DC}/σ_{OP} were approximately equal to 1. Accordingly, the NbSe₂ films on PET exhibited the properties of typical graphene networks, making such films promising materials for use as transparent conductors.

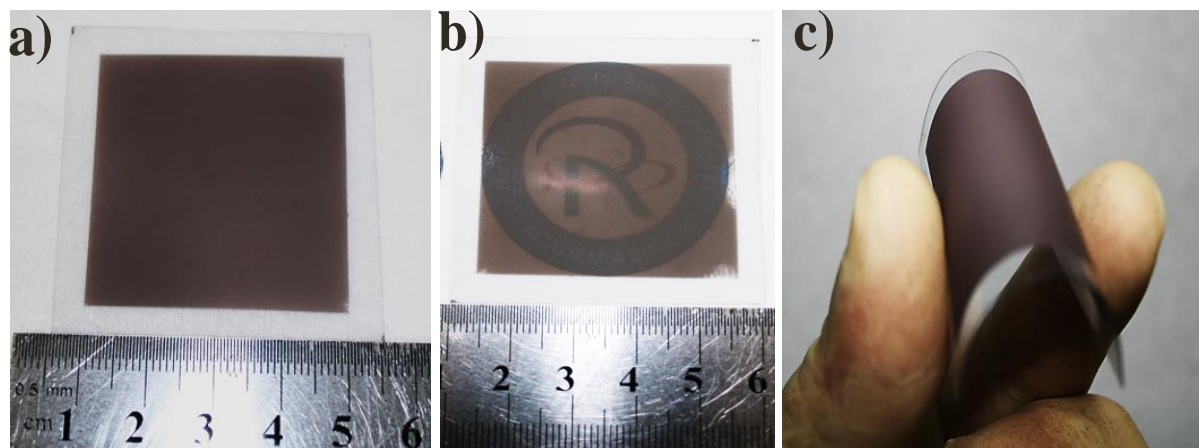


Figure S10. Large area thin film: Thin film of NbSe₂ prepared by spray-coating on a PET substrate. (a) 4 × 4 cm thin film; (b) semi-transparency of the film in (a); (c) flexibility of the thin film, with a bending angle of greater than 60°.

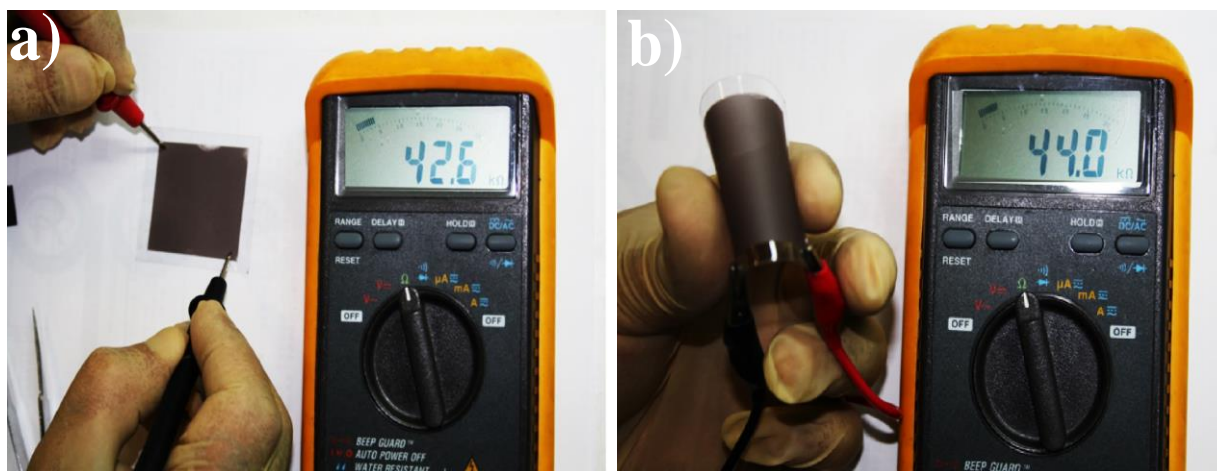


Figure S11. Bending tests of thin film: Resistance measurements of a large-area (4 × 4 cm) thin film of NbSe₂ prepared by spray-coating on a PET substrate. (a) Flat thin film; (b) thin film bent at an angle of greater than 60°. The resistance measured with an AVO meter, of the thin film before and after bending remained relatively constant (ca. 44 kΩ).

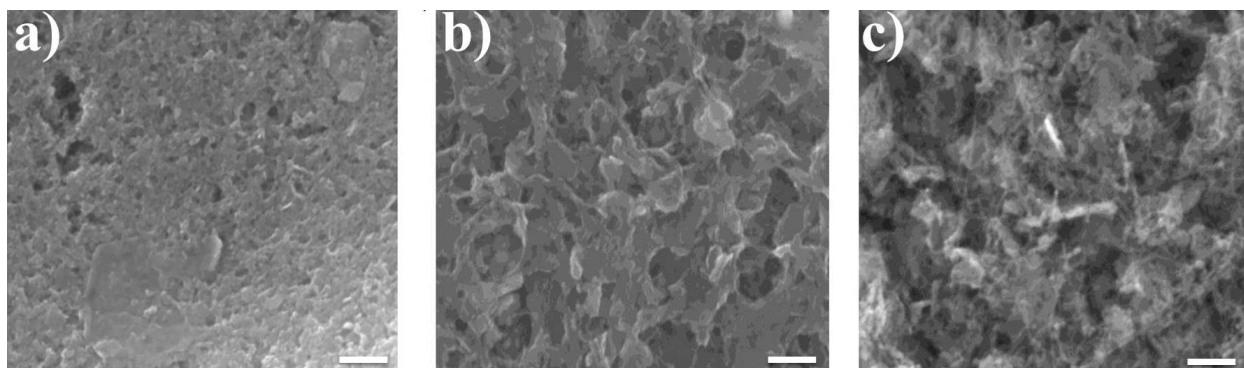
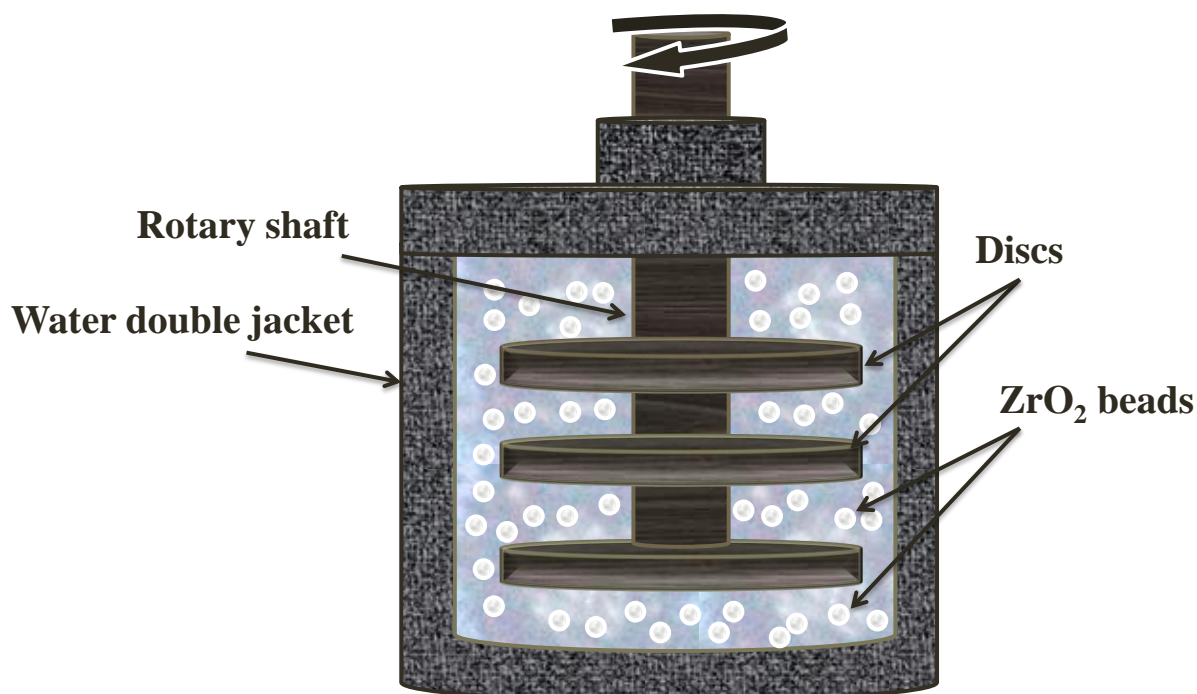


Figure S12. SEM images of NbSe₂ electrodes (a) nanoparticles, (b) nanosheets, and (c) nanorods prepared on FTO substrates. The scale bar is 100 nm.

S4 Equipment and Experimental Conditions:

Mechanical comminuting through bead-milling in a homemade instrument was performed for different time intervals. The stirrer comprised three perforated discs mounted on a drive shaft (Sachem S1). The beads and the suspension were stirred by a central rotating agitator, typically at 1000–3000 rpm. The fluid rotation between the disc tip and the fixed wall of the mill generated an instability, driven by the centrifugal force, similar to Couette–Taylor instability occurring in the gap between two cylinders (the inner cylinder is rotating while the outer one stays at rest).¹ The apparent volume of grinding micronized beads was approximately 50%; the remaining volume was a suspension of the particles to be ground. The carrying fluid was an aqueous solution with grinding aids. Particle fragmentation resulted from the capture of particles inside zones subjected to the strong stress arising from two grinding beads colliding (frontal collisions in straining regions or oblique collisions induced by local shear). The comminution of the feed particles in a stirred media mill may occur through two distinct mechanical phenomena: the particles are either squeezed between two colliding grinding beads or between a bead and a wall. The centrifugal force of the suspension was related to the tangential mean flow

induced by the high speeds of the rotating shaft and discs. Particles and beads in the vicinity of the outer wall of the mill chamber experienced a high pressure.



Scheme S1. Schematic representation of milling equipment: Schematic representation of the grinding machine; the apparent volume of the grinding beads was approximately 50%, with the remaining volume comprising a suspension of particles.

Supplementary References

- 1 Gers, R., Climent, E., Legendre, D., Anne-Archard, D. & Frances, C. *Chem Eng Sci* , 2010, **65**, 2052–2064
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- 4 Reynolds, K. J., Frey, G. L. & Friend, R. H. *Appl Phys Lett*, 2003, **82**, 1123–1125 (2003).