Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A. This journal is © The Royal Society of Chemistry 2014

Electronic Supplementary Information

Effect of Bulk and Surface Structural Changes in Li₅FeO₄ Positive Electrodes during First Charging on Subsequent Lithium-Ion Battery Performance

Toyoki Okumura, * Masahiro Shikano, and Hironori Kobayashi*

Research Institute for Ubiquitous Energy Devices, National Institute of Advanced Industrial Science and Technology (AIST), 1-8-31 Midorigaoka, Ikeda, Osaka 563-8577, Japan

*E-mail: toyoki-okumura@aist.go.jp; hironori-kobayashi@aist.go.jp

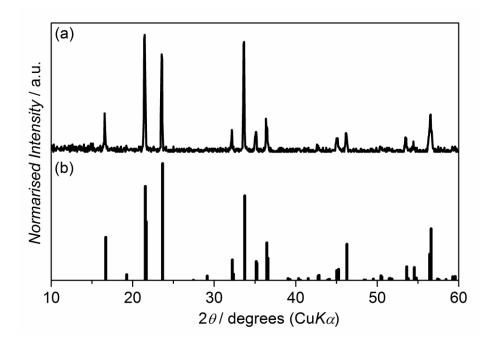


Fig. S1 (a) X-ray diffraction pattern of a synthesised compound. (b) Reference pattern for a Li₅FeO₄ low-temperature phase (JCPDS card #01-075-1253).

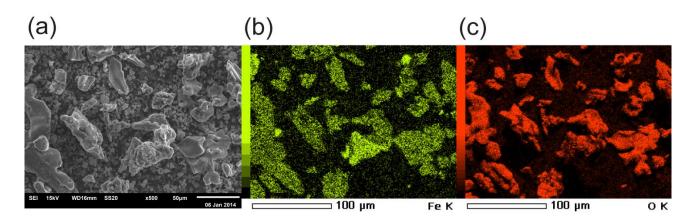


Fig. S2 (a) SEM image of synthesised Li₅FeO₄. EDX maps of (b) Fe *K*-edge and (c) O *K*-edge.

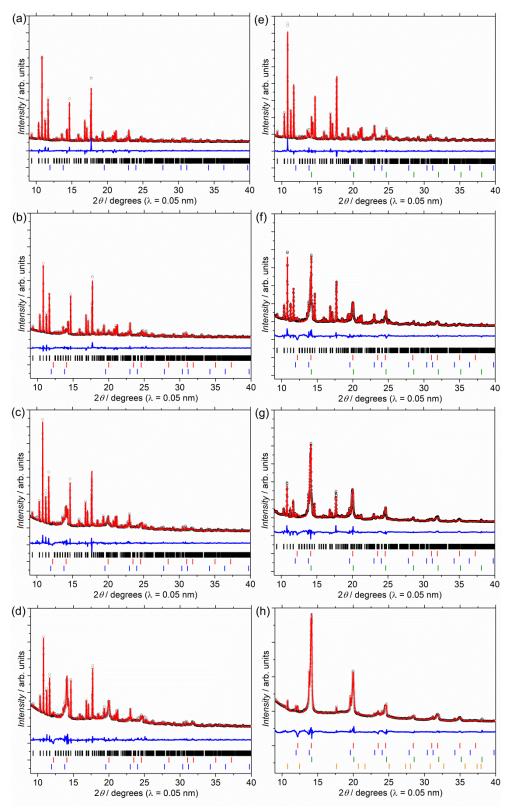


Fig. S3 Rietveld refinements of ex situ synchrotron X-ray diffraction patterns of electrochemically charged/discharged samples. The differences between the two batches of #1 and #2 are also shown. The number of charged electrons per positive electrode is (a) e = 0 (pristine, #1), (b) 0.33 (charged until 58 mAh g⁻¹, #1), (c) 0.66 (charged until 116 mAh g⁻¹, #1), (d) 1 (charged until 173 mAh g⁻¹, #1), (e) e = 0 (pristine, #2), (f) 1 (charged until 173 mAh g⁻¹, #2), (g) 1.5 (charged until 260 mAh g⁻¹, #2) and (h) 1.85 (fully charged until 320 mAh g⁻¹, #2). Observed, calculated and difference plots are shown by black open circles, red solid lines and blue solid lines, respectively. Diffraction positions are also shown by several colour bars; black (Li₅FeO₄), red (generated PC-LFO; Li_αFeO₂), blue (LiFeO₂ impurity), green (Fe impurity), and orange (Li₂O).

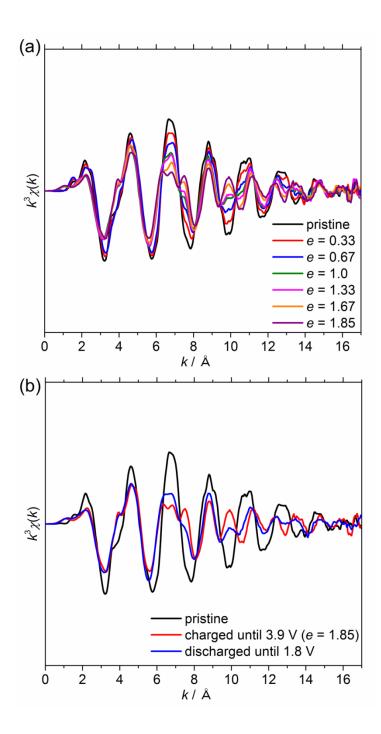


Fig. S4 Pseudo-radial structure functions of *ex situ* Fe *K*-edge EXAFS oscillations for (a) electrochemical samples halted during the first charging process, (b) electrochemically charged/discharged samples.