

Electronic Supplementary Information

MoO₂ Nanobelts@Nitrogen Self-doped MoS₂ Nanosheets as Electrocatalyst for the Enhanced Hydrogen Evolution Reaction

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Table S1 Comparison of the HER performance among different literatures

Samples	Onset overpotential (mV)	Tafel Slope (mV/decade)	literatures
① MoS ₂ /reduced graphene	-100	41	8b
② MoS ₂		94	
① defect-rich MoS ₂	-120	50	21
② defect-free MoS ₂	-180	90	
① MoO ₂ @N-doped MoS ₂	-156	47.5	Our experiment
② MoS ₂	-261	77.5	

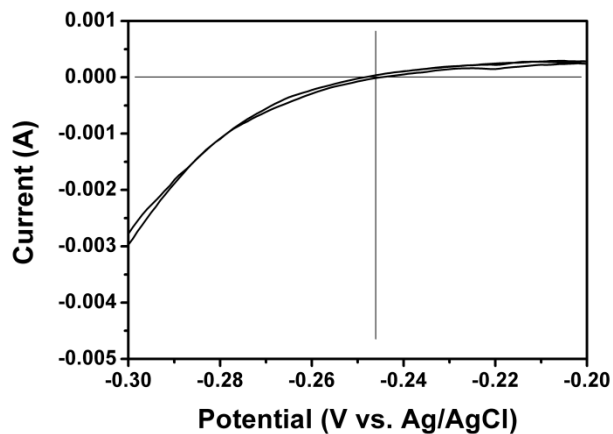


Figure S1. Reversible hydrogen electrode (RHE) was tested in our experiment.

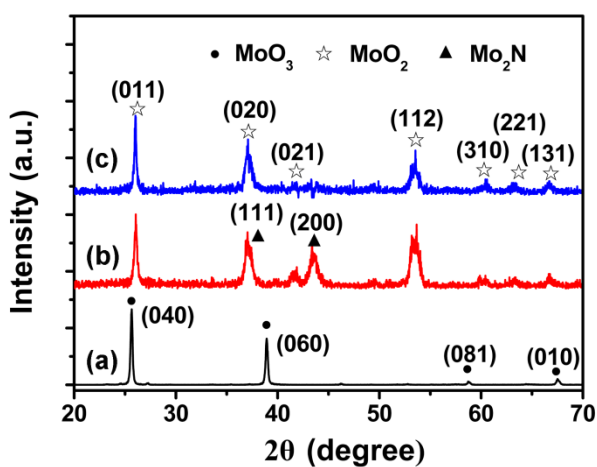


Figure S2. XRD result of different samples obtained different process: (a) MoO_3 nanobelts, (b) $\text{MoO}_2@ \text{Mo}_2\text{N}$ nanobelts and (c) MoO_2 nanobelts@nitrogen-doped MoS_2 nanosheets.

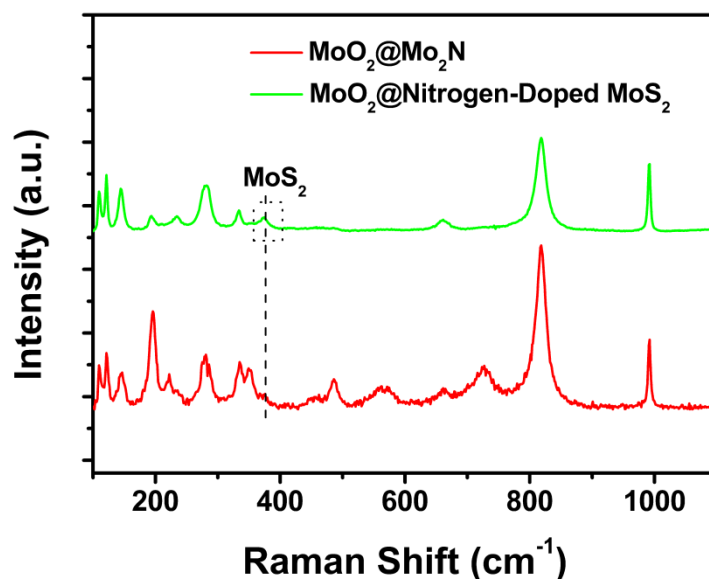


Figure S3. Raman spectra of $\text{MoO}_2@ \text{Mo}_2\text{N}$ nanobelts and MoO_2 nanobelts@nitrogen-doped MoS_2 nanosheets.

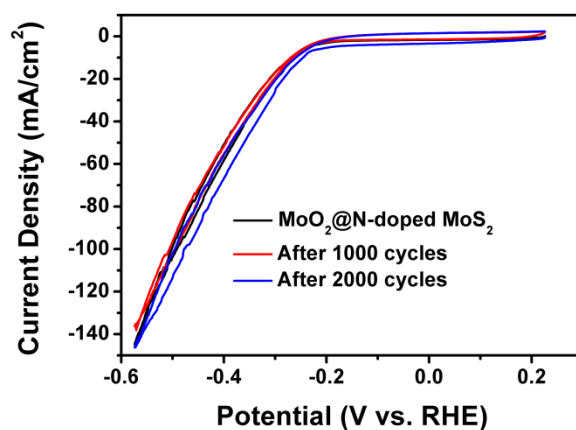


Figure S4. Reverse scans of MoO₂@N-doped MoS₂ electrode.

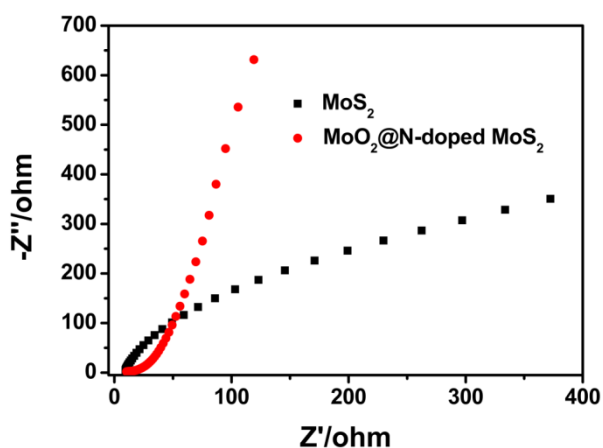


Figure S5. Nyquist plots of MoS₂ and MoO₂@N-doped MoS₂. It can be seen that the MoS₂ sample showed an arc with a large diameter suggesting a large charge–transfer resistance for HER, whereas the MoO₂@ MoS₂ sample showed almost no arc but a linear profile which suggests significantly reduced charge-transfer resistance and hence the HER was under diffusion control. This indicates that the electronic conductivity of MoO₂@ MoS₂ is indeed markedly enhanced as compared to MoS₂ alone.

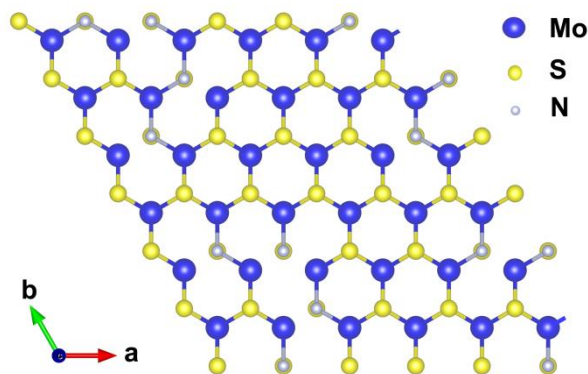


Figure S6. Schematic representation of N-doped MoS₂ nanosheets. Blue for Mo atoms, yellow for S atoms, and gray for N atoms.

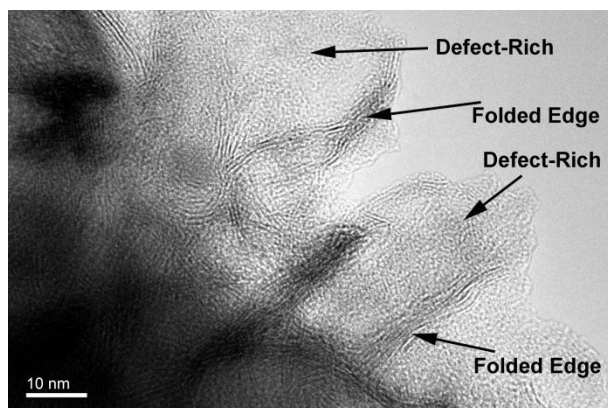


Figure S7. HRTEM image N-doping MoS₂ nanosheets.

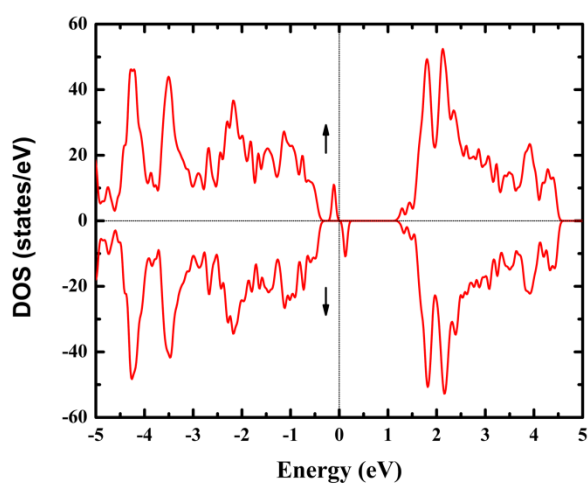


Figure S8. DOS of N-doped MoS₂ monolayer. The positive and negative values indicate the spin up and down DOS respectively. The Fermi energy is set to 0.