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Electronic Supplementary Information

Correlation between nanoparticle location and graphene nucleation in chemical vapour

deposition of graphene

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Figure S1. Elemental analysis composition of nanoparticles located on the center of graphene domains. (a,d) SEM images of nanoparticles on the center of graphene domains. (b,e) Corresponding EDS results and (c,f) elemental compositions.



Figure S2. (a) SEM image of a large particle. (b) Elemental composition. (c~f) Elemental mappings. The images show the shapes of corresponding components. Only c and e are consistent with that in a. Hence, the nanoparticles are consisted of Fe and O.

		-	-
(a)	Element	Weight%	Atom%
	С	4.85	82.96
100000	AI	0.05	0.41
1000	Si	0.13	0.98
500	Ca	0.09	0.48
500nm	Cu	4.69	15.17
(b)			
	Element	Weight%	Atom%
	С	6.49	68.56
	Cu	15.74	31.44
500nm			

Figure S3. Elemental composition of two typical large particles. (a) Irregular shape (Al, Si, and Ca). (b) Round shape (C).



Figure S4. (a) SEM image of graphene domains with uniform thickness. (b) SEM image of multilayer graphene domains.



Figure S5. Deposition of Ag particles on graphene domains. (a,b) Ag particles gathering on the edges of graphene domains. (c) Deposition of Ag particles on cracked graphene domains; (d) Ag particles deposited at the centers of multilayer graphene domains.

Graphene domains on copper substrates were immersed in AgNO₃ solution (0.01~0.1 mM) for 15 min for the deposition of Ag particles. Generally, Ag particles tend to deposit on the edges

and defects of graphene domains (Figs. S5a-c) [S1]. In Fig. S5d, Ag particles locate in the centre of multilayer graphene domains similar to the nanoparticles in this work, demonstrating that the centre parts are active sites for multiple particles. Besides, the deposition of Ag particles depends on post-treatment process, which also demonstrates that the nanoparticles can form later than graphene, and share the same nucleation sites with it.

Reference:

[S1] Li, Z., et al., Graphene buffered galvanic synthesis of graphene-metal hybrids. JOURNAL OF MATERIALS CHEMISTRY, 2011. 21(35): p. 13241-13246.