Electronic supplementary information

Organic Linker Geometry Controlled Synthesis of Coordination Polymer Spheres and Their Thermal Transformation to Yolk-Shell Metal Oxides

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Fig. S1 XRD pattern of Eu-CPs.



Fig. S2 TEM image and size-distribution diagram of Eu-CPs prepared under 15 mM precursor concentration at 160 $^{\circ}$ C for 10 min. Scale bar: 500 nm.



Fig. S3 FT-IR spectra of the Eu-CPs prepared with different precursor concentrations (a) 2.5 mM, (b) 5 mM, (c) 10 mM and (d) 15 mM. (e) 1,3-Benzenedicarboxylic acid (L5).



Fig. S4 TGA curve of the Eu-CPs.

Sample	M / mM	Linker / mM	T / °C	t / min	Diameter / nm
Ni-CPs	10	10	140	360	303
Zn-CPs	10	10	140	30	1366
Mg-CPs	5	5	160	360	55
In-CPs	10	10	140	30	155
Y-CPs	2.5	2.5	160	30	374
Sm-CPs	2.5	2.5	160	30	342
Gd-CPs	2.5	2.5	160	30	134
Ho-CPs	2.5	2.5	160	30	300
Er -CPs	2.5	2.5	160	30	417
Yb-CPs	5	5	160	30	250
Lu-CPs	2.5	2.5	160	30	370
EuGd-CPs (6:4)	5	5	160	30	261
EuGd-CPs (4:6)	5	5	160	30	243
EuYb-CPs	5	5	160	30	256
EuYYb-CPs	7.5	7.5	160	30	300
YEuErLu-CPs	10	10	160	30	520

 Table S1 The preparation conditions of the M-CPs.



Fig. S5 TEM images of M-CPs prepared with different metal cations.









Fig. S7 EDS of YEuErLu-CPs.



Fig. S8 SEM and TEM images of the metal oxide spheres obtained by calcination of their respective solid precursors (M-CPs) at 600 $\,^{\circ}$ C for 3 h. Scale bars: 200 nm.



Fig. S9 XRD patterns (a, b) of SmYb-MOs with different molar percentage of Yb. (c) The changes of (222) peak degree with Yb content.



Fig. S10 Size effect of Eu-CPs on the cyanosilylation of benzaldehyde and filtration experiment. Reaction conditions: 5 mg catalyst, 0.25 mmol benzaldehyde, 0.5 mmol Me₃SiCN, 1.5 mL CH₂Cl₂, room temperature, under Ar.



Fig. S11 Recyle of the Eu-CPs. Reaction conditions: 5 mg catalyst, 0.25 mmol benzaldehyde, 0.5 mmol Me_3SiCN , 1.5 mL CH_2Cl_2 , room temperature, 6 h, under Ar.



Fig. S12 Fluorescence images (excited at 245 nm) of diluted reaction solution (a) and after filtration (b).









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