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## **Supplementary Information**

## Microstructural Optimization of MFI-type Zeolite Membrane for Ethanol-Water Separation

Yong Peng,<sup>a</sup> Huibin Lu,<sup>a</sup> Zhengbao Wang\*<sup>a</sup> and Yushan Yan<sup>a,b</sup>

<sup>a</sup> Department of Chemical and Biological Engineering, and MOE Engineering Research Center of

Membrane and Water Treatment Technology, Zhejiang University, Hangzhou 310027, PR China

<sup>b</sup> Department of Chemical and Biomolecular Engineering, University of Delaware, Newark, DE

19716, USA

Corresponding author Fax/Tel.: +86-571-8795-2391; E-mail: zbwang@zju.edu.cn

zeolite membranes <sup>a</sup>								
Membrane No.	ТРАОН	content	Time (h)	Flux (kg m <sup>-2</sup> h <sup>-1</sup> )	Separation factor ( - )			
	( <i>x</i> )							
MS1	0.17		8	1.62	65			
MS2	0.17		12	1.66	63			
MS3	0.10		8	1.75	66			
MS4	0.05		8	1.80	59			
<sup><i>a</i></sup> Molar composition of the synthesis solution: <i>x</i> TPAOH : 1TEOS : 165H <sub>2</sub> O.								

 Table S1. Effects of TPAOH content and crystallization time on the PV performance of MFI

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**Table S2**. Effects of OH<sup>-</sup>/Si ratio on the PV performance of MFI-type zeolite membranes whenthe TPA<sup>+</sup>/Si ratio is fixed at  $0.05^a$ 

Membrane No.	NaOH content (y)	OH-/Si	Flux (kg m <sup>-2</sup> h <sup>-1</sup>	) Separation factor ( - )			
M7	0	0.05	1.96	55			
M9	0.05	0.10	1.57	77			
M15	0.12	0.17	1.09	77			
M16	0.27	0.32	0.59	66			
<sup>a</sup> Molar composit	ion of the synthesis	solution:	0.05TPAOH : 1TI	EOS : $165H_2O$ : $yNaOH$ ,			
synthesis time: 4 h.							



Fig. S1 Diagram of gas permeation setup.



Fig. S2 Measure points of EDX analysis of membranes M2 (a), M5 (b), and M6 (c).



Fig. S3 XRD patterns of membranes M5 and M6.



Fig. S4 SEM top-view images of the membrane synthesized at 175 °C for 2 h when the ratio of  $TPA^+/Si$  is 0.17 (a) and 0.05 (b).