

Supplementary Information

Effect of Processing Additive on Morphology and Charge Extraction in Bulk-Heterojunction Solar Cells

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Characterization: ¹H and ¹³C NMR spectra were recorded using a Varian AS400 in deuterated chloroform or acetone solution at 298 K. Chemicals shifts were reported as δ values (ppm) relative to chloroform value of 7.26 ppm. For our size-exclusion chromatography (SEC) measurements, we used a Varian Polymer Laboratories GPC220 equipped with a RI detector, a PL BV400 HT Bridge Viscometer, 2 PLgel Mixed C (300 x 7.5 mm) columns and a PLgel Mixec C Guard column. The temperature of the system was set to 110°C and the 1,2,4-trichlorobenzene (TCB) (with 0.0125% BHT w/v) flow was set to 1mL/min. The samples (2 mg) were dissolved in 2 mL of TCB in a 5 mL chromatography vial then stirred and heated to 110°C for 1 hour to let aggregates completely dissolve. Then, a filtration through a .45 μ m cellulose fiber film in a 5 mL chromatography vial lead to a homogenous polymeric solution. The sample was injected though a loop of 200 μ L with a Varian Polymer Laboratories PL-SP 260VC sample preparation system. Calibration was set with narrow polystyrene standards. UV-vis-NIR absorption spectra were recorded using a Varian Cary 500 UV-vis-NIR spectrophotometer using 1 cm path length quartz cells. A polymeric solution was spin-coated on untreated glass substrate in order to perform the solid-state measurement. The optical bandgap was calculated from the onset of the absorption band. Cyclic

voltamograms were recorded from a Solartron 1287 Potentiostat using a three-electrode set-up; a working and counter electrodes in platinum and an Ag/Ag⁺ (0.1 M of AgNO₃ in anhydrous acetonitrile) reference electrode. The scan rate was 50 mVs⁻¹ and the electrolyte was an anhydrous argon saturated solution of 0.1 M of tetrabutylammonium tetrafluoroborate (Bu₄NBF₄). Thin films were made by casting a diluted solution of a copolymer on the working electrode, dried under ambient conditions. For each measurement, a new film was cast. The HOMO and LUMO energy levels were determined from the oxidation and reduction onsets, assuming an SCE electrode to be at -4.7 eV from vacuum.

Table S1. Information of synthesized polymer of PBnDT-FTAZ.

Polymer	Mn (kDa)	Mw (kDa)	PDI	Yield (%)	HOMO (eV)	LUMO (eV)	E _g ^{CV}	E _g ^{Opt}
PBnDt-FTAZ	73	146	2.0	91	-5.32	-3.38	1.94	1.91

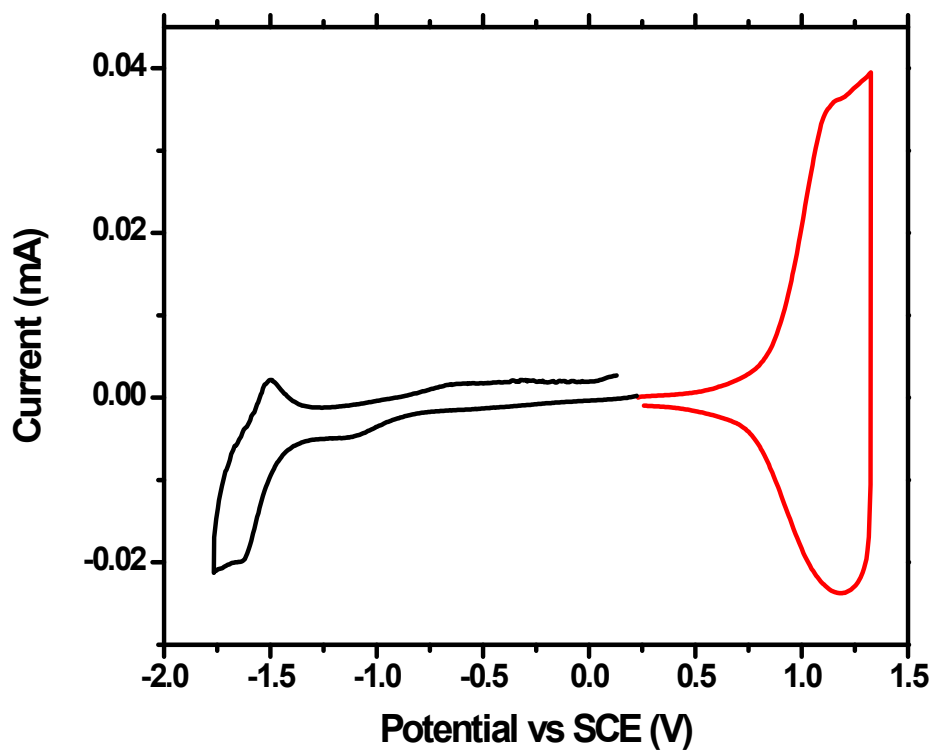


Figure S1. Cyclic voltammetry (black line: reduction and red line: oxidation) of PBNdT-FTAZ. The HOMO and LUMO level of PBNdT-FTAZ shows -5.32 eV and -3.38 eV, respectively.

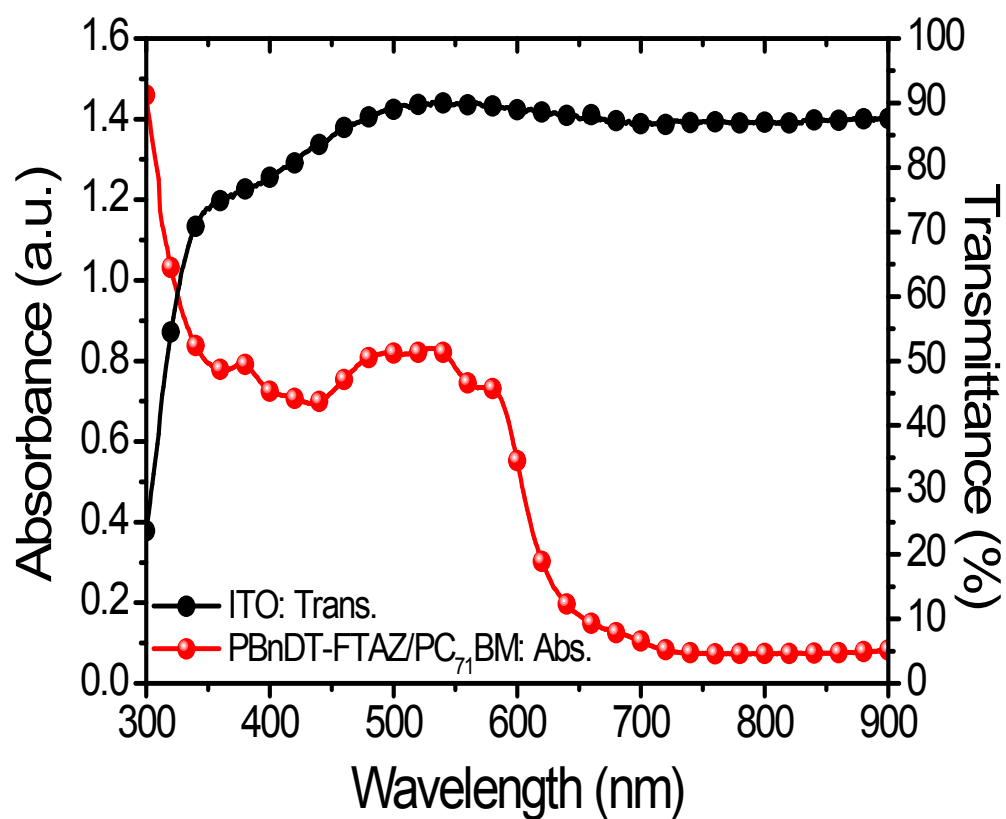


Figure S2. Normalized UV-vis absorption spectra of PBnDT-FTAZ:PC₇₀BM BHJ film (red curve) and transmittance of the ITO substrate (black curve).

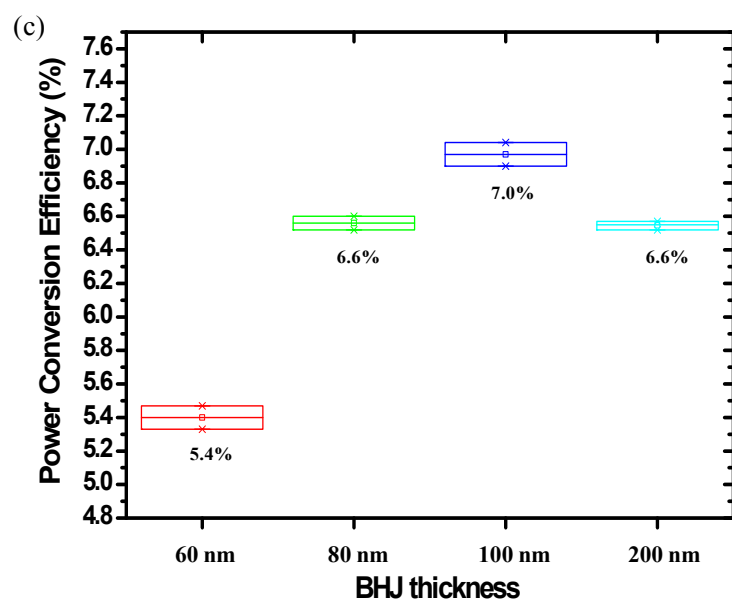
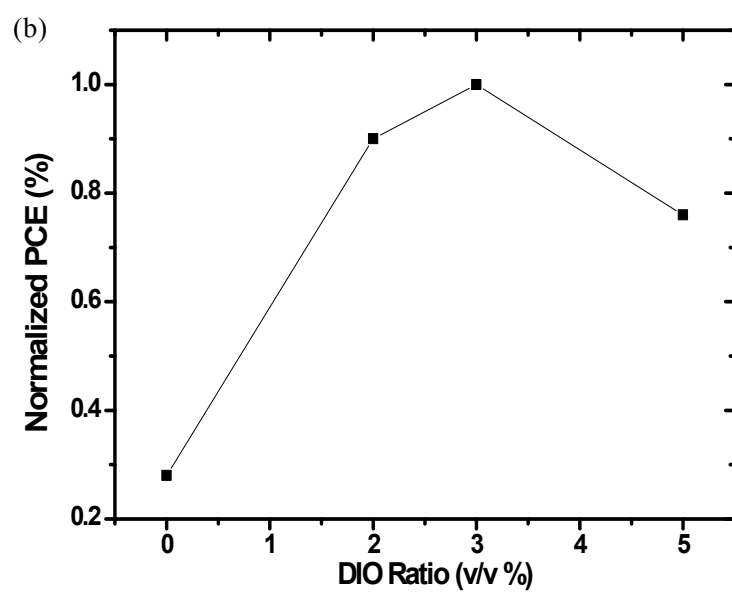
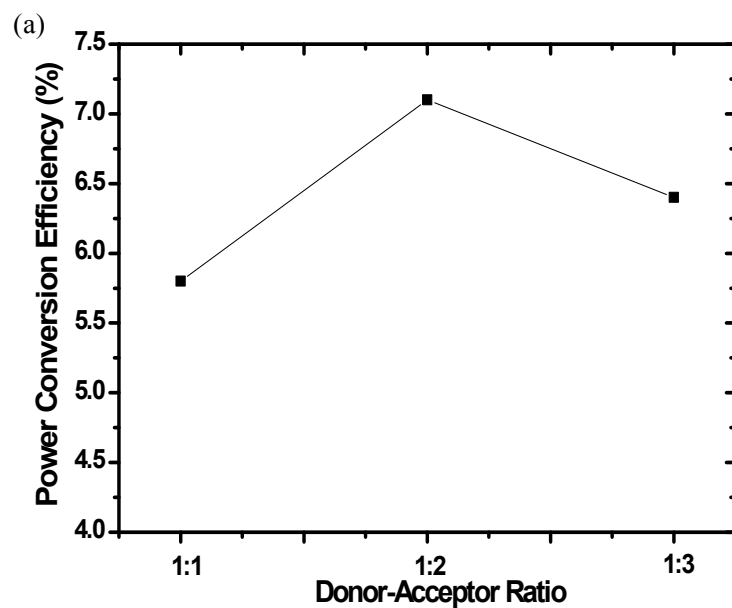


Figure S3. PCE ranges of the PBnDT-FTAZ:PC₇₀BM BHJ device fabricated with (a) donor:acceptor ratio from 1:1 to 1:3, (b) Normalized PCE of DIO ratio from 0% to 5%, and (c) Average PCE depending on BHJ thicknesses from 60 nm to 200 nm (DIO 3%).

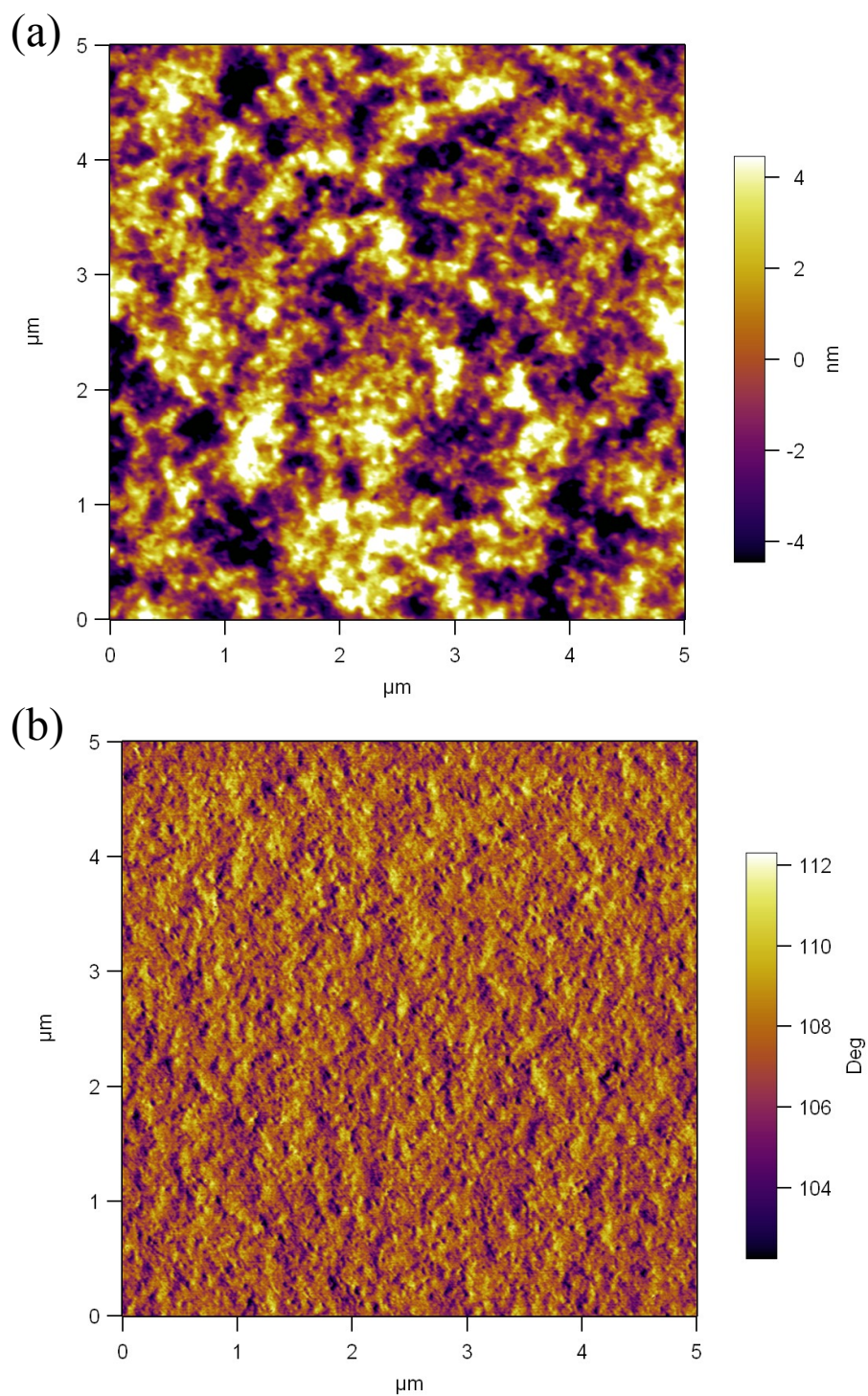


Figure S4. AFM two dimensional images of PBnDT-FTAZ:PC₇₀BM BHJ fabricated with DIO (3%); height images (a) and phase images (b). The D-A ratio of BHJ is 1:2.