

Electronic Supplementary Information

pH-Controlled growth of ultrathin iron vanadium oxide (FeV₃O₈) nanoplatelets with high visible-light photo-catalytic activity

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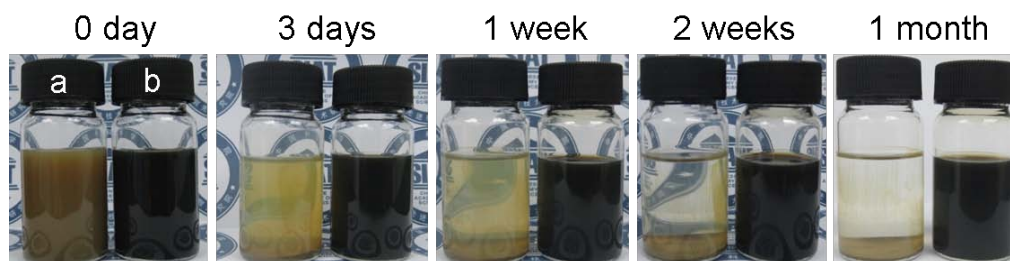


Fig. S1 Stability of ultrathin FeV₃O₈ nanoplatelets as a function of PVP. (a) The product synthesized in the absence of PVP gradually deposited. (b) The product prepared in the presence of PVP was stable for at least 1 month.

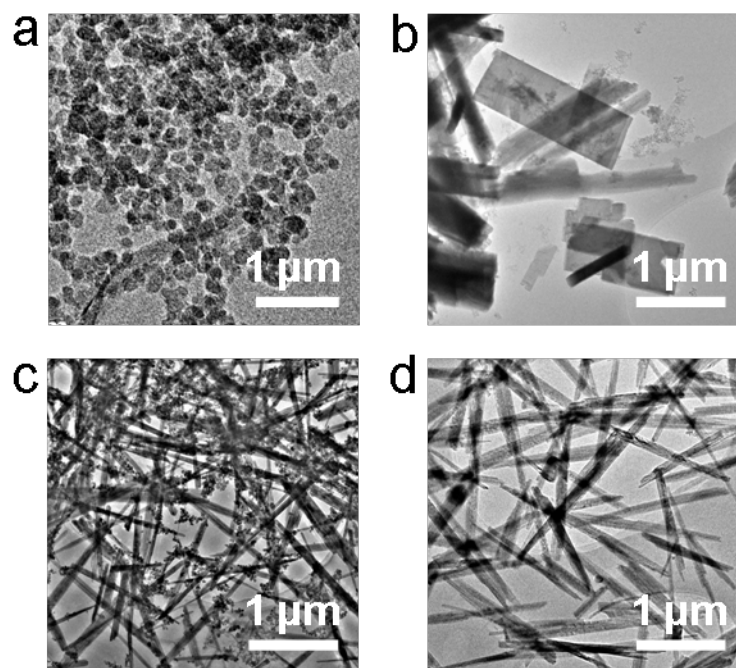


Fig. S2 TEM images of the products synthesized at different pH values: (a) pH 3, (b) pH 5, (c) pH 2, and (d) pH 1.

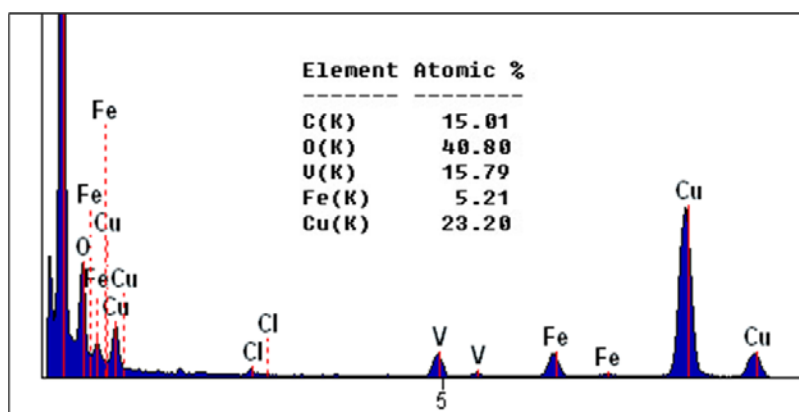


Fig. S3 EDS analysis of ultrathin FeV₃O₈ nanoplatelets

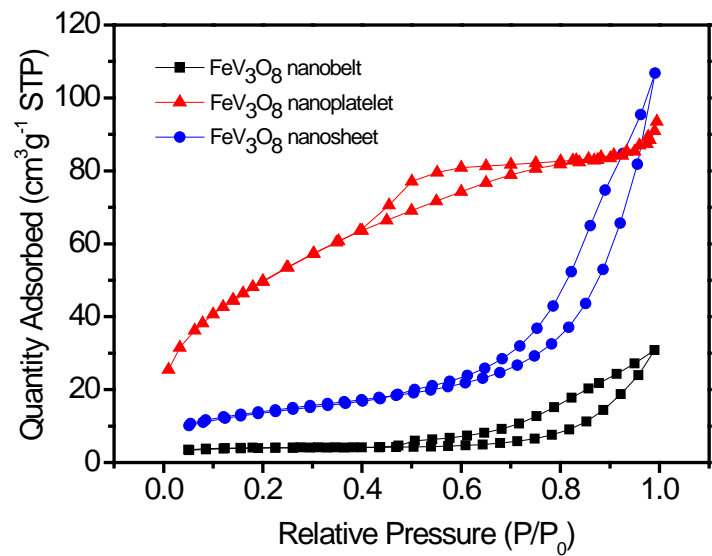


Fig. S4 The N₂ adsorption and desorption isotherm of three samples: FeV₃O₈ nanobelt (black square), FeV₃O₈ nanoplatelet (red triangle) and FeV₃O₈ nanosheet (blue circle).

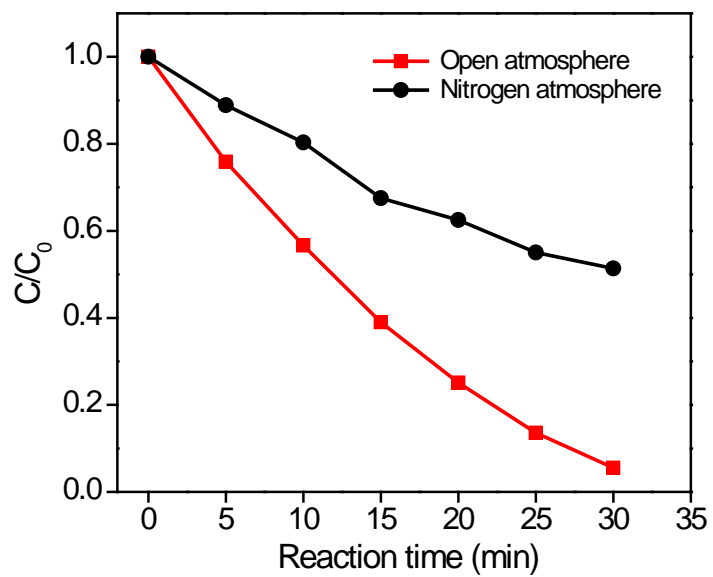


Fig. S5 Photodegradation efficiencies of ultrathin FeV_3O_8 nanoplatelets within 30-min visible-light irradiation under different reaction conditions: open atmosphere (red square) and nitrogen atmosphere (black circle).