

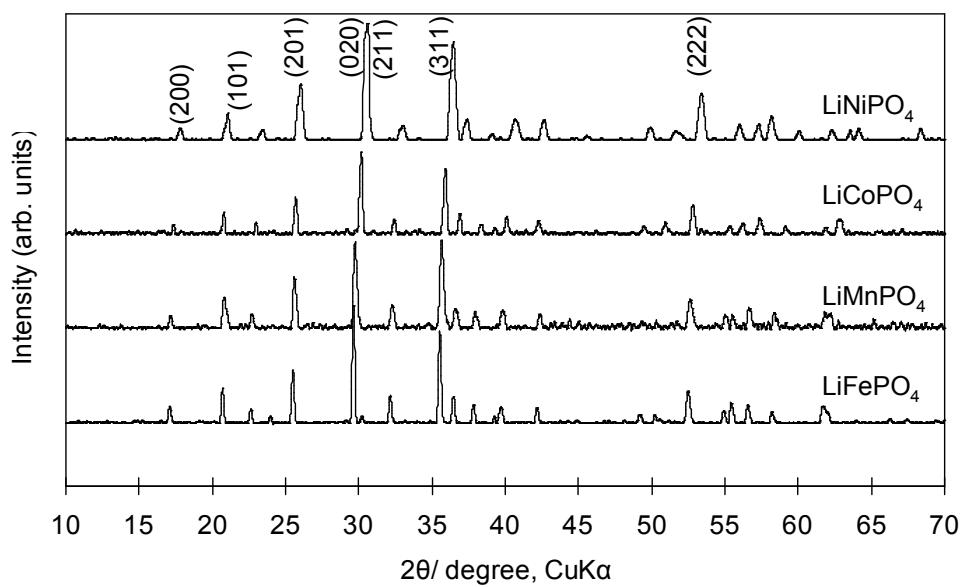
## **Supplementary Information**

# **Benzylamine-Directed Growth of Olivine-type LiMPO<sub>4</sub> Nanoplates by Supercritical Ethanol Process for Lithium-ion Battery**

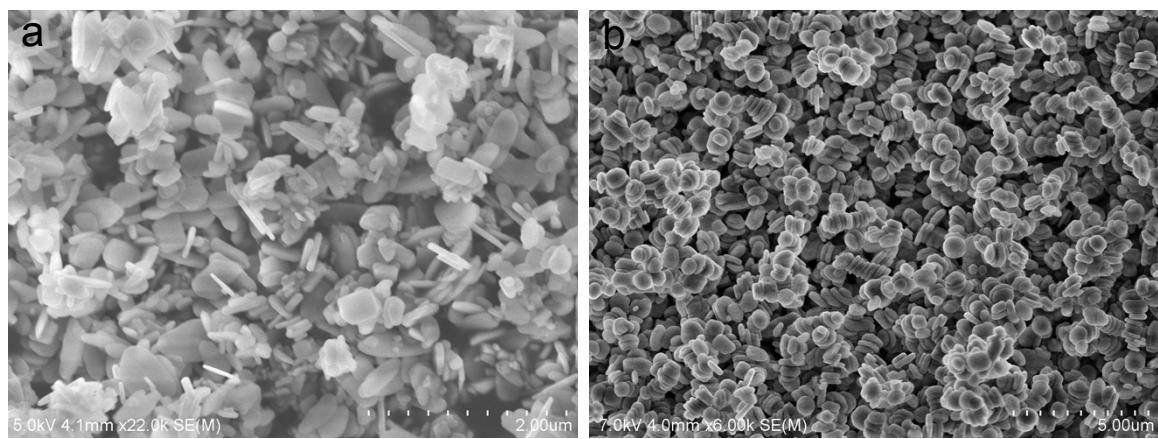
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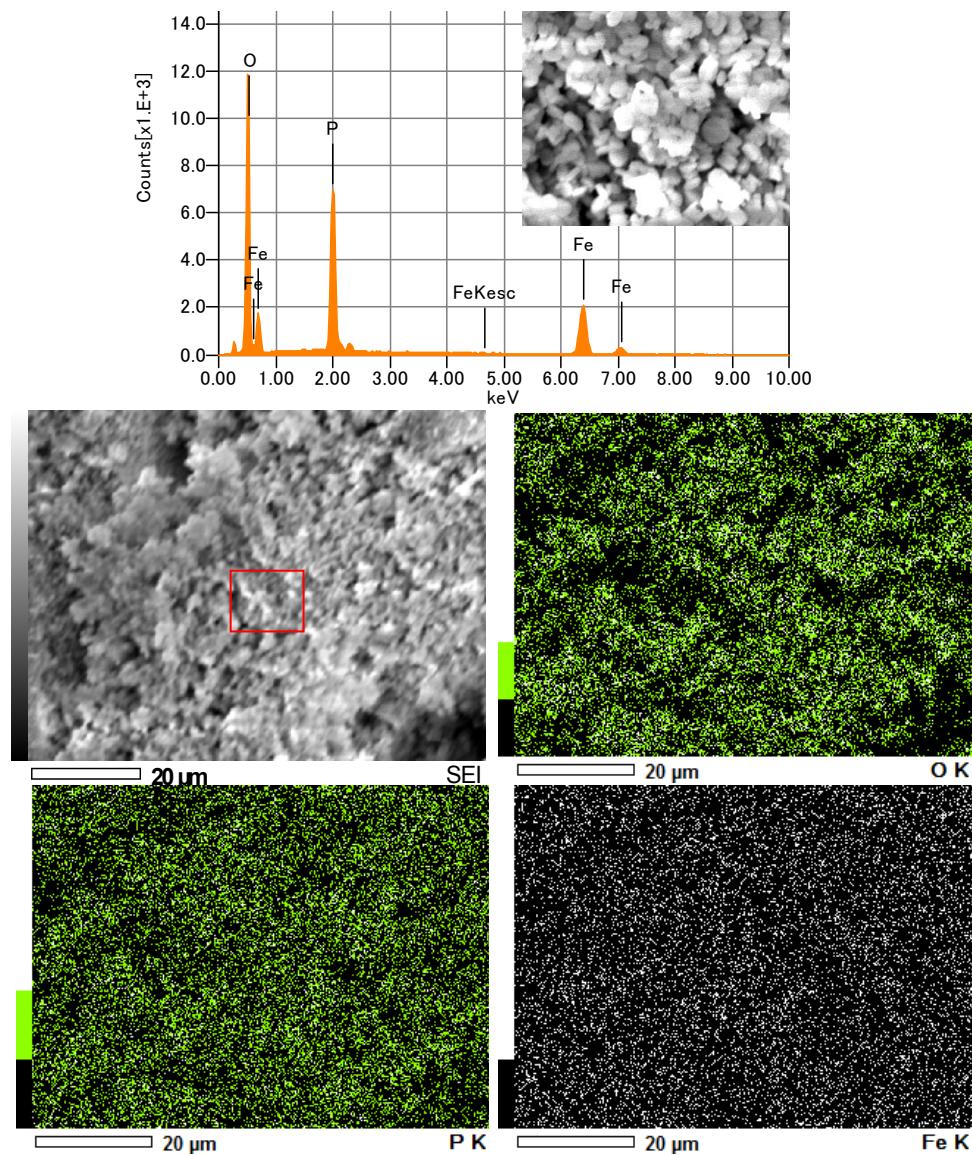
Email: tqduc@mail.tagen.tohoku.ac.jp; i.honma@tagen.tohoku.ac.jp



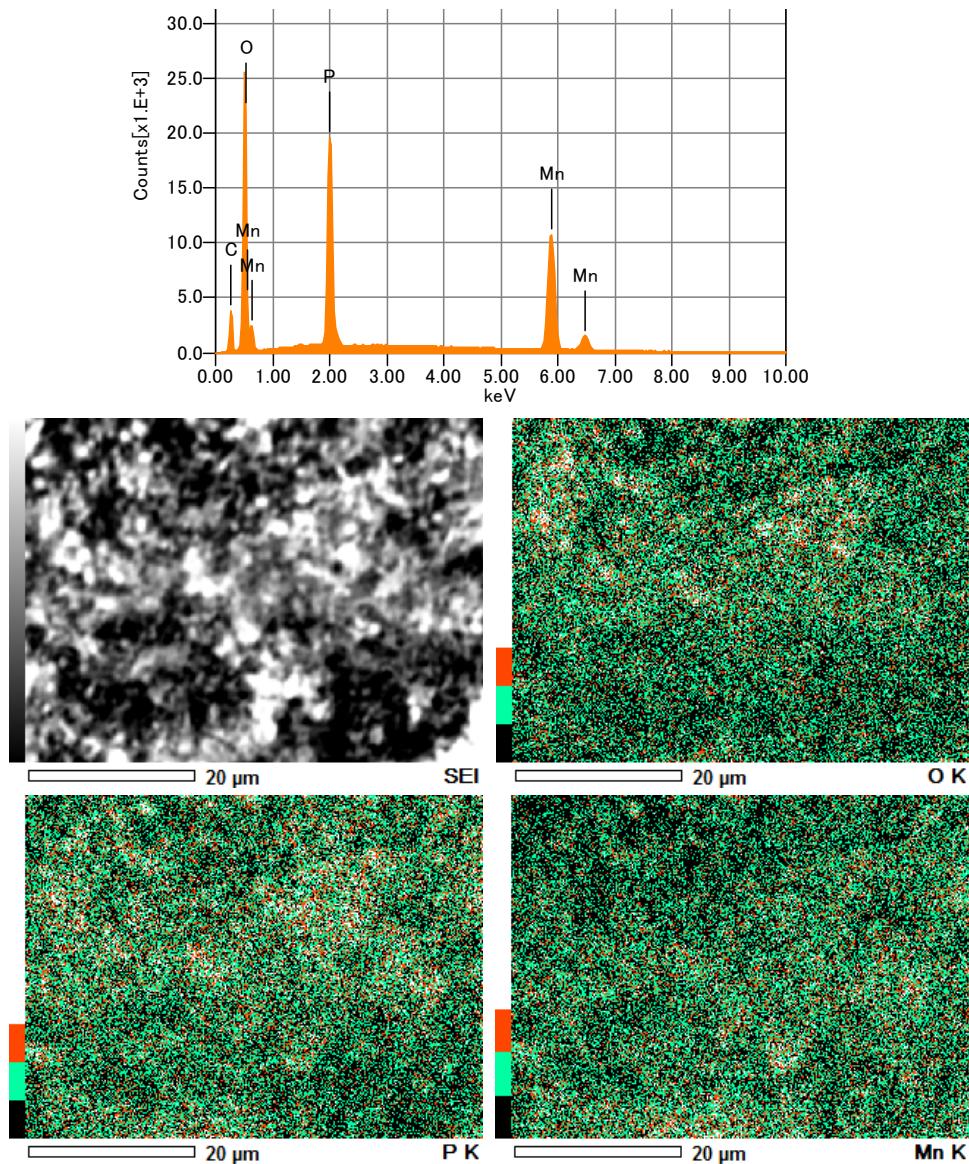
**Fig. S1** XRD patterns of the synthesized lithium metal phosphates by supercritical ethanol processing.



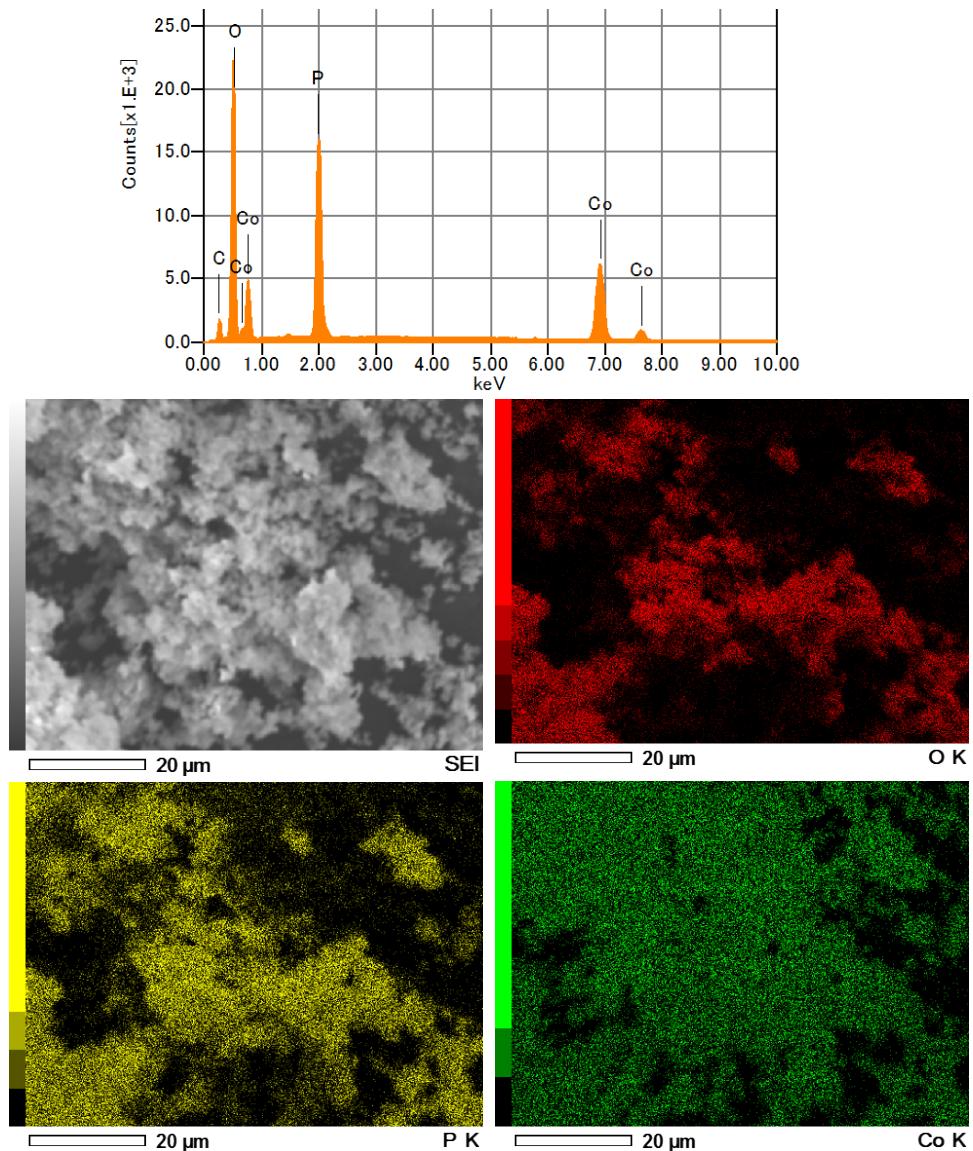
**Fig. S2** SEM images of a)  $\text{LiMnPO}_4$  and b)  $\text{LiCoPO}_4$ .



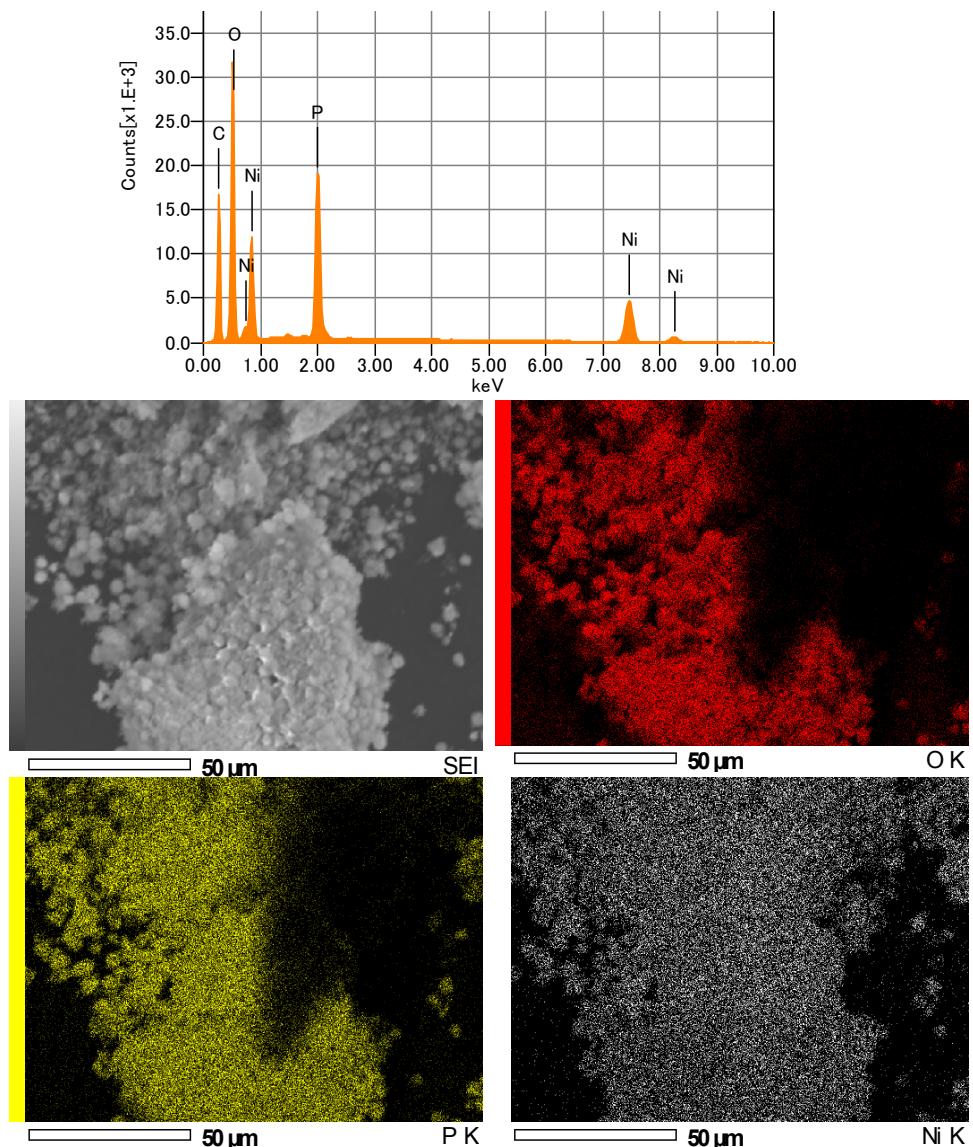
**Fig. S3** EDS spectrum, SEM image and elemental mapping of Fe, P, O of the synthesized LiFePO<sub>4</sub> nanoplates.



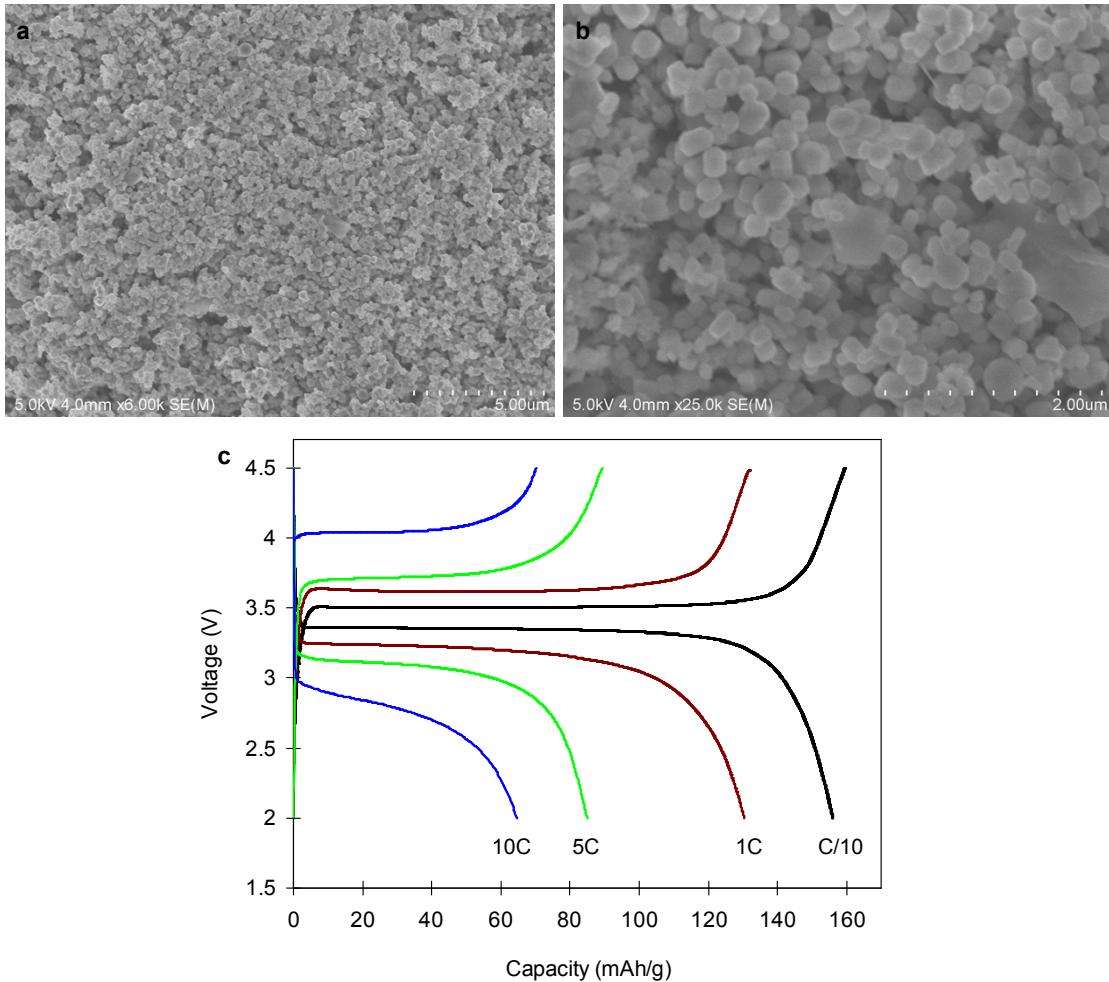
**Fig. S4** EDS spectrum, SEM image and elemental mapping of Mn, P, O of the synthesized LiMnPO<sub>4</sub> nanoplates.



**Fig. S5** EDS spectrum, SEM image and elemental mapping of Co, P, O of the synthesized  $\text{LiCoPO}_4$  nanoplates.

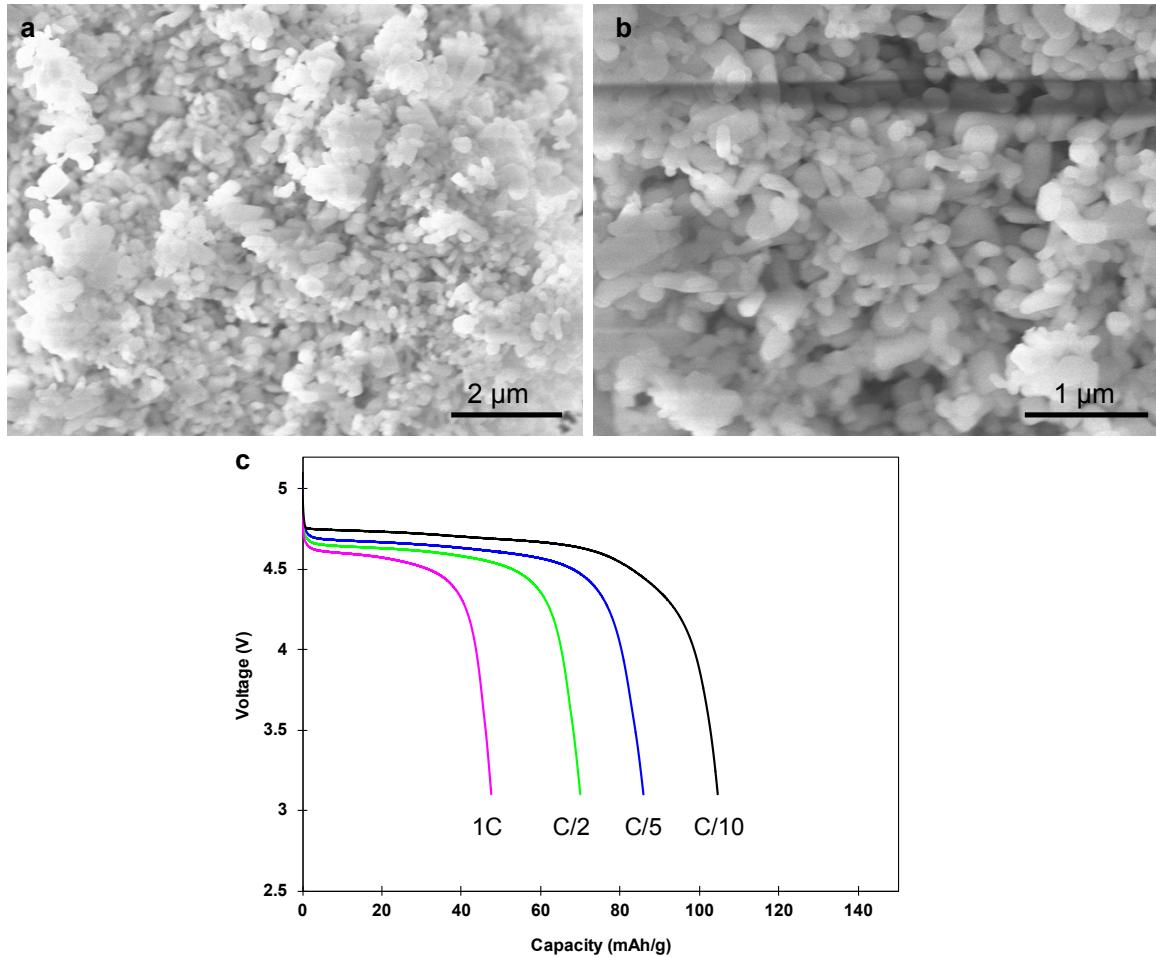


**Fig. S6** EDS spectrum, SEM image and elemental mapping of Ni, P, O of the synthesized LiNiPO<sub>4</sub> nanoplates.



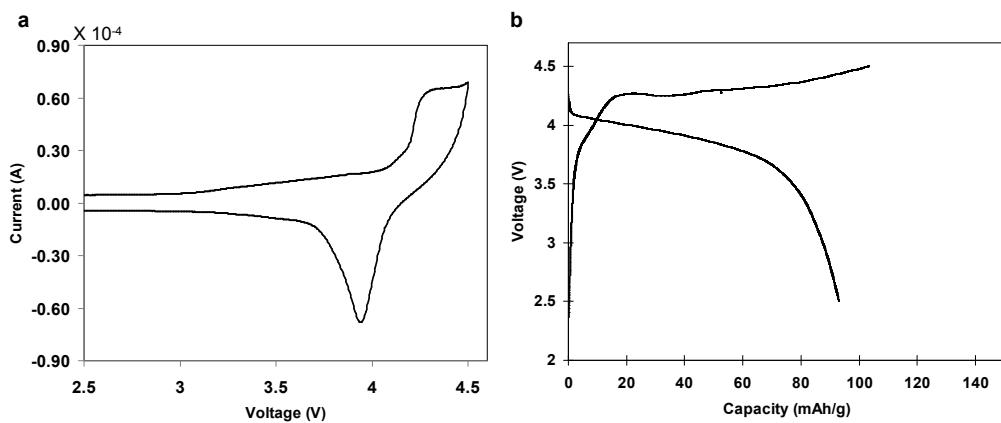
**Fig. S7** a, b) SEM images of the synthesized LiFePO<sub>4</sub> nanoparticles, c) typical first charge/discharge profiles of the LiFePO<sub>4</sub> nanoparticles at current rates of C/10-10 C.

Fig. S7 shows SEM images and typical charge/discharge profiles of the LiFePO<sub>4</sub>/C nanoparticles synthesized by supercritical ethanol at current rates ranging from C/10-10 C as a control sample. The LiFePO<sub>4</sub> nanoparticles were synthesized by the same method without the addition of benzylamine. SEM images indicate the presence of uniform nanoparticles with size of 150-200 nm. The nanoparticles exhibited high reversible capacities at low current rate, however, at higher current rate of 5 C, the nanoparticles showed discharge capacity of 85.0 mA h g<sup>-1</sup> and 64.7 mA h g<sup>-1</sup> which are much lower than that of nanoplates at same current rates (Fig. 8).

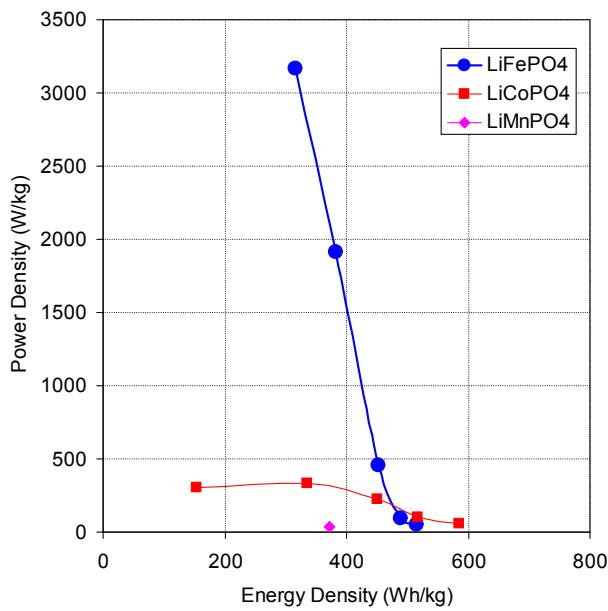


**Fig. S8** a, b) SEM images of the synthesized LiCoPO<sub>4</sub> nanoparticles, c) typical first charge/discharge profiles of the LiCoPO<sub>4</sub> nanoparticles at current rates of C/10-10 C.

Fig. S8 shows SEM images of the LiCoPO<sub>4</sub> nanoparticles synthesized by supercritical ethanol with small amount of benzylamine (2 mL) as *in situ* OH<sup>-</sup> source. It can be cleanly seen that particles are well-dispersive with size of 100-150 nm. Fig. S8c shows the typical discharge profiles of the LiCoPO<sub>4</sub>/C nanoparticles at current rates ranging from C/10-1 C. The nanoparticles exhibited initial discharge capacity of 105, 86, 70, 48 mA h g<sup>-1</sup> at C/10, C/5, C/2, 1 C, respectively which are lower than that of nanoplates at same current rates (Fig. 9).



**Fig. S9** Electrochemical performances of the synthesized  $\text{LiMnPO}_4$  nanoplates in Li-ion batteries tested in the potential range of 2.0–4.5 V. a) cyclic voltammograms of the cells containing  $\text{LiMnPO}_4$  nanoplates at a sweeping rate of  $0.1 \text{ mV s}^{-1}$ . b) typical first charge/discharge profiles at current rate of C/10.



**Fig. S10** Performances, in terms of energy density and power density of different cathode materials for Li-ion batteries.