Supporting Information

A carbon Modified NaTaO₃ Mesocrystal Nanoparticle with

Superior Efficiency of Visible Light Induced Photocatalysis

Xiaoyong Wu, Shu Yin *, Bin Liu, Makoto Kobayashi, Masato Kakihana, Tsugio Sato

Institute of Multidisciplinary Research for Advanced Materials, Tohoku University, 2-1-1,

Katahira, Aoba-ku, Sendai 980-8577, Japan.

*****Corresponding Authors:

Shu Yin

Institute of Multidisciplinary Research for Advanced Materials, Tohoku University, Japan.

Tel: + (00)81-22-217-5599; fax: + (00)81-22-217-5599; e-mail: <u>shuyin@tagen.tohoku.ac.jp</u>.



Figure S1. XRD patterns of samples NaTaO₃-E-G0, NaTaO₃-E-G2 and NaTaO₃-W-G2 in the degree range of 20-35.



Figure S2. XRD patterns of carbon modified NaTaO₃ samples as a function of glucose content.





Figure S3. TEM images of NaTaO₃-E-G0 (a), NaTaO₃-E-G1 (b), NaTaO₃-E-G2 (c) and NaTaO₃-E-G2.5 (d).



Figure S4. TEM images of $NaTaO_3$ -W-G0 (a) and $NaTaO_3$ -E-O2-300 (0).



Figure S5. XRD pattern and TEM image of NaTaO₃-PE-G0.



Figure S6. DRS (a) and FTIR spectra (b) of samples as a function of glucose content in the reaction solution.



Figure S7. XPS survey spectra of samples NaTaO₃-E-G0 and NaTaO₃-E-G2.



Figure S8. Raman spectra of samples.



Figure S9. DeNO_x ability of samples NaTaO₃-E-G0, NaTaO₃-E-G1, NaTaO₃-E-G2 and NaTaO₃-E-G2.5 varied with the addition of glucose contents as well as commercial P25.



Figure S10. Photoluminescence spectra of samples NaTaO₃-E-G0, NaTaO₃-E-G2 and NaTaO₃-W-G2 with the excitation of 304 nm.