

Electronic Supplementary Material (ESI) for
This journal is © The Royal Society of Chemistry 2014

Supporting Information

Bulk $\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$ as Long-Life and High-Power Li-ion Battery Anodes

Qiushi Cheng,^a Jianwen Liang,^a Yongchun Zhu,^{*a} Lulu Si,^a Cong Guo^a and Yitai Qian^{*ab}

^a Hefei National Laboratory for Physical Science at Micro-scale, Department of Chemistry, University of Science and Technology of China, Hefei, Anhui 230026, P. R. China. E-mail: ychzhu@ustc.edu.cn, ytqian@ustc.edu.cn; Tel: +86-551-63601589

^b School of Chemistry and Chemical Engineering, Shandong University, Jinan, 250100, P. R. China. E-mail: ytqian@ustc.edu.cn; Tel: +86-551-63607234

Experimental section

Characterizations. The as-prepared products were characterized by XRD (Philips X' Pert Super diffractometer, Cu Ka, $\lambda=1.54178$ Å), SEM (JEOL-JSM-6700F), HRTEM and SAED (JEOL-2010), Differential thermal analysis (DTA) was measured in air from room temperature to 1300 °C with a heating rate of 10 °C min⁻¹. Raman spectra was tested by using a JYLABRAM-HR Confocal Laser Micro-Raman spectrometer. Inductively coupled plasma (ICP) atomic emission spectrometer (PerkinElmer Instruments) was conducted to measure the ratio of metal atoms. The elements distribution of the sample was detected by energy-dispersive X-ray spectrometry (EDX) elemental mapping analysis (JEM-ARM 200F).

Electrochemical measurements. Electrochemical experiments were tested by using CR2016 coin cells. The electrodes were prepared by mixing the obtained materials, super P and PVDF binder dispersed in N-methylpyrrolidone (NMP) at a weight ratio of 70:20:10. Firstly, the active material and super P were mixed thoroughly, then the binder was added and the mass mixed again. The obtained slurry was coated on Cu foil and dried at 110 °C for 10 h in vacuum. The loading of active materials was 1.0-2.5 mg cm⁻². The LiFePO₄ cathode was coated on Al foil. For the $\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$ /LiFePO₄ full-cell, the anode is limited and the weight ratio of LiFePO₄ and $\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$ is 2.9:1. The capacity of the full-cell was calculated according to the mass of the $\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$ electrode. The electrolyte used for testing was 1 M LiPF₆ in 1:1 EC/DEC and the separator was Celgard 2400. The cells were assembled in an argon-filled glove, and then aged for 12 h before

testing to guarantee full access of the electrolyte with electrode.

Galvanostatic discharge-charge experiments were conducted on a battery-testing system (Land-CT2001A) in the voltage range of 1.0-2.4 V at room temperature.

Cyclic voltammograms (CVs) were tested by CHI660D electrochemical workstation.

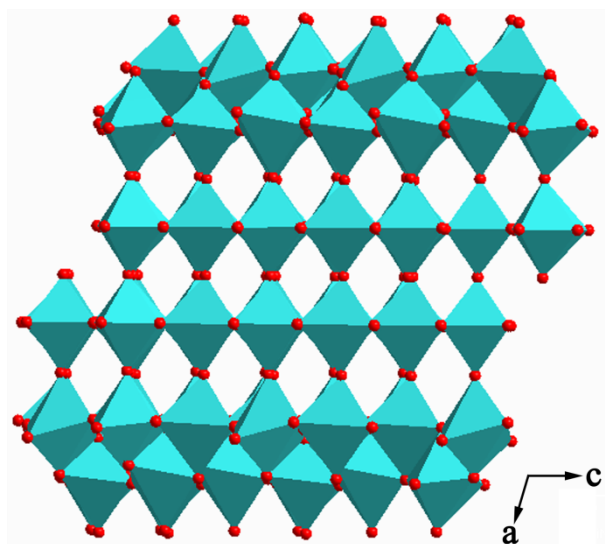


Fig. S1 Crystal structure view of $\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$.

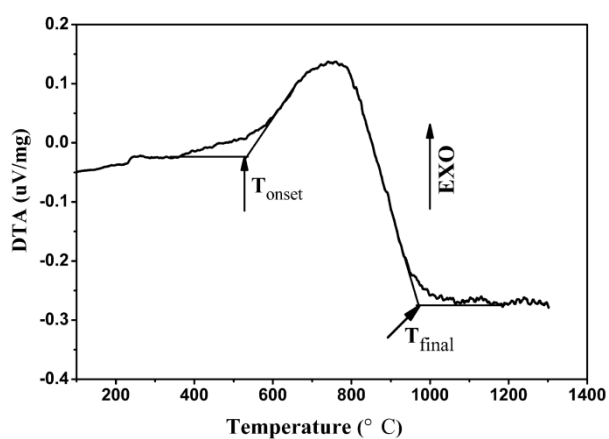


Fig. S2 DTA of the $\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$ with a heating rate $10^\circ\text{C min}^{-1}$ in a flow of air.

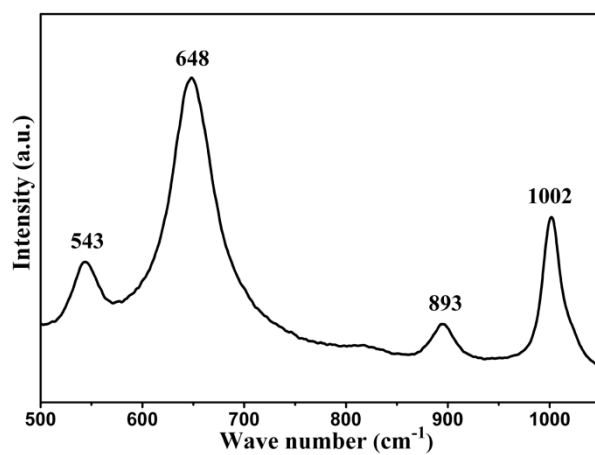


Fig. S3 Raman spectrum of $\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$

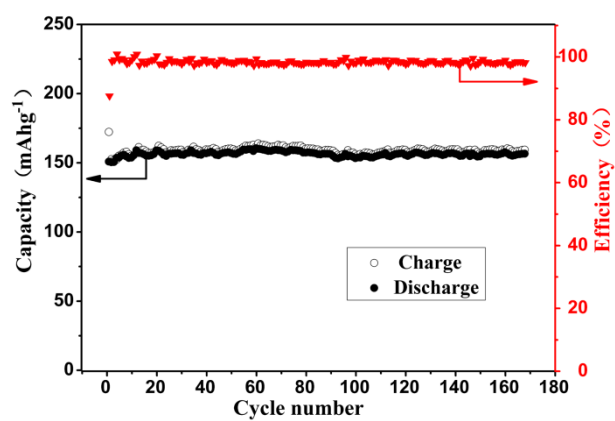


Fig. S4 Cyclic performance of LiFePO_4 at a current rate of 1C cycled between 2.5 and 4.2 V