

Supporting Information

Hierarchically porous anatase TiO₂ microspheres composed of tiny octahedra with enhanced electrochemical properties in lithium-ion batteries †

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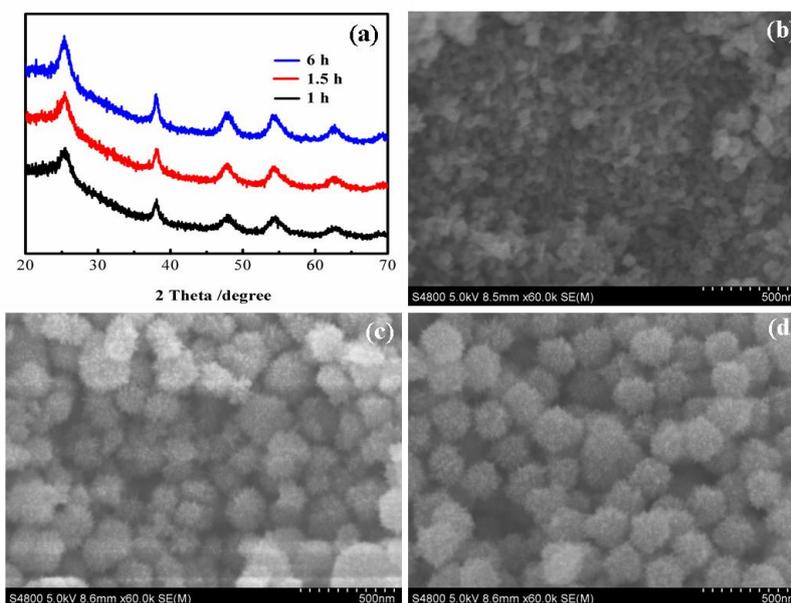


Fig. S1 Anatase TiO₂ obtained at 120 °C for different times: (a) XRD patterns, and SEM images of (b) 1, (c) 1.5 and (d) 6 h.

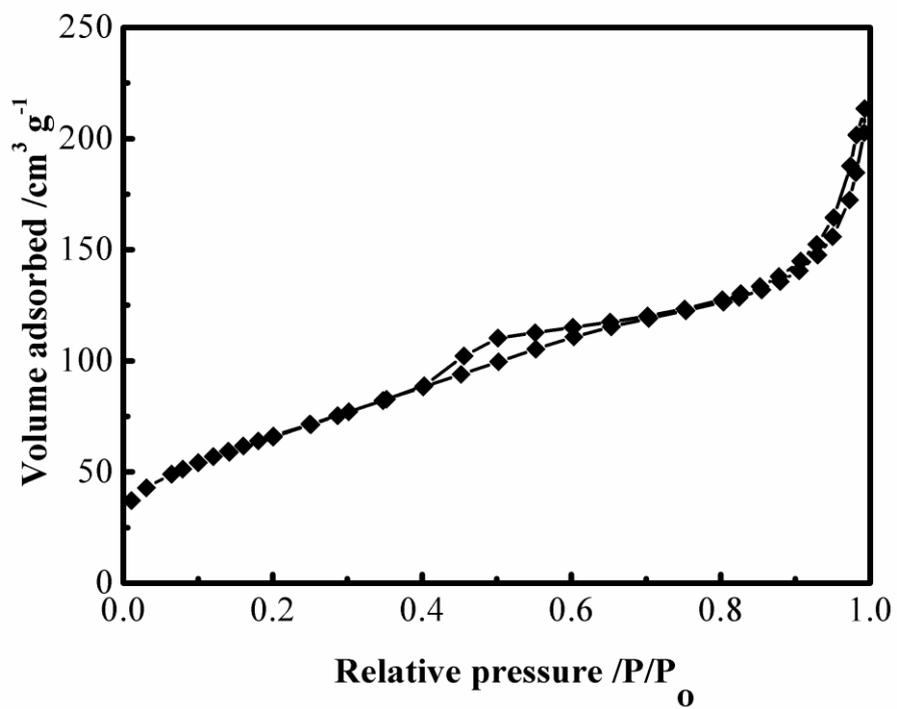


Fig. S2 N_2 adsorption-desorption isotherms of anatase TiO_2 obtained at 120°C for 12 h.

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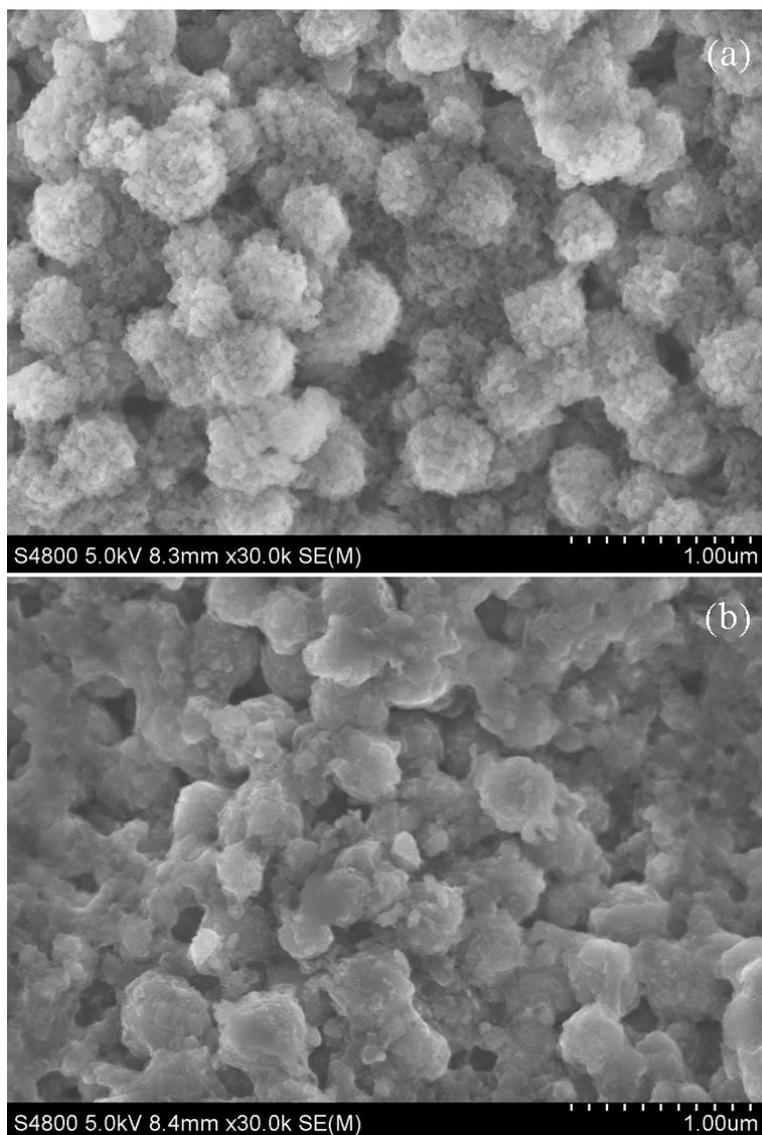


Fig. S3 SEM images of hierarchical TiO₂ microspheres (a) before discharge-charge and (b) after 200 discharge-charge at 10 C.

5 In order to further understand the enhanced electrochemical properties of hierarchical TiO₂ microspheres, the morphology of hierarchical TiO₂ microspheres after 200 cycles charge/discharge was investigated. As showed in Fig. S3, the microspherical structures mainly can be retained.

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Table 1 Summary of discharge capacity for various hierarchical TiO₂ as an anode for LIBs.

Hierarchical TiO ₂	Discharge capacity (mA h g ⁻¹)	Current density (mA g ⁻¹)	Reference
Hierarchical TiO₂ composed of tiny octahedra	142.3 (after 200 cycles)	1680	Present work
Hierarchical TiO ₂ composed of nanosheets	136 (after 100 cycles)	850	S1
TiO ₂ composed of nano-grains	135 (after 5 cycles) 108 (after 500 cycles)	1675	S2
Spherical TiO ₂ composed of nanowires	102 (after 50 cycles)	1000	S3
Hierarchical TiO ₂ composed of nanorods	129.1 (after 100 cycles)	850	S4
Hierarchical TiO ₂ with high surface area of 221.9 m ² g ⁻¹ **	229 (after 100 cycles)	1685	S5

** the sample contains brookite TiO₂.

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