

Supporting Information

Photocatalytic Performance Enhanced via P3HT-g-C₃N₄ Heterojunction

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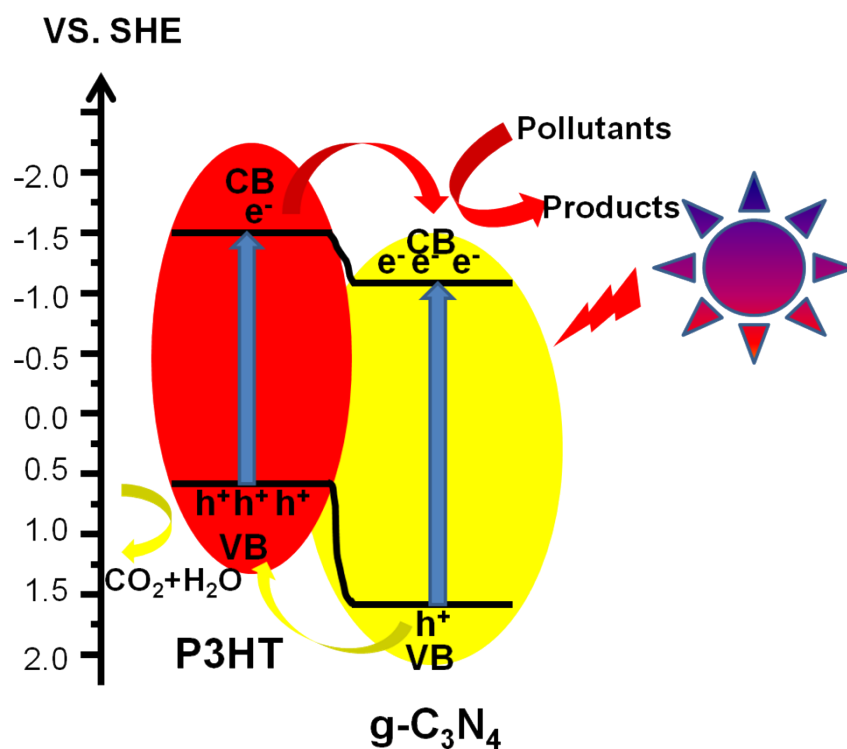


Figure S1 Schematic illustration of organic heterojunction formed between P3HT and g-C₃N₄.

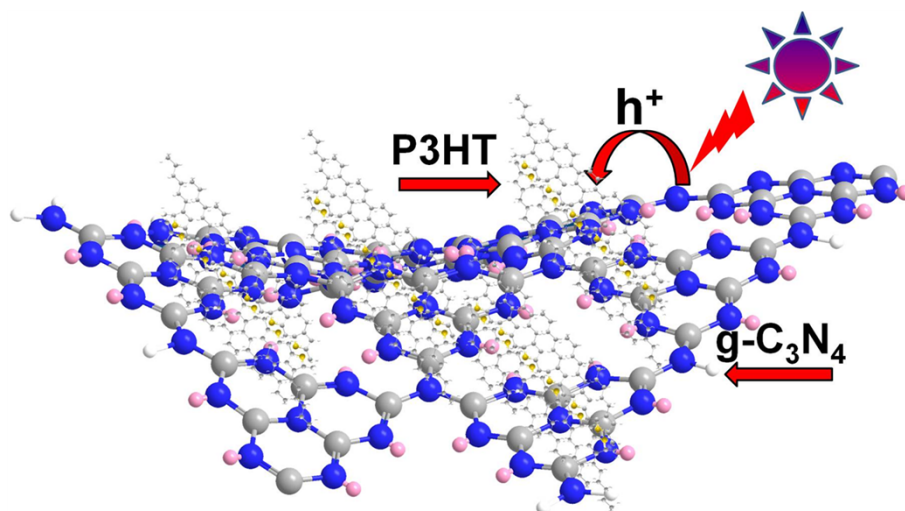


Figure S2 Schematic illustration of charge separation and photocatalytic process over g-C₃N₄ and P3HT-g-C₃N₄ photocatalysts under visible light irradiation.

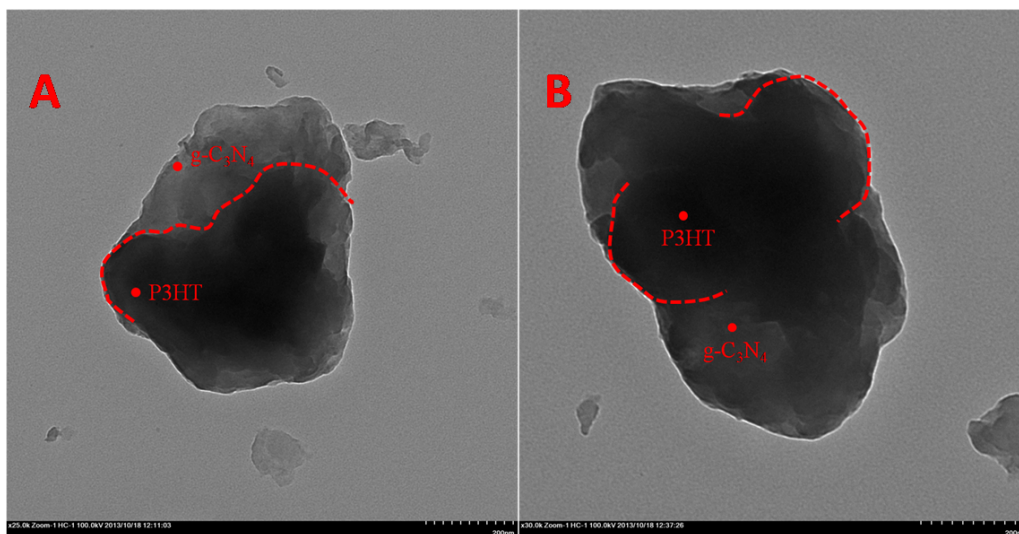


Figure S3. (A. B) TEM images of P3HT-g-C₃N₄ composites from a different perspective.

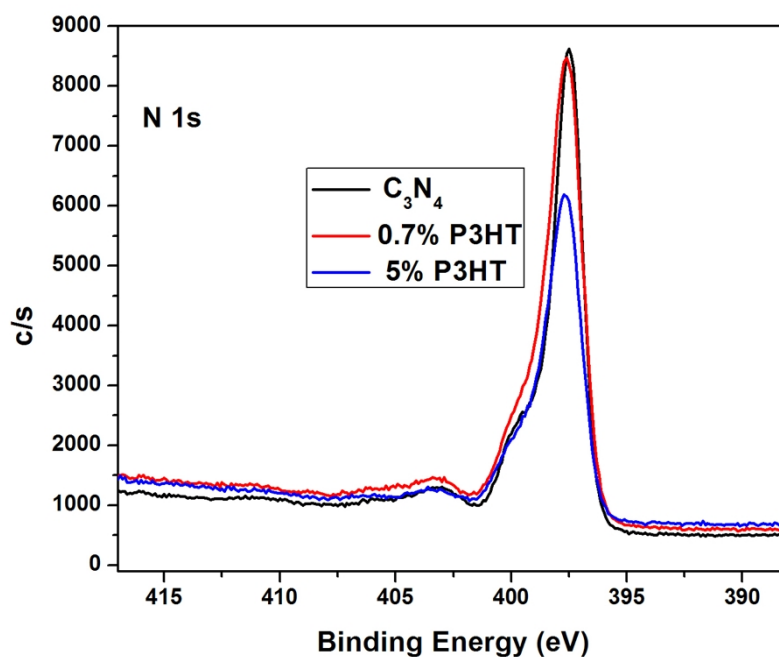


Figure S4. High-resolution XPS spectra of N1s recorded from P3HT, g- C_3N_4 and P3HT-g- C_3N_4 .

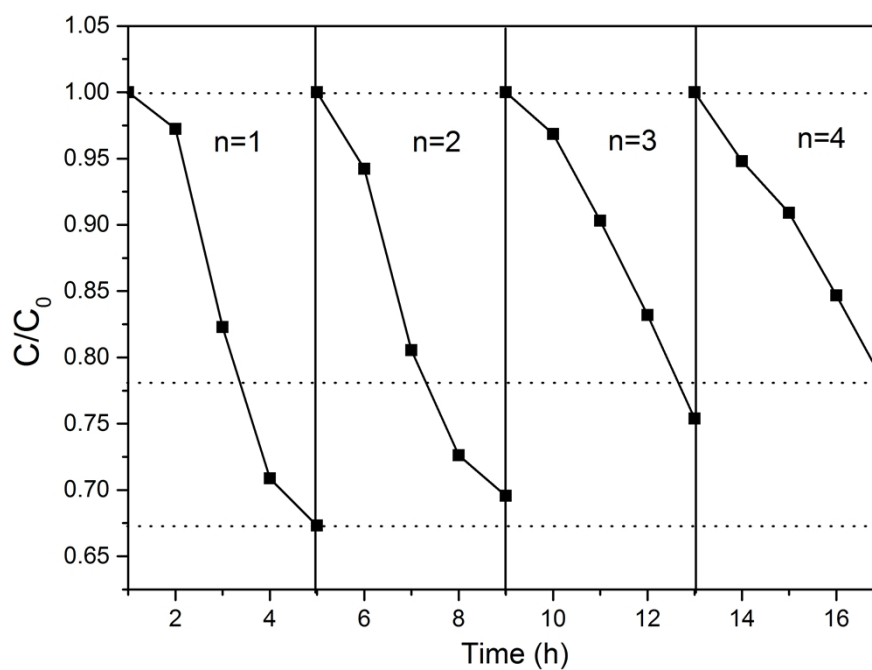


Figure S5. Photostability experiments for photocatalytic degradation of MB over 0.7 wt.% P3HT-g- C_3N_4 photocatalyst under visible light irradiation. ($[MB] = 3 \times 10^{-5} \text{ mol/L}$, $\lambda \geq 420 \text{ nm}$).

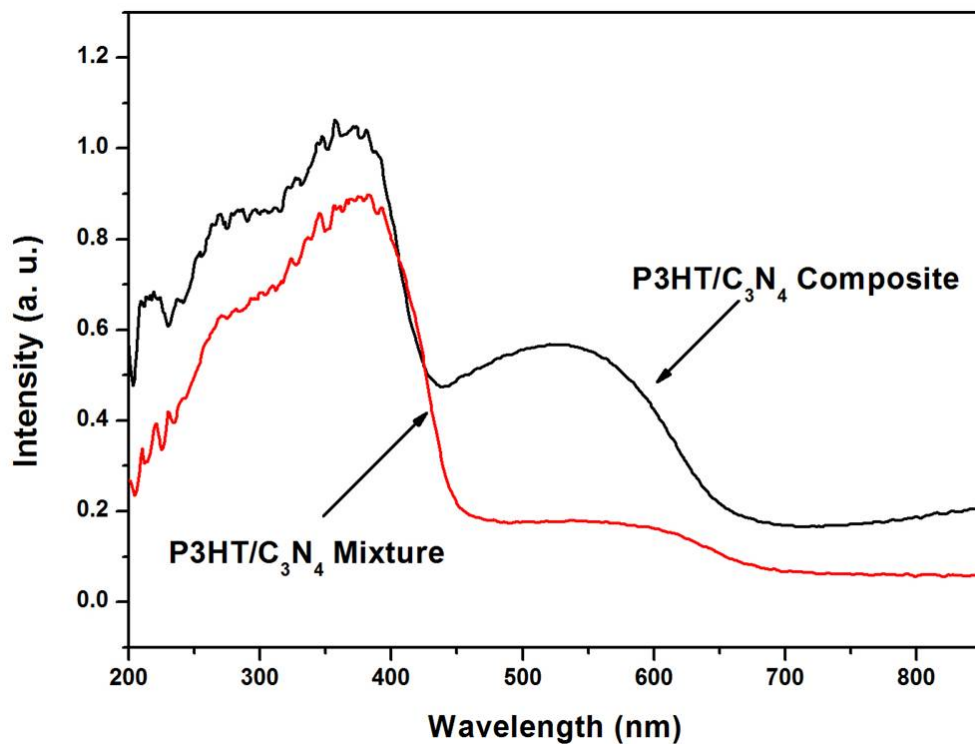


Figure S6 DRS of P3HT-g-C₃N₄ polymer composite and mixture

The N₂ adsorption/desorption isotherms are demonstrated as shown in Figure S3. The BET surface area (S_{BET}) of the pure g-C₃N₄ and 5.0 wt.% composites was 15.8 and 4.8 m² g⁻¹ respectively, while that of 0.7 wt.% composites was 3.8 m² g⁻¹, which is lower than pure g-C₃N₄ and 5.0 wt.% composites. This result demonstrated that the S_{BET} of the P3HT-g-C₃N₄ composite catalysts decreased with the addition of P3HT, indicating that the specific surface area does not contribute to photocatalytic activity of composite photocatalysts.

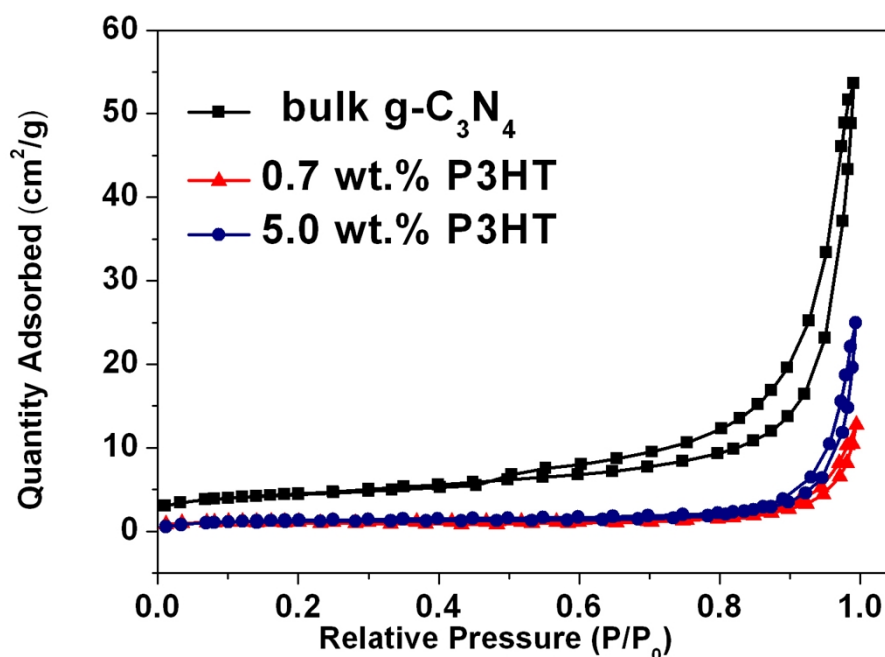


Figure S7. N₂ adsorption-desorption isotherms of the prepared bulk g-C₃N₄, 0.7 wt.% P3HT and 5.0 wt.% P3HT samples.

Reference

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