

## Electronic Supporting Information

### Scalable synthesis of $\text{CuInS}_2$ nanocrystal inks for photovoltaic applications

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#### 1. TEM analyses of $\text{CuInS}_2$ samples obtained after different reaction times

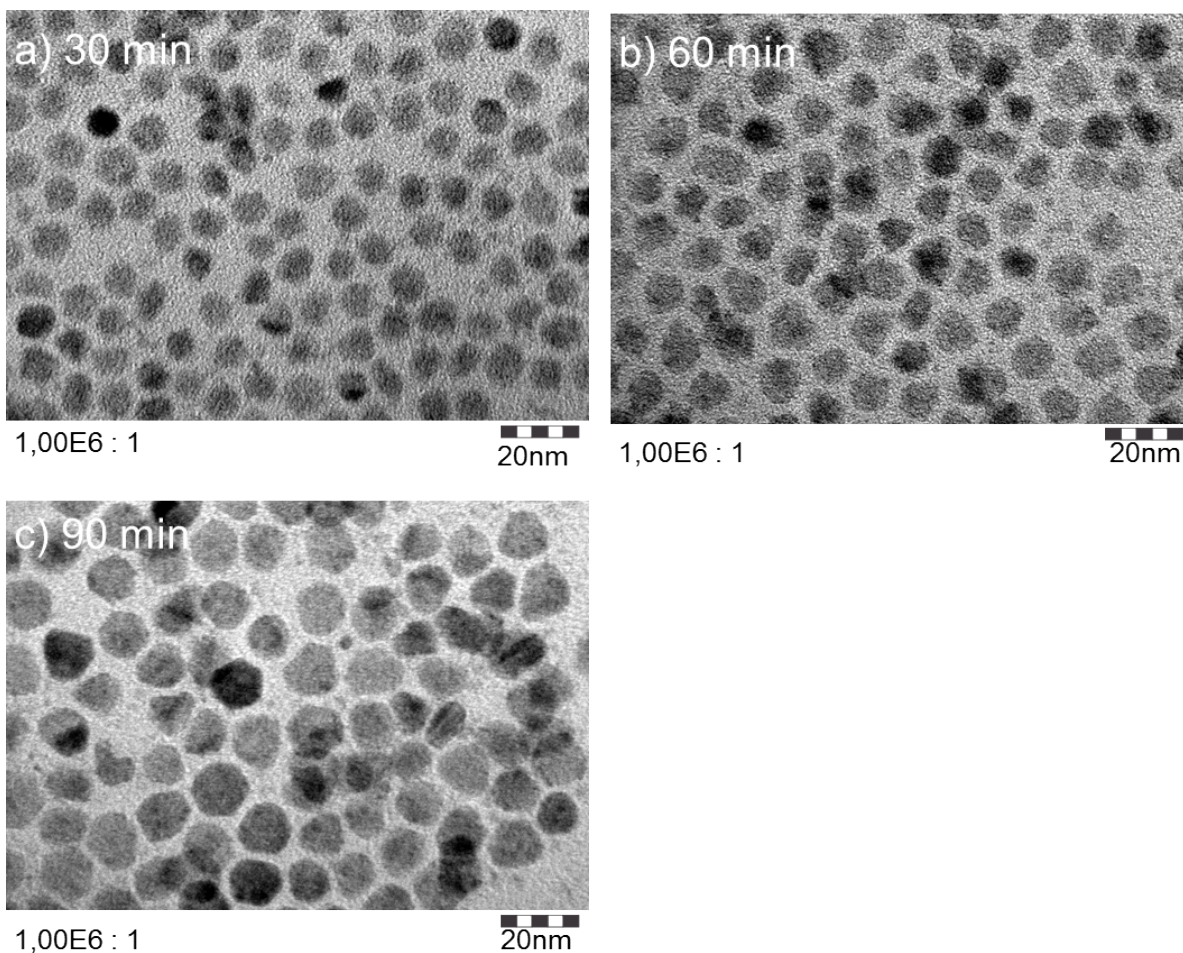


Figure S1: TEM analyses of samples obtained after different reaction times (a: 30 min, b: 60 min, c: 90 min; see main article for related TEM images taken after 1, 5 and 15 min reaction time).

## 2. HRTEM analysis of $\text{CuInS}_2$ nanoparticles

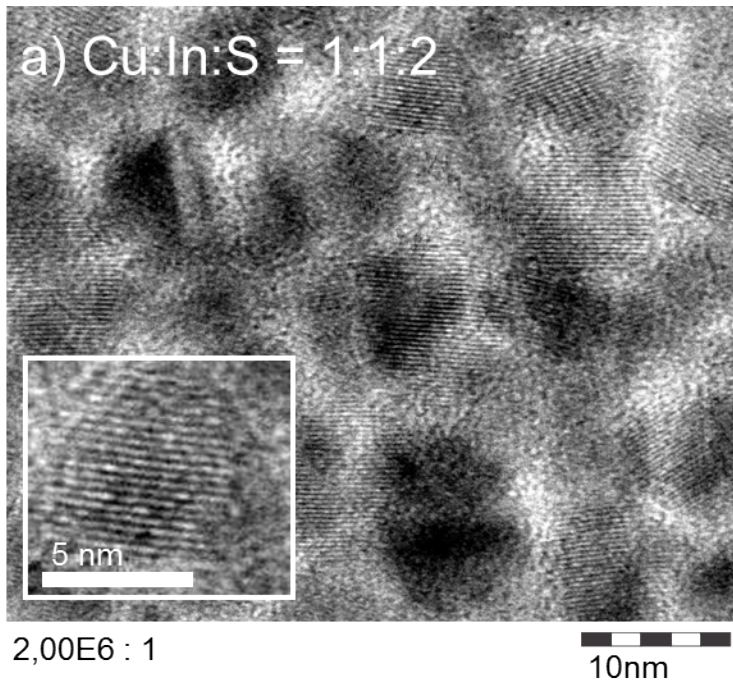
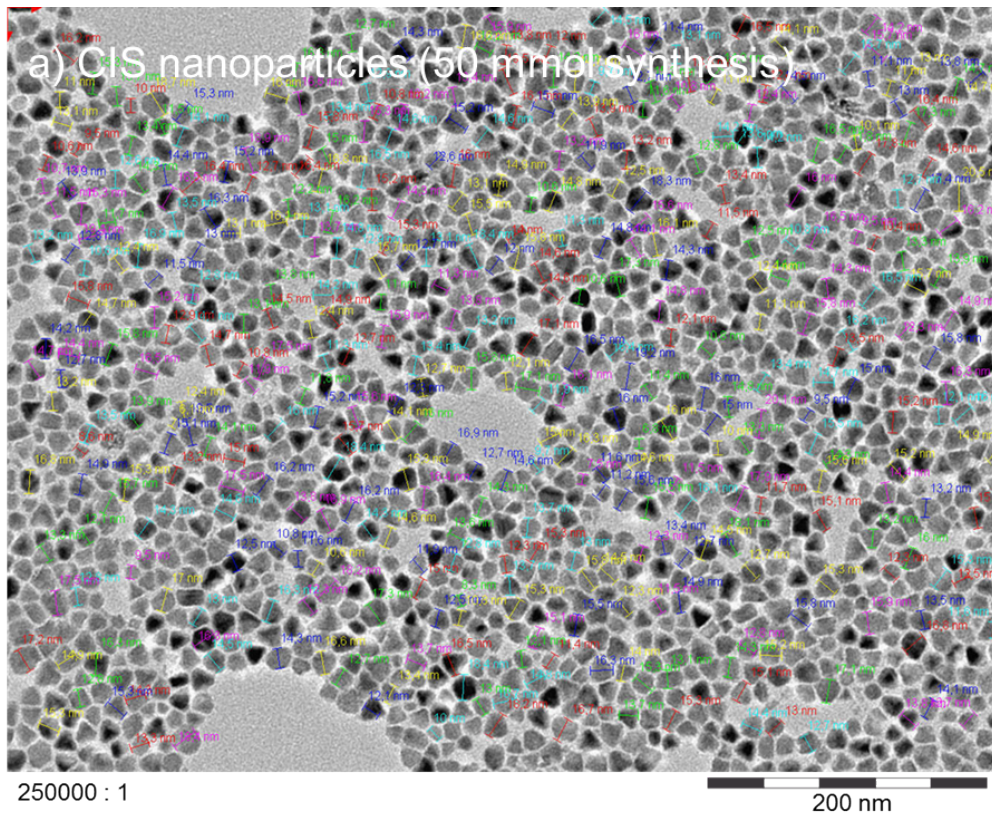


Figure S2: HRTEM analysis of  $\text{CuInS}_2$  sample prepared with  $\text{Cu:In:S} = 1:1:2$  precursor ratio. The observed spacing  $d = 3.2 \text{ \AA}$  corresponds to the (112) lattice planes of tetragonal (chalcopyrite)  $\text{CuInS}_2$ .

## 3. Size distribution of $\text{CuInS}_2$ nanoparticles obtained at 11g scale



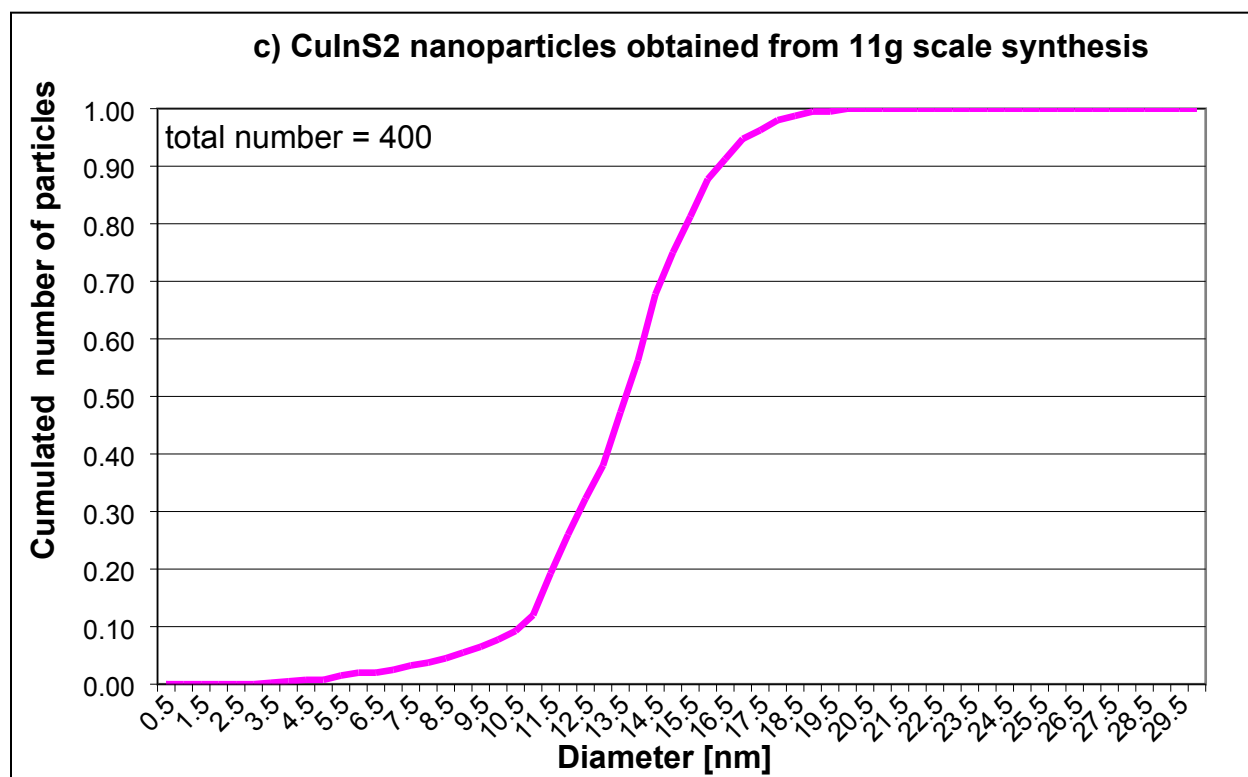
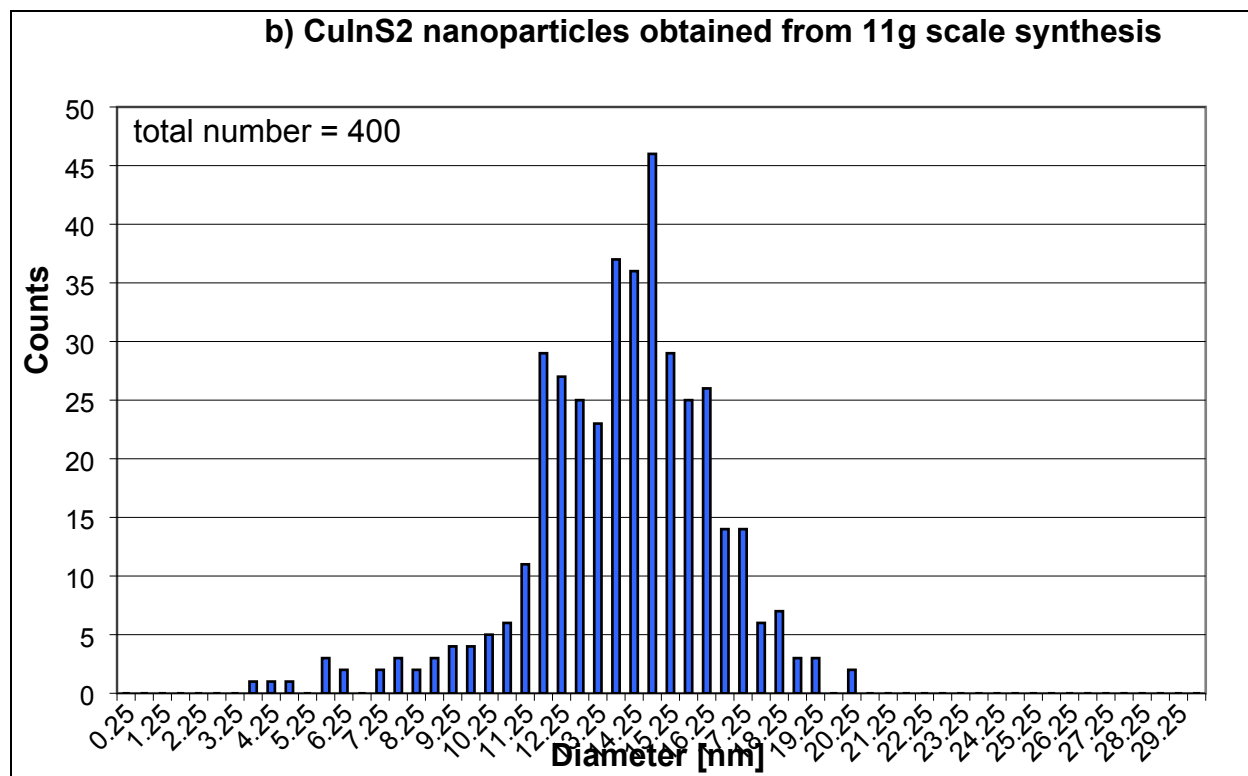


Figure S3: Size distribution of CIS nanoparticles obtained at 11g scale (see also Figure 3 in main article). a) TEM image showing the typical particle shapes and sizes, b) distribution of the diameter, and c) cumulated number of particles at different diameters.

#### 4. Effect of surface ligand exchange from oleylamine to pyridine:

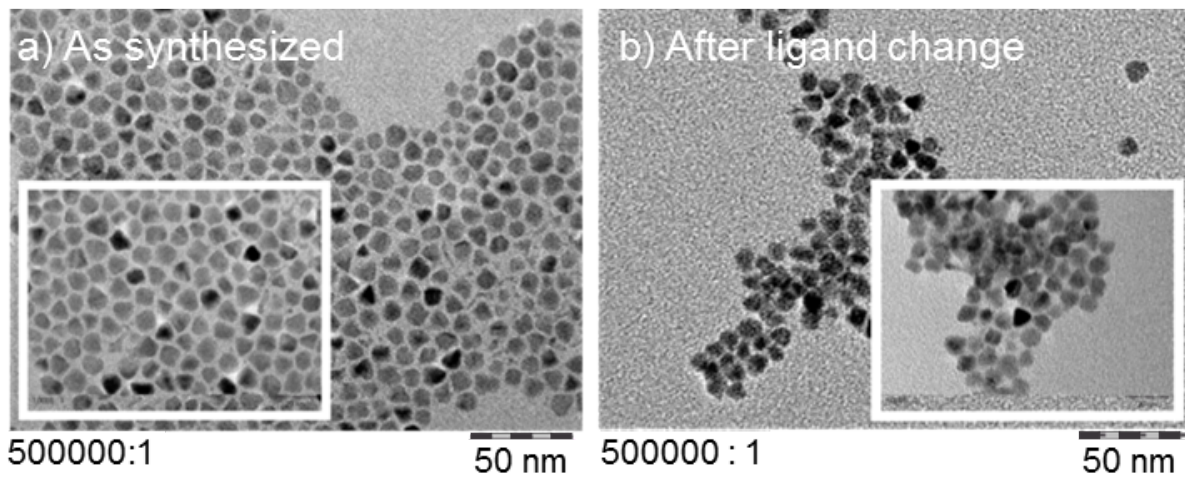


Figure S4: TEM analyses of samples obtained directly after synthesis (a) and after ligand exchange (b) from oleylamine to pyridine. The nanoparticles are in closer contact after surface modification, which hints at a successful exchange of long chain oleylamine surface ligand by pyridine.