

Supporting Information

**Low temperature synthesized carbon nanotube superstructures
with superior CO₂ and hydrogen storage capacity**

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Excess and Total hydrogen uptake

Our measurements provided the excess hydrogen uptake, which is the amount of hydrogen adsorbed in the carbons above that which would have been stored in the pores under similar conditions (temperature and pressure) assuming that there is zero energy of interaction between the hydrogen and the carbon pore walls. The total uptake is calculated from the excess storage by taking into account the amount of hydrogen compressed into the carbon pore volume space. Our gravimetric methods measured the excess hydrogen uptake (θ_{Exc}) from which the total storage (θ_T) was calculated from the equation:

$$\theta_T = \theta_{Exc} + \frac{100 \times d_{H_2} \times V_T}{(1 + d_{H_2} \times V_T)}$$

Where;

θ_T = total hydrogen uptake (wt%)

θ_{Exc} = excess hydrogen uptake (wt%)

d_{H_2} = density (g cm⁻³) of compressed hydrogen gas at the relevant temperature and pressure. The density was obtained from the National Institute of Standards and Technology (NIST) website (<http://www.nist.gov/>)

V_T = Pore volume (cm³ g⁻¹) of the carbon from nitrogen sorption analysis

Supporting Table S1. Textural properties and CO₂ uptake at 25 °C and 1 bar for activated CN2T samples prepared between 500 and 700 °C.

Sample	Surface area (m ² g ⁻¹)	Micropore surface area (m ² g ⁻¹) ^a	Pore volume (cm ³ g ⁻¹)	Micropore volume (cm ³ g ⁻¹) ^b	Pore size ^c (Å)	CO ₂ uptake at 25 °C and 1 bar (mmol g ⁻¹)
CN2500	1026	940 (92%)	0.56	0.41 (73%)	5.5/7	3.7
CN2550	1175	1091 (93%)	0.63	0.48 (76%)	5.5/7	4.2
CN2600	1479	1400 (95%)	0.83	0.67 (81%)	6/9/11	4.5
CN2650	1628	1577 (97%)	0.83	0.69 (83%)	7/8/9.5/12	4.8
CN2700	2102	1987 (95%)	1.18	0.95 (81%)	6/9/13	4.0

Supporting Table S2. CO₂ uptake at 0 °C of activated CNxT samples.

Sample	CO ₂ uptake at 0 °C (mmol g ⁻¹)			
	1 bar	5 bar	10 bar	20 bar
CN2600	6.1	10.3	11.6	12.4
CN2650	8.4	13.4	14.8	15.5
CN2700	6.4	13.8	16.4	17.8
CN2800	4.6	13.9	19.8	25.4
CN2900	3.8	11.8	17.5	25.7
CN4800	3.3	11.0	16.8	26.4

Table S3. Total gravimetric hydrogen uptake for sample CN4800 under cryogenic (-196 °C) or room temperature conditions.

Pressure (bar)	Hydrogen uptake	
	<u>Cryogenic (wt%)</u>	<u>Room temperature (wt%)</u>
1	2.7	0.12
2	3.6	0.16
4	4.5	0.23
8	5.6	0.37
10	6.0	0.44
20	7.5	0.80
30	8.7	1.13
40	9.7	1.45
50	10.6	1.76
60	11.3	2.08
70	11.9	2.36
80	12.5	2.65
90	13.0	2.92
100	13.4	3.20
110	13.8	3.45
120	14.1	3.70
130	14.5	3.95
140	14.7	4.20
150	14.9	4.43

Table S4. Surface area, pore volume and hydrogen storage (at -196 °C) for sample CN4800 compared to the best benchmark metal organic frameworks (MOFs).

Sample	Surface area (m ² g ⁻¹)	Pore Volume (cm ³ g ⁻¹)	Gravimetric H ₂ uptake (wt%)		Reference
			Excess ^a	Total ^b	
CN4800	3802	2.98	6.0	12.5 (14.9) ^c	This work
MOF-5	3800	1.55	7.1	9.6 (11.1) ^c	1
MOF-177	4740	1.89	7.1	11.3	2
NOTT-112	3800	1.62	7.1	10.0	3
NU-100	6143	2.82	9.2	16.4	4
MOF-210	6240	3.60	8.3	17.6	5

^aExcess H₂ uptake at ca. 40 bar. ^bTotal H₂ uptake at 80 bar. ^cValues in parenthesis are total H₂ uptake at 150 bar.

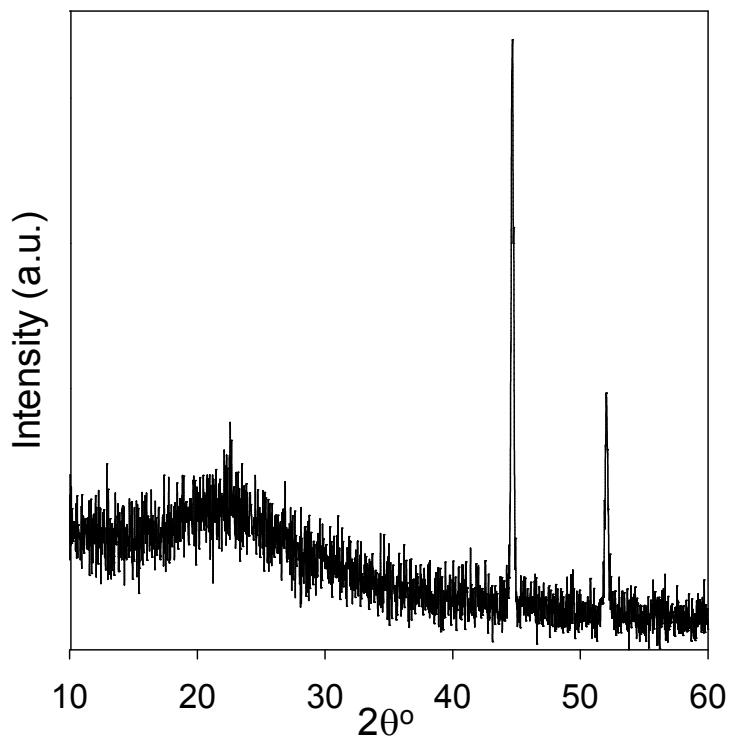
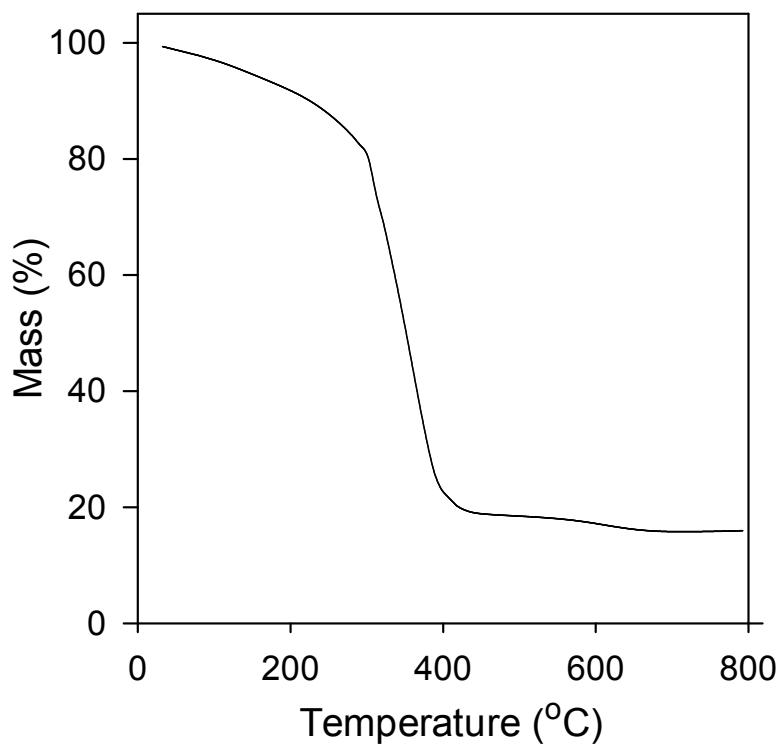
1. S. S. Kaye, A. Dailly, O. M. Yaghi and J. R. Long, *J. Am. Chem. Soc.* 2007, **129**, 14176–14177.
2. H. Furukawa, M. A. Miller and O. M. Yaghi, *J. Mater. Chem.* 2007, **17**, 3197–3204.
3. Y. Yan, X. Lin, S. Yang, A. J. Blake, A. Dailly, N. R. Champness, P. Hubberstey and M. Schroder, *Chem. Commun.* 2009, 1025–1027.
4. O. K. Farha, A. O. Yazaydin, I. Eryazici, C. D. Malliakas, B. G. Hauser, M. G. Kanatzidis, S. T. Nguyen, R. Q. Snurr and J. T. Hupp, *Nat. Chem.* 2010, **2**, 944–948.
5. H. Furukawa, N. Ko, Y. B. Go, N. Aratani, S. B. Choi, E. Choi, A. Ö. Yazaydin, R. Q. Snurr, M. O’Keeffe, J. Kim and O. M. Yaghi, *Science* 2010, **329**, 424–428.

Table S5. Surface area, pore volume and hydrogen storage at room temperature (25 °C) for sample CN4800 compared to the best benchmark carbon and metal organic frameworks (MOFs).

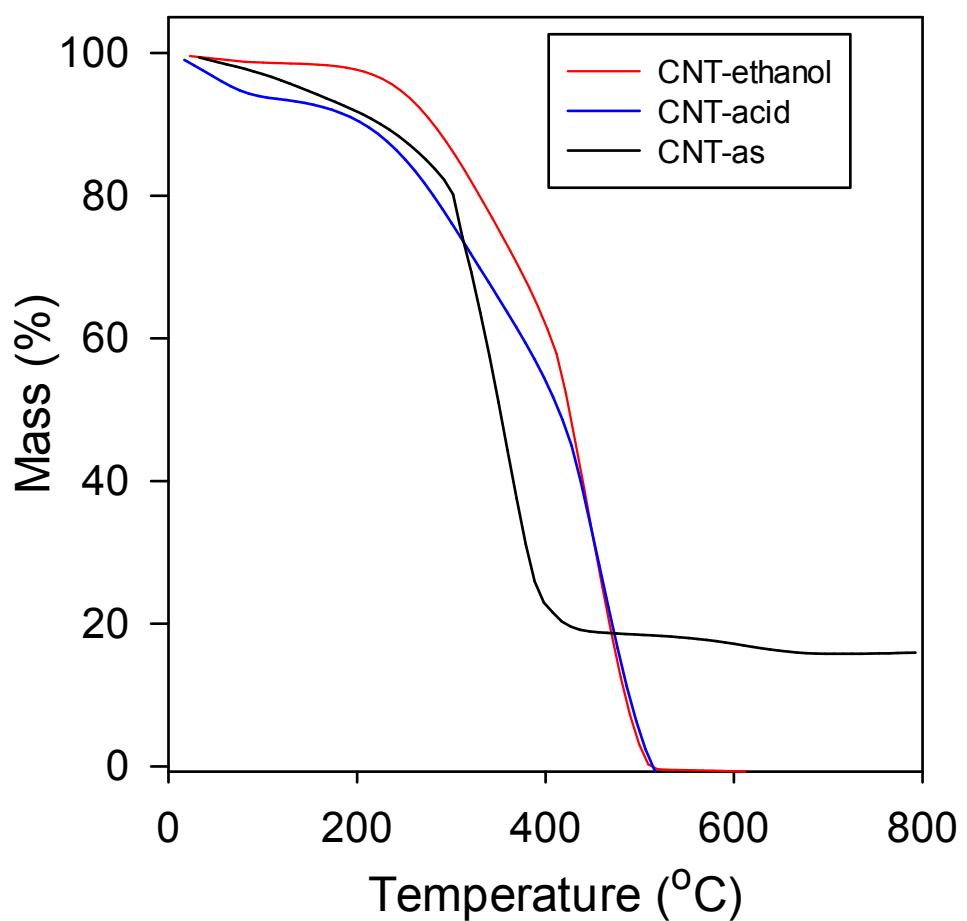
Sample	Surface area (m ² /g)	Pore Volume (cm ³ /g)	H ₂ uptake (wt%)	Reference
			Excess Total ^a	
CN4800	3802	2.98	1.3 4.4	This work
Carbon KUA5	3183	1.25	1.0 2.7	1,2
ZTC	3800	1.70	1.2 2.9	3
ZTC	3600	1.66	1.3 2.8	4
MOF-5	3800	1.55	0.7 2.1	5
MOF-210	6240	3.60	0.5 ^b 2.7 ^b	6,7

^aTotal H₂ uptake at 25 °C and 150 bar. ^buptake at 80 bar.

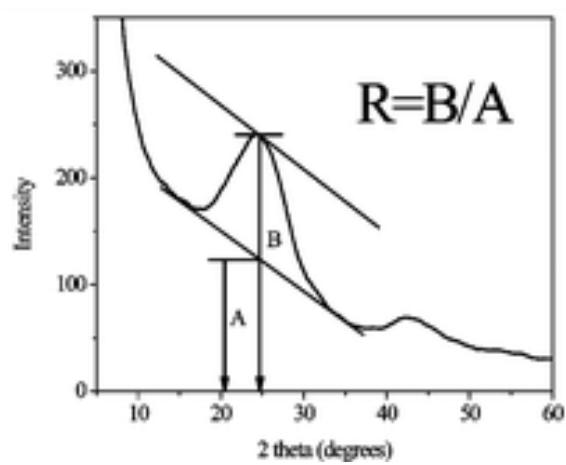
1. M. Jordá-Beneyto, F. Suárez-García, D. Lozano-Castelló, D. Cazorla-Amorós, A. Linares-Solano, *Carbon* 2007, **45**, 293–303.
2. M. Jorda-Beneyto, D. Lozano-Castello, F. Suarez-Garcia, D. Cazorla-Amoros and A. Linares-Solano, *Microporous Mesoporous Mater.* 2008, **112**, 235–242.
3. H. Nishihara, P. X. Li, L. X. Hou, M. Ito, M. Uchiyama, T. Kaburagi, A. Ikura, J. Katamura, T. Kawarada, K. Mizuuchi and T. Kyotani, *J. Phys. Chem. C* 2009, **113**, 3189–3196.
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5. S. S. Kaye, A. Dailly, O. M. Yaghi and J. R. Long, *J. Am. Chem. Soc.* 2007, **129**, 14176–14177.
6. O. M. Yaghi, http://www.hydrogen.energy.gov/pdfs/review11/st049_yaghi_2011_p.pdf
7. O. M. Yaghi, http://www.hydrogen.energy.gov/pdfs/progress11/iv_c_9_yaghi_2011.pdf



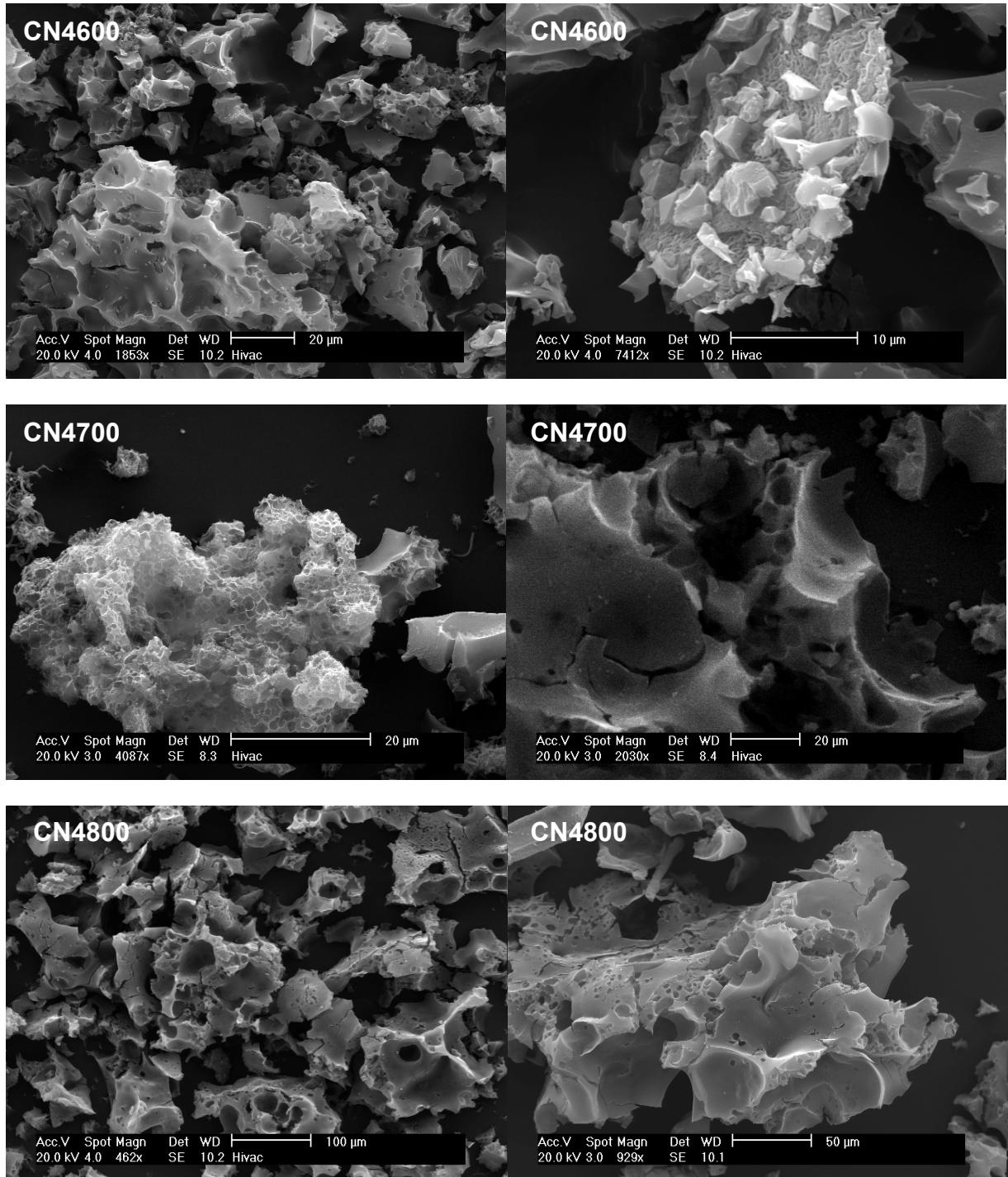
Supporting Figure S1. Thermogravimetric analysis (TGA) curve (top) and XRD pattern (bottom) of as-synthesised (CNT-as) sample.



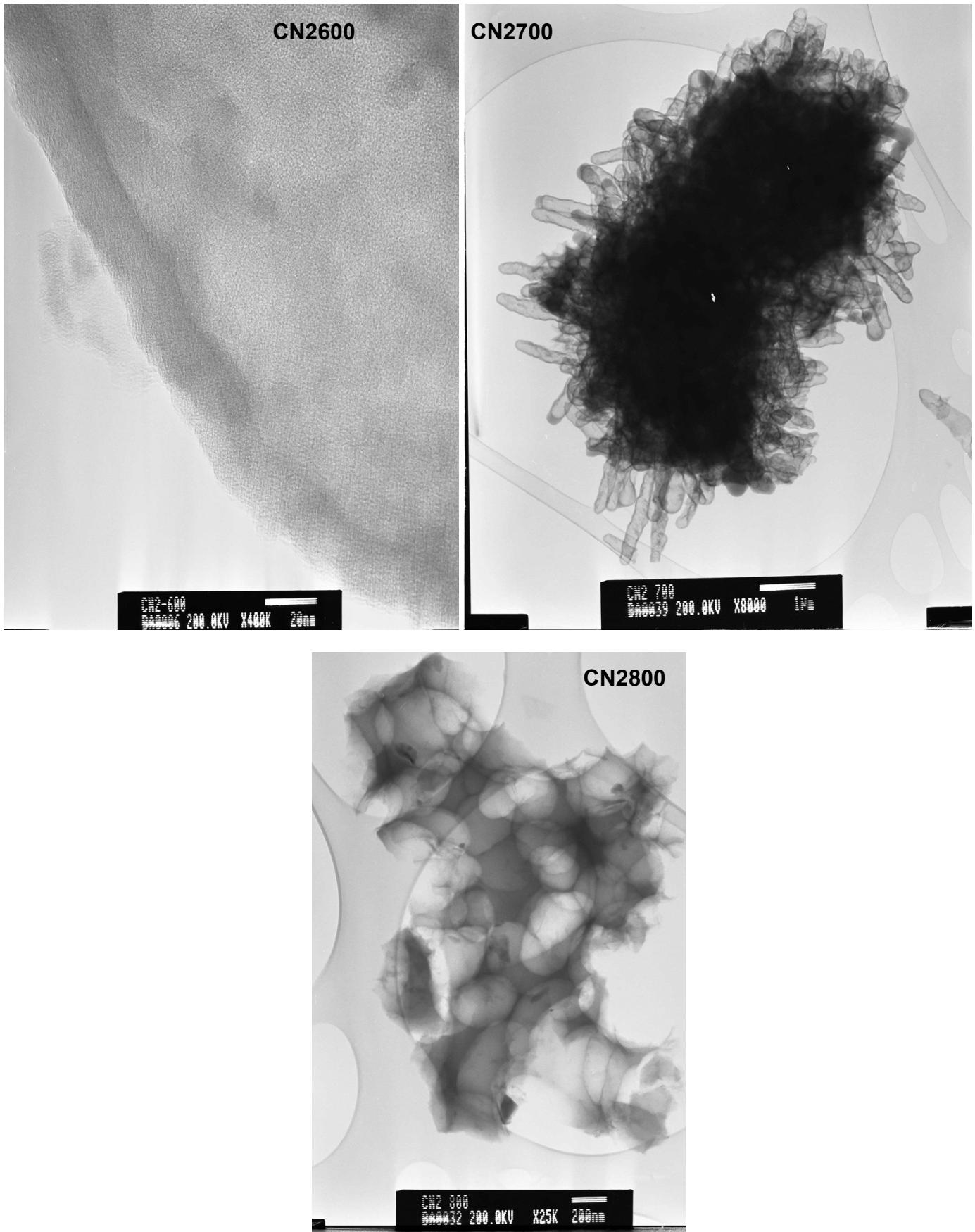
Supporting Figure S2. Thermogravimetric analysis curves of as-synthesised (CNT-as) sample, and after purification via washing in acid (CNT-acid) or ethanol (CNT-ethanol).



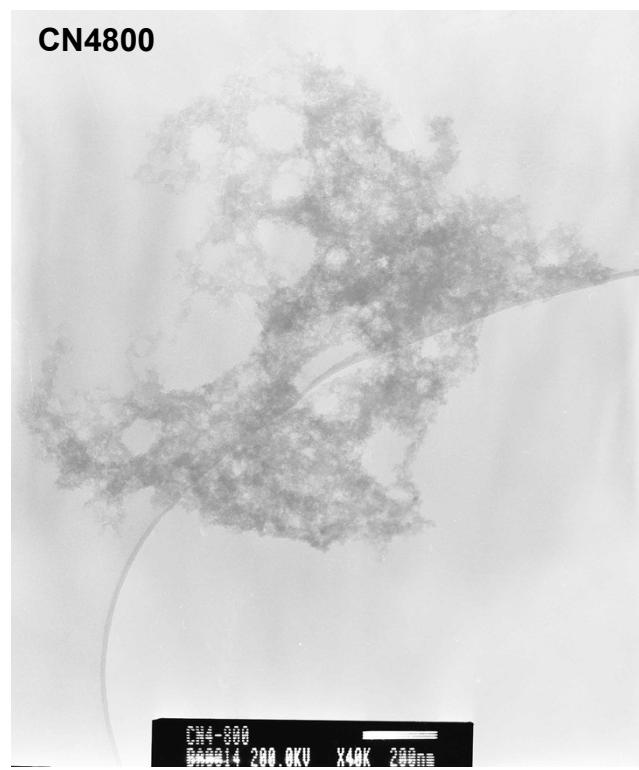
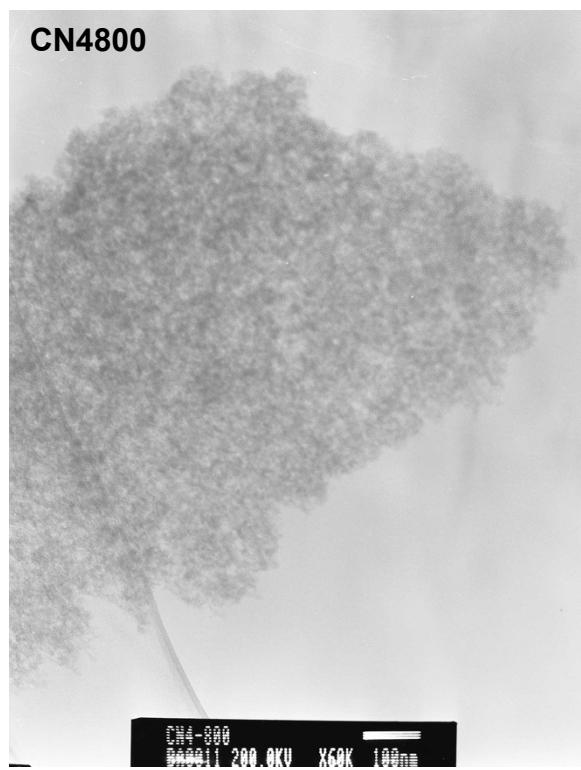
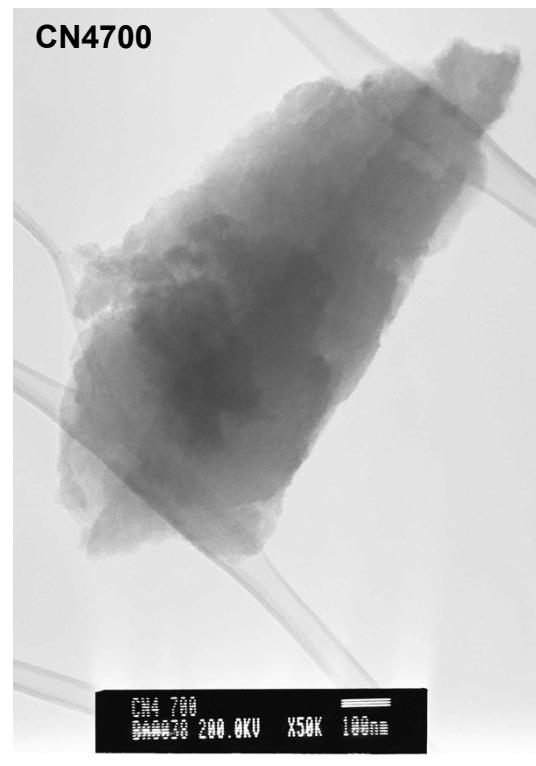
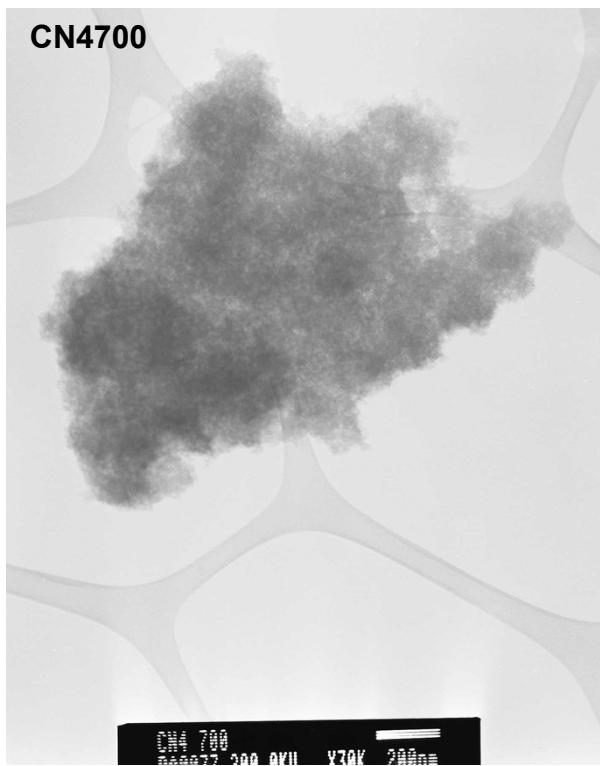
Supporting Figure S3. Illustration of calculating the R value from XRD patterns.
Adapted from ref. 37.



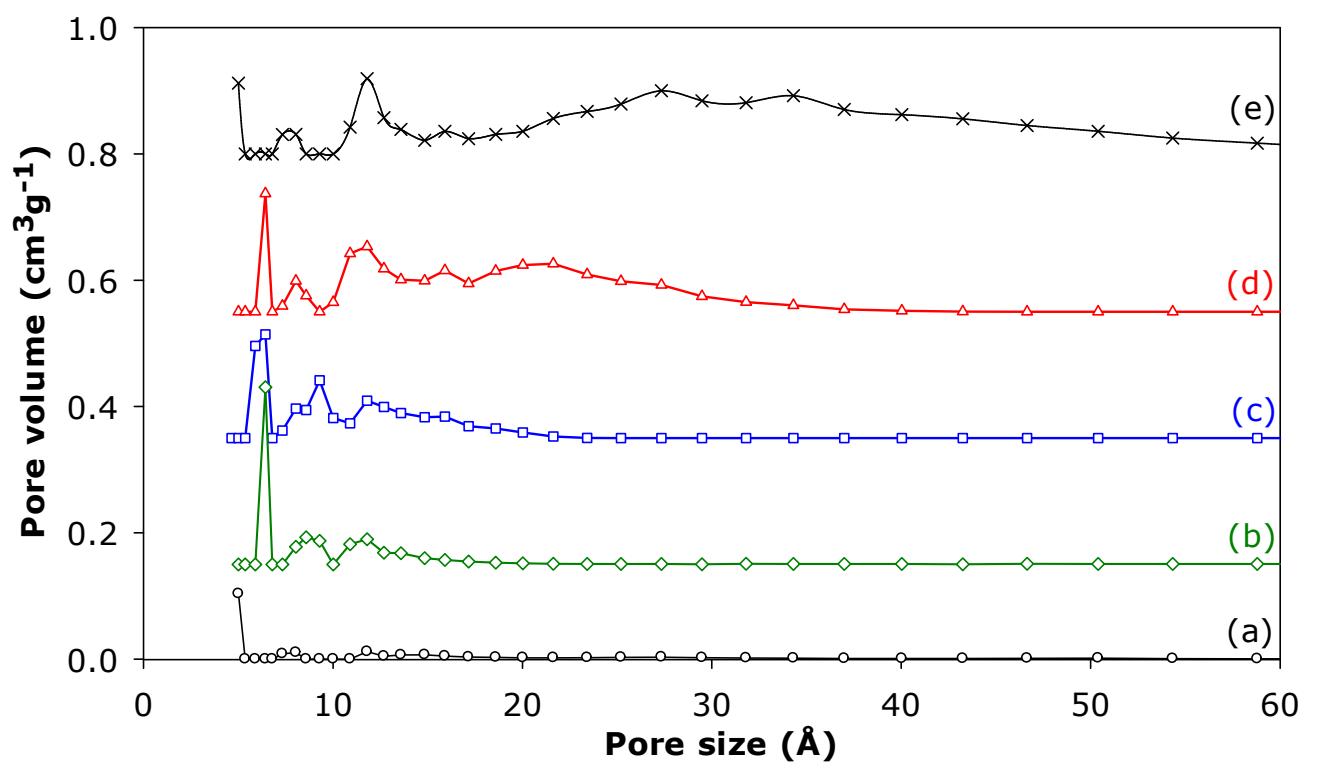
Supporting Figure S4. SEM images of activated CNT samples prepared at KOH/carbon ratio of 4 at various temperatures.



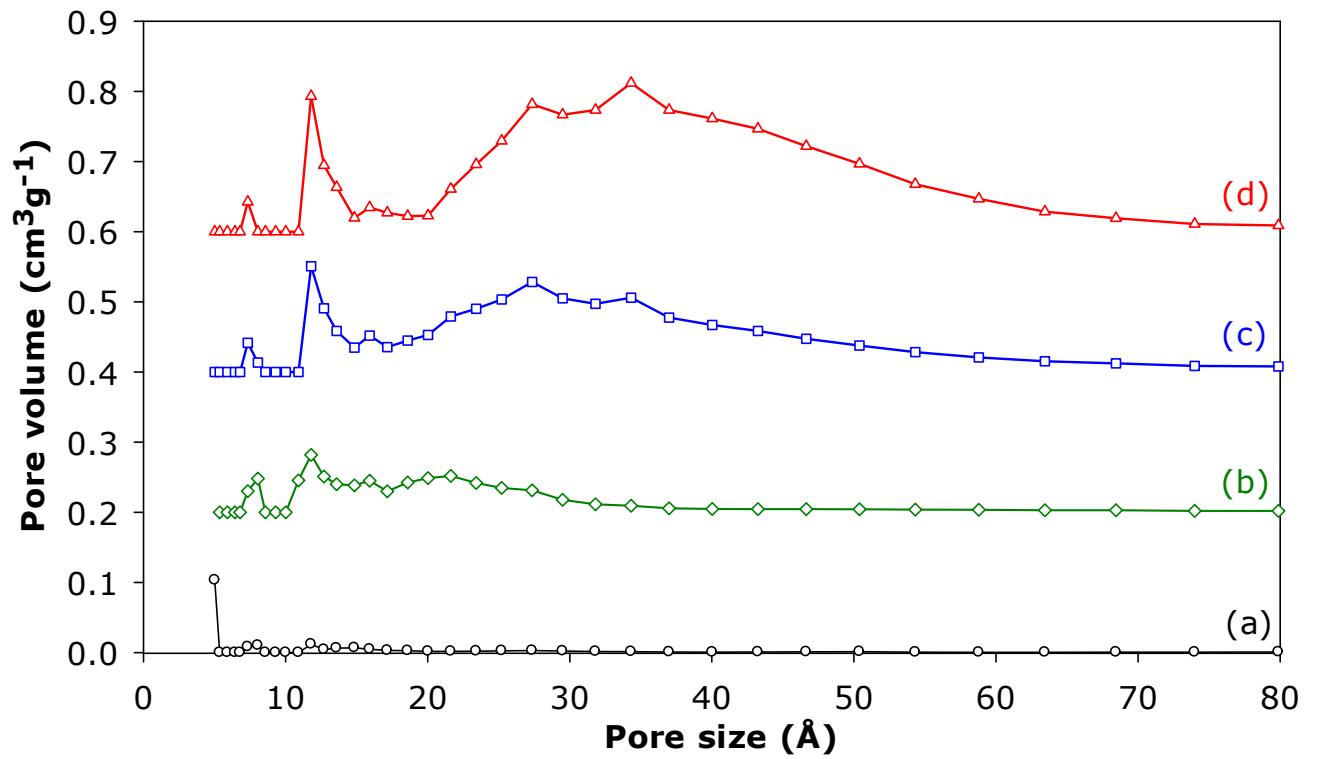
Supporting Figure S5. TEM images of activated CN2T samples.



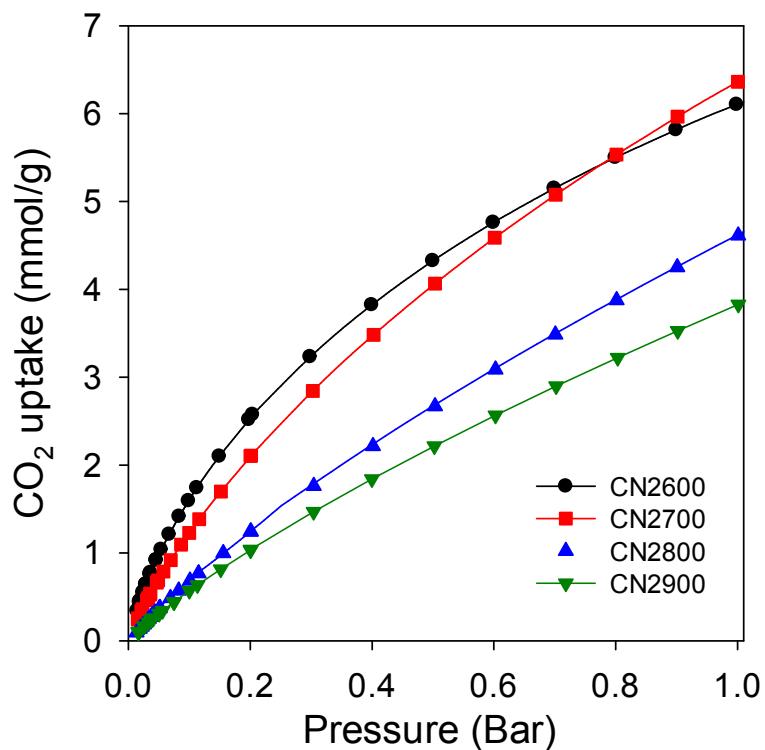
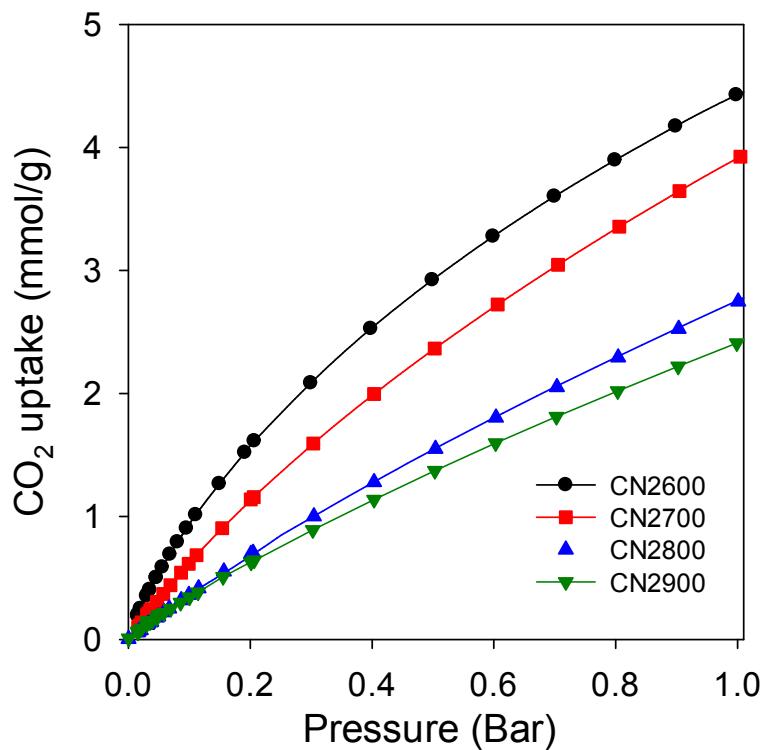
Supporting Figure S6. TEM images of activated CN4T samples.



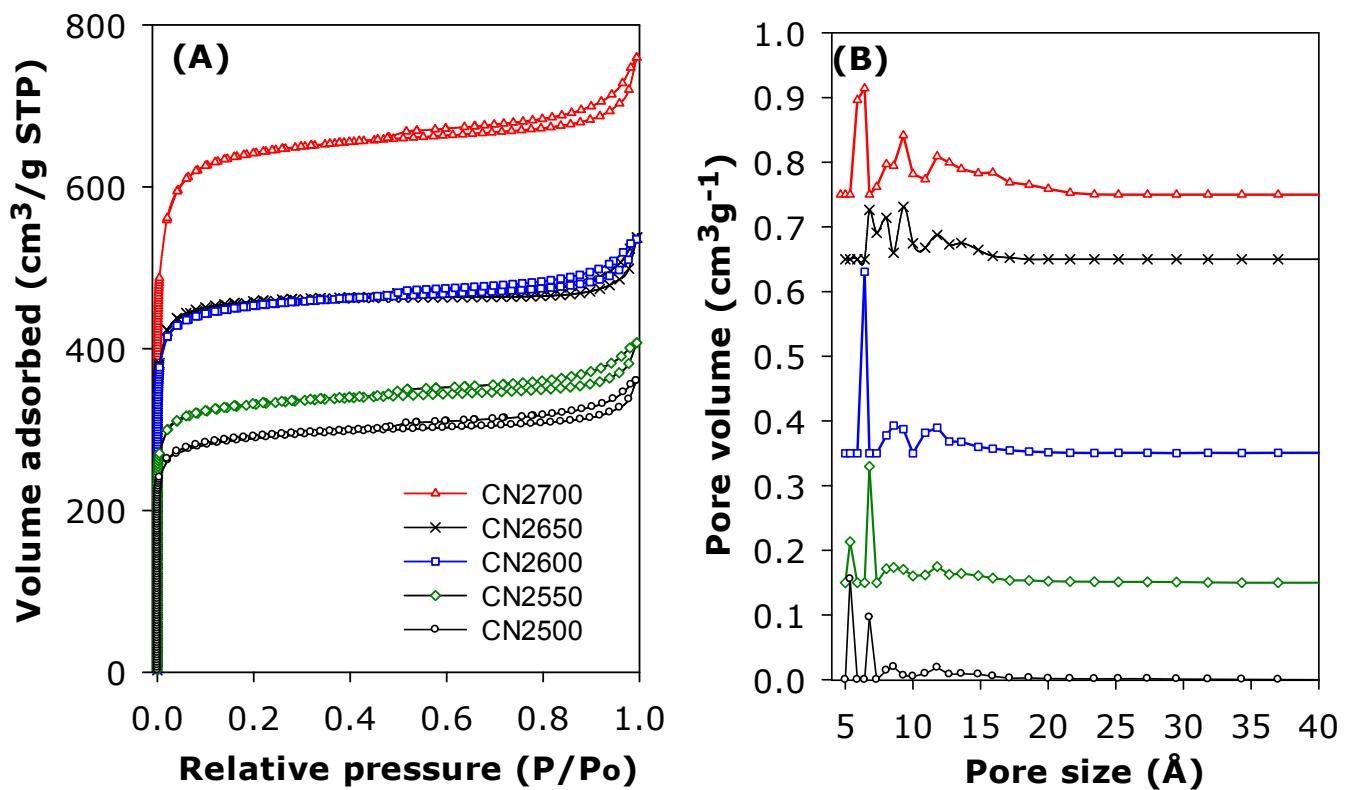
Supporting Figure S7. Pore size distribution curves of (a) pristine (CNT-as) sample, and activated CN2T samples; (b) CN2600, (c) CN2700, (d) CN2800 and (e) CN2900.



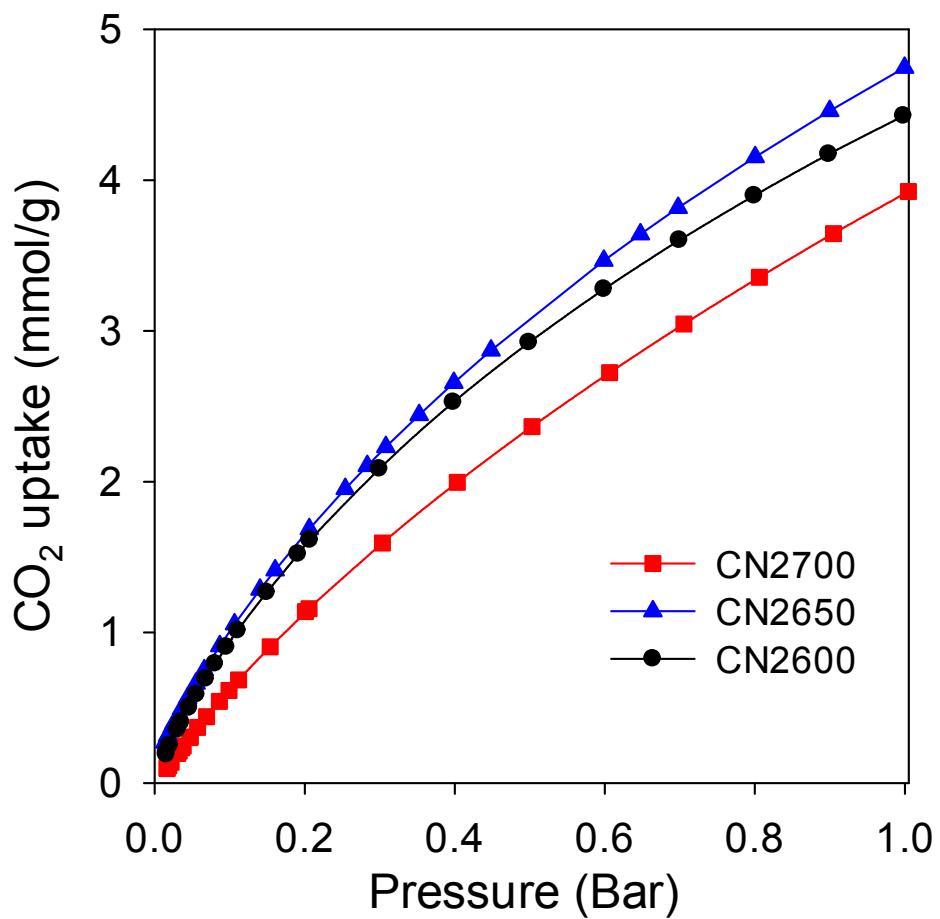
Supporting Figure S8. Pore size distribution curves of (a) pristine (CNT-as) sample and activated CN4T samples; (b) CN4600, (c) CN4700 and (d) CN4800.



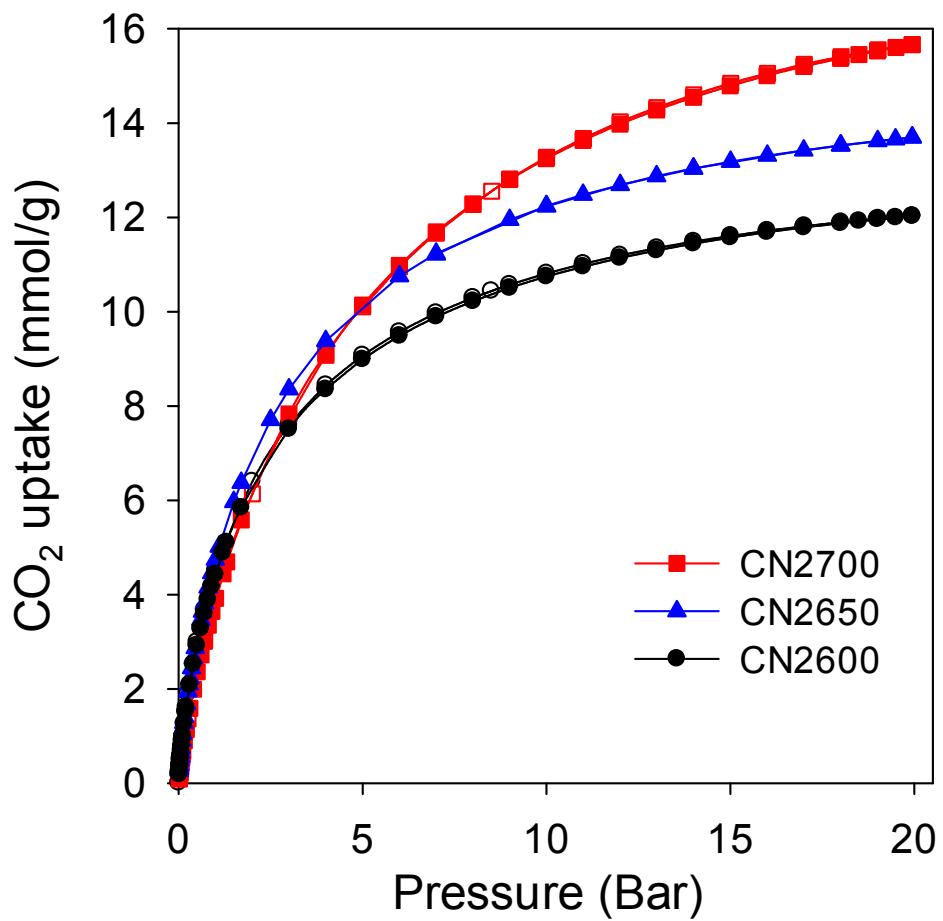
Supporting Figure S9. Low pressure CO₂ uptake isotherms for activated CN2T carbons at 25 °C (top panel) and 0 °C (bottom panel) in the pressure range 0 - 1 bar.



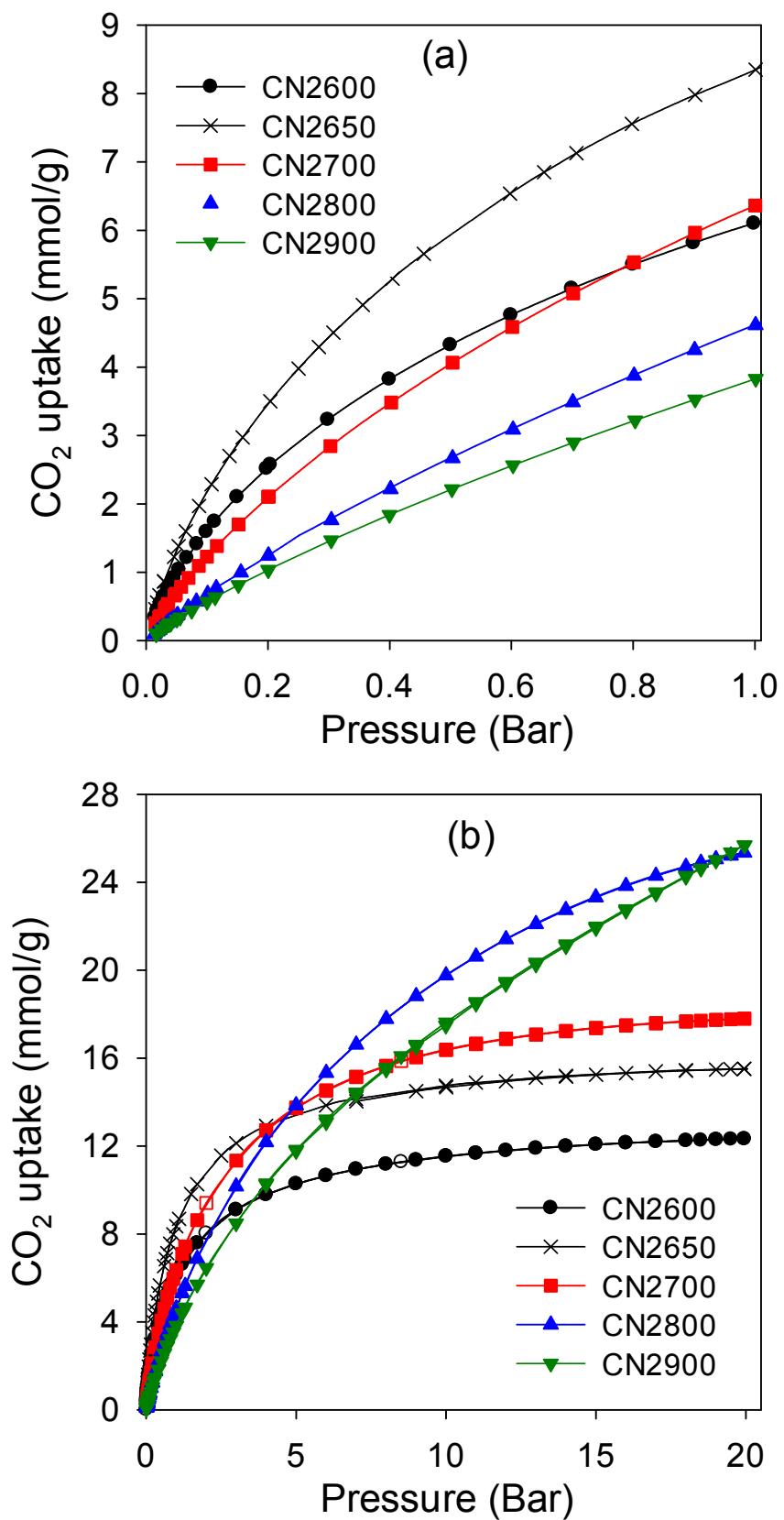
Supporting Figure S10. Nitrogen sorption isotherms (A) and pore size distribution curves (B) of activated CN2T samples prepared between 500 and 700 °C.



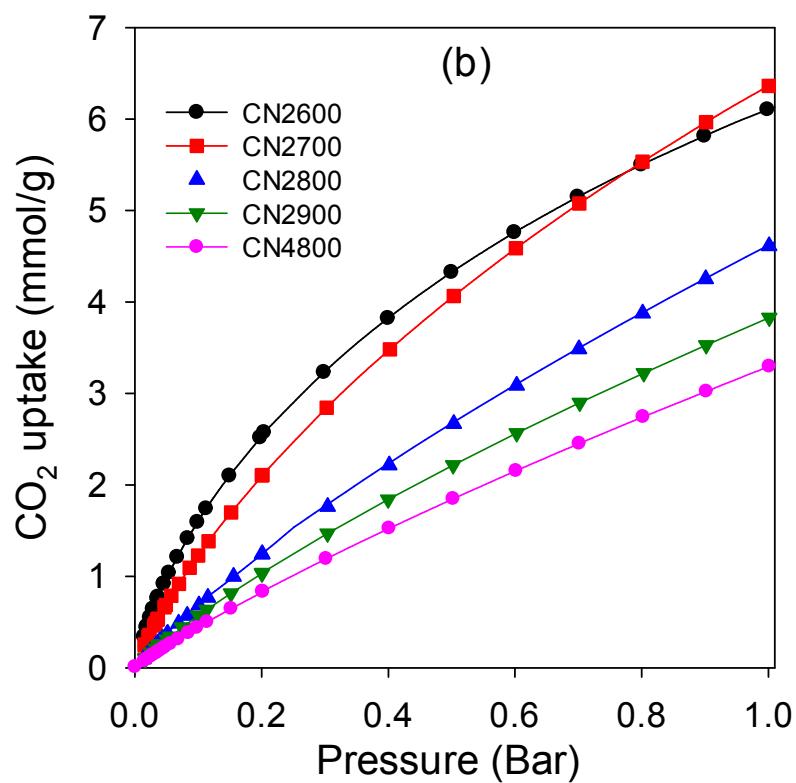
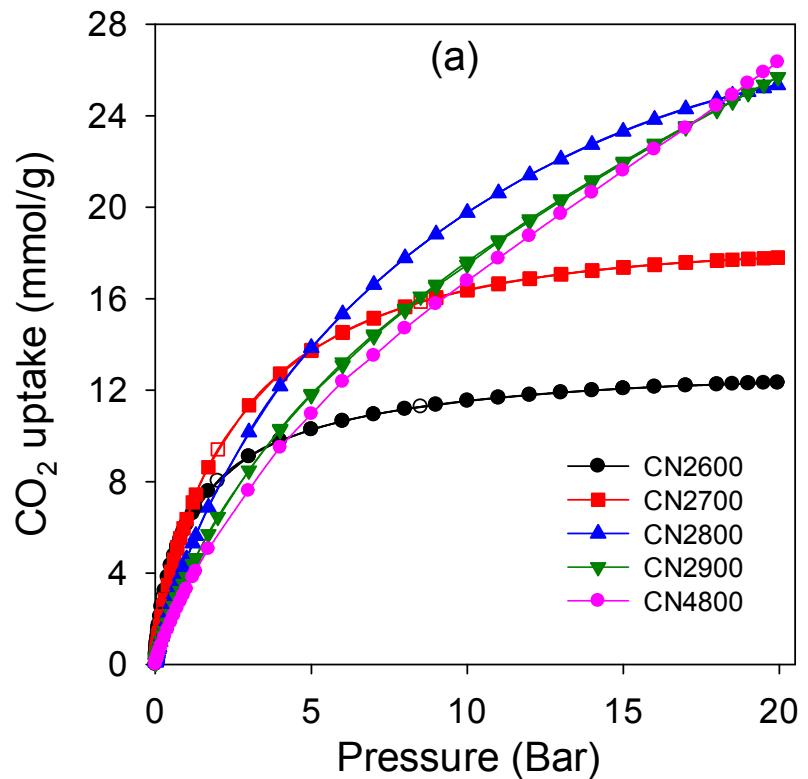
Supporting Figure S11. Low pressure CO₂ uptake isotherms at 25 °C for activated CN2T samples for T = 600, 650 and 700 °C.



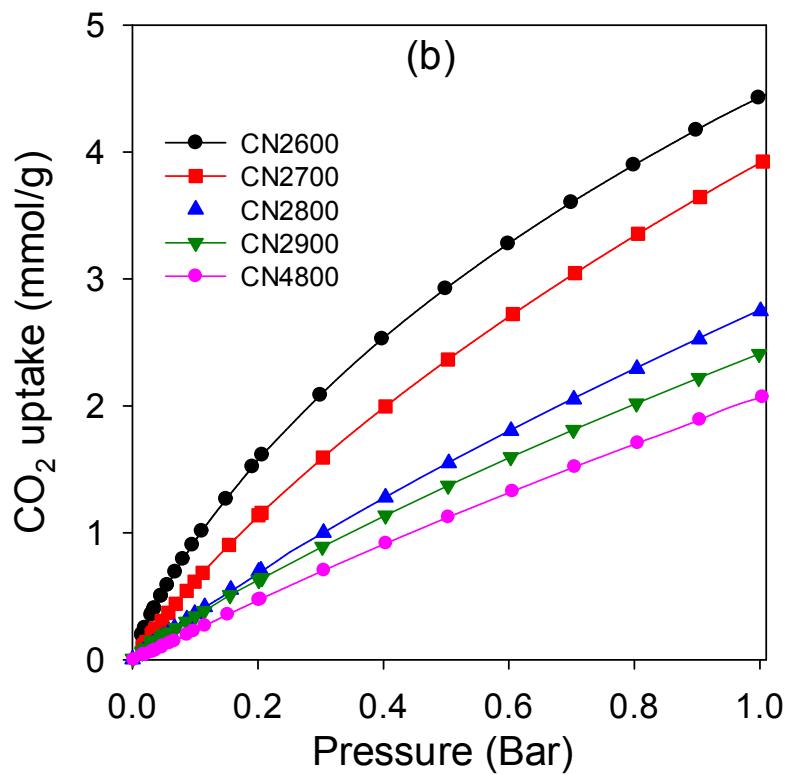
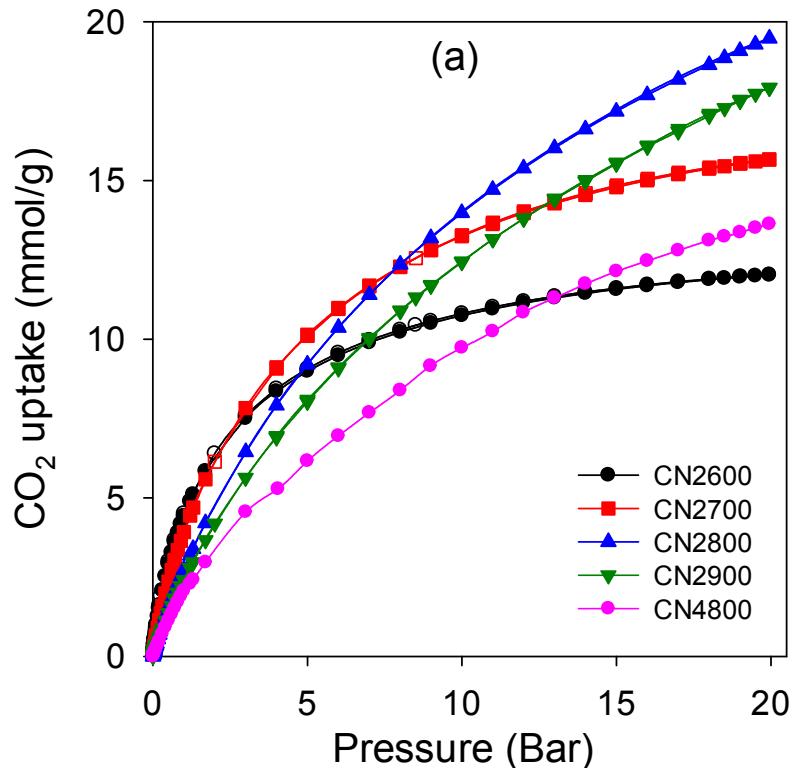
Supporting Figure S12. CO₂ uptake isotherms at 25 °C and 0 – 20 bar for activated CN2T samples for T = 600, 650 and 700 °C.



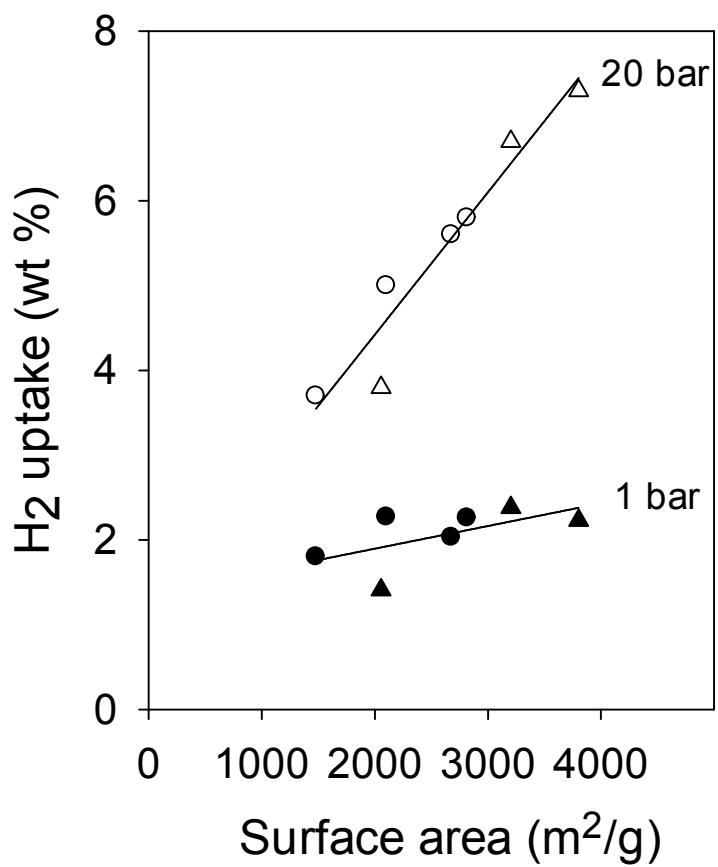
Supporting Figure S13. CO₂ uptake isotherms for activated CN2T samples at 0 °C and (a) low pressure (0 – 1 bar) and (b) 0 – 20 bar.



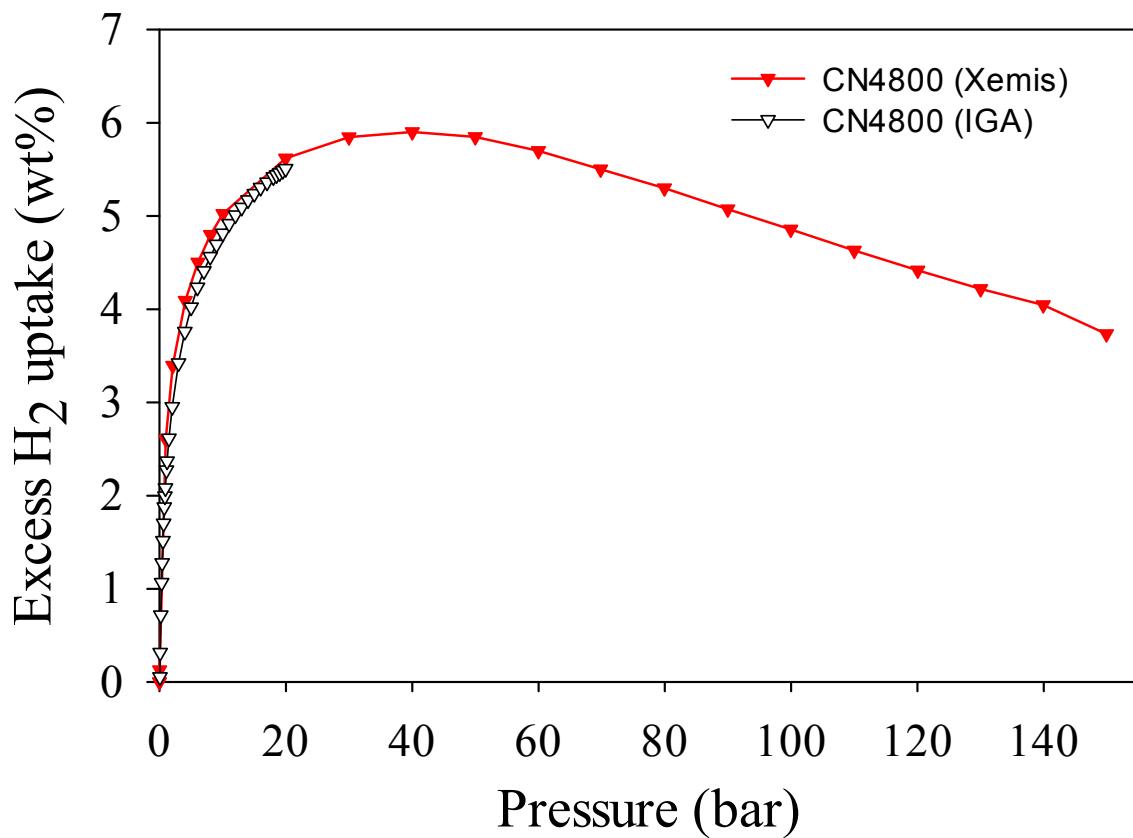
Supporting Figure S14. CO₂ uptake isotherms at 0 °C for sample CN4800 compared to CN2T samples; (a) 0 – 20 bar and (a) low pressure (0 – 1 bar).



Supporting Figure S15. CO₂ uptake isotherms at 25 °C for sample CN4800 compared to CN2T samples; (a) 0 – 20 bar and (a) low pressure (0 – 1 bar).



Supporting Figure S16. Plot of hydrogen uptake of CN2T (\circ, ∞) and CN4T (Δ, \blacktriangle) samples, at 1 bar and 20 bar, as a function of surface area.



Supporting Figure S17. Excess hydrogen uptake data for sample CN4800 at -196 °C obtained on a Hiden IGA compared to that obtained on a Hiden Xemis analyser. There is close agreement between hydrogen uptake data from IGA and Xemis analysers in the range 0 – 20 bar.