

## Supporting Information

### Parametric Investigation of Room-temperature Fluoride-ion Batteries: Assessment of Electrolytes, Mg-based Anodes, and BiF<sub>3</sub>-Cathodes

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Figure S1: SEM/EDX of separator fibers (Whatman filter)

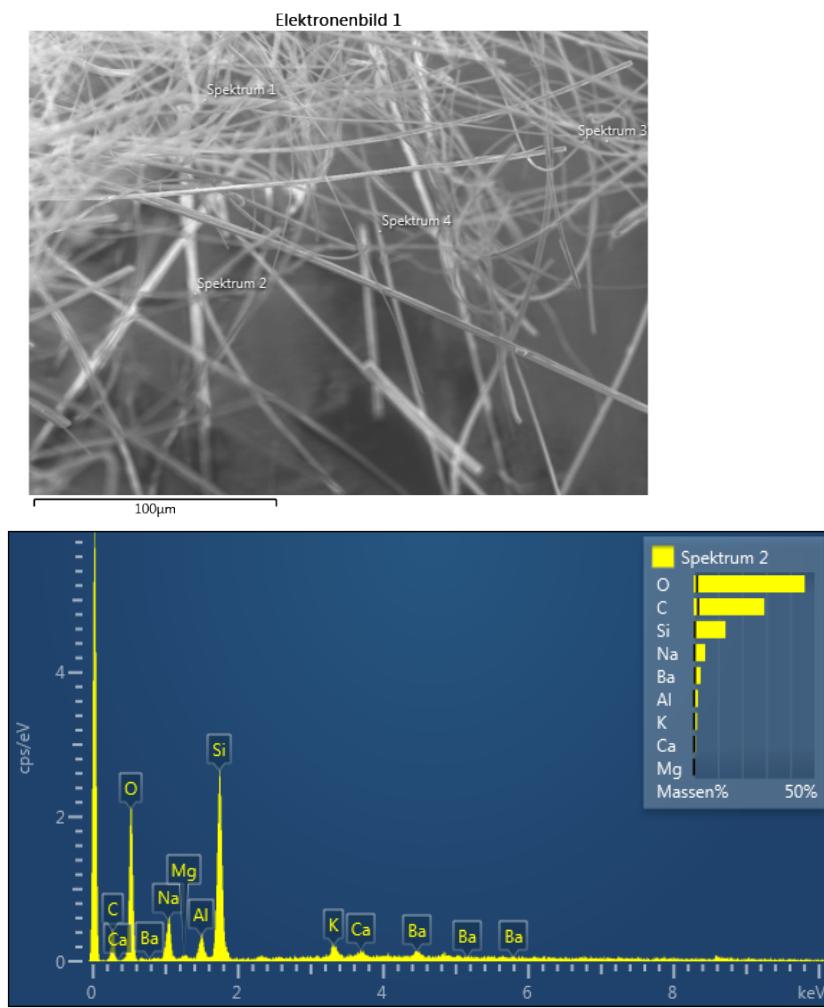


Figure S2: IR spectra of the three electrolyte; **TG (1)**, **Hi (2)** and **PEO (3)** with ammonium bifluoride as reference

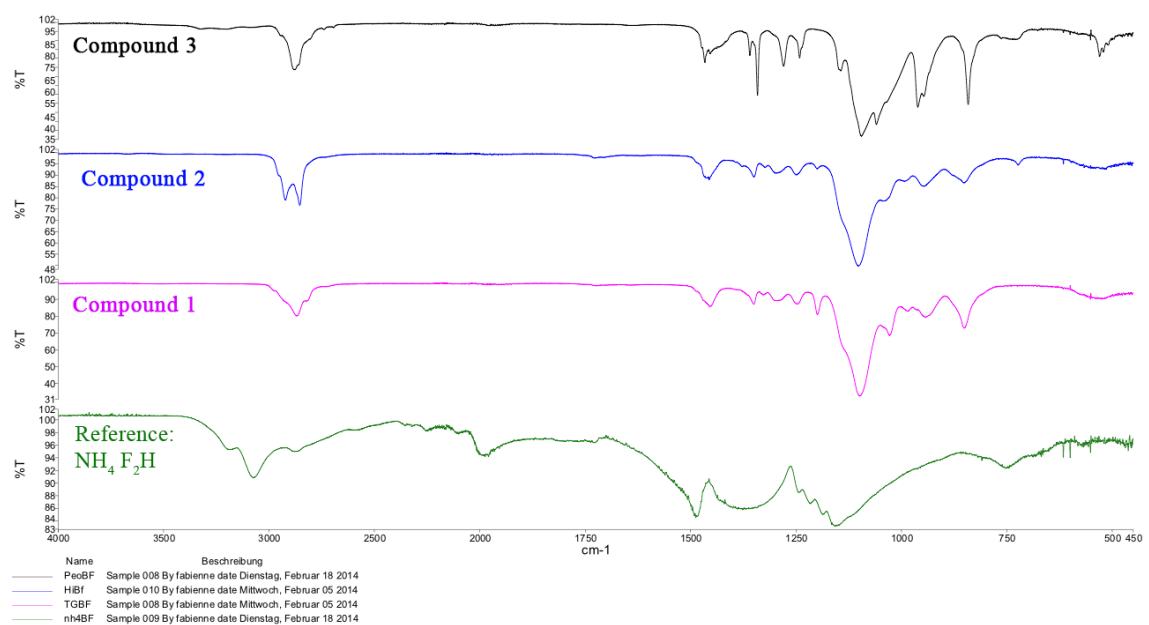


Figure S3: PXRD of two typical  $\text{BiF}_3$ -cathodes after discharge

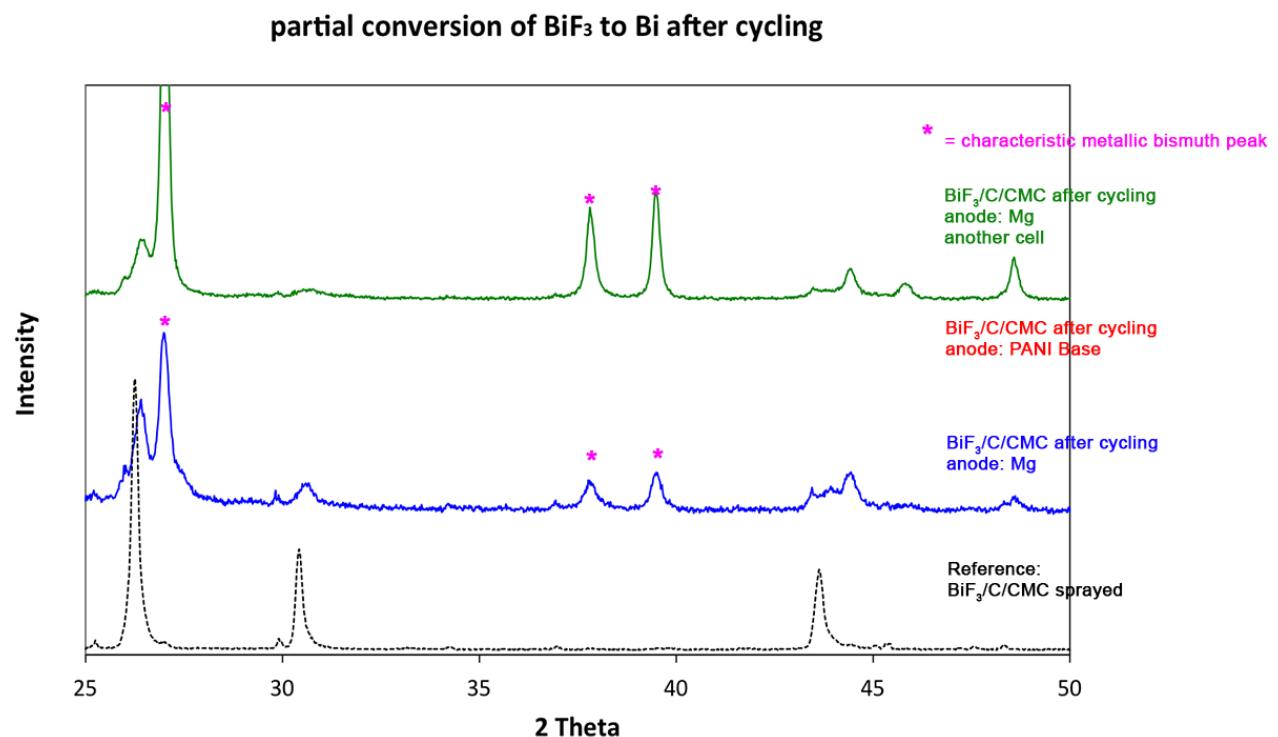
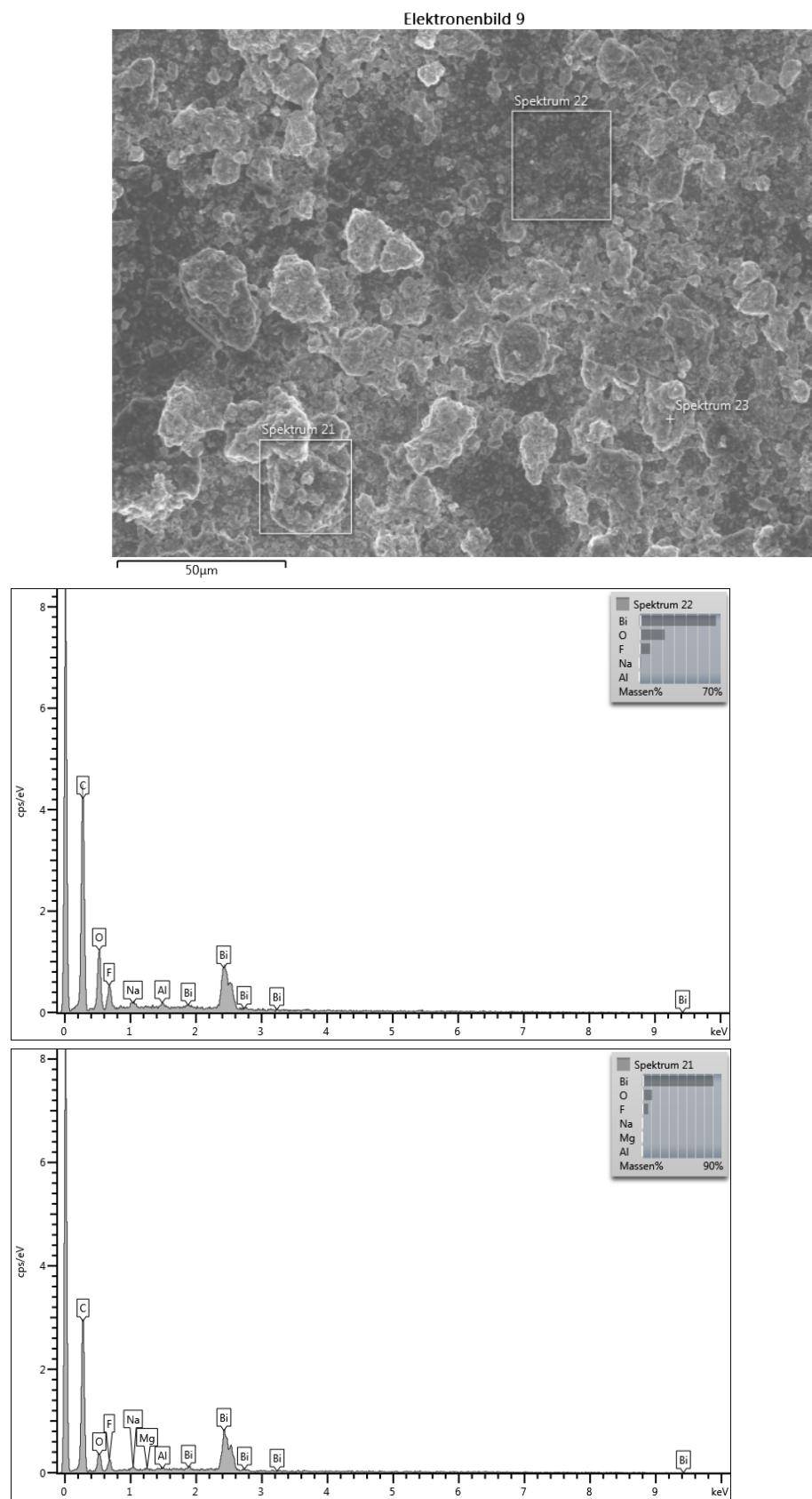


Figure S4: BiF<sub>3</sub>-cathode supplemental SEM/EDX measurement



Elektronenbild 7

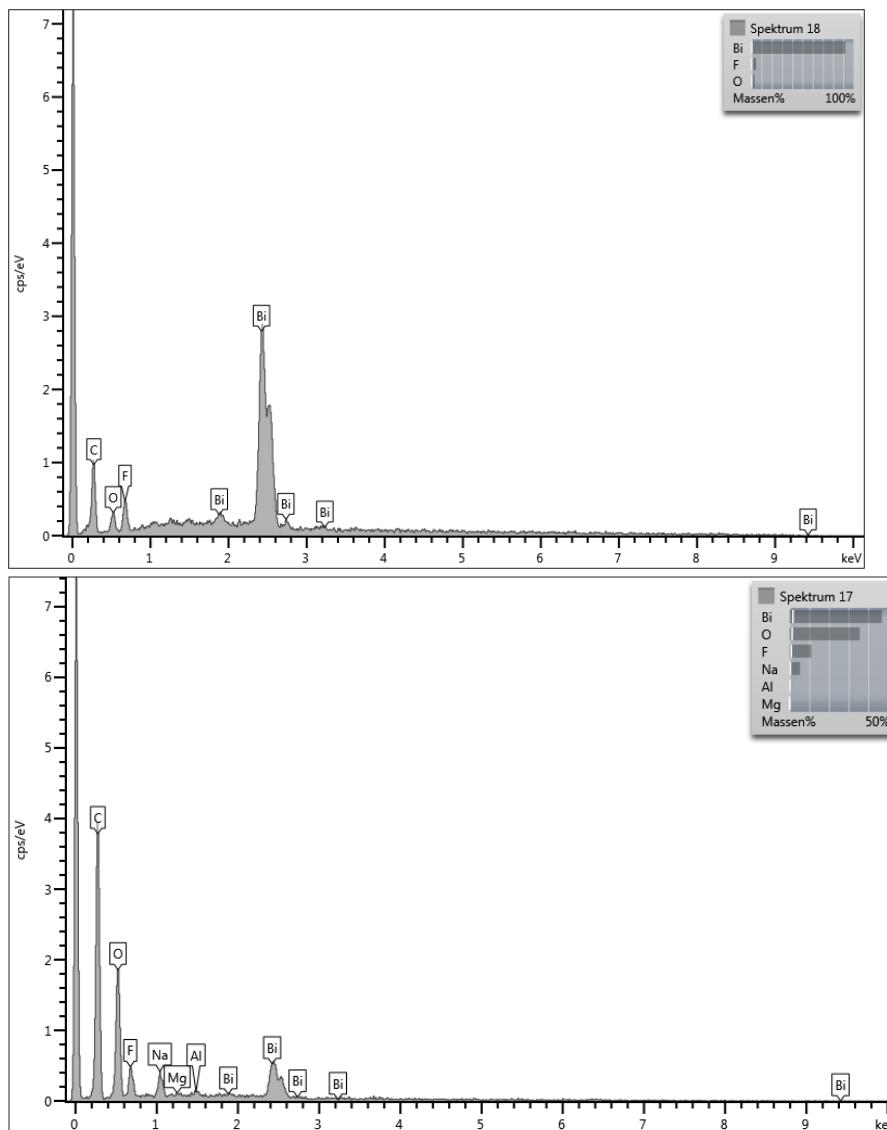
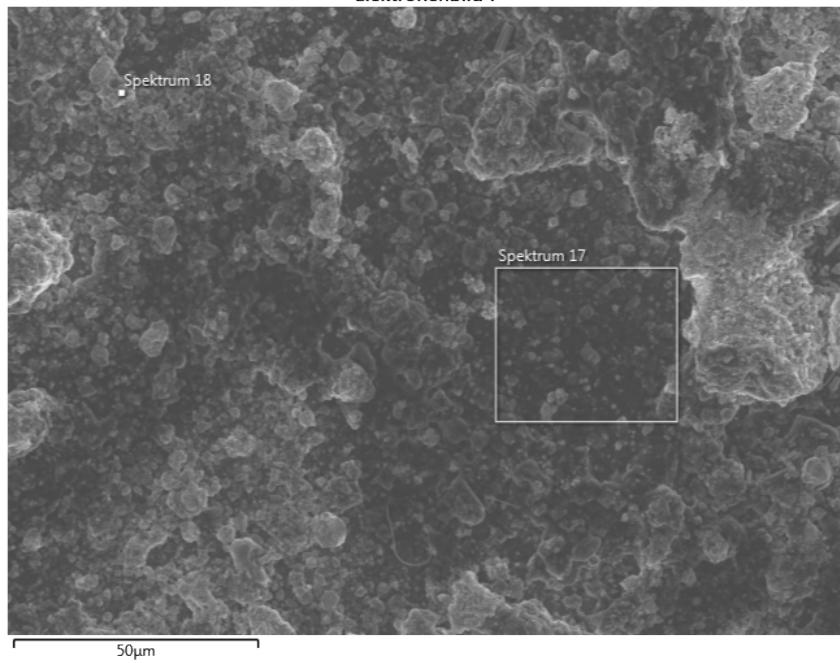
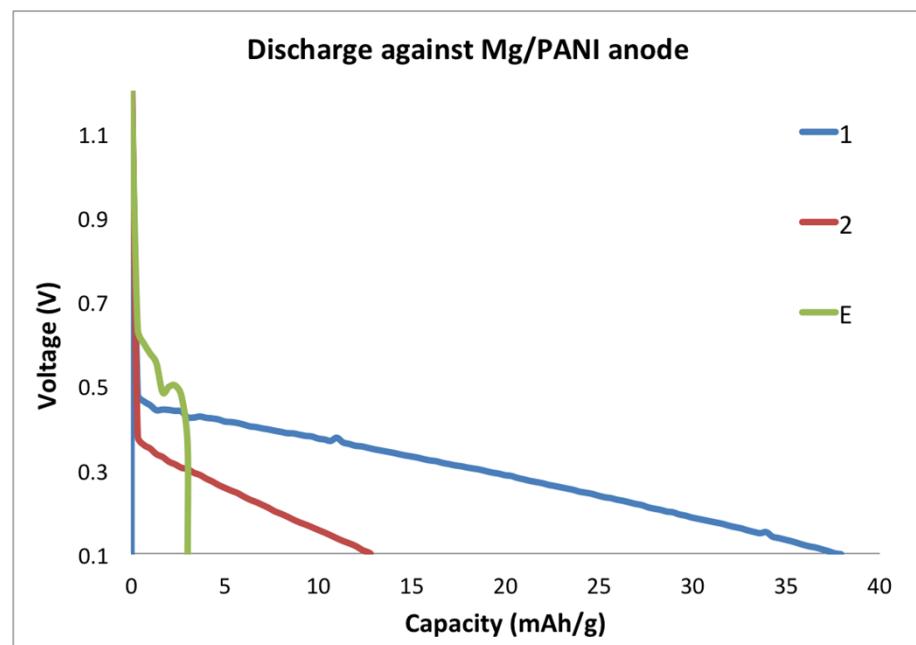


Figure S6: Mg/PANI composite electrode, first discharge capacity, with three different electrolytes; TG(1), Hi(2) and PEG6000(E)



Results from ICP-OES measurement, content of Mg ions and Na ion in the separators

**Prüfergebnisse:**

Probenbezeichnung		Magnesium	Natrium
Auftraggeber	Labornummer	mg/l	mg/l
Referenz II 2	SP 346/14	2.5	14.1

DIN EN ISO

Kationen: 11885

Bestimmungsgrenze: 0,1 mg/l

<sup>19</sup>F-NMR Data for the new compounds and for the diluted electrolyte Hi and TG

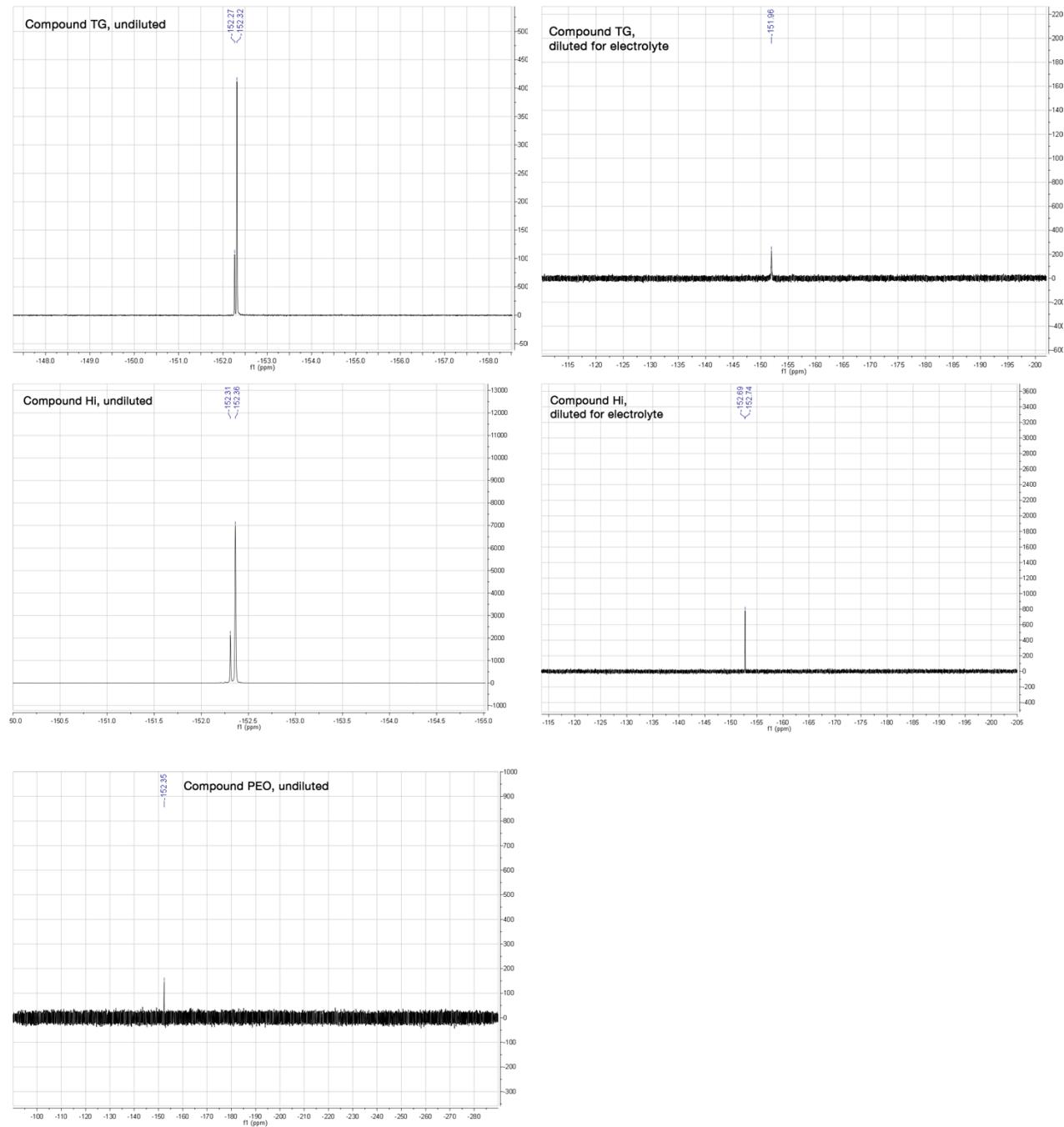
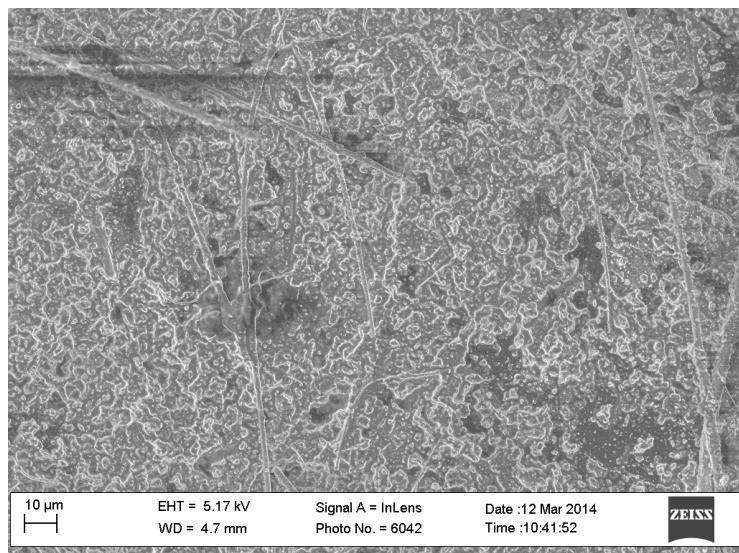


Fig. S9: SEM picture of passivated Mg anode: Heavy crust formation



Different Anode:  
Mg foil:  
Heavy crust formation of  
MgF<sub>2</sub> onto the anode  
surface