

Shape-controlled octahedral cobalt disulfide nanoparticles supported on nitrogen and sulfur-doped graphene/carbon nanotube composites for oxygen reduction in acidic electrolyte

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Supporting information

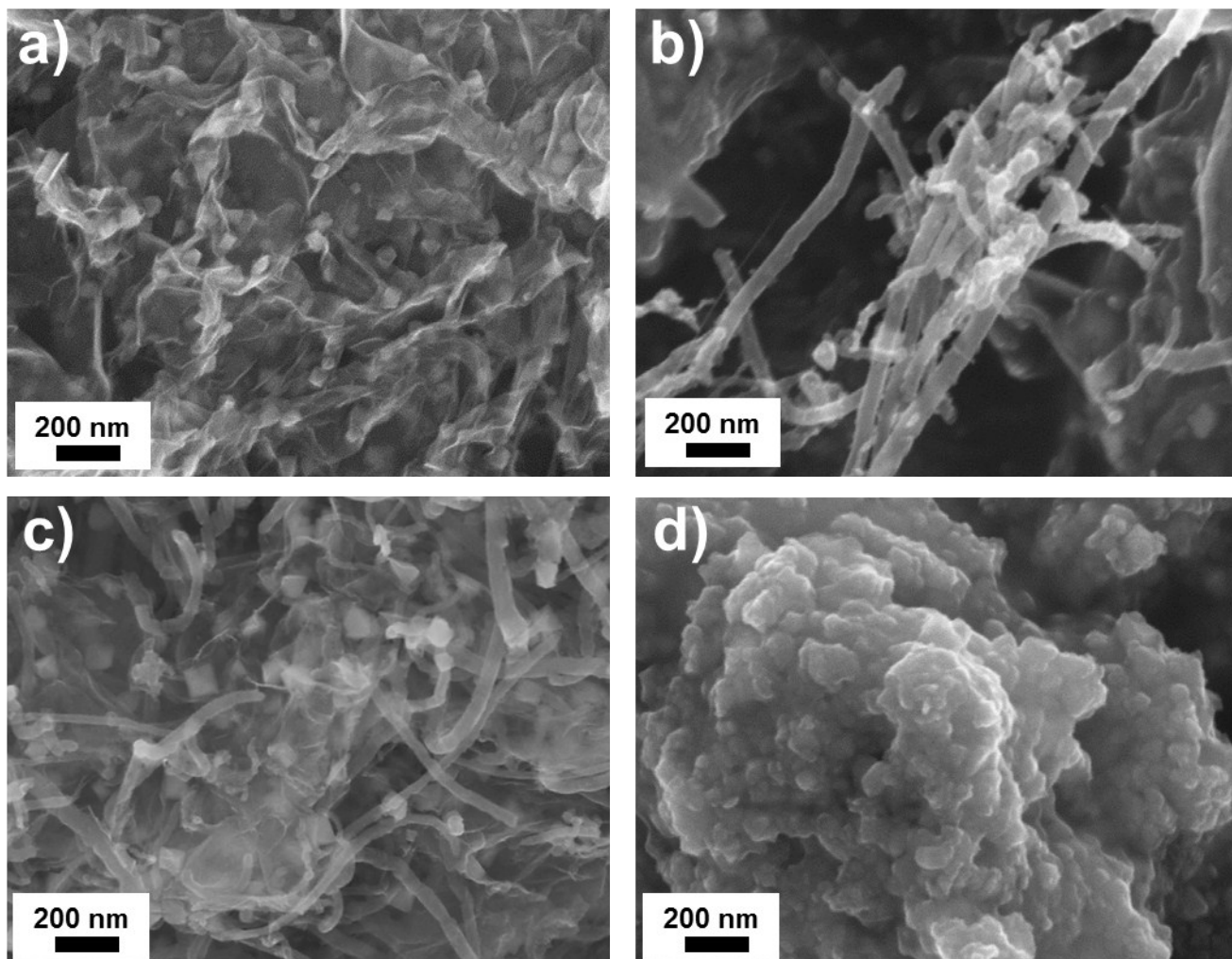


Figure S1. SEM images of (a) CoS₂-G, (b) CoS₂-C, (c) CoS₂-CG and (d) as-prepared, unsupported CoS₂.

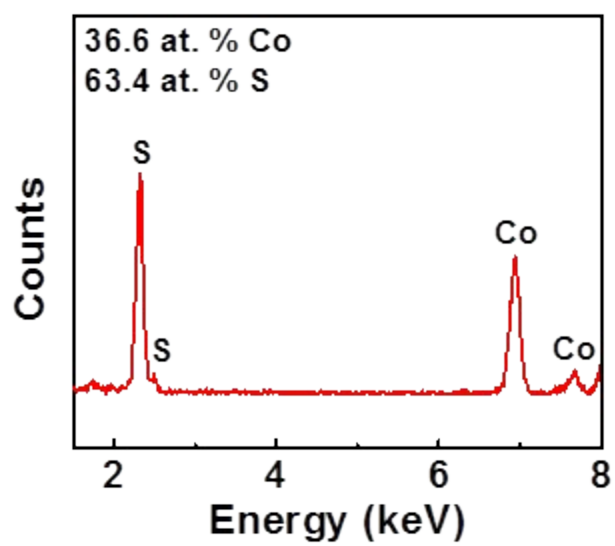


Figure S2. Energy dispersive x-ray spectrum of a single CoS₂ octahedral nanoparticle from the CoS₂-CG sample.

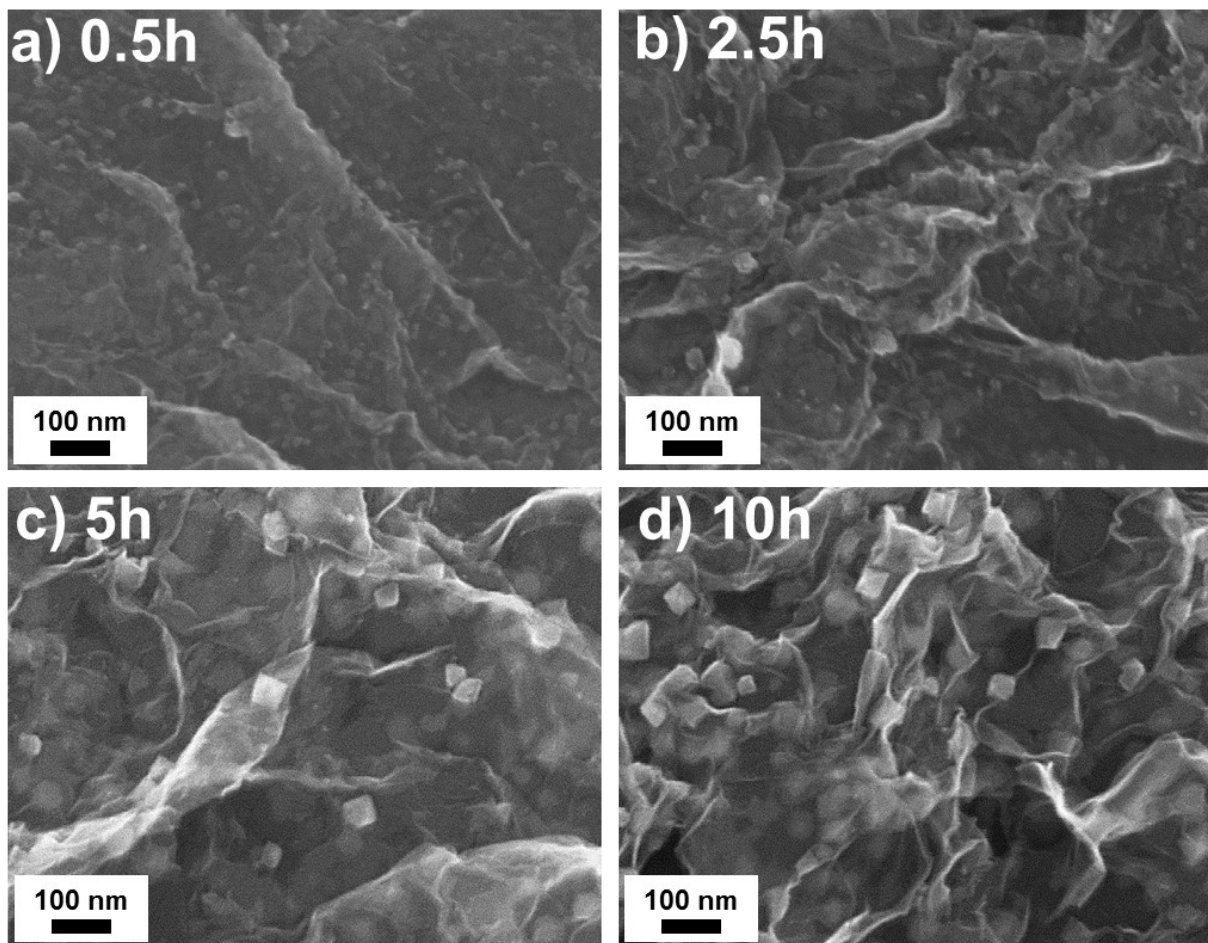


Figure S3. SEM images of the time dependent growth of CoS₂ supported on graphene held at 220 °C for (a) 0.5 h, (b) 2.5 h, (c) 5 h and (d) 10 h.

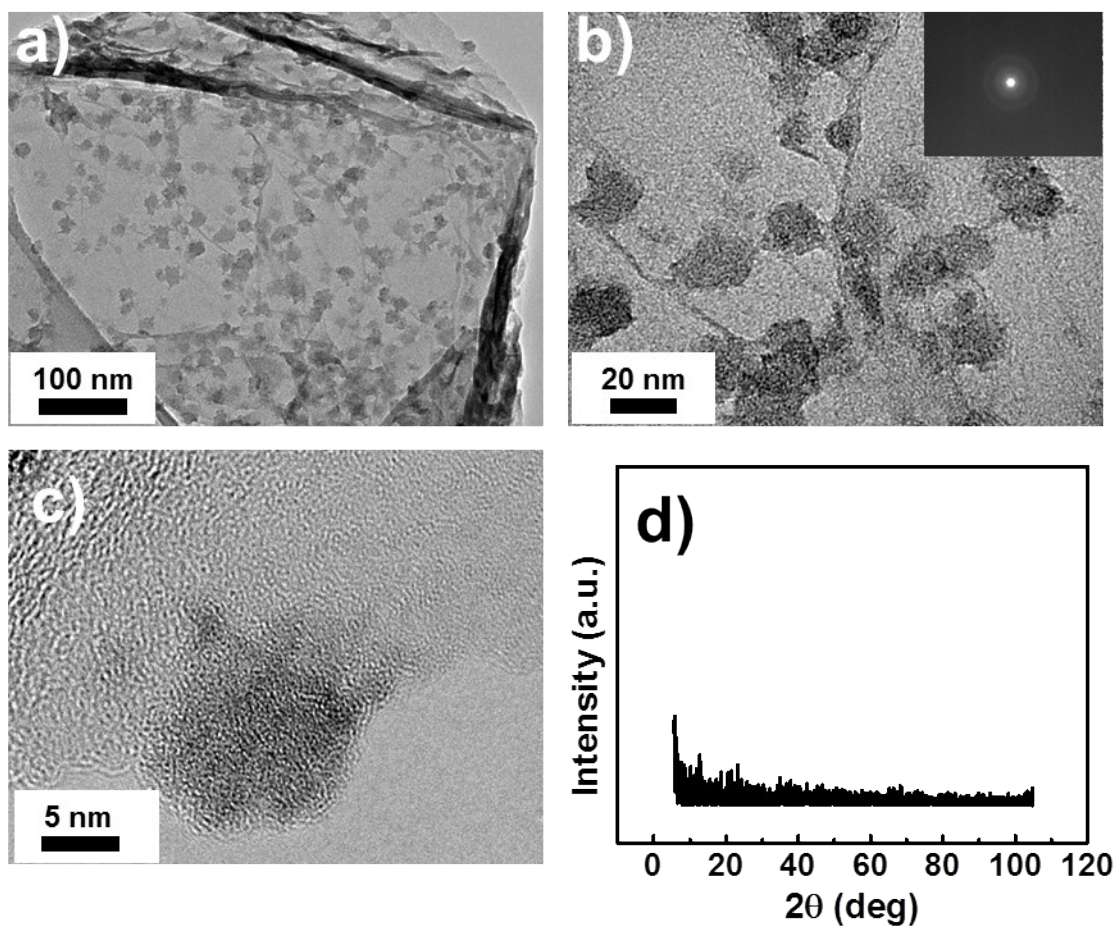


Figure S4. (a,b) TEM images, (c) high resolution TEM image, (d) XRD pattern and (b-inset) SAED pattern of CoSG-120.

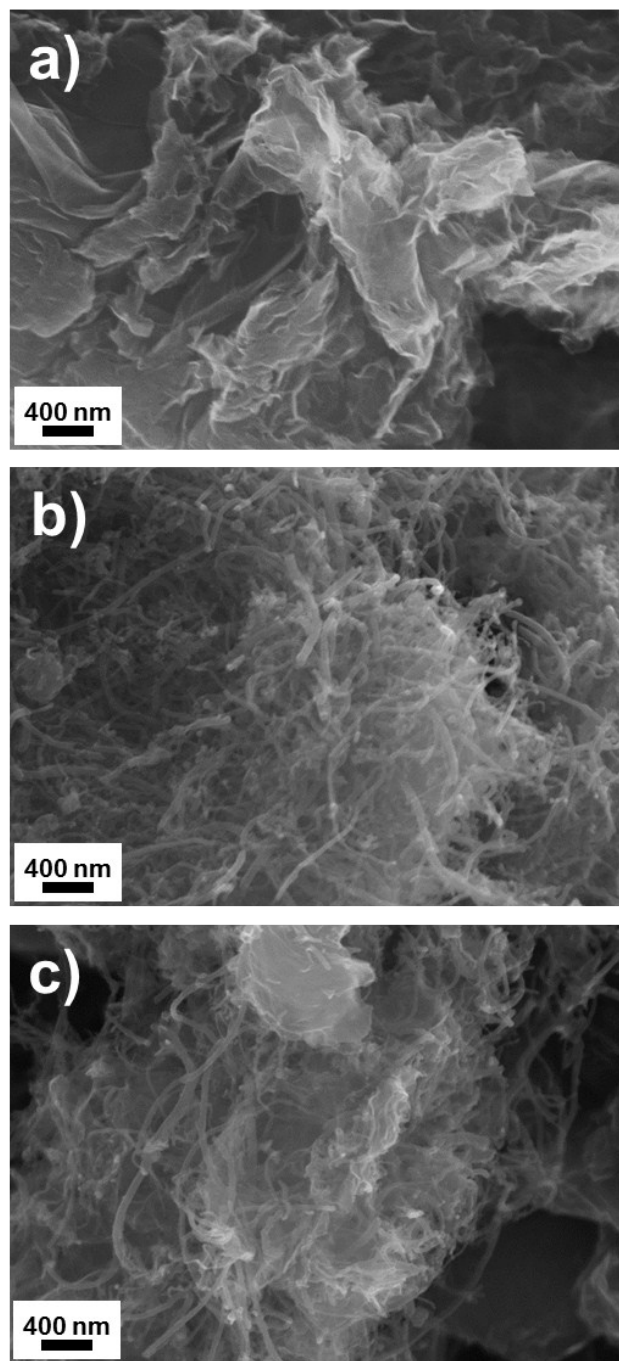


Figure S5. SEM images of (a) NS-G, (b) NS-C and (c) NS-CG.

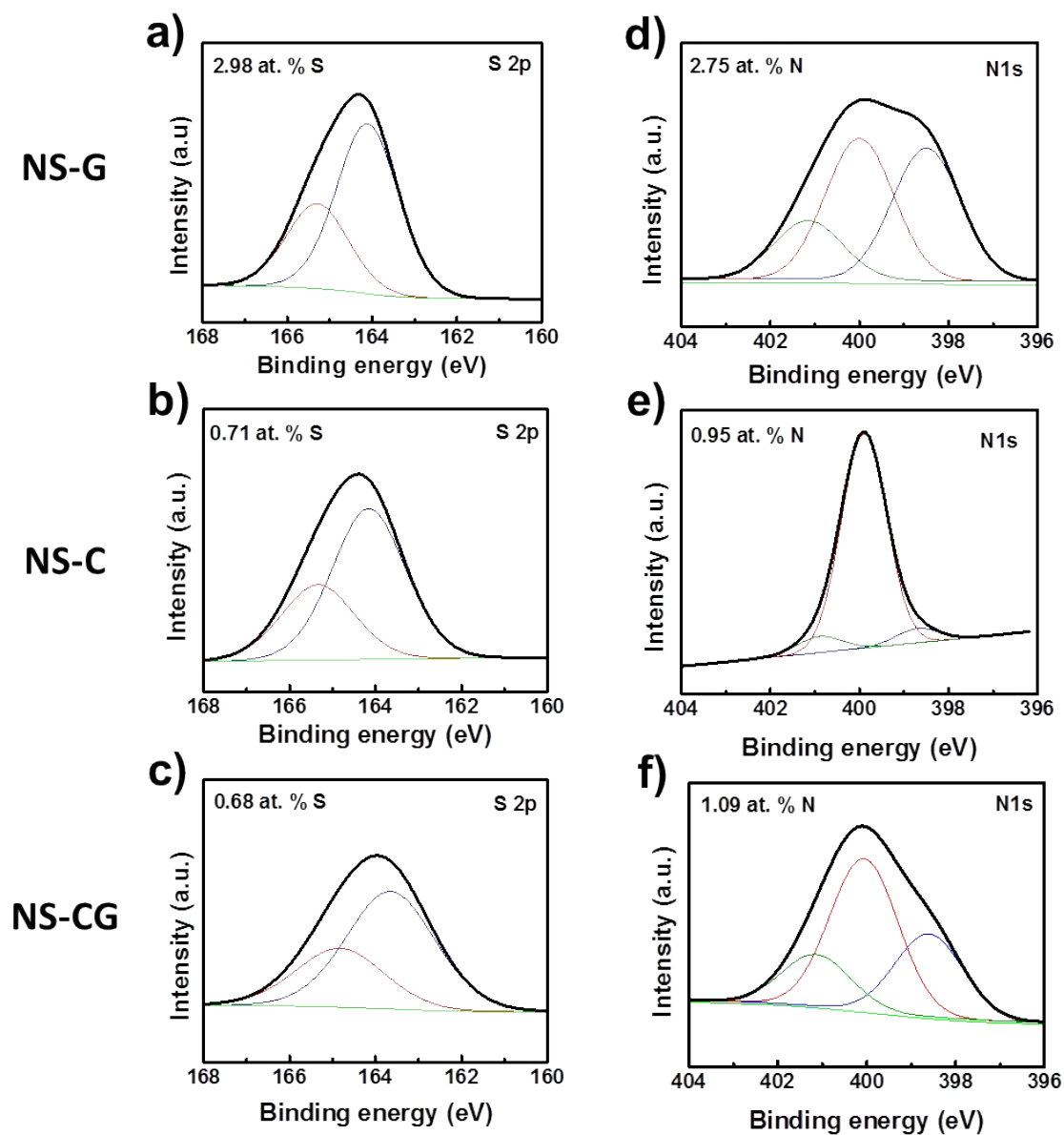


Figure S6. High resolution XPS (a-c) S2p and (d-f) N1s spectra for NS-G, NS-C and NS-CG.

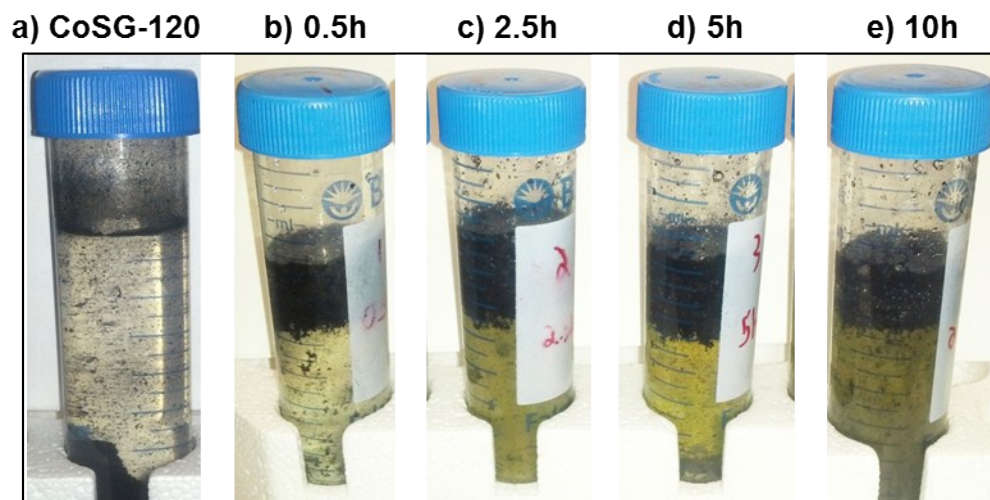


Figure S7. Solution obtained after solvothermal synthesis for (a) CoSG-120 and CoS₂ supported on graphene reaction mixture materials held at 220 °C for (b) 0.5 h, (c) 2.5 h, (d) 5 h and (e) 10 h.

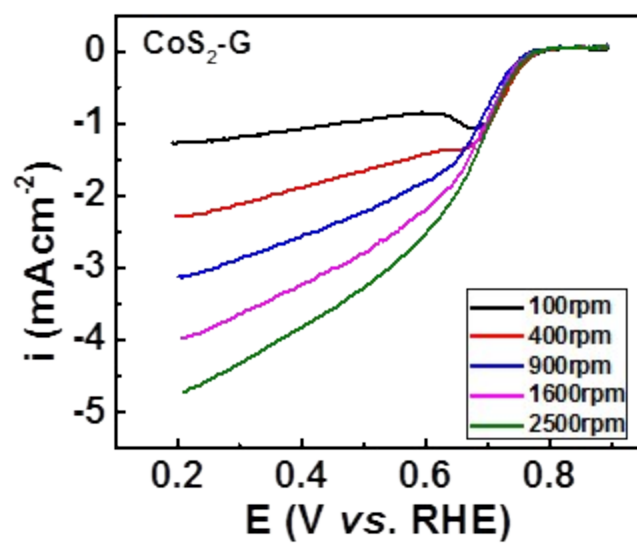


Figure S8. ORR polarization curve for CoS₂-G at various electrode rotation rates.

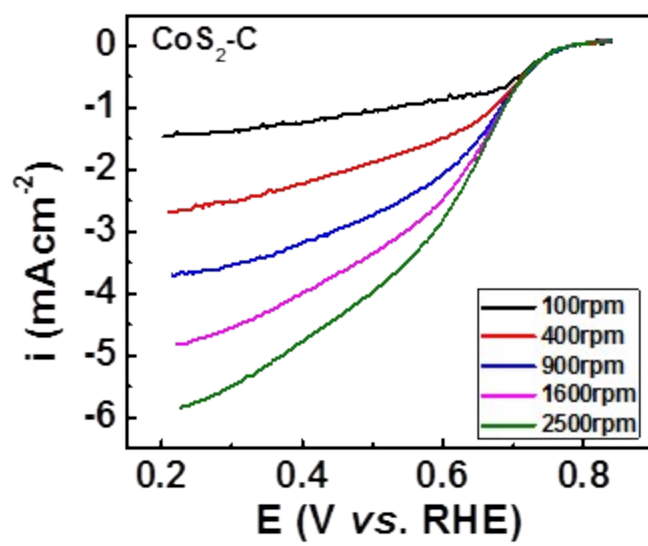


Figure S9. ORR polarization curve for CoS₂-C at various electrode rotation rates.

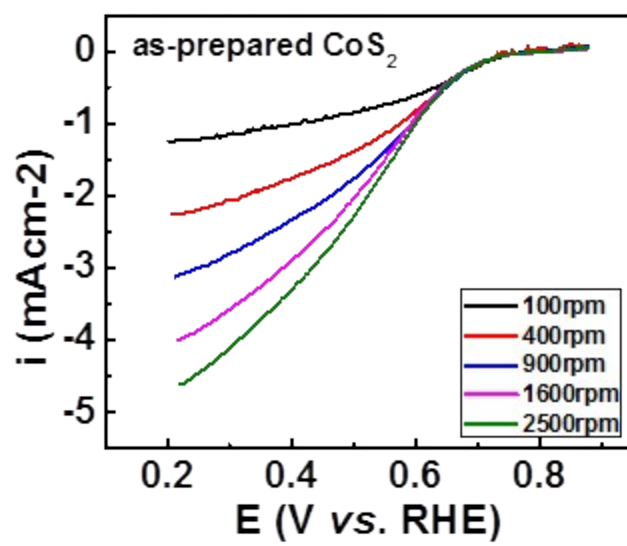


Figure S10. ORR polarization curve for as-prepared CoS_2 at various electrode rotation rates.

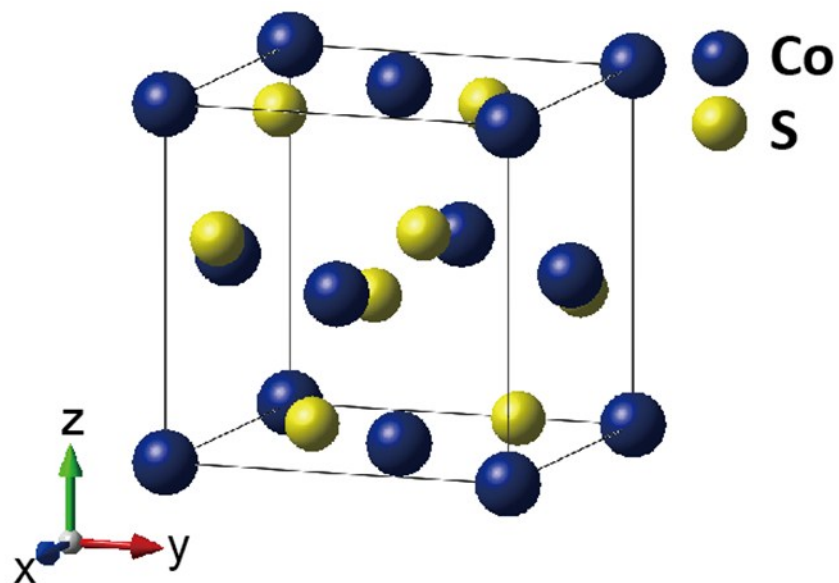


Figure S11. Fully relaxed bulk CoS_2 structure.