## **Electronic Supplementary Information (ESI) for:**

## Wet-Chemical Nitrogen-Doping of Graphene Nanoplatelets as Electrocatalysts for Oxygen Reduction Reaction

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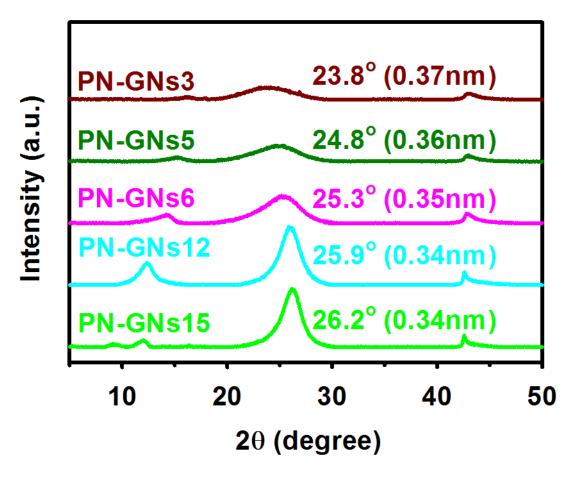
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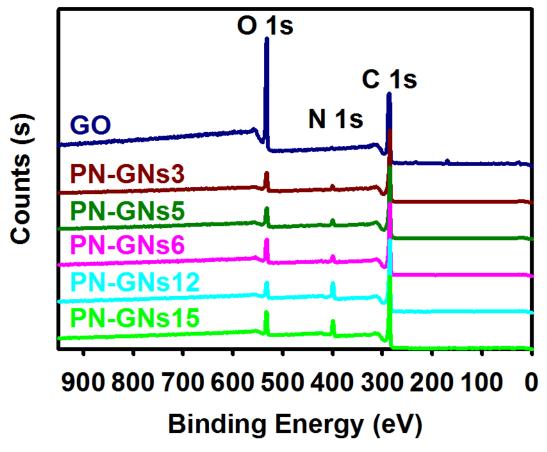
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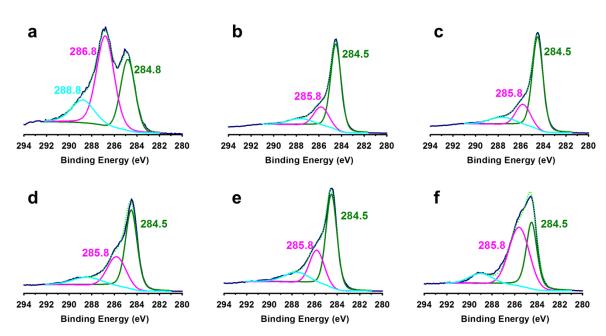
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**Figure S1.** Magnified XRD diffraction patterns from the rectangle area in Figure 2d with the numbers of *d*-spacing in nanometer (nm) of PN-GNs3, PN-GNs5, PN-GNs6, PN-GNs12 and PN-GNs15.



**Figure S2.** XPS survey spectra of GO, PN-GNs3, PN-GNs5, PN-GNs6, PN-GNs12 and PN-GNs15.



**Figure S3.** High-resolution XPS spectra of C 1s peaks: (a) GO; (b) PN-GNs3; (c) PN-GNs5; (d) PN-GNs6; (e) PN-GNs12; (f) PN-GNs15.

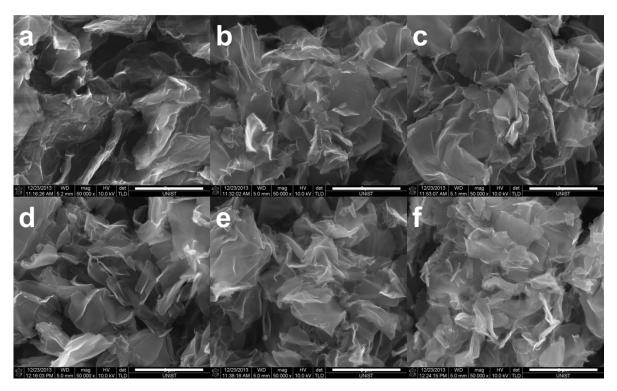
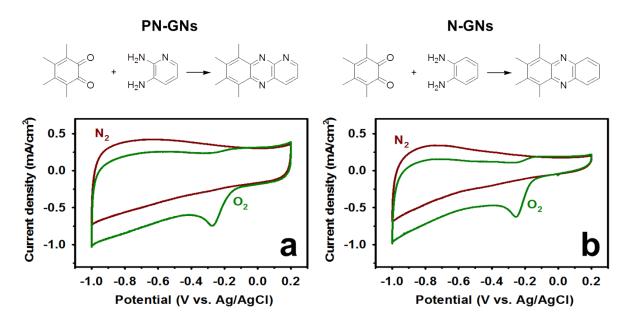
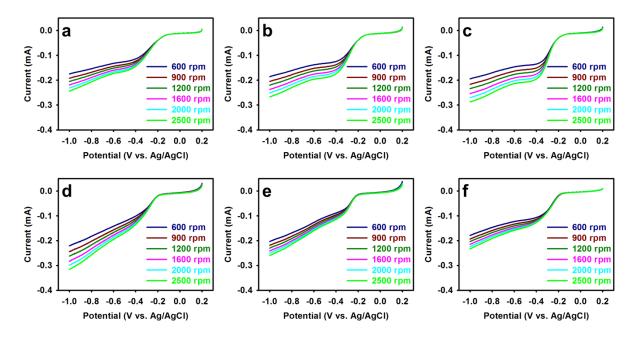


Figure S4. SEM images: (a) GO; (b) PN-GNs3; (c) PN-GNs5; (d) PN-GNs6; (e) PN-GNs12; (f) PN-GNs15. Scale bars are 2  $\mu$ m.



**Figure S5.** Reaction scheme and cyclic voltammograms (CV) of samples on glassy carbon (GC) electrodes in N<sub>2</sub>- and O<sub>2</sub>- saturated 0.1 M aqueous KOH solution at a scan rate of 10 mV s<sup>-1</sup>: (a) PN-GNs6; (b) N-GNs.



**Figure S6**. RDE voltammograms in O<sub>2</sub>- saturated 0.1 M aqueous KOH solution with a scan rate of 10 mVs<sup>-1</sup> at different rotating rates of 600, 900, 1200, 1600, 2000 and 2500 rpm: (a) PN-GNs3; (b) PN-GNs5; (c) PN-GNs6; (d) PN-GNs12; (e) PN-GNs15; (f) N-GNs.

## Calculation of electron transferred number (n)

The kinetic analysis for ORR was conducted according to Koutecky-Levich plots:

$$\frac{1}{j} = \frac{1}{j_k} + \frac{1}{B\omega^{0.5}} \tag{1}$$

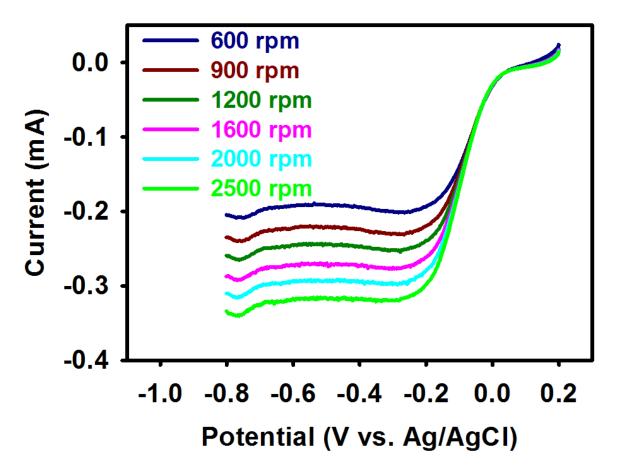
where  $j_k$  is the kinetic current and B is Levich slope which is given by:

$$B = 0.2nF(D_{O_2})^{2/3}v^{-1/6}C_{O_2}$$
(2)

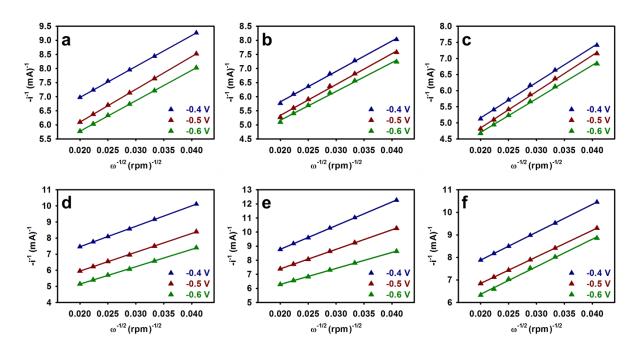
The constant 0.2 is generally adopted when the rotation speed is expressed in rpm. n is the number of electrons transferred for the reduction of one  $O_2$  molecule, F is the Faraday constant (F = 96485 C/mol),  $D_{O2}$  is the diffusion coefficient of  $O_2$  ( $D_{O2} = 1.9 \times 10^{-5}$  cm<sup>2</sup> s<sup>-1</sup>), v

is the kinematics viscosity for KOH (v = 0.01 cm<sup>2</sup> s<sup>-1</sup>) and  $C_{\rm O2}$  is concentration of  $O_2$  in the solution ( $C_{\rm O2}$  = 1.2 × 10<sup>-6</sup> mol cm<sup>-3</sup>).

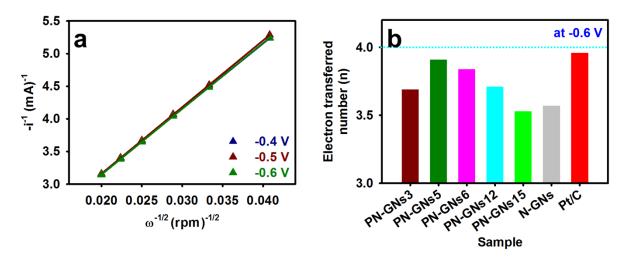
According to equations (1) and (2), the number of electrons transferred (n) can be obtained from the slope of Koutecky-Levich plot of  $i^{-1}$  vs.  $\omega^{-1/2}$ .



**Figure S7**. RDE voltammograms of Pt/C in  $O_2$ - saturated 0.1 M aqueous KOH solution with a scan rate of 10 mVs<sup>-1</sup> at different rotating rates of 600, 900, 1200, 1600, 2000 and 2500 rpm.



**Figure S8**. Koutechy-Levich plots derived from RDE measurements at different electrode potentials (blue line: -0.4 V, brown line: -0.5 V and green line: -0.6 V): (a) PN-GNs3; (b) PN-GNs5; (c) PN-GNs6; (d) PN-GNs12; (e) PN-GNs15; (f) N-GNs.



**Figure S9**. (a) Koutechy-Levich plots derived from RDE measurements of Pt/C at different electrode potentials (blue line: -0.4 V, brown line: -0.5 V and green line: -0.6 V); (b) electron transferred number (n) of all samples at -0.6 V.

**Table S1**. Elemental analyses of graphite, GO, PN-GNs3, PN-GNs5, PN-GNs6, PN-GNs12, PN-GNs15 and N-GNs

Materials	С	0	Н	N	Sum	C/O	C/N
Graphite	98.74	BDLa	0.11	BDLa	98.86	$\infty_{\mathrm{p}}$	$\infty_{p}$
GO	52.21	43.34	2.02	$BDL^a$	97.57	1.61	$\infty_{p}$
PN-GNs3	75.12	17.69	1.69	3.27	97.77	5.66	26.72
PN-GNs5	74.33	17.12	1.71	4.82	97.98	5.79	17.99
PN-GNs6	72.89	15.63	1.93	6.49	96.94	6.22	13.10
PN-GNs12	70.18	13.38	2.02	11.64	97.22	6.99	7.03
PN-GNs15	68.19	12.31	2.09	15.28	97.87	7.39	5.21
N-GNs	79.01	10.49	1.62	6.45	97.57	10.04	14.29

<sup>&</sup>lt;sup>a</sup> BDL = Below detection limit

 $b \infty = Unlimited$ 

**Table S2**. Capacitance (F/g) of samples on glassy carbon (GC) electrodes in  $N_2$ -and  $O_2$ -saturated 0.1 M aqueous KOH solution at a scan rate of 10 mV s<sup>-1</sup>

Sample	Capacitance (F/g) in O <sub>2</sub>	Capacitance (F/g) in N <sub>2</sub>		
PN-GNs3	130.1	120.2		
PN-GNs5	132.5	120.6		
PN-GNs6	139.7	122.5		
PN-GNs12	129.4	118.6		
PN-GNs15	108.5	104.8		
N-GNs	97.3	81.4		
Pt/C	88.8	82.3		

**Table S3**. Onset potentials and limiting diffusion current at -0.6 V of all samples on a rotating disk electrode at a rotation speed of 1600 rpm and a scan rate of 10 mV s<sup>-1</sup>

	PNGNs3	PNGNs5	PNGNs6	PNGNs12	PNGNs15	N-GNs	Pt/C
Onset	-0.17	-0.19	-0.20	-0.20	-0.21	-0.22	0.01
potential (V)							
Current	-0.16	-0.18	-0.19	-0.18	-0.15	-0.14	-0.27
(mA)							

Sample	-0.4 V	-0.5 V	-0.6 V
PN-GNs3	3.63	3.43	3.70
PN-GNs5	3.70	3.65	3.91
PN-GNs6	3.64	3.55	3.84
PN-GNs12	3.15	3.42	3.71
PN-GNs15	2.38	3.88	3.53
N-GNs	3.24	3.40	3.57
Pt/C	4.00	3.92	3.96