Supporting information for

Glucose-, pH- and thermo-responsive Nanogels Crosslinked by Functional Superparamagnetic Maghemite Nanoparticles as Innovative Drug Delivery Systems

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## 1. Supplemental experimental part

*Scanning electron microscopy (SEM)* was used to study the surface of the RCL nanogels with a JSM 840A microscope at 4 kV in high vacuum conditions. A drop of the RCL nanogels aqueous solution was deposited on a glass substrate, and the sample was dried at RT under air overnight. Then the sample was sputtered with gold in a cathode evaporator under argon atmosphere.

<sup>1</sup>*H* nuclear magnetic resonance (*NMR*) spectra of the PVAc-*b*-PNVCL and PVOH-*b*-PNVCL macromolecules were recorded with a 250 MHz Bruker spectrometer in DMSO- $d_6$  at 50 °C.

*Size-exclusion chromatography (SEC)* in THF (flow rate of 1 mL/min) was performed at 40 °C using a Waters 600 liquid chromatograph equipped with a 410 refractive index detector and styragel HR columns (four columns HP PL gel 5  $\mu$ m, 10<sup>5</sup> Å, 10<sup>4</sup> Å, 10<sup>3</sup> Å and 10<sup>2</sup> Å), calibrated with polystyrene standards. SEC in DMF (0.025 M LiBr, flow rate of 1 mL/min) was performed at 55 °C with a Waters 600 liquid chromatograph equipped with a 410 refractive index detector and styragel HR columns (HR1, 100-5000; HR3, 500-30,000; HR4, 5000-500,000; HR5, 2000-4000,000), calibrated with polystyrene standards.

*Fourier transform infrared spectra* (*FTIR*) of the bare  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs,  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub>-APTES NPs and  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub>-CNPBA NPs were performed with a Perkin Elmer FTIR instrument. Samples were mixed and grinded with potassium bromide, and then compressed for IR analysis.

*X-ray photoelectron spectroscopy (XPS)* was used to characterize the surface-functionalized  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs with a VG Scientific 220 i-XL ESCALAB spectrometer at 100 W (10 kV and 10 mA), which is equipped with a non-monochromatised MgK $\alpha$  source (hv = 1253.6 eV). A pressure of 10<sup>-7</sup> Pa was maintained in the chamber during analysis. The analyzed area was *ca*. 150 µm in diameter. Full spectra (0-1150 eV) were obtained with constant pass energy of 150 eV and high-resolution spectra at 40 eV. Charge neutralization was required for insulating samples. The peaks were referenced to C1s peak at 284.7 eV.

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## 2. Supplemental results



**Figure S1.** <sup>1</sup>H NMR spectrum of PVAc-*b*-PNVCL copolymer (a, DMSO-*d*<sub>6</sub>, 353 K) and PVOH-*b*-PNVCL (b, DMSO-*d*<sub>6</sub>, 353 K), SEC traces of PVAc-Co(II) macro-initiator (blue,  $M_n = 4.42 \times 10^4$  g/mol,  $M_n/M_n = 1.04$ ) and PVAc-*b*-PNVCL copolymer (red,  $M_n = 5.67 \times 10^4$  g/mol,  $M_n/M_n = 1.06$ ) in DMF (c, PS calibration standard).



**Figure S2.** XPS spectrum of the  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub>-CNPBA NPs, inset: partially-magnified spectrum in the range of 220~170 eV, and B1s signal could be detected at *ca*. 192 eV (a), and TGA traces of the  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs at each step of surface modification (b)



**Figure S3.** Pictures of RCL nanogels (a) and NR-encapsulated RCL nanogels (b), and UV/*vis* spectra of the  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub>-CNPBA NPs, RCL nanogels and NR-uploaded RCL nanogels (c)





**Figure S4.** Representative SEM image of the RCL nanogels (scale bar: 2000 nm), magnified SEM image of the RCL nanogels (scale bar: 500 nm) (a); intensity-averaged size distribution of the RCL nanogels from DLS analysis (b).



**Figure S5.** TEM images of the RCL nanogels after 6-h treatment of glucose (10 mM, pH 7.4, 25°C) (a, scale bar: 100 nm), and 6-h treatment of acidic pH (pH 5.0, no glucose, 25°C) (b, scale bar: 100 nm).



**Figure S6.** fluorescence-activated cell sorting (FACS) histogram of the untreated MEL-5 cells (red) and treated cells (blue) after 12-h incubation with NR-loaded RCL nanogels (100  $\mu$ g/mL), and the log of red fluorescence intensity (Nile Red-A on *x*-axis) is plotted against the number of cells (counts on *y*-axis).



**Figure S7.** UV/*vis* absorption spectra of tamoxifen/methanol solutions with tamoxifen different concentrations (a); and linear fitting of absorbance at 280 nm *vs*. concentration from 0 to 100  $\mu$ g/mL (b).



**Figure S8.** Cell viability of MEL-5 cells in DMEM complete medium (pH 7.4) with different tamoxifen concentrations (a), and cell viability of MEL-5 cells in DMEM buffer (pH 7.4) with different glucose concentrations (b); percentage viability was expressed relatively to the un-treated cells (control at pH 7.4, 5 mM glucose, 100% viability). Results were all presented as mean value  $\pm$  standard deviation (n = 5).



**Figure S9.** Cell viability of MEL-5 cells in DMEM buffer at pH 5.0 (a) and pH 7.4 (b) with different glucose concentrations; percentage viability was expressed relatively to the un-treated cells (control at pH 7.4 and 5 mM glucose, 100% viability). Results are presented as mean value  $\pm$  standard deviation (n = 5).