

## Supporting Information

### **Ce<sup>3/4+</sup> Cation-Functionalized Maghemite Nanoparticles Towards siRNA-Mediated Gene Silencing**

*Liron L. Israel<sup>a</sup>, Emmanuel Lellouche<sup>b</sup>, Ron S. Kenett<sup>c</sup>, Omer Green<sup>a</sup>, Shulamit Michaeli<sup>b,8</sup>, and Jean-Paul Lellouche<sup>a,\*</sup>*

<sup>a</sup> Department of Chemistry, <sup>b</sup> Faculty of Life Sciences, Institute of Nanotechnology & Advanced Materials, Institute of Nanotechnology & Advanced Materials, Bar-Ilan University, Ramat-Gan, 5290002, Israel

<sup>c</sup> KPA Ltd., Raanana 43100, Israel & University of Turin, Italy

#### **A. Experimental procedure for nanocarrier fabrication - Starting neat neutral magnetite (Fe<sub>3</sub>O<sub>4</sub>) nanoparticles (Fe<sub>3</sub>O<sub>4</sub> NPs, basic Massart hydrolytic method)**

A solution of FeCl<sub>3</sub>•6H<sub>2</sub>O (240.0 mg, 0.9 mmol) dissolved in deoxygenated milliQ purified H<sub>2</sub>O (4.5 mL) was mixed with an aqueous solution of FeCl<sub>2</sub>•4H<sub>2</sub>O (97.5 mg, 0.45 mmol, 4.5 mL H<sub>2</sub>O). This solution was kept under N<sub>2</sub> and ultrasonicated (Bransonic® ultrasonic cleaner bath, 2510E MTH model, 42 KHz at full power) for 5-10 min at room temperature. Then, a concentrated 24% weight aqueous NH<sub>4</sub>OH (0.75 mL) was introduced in one shot, resulting in an immediate black precipitation of magnetite (Fe<sub>3</sub>O<sub>4</sub>) NPs. Sonication was continued for 10 additional minutes. Resulting Fe<sub>3</sub>O<sub>4</sub> NPs were transferred into a glass bottle (100 mL), magnetically decanted (using a strong external magnet), and washed with ddH<sub>2</sub>O (3 x 40 mL) until neutrality. Then, brilliant black free flowing magnetite NPs were stored as a 30 mL NP suspension in ddH<sub>2</sub>O before any further processing. Then, an ageing process must be at least executed for a minimum storage time in these conditions of 2h (room temperature).

**B. Experimental procedure for DoE-optimized nanocarrier fabrication (MINITAB® 16 DoE software, version 16.2.4, Minitab Inc., *profile optimizer tool*) - CAN<sub>DOE</sub>-stabilized maghemite nanoparticles (CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs)**

First, the former aqueous magnetite NP suspension (30 mL, 4.22 h ageing time) was magnetically decanted to separate the magnetite NPs from its aqueous storage phase. Ceric ammonium Nitrate (CAN, (NH<sub>4</sub>)<sub>2</sub>Ce(IV)(NO<sub>3</sub>)<sub>6</sub>, 500.0 mg, 0.912 mmol) dissolved in 6.0 mL MeCOMe was introduced onto decanted magnetite NPs, followed by the addition of degassed milliQ purified H<sub>2</sub>O (18.0 mL). The corresponding mixture was ultrasonicated using a high-power sonicator (Sonics®, Vibra cell, 750 Watt, power modulator set-up at 25%) equipped with a titanium horn (0.5 h, 0°C) under an inert argon atmosphere.

At this stage and optionally, the resulting highly stabilized hydrophilic CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs might be purified for other purposes [washing with ddH<sub>2</sub>O (3 x 10 mL) using an Amicon® Ultra-15 centrifugal filter devices (100K) processed at 4,000 rpm during 5-6 min (rt, 18°C) and re-dispersed in ddH<sub>2</sub>O (15 mL)] or alternatively processed for *b*-PEI (25.0 kDa branched PEI polymer) functionalization/decoration using mild aqueous contacting (corresponding experimental details mentioned below, section C).

NP cleaning/washing using an Amicon® Ultra-15 centrifugal filter device (100KDa)



**Selected characterization data:**

- Average NP TEM/DLS diameters: 6.61±2.04 and 60.00±1.98 nm (DLS, PDI: 0.136) respectively
- $\zeta$  potential: +45.7±1.88 mV
- w/w Ce/Fe ratio: 0.1000
- Average NP concentration: 5.8 mg/mL) before any further processing

- Kaiser test (free reactive primary amine quantification) - Experimental protocol for NP derivatization (coordination/contacting mode) & reactive primary amine quantification by UV spectroscopy.

To 1.0 mL of an aqueous suspension of  $\text{CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs (5.8 mg/mL) was added 0.2 mL (2.58 mmol) of 1,4-diaminobutane ( $\text{H}_2\text{N}-(\text{CH}_2)_4\text{-NH}_2$ ), and the reaction medium was gently shaken overnight at room temperature (20°C). At reaction completion, poly $\text{NH}_2$ -modified  $\text{CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs were cleaned before UV Kaiser test/ninhydrin reaction by three sequential centrifuge precipitation-dd $\text{H}_2\text{O}$  washing steps (12,500 rpm, 5°C) for removal of the 1,4-diamine in excess.

Resulting poly $\text{NH}_2$ -modified  $\text{CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs were equally divided into two 13x100 mm-sized test tubes and freeze dried (lyophilization, -50°C, 0.06 mbar). Each weighed product was sequentially added with (i) 75.0  $\mu\text{L}$  of an ethanol solution of phenol (PhOH, 40.0 g, 10 mL EtOH), (ii) 75.0  $\mu\text{L}$  of an ethanol solution of ninhydrin (2.5 g dye, 14.0 mmol, 50.0 mL EtOH) and 100.0  $\mu\text{L}$  of an aqueous KCN pyridine-containing solution (65.0 mg/1.0 mmol KCN dissolved in 100.0 mL dd $\text{H}_2\text{O}$  = solution A - then 2.0 mL of solution A were diluted with 100 mL of pyridine). All the test tubes were then placed on a heating block pre-adjusted to 100°C for 10 min (ninhydrin reaction with accessible primary outer  $\text{NH}_2$  groups). Then, each test tube was added with an aqueous 60% v/v solution of EtOH in dd $\text{H}_2\text{O}$  (4.8 mL) followed by medium filtration (0.22  $\mu\text{m}$  Millipore filter). A 0.5 mL aliquot of the filtered solution was then diluted with the former aqueous 60% v/v solution of EtOH in dd $\text{H}_2\text{O}$  (4.5 mL). Finally, this diluted medium was UV assayed (UV spectrophotometer) at a  $\lambda_{\text{max}}$  of 570 nm (ninhydrin dye) to quantify the number of accessible reactive primary amine functional groups present onto the NP surface (mmol accessible  $\text{NH}_2$  groups/g NPs). This test afforded a value of  $0.272 \pm 0.023$   $\mu\text{mol}$  accessible  $\text{NH}_2$  groups/mg  $\text{CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs.

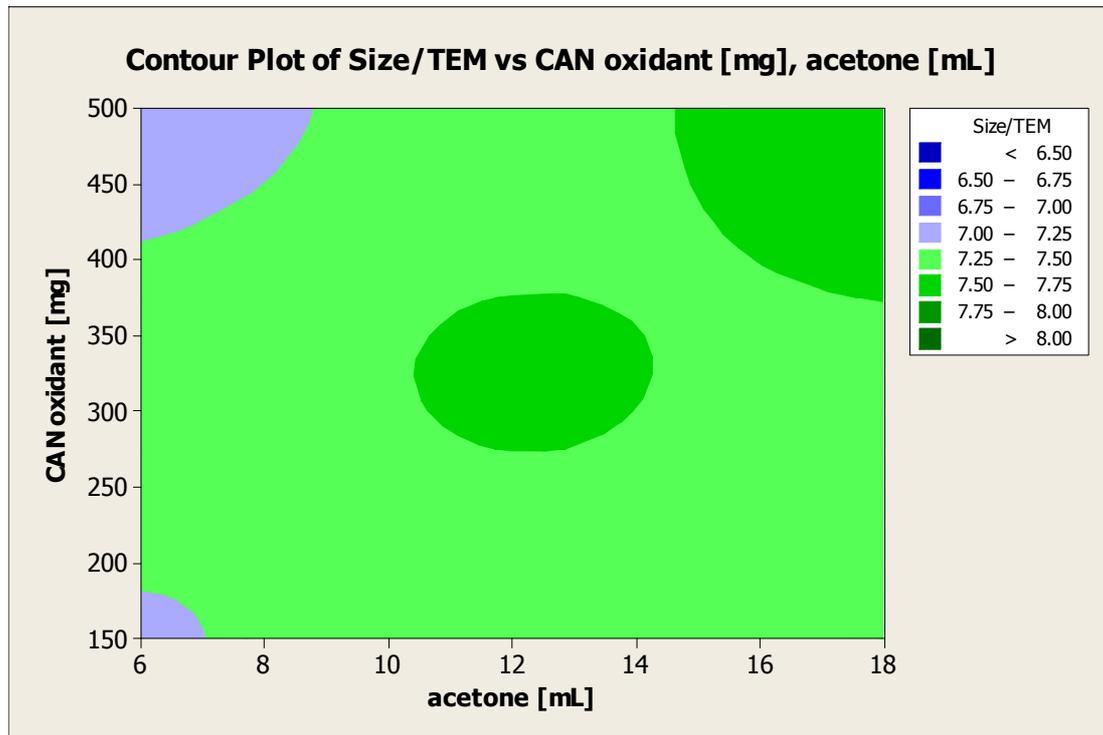
### **C. Experimental procedure for DoE-optimized nanocarrier fabrication – *b*-PEI-decorated CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs (*b*-PEI<sub>25</sub>-CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs) - Optimal aqueous PEI contacting process/experimental protocol**

The former corresponding CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs aqueous suspension (section B, 1.0 mL, Fe: 1.93 mg/mL – 1.93 mg total Fe, 0.0346 mmol Fe, ICP-AES measurement) was diluted to 25.0 mL using milliQ purified H<sub>2</sub>O. Then, 10.13 mg of 25kDa branched *b*-PEI (0.4053  $\mu$ mol, PEI/Fe Wt ratio: 5.25) were added at room temperature to the NP suspension as an aqueous solution (10.0 mg *b*-PEI/mL) and the medium was shaken overnight at room temperature (orbital shaker). At completion of such a mild *b*-PEI contacting/NP surface decoration, resulting crude *b*-PEI<sub>25</sub>-CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs were washed 3 times (3 x 10 mL ddH<sub>2</sub>O) using an Amicon® Ultra-15 centrifugal filter device (100K) operated at 4,000 rpm (5 min) followed by a size exclusion process performed using centrifugation (8,000 rpm/min, 16 min, 18°C and 7,000 rpm, 10 min, 18°C) to afford cleaned *b*-PEI<sub>25</sub>-CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs.

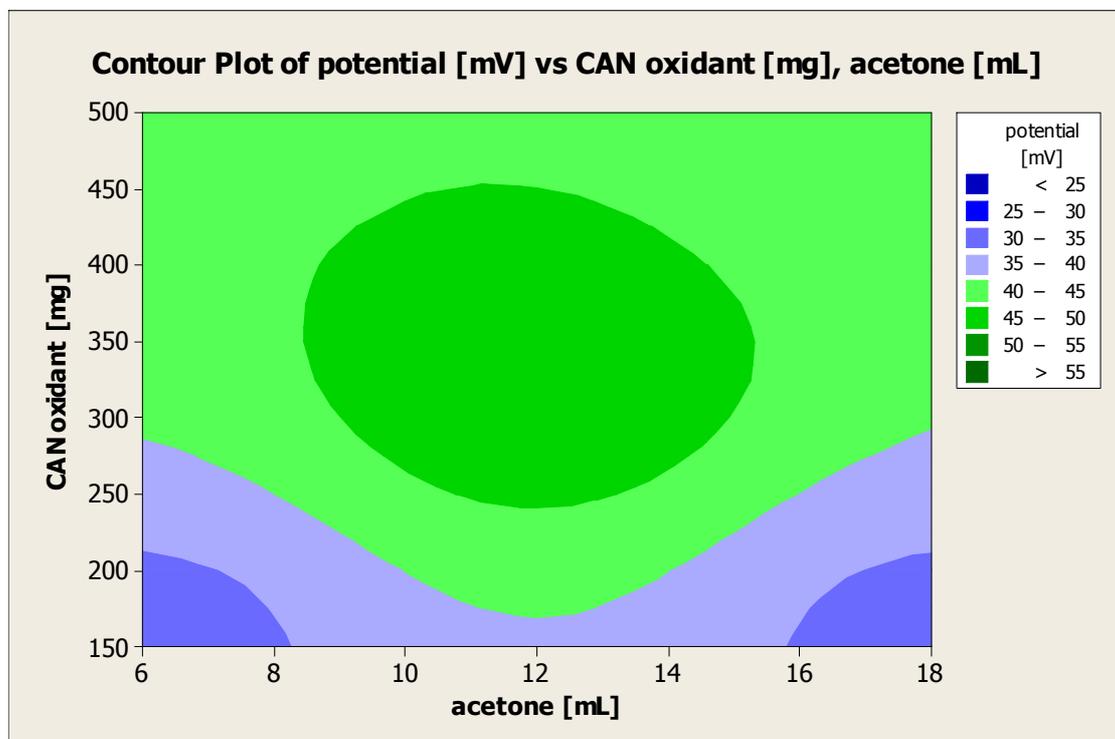
#### **Selected characterization data:**

- Average NP TEM/DLS diameters: 6.86 $\pm$ 1.55 nm and 82.90 $\pm$ 1.26 nm (DLS, PDI: 0.195) respectively
- $\zeta$  potential: +31.1 mV
- w/w Ce/Fe ratio: 0.1000
- TGA weight loss (200-410°C temperature range): 73.62%

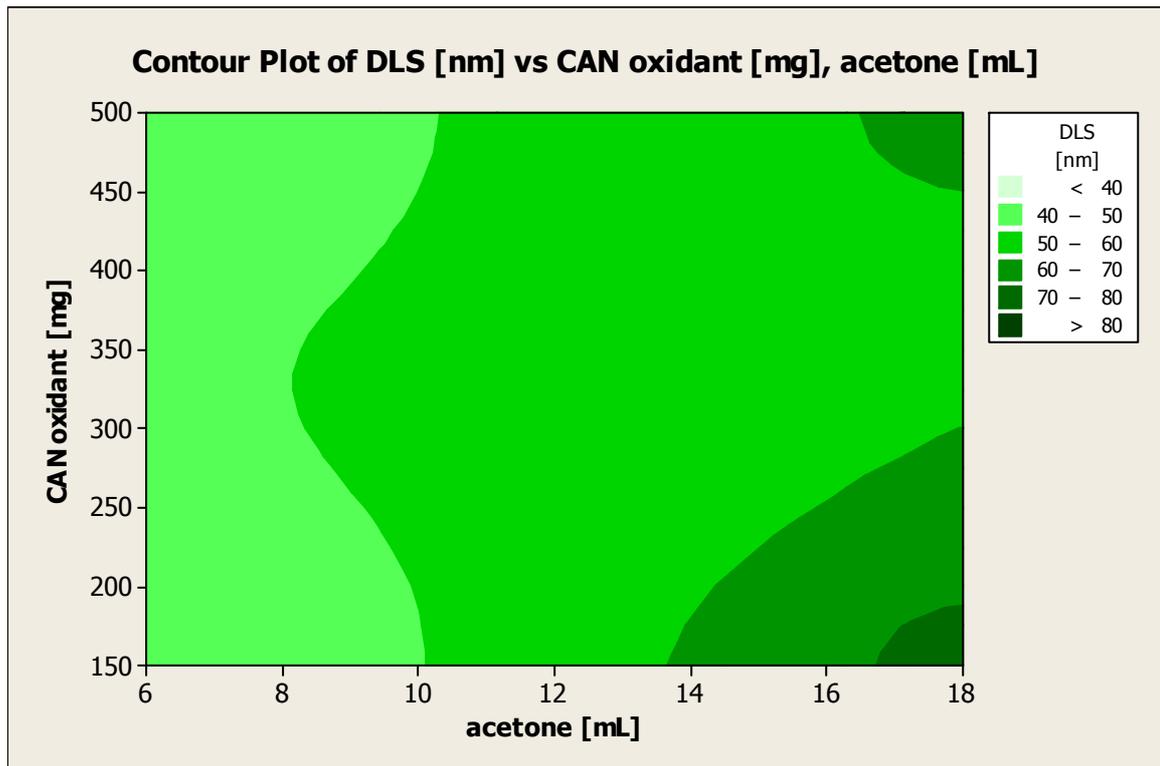
**D. DoE-optimized fabrication of CAN- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs – Complementary graphs & data relating to the fabrication of optimized CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs**



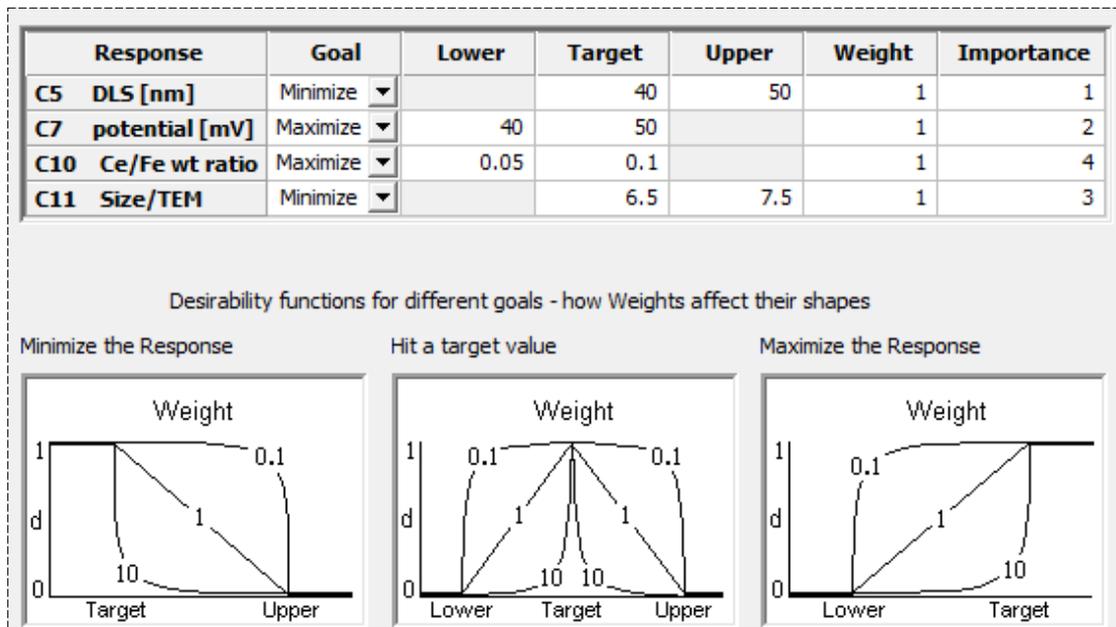
**Figure SI-1.** DoE-optimized fabrication of CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs - 2D Contour plot of NP TEM sizes vs. CAN oxidant (mg) & MeCOMe (mL) amounts



**Figure SI-2.** DoE-optimized fabrication of CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs - 2D Contour plot of NP  $\zeta$  potential values vs. CAN oxidant (mg) & MeCOMe (mL) amounts

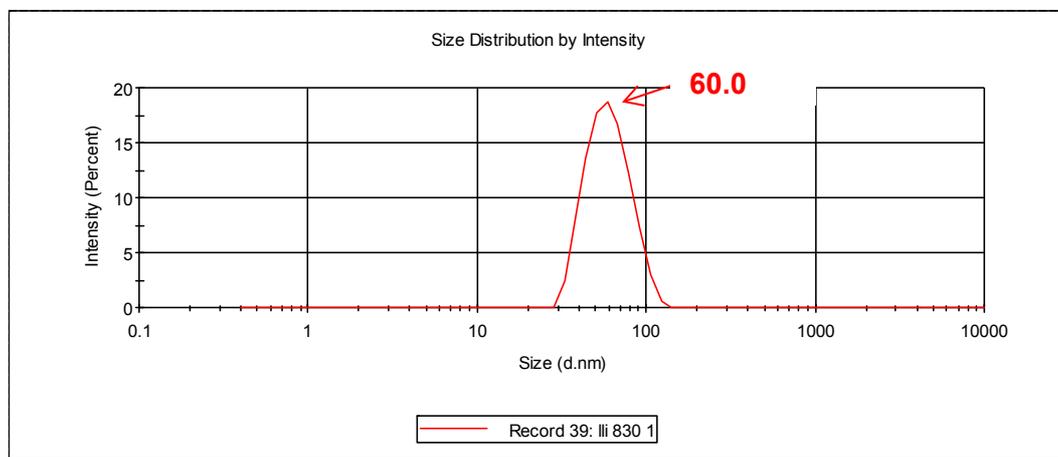


**Figure SI-3.** DoE-optimized fabrication of  $CAN_{DOE}-\gamma-Fe_2O_3$  NPs - 2D Contour plot of NP hydrodynamic size (DLS) values vs. CAN oxidant (mg) & MeCOMe (mL) amounts

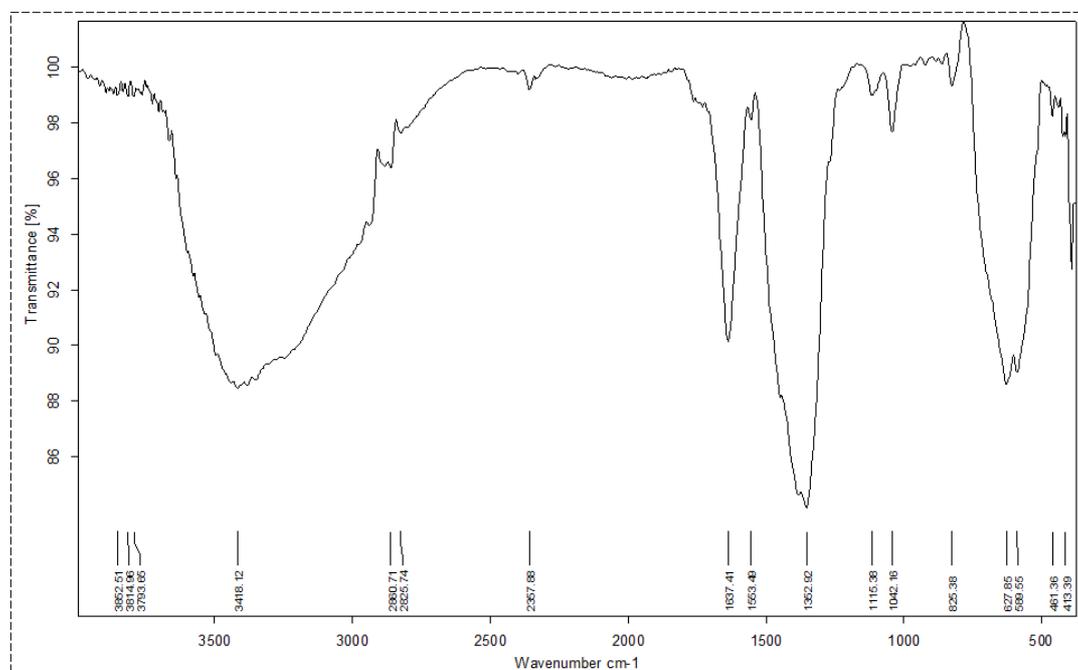


**Figure SI-4.** DoE-optimized fabrication of  $CAN_{DOE}-\gamma-Fe_2O_3$  NPs - NP specifications for MINITAB® 16 software profile optimizer tool calculations

### E. Optimized $CAN_{DOE-\gamma-Fe_2O_3}$ NPs – Additional characterization data

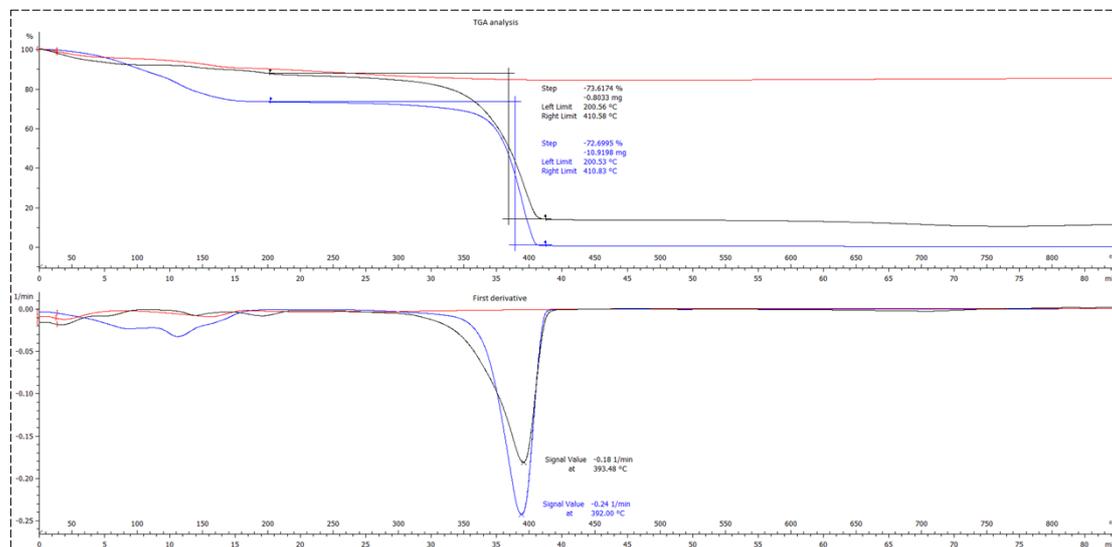


**Figure SI-5.** DLS analysis of DoE-optimized ultra-small  $6.61 \pm 2.04$  nm-sized core  $CAN_{DOE-\gamma-Fe_2O_3}$  NPs (MINITAB<sup>®</sup> 16 software profile optimizer tool output)

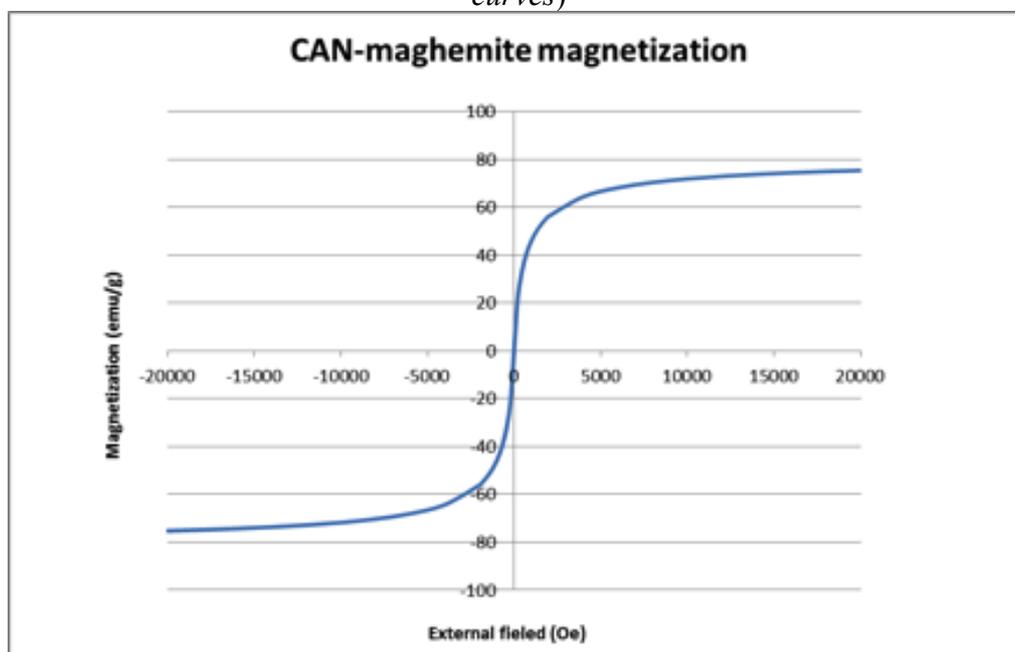


**Figure SI-6.** FT-IR spectrum of DoE-optimized ultra-small core  $6.61 \pm 2.04$  nm-sized  $CAN_{DOE-\gamma-Fe_2O_3}$  NPs (MINITAB<sup>®</sup> 16 software profile optimizer tool output)

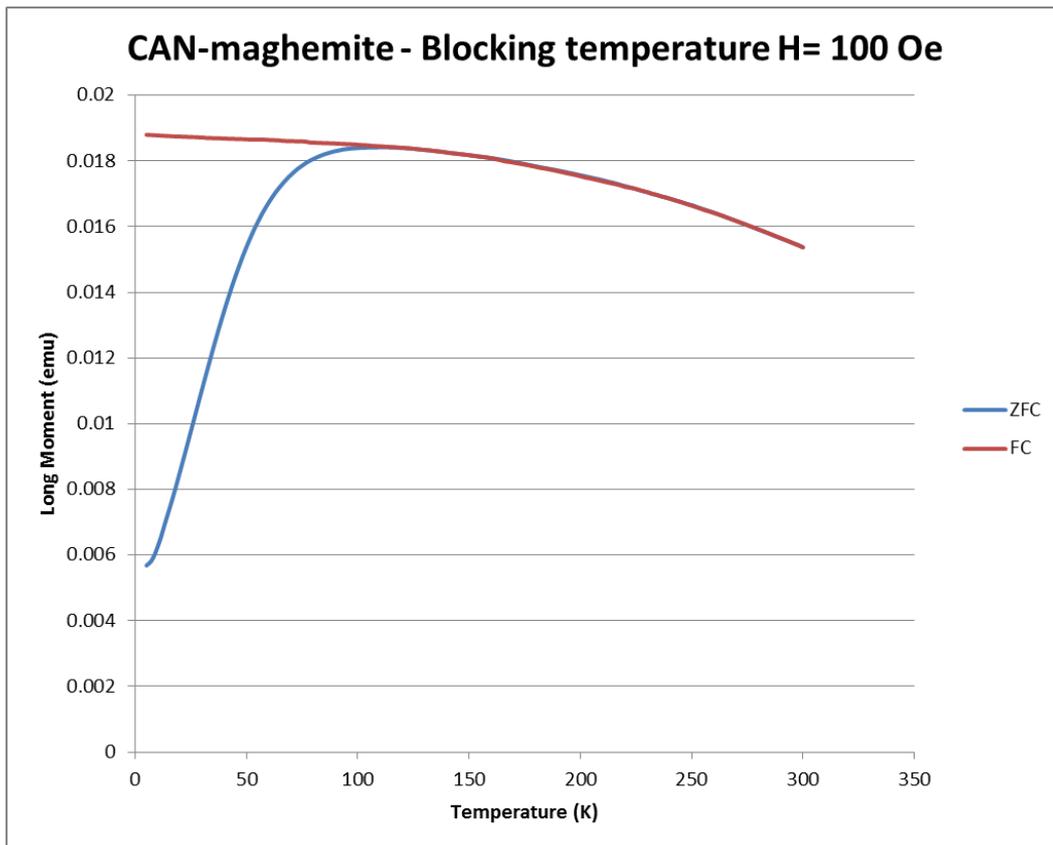
**CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs**  
***b*-PEI-CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs**  
**25kDa *b*-PEI**



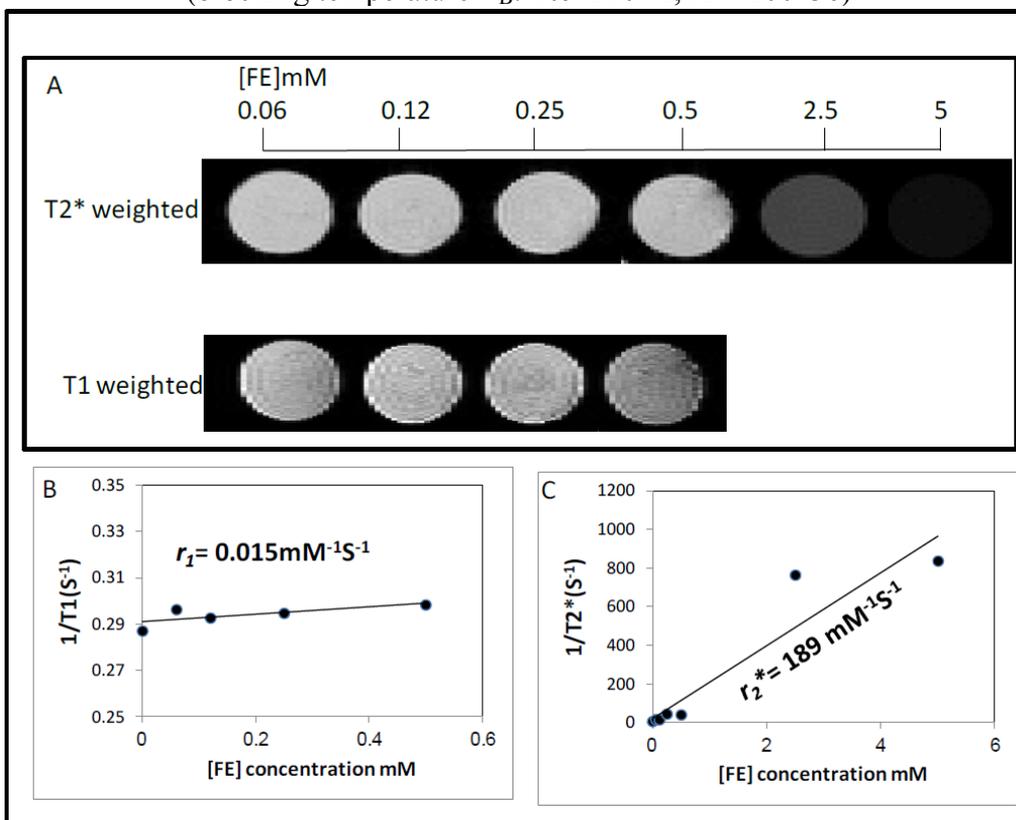
**Figure SI-7.** TGA thermogram (*top*) & weight loss derivative function (*bottom*) graphs of (i) DoE-optimized ultra-small  $6.61 \pm 2.04$  nm-sized core CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs (MINITAB<sup>®</sup> 16 software profile optimizer tool output, *red curves*), (ii) *b*-PEI<sub>25</sub>-CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs (*black curves*), & (iii) pure 25kDa *b*-PEI polymer (*blue curves*)



**Figure SI-8.** DoE-optimized ultra-small  $6.61 \pm 2.04$  nm-sized core CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs (MINITAB<sup>®</sup> 16 software profile optimizer tool output): SQUID magnetization profile (saturation magnetization  $M_s$ : 75.2 emu/g)

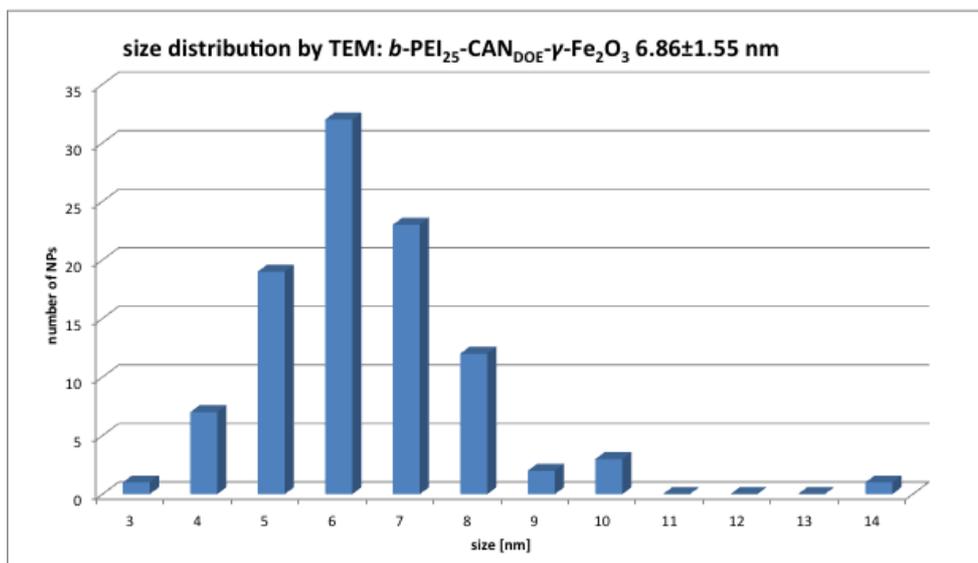


**Figure SI-9.** DoE-optimized ultra-small  $6.61 \pm 2.04$  nm-sized core  $\text{CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs (MINITAB® 16 software profile optimizer tool output): ZFC/FC graphs (blocking temperature  $T_B$ : 109-110°K, H = 100 Oe)

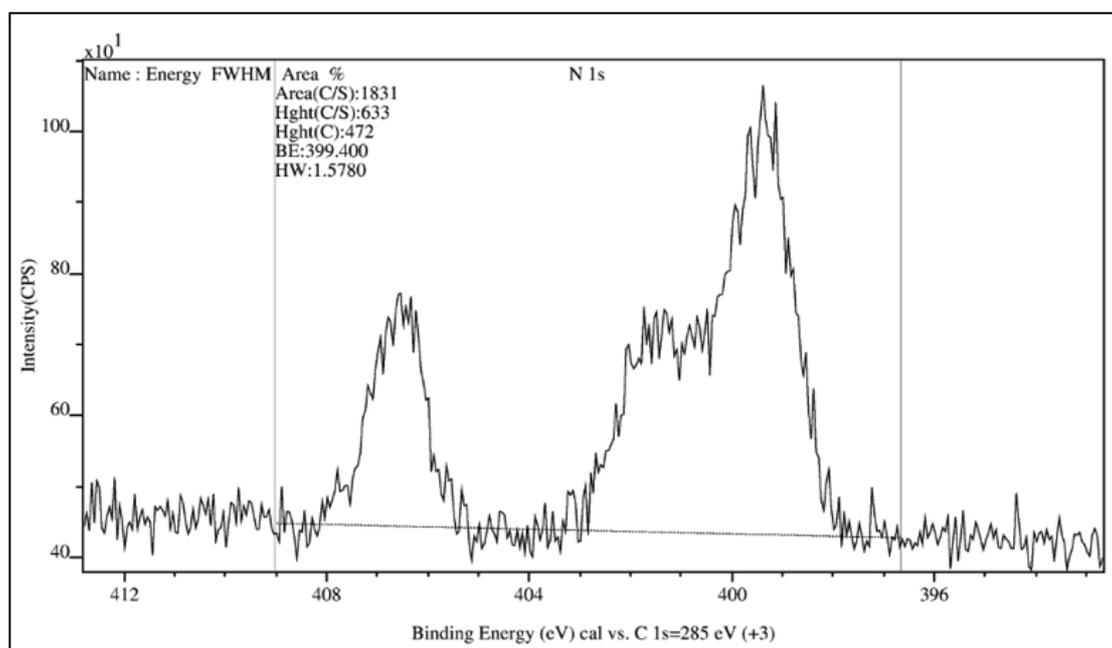


**Figure SI-10.** MRI relaxivities images & measurements for DoE-optimized ultra-small  $6.61 \pm 2.04$  nm-sized core  $\text{CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs (MINITAB® 16 software profile optimizer tool output):

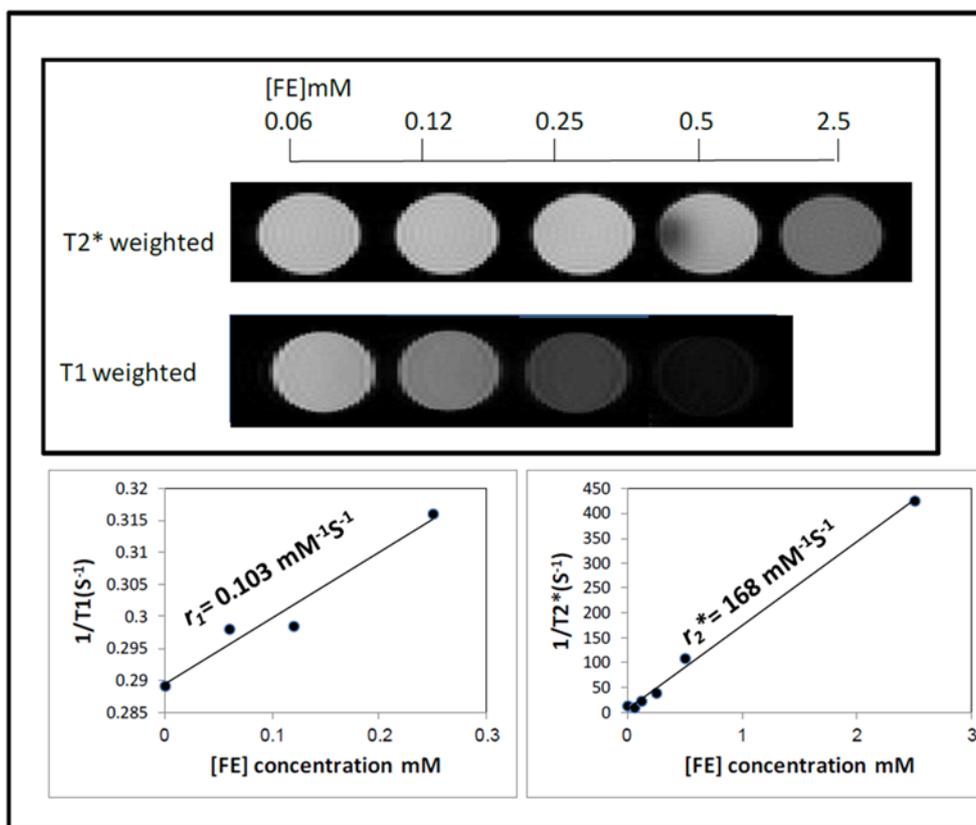
**F. Selected characterization data of optimal  $b\text{-PEI}_{25}\text{-CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs**



**Figure SI-11.** Size distribution histogram of ultra-small averaged  $6.86 \pm 1.55$  nm-sized optimized  $b\text{-PEI}\text{-CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs



**Figure SI-12.** XPS analysis of optimal  $b\text{-PEI}_{25}\text{-CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs, N 1s peaks (nitrate anions & amine species, BE: 406.0-407.0 & 398.0-402.0 eV respectively)

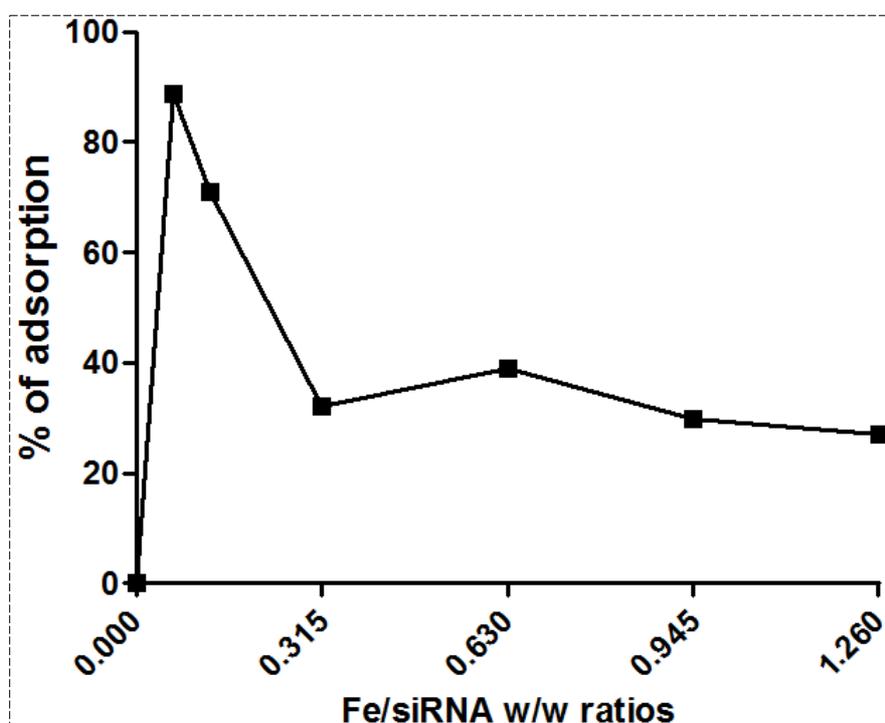


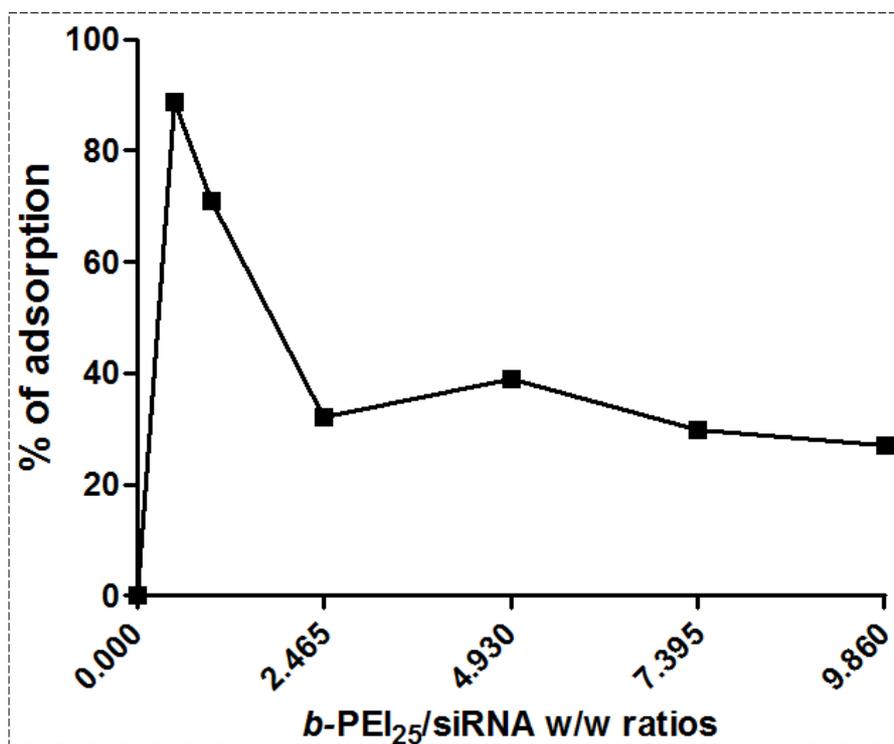
**Figure SI-13.** MRI relaxivity images & measurements for **optimized *b*-PEI-CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs**

## G. Selected experimental procedures for nanocarrier functionalization using siRNA species for cell delivery/gene silencing

### G1. Nanoparticle-mediated capture of siRNA - Illustrating example

Aqueous suspensions of former *b*-PEI<sub>25</sub>-CAN<sub>DOE</sub>- $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> NPs were diluted in ddH<sub>2</sub>O at different concentrations to reach different w/w Fe/siRNA mass ratios (0.063, 0.126, 0.315, 0.630, 0.945, and 1.260). Following appropriate calculations based on former TGA & ICP-AES data (NP batch: 0.079 mg NPs/mL suspension), these Fe/siRNA w/w ratios are equivalent to the following *b*-PEI<sub>25</sub>/siRNA w/w ratios: 0.493, 0.986, 2.465, 4.930, 7.395, and 9.860. To each nanoparticle suspension and to the control tube (absence of particles), 3.0  $\mu$ g of siRNA were added and incubated at room temperature for 15 min toward complex formation. After 15 min of incubation, suspensions were centrifuged at 11,000 rpm for 10 min and free siRNA was measured in supernatant with a spectrophotometer (Nanodrop 1000, Fisher Scientific,  $\lambda$ : 260 nm). The amount of free siRNA at each ratio was normalized to control tube (Fig. SI-15 reported below).





**Figure SI-14.** siRNA adsorption graphs by optimal  $b\text{-PEI}_{25}\text{-CAN}_{\text{DOE}}\text{-}\gamma\text{-Fe}_2\text{O}_3$  NPs at various increasing (i) Fe/siRNA (*top*) & (ii)  $b\text{-PEI}_{25}/\text{siRNA}$  (*bottom*) w/w ratios

### G2. Gene silencing experiments - Cell culture medium

U2OS human osteosarcoma cells have been obtained from the American Type Culture Collection (ATCC; Manassas, VA, USA). Cells were cultured in a Dulbecco's Modified Eagle Medium supplemented with 10% fetal bovine serum, 100  $\mu\text{g}/\text{mL}$  penicillin, 100 U/mL streptomycin, 2mM *L*-glutamine and 25mM HEPES (Biological Industries Ltd., Israel). Cells were grown at 37°C in 5%  $\text{CO}_2$ .

### G3. Luciferase U2OS human osteosarcoma cell line

U2OS cells were transfected by electroporation with (i) 5  $\mu\text{g}$  psiCHECK-2 *Firefly* and *Renilla* luciferase expression vector (Promega), and (ii) 0.25  $\mu\text{g}$  pPUR puromycin resistance vector (Clontech). Plasmids were mixed with U2OS cells ( $5 \times 10^6$  cells in 500  $\mu\text{L}$  phosphate-buffered saline), and the mixture was incubated at 4°C for 5 min and, then electro-pulsed with a Gene Pulser Xcell apparatus (Bio-Rad) at 170 V and 450  $\mu\text{F}$ . After electroporation, cells were re-suspended in a complete medium and incubated at 37°C in 5%  $\text{CO}_2$ . Subsequently, cells with incorporated plasmids were selected with 1  $\mu\text{g}/\text{mL}$  puromycin (Invivogen) and maintained in the same complete

medium for several weeks. Finally, clones were isolated and maintained in a complete medium with 1  $\mu\text{g}/\text{mL}$  puromycin for processing during additional supplementary weeks.