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## **Supporting Information to**

Cell Membrane Permeable Fluorescent Ca<sup>2+</sup> Probe Based on Bis-BODIPY with Branched PEG

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### **1. General Information**

#### Materials

Ionomycin, bis(pinacolato)diboron were purchased from Jiangsu Sukailu Chemical Co., Ltd; 3-(Nmorpholino)propanesulfonic acid (MOPS), ethyleneglycol tetraacetic acid (EGTA), phosphorus oxychloride and trifluoroacetic acid (TFA) were purchased from J&K Company; 2,3-dichloro-5,6dicyano-p-benzoquinone (DDQ) were purchased from Sinopharm Chemical Reagent Co., Ltd; 2,4dimethylpyrrole were purchased from Tianjin Kemiou Chemical Reagent Co., Ltd; N-(3-Dimethylaminopropyl)-N'-ethylcarbodiimide hydrochloride (EDCI) and 1-Hydroxybenzotrizole were purchased from Shanghai Medpep Co. Ltd; Hank's Balanced Salts Solution (HBSS) were purchased from Life Technologies Co., Ltd, and used without any further purification. Solvents were either employed as purchased or dried according to procedures described in the literature. Deionized water was obtained from a Milli-Q water purification system (Millipore).

The <sup>1</sup>H-NMR spectra were recorded at 20 °C on 600 MHz NMR spectrometer (Bruker). The<sup>13</sup>C-NMR spectra were recorded at 20 °C on 150 MHz NMR spectrometer (Bruker). Chemical shifts are reported in ppm at room temperature using CDCl<sub>3</sub> and DMSO-d<sub>6</sub> as solvent, tetramethylsilane as internal standard unless indicated otherwise. Abbreviations used for splitting patterns are s = singlet, d = dublett, t = triplet, qui = quintet, m = multiplet. Mass spectra were carried out using MALDI-TOF/TOF matrix assisted laser desorption ionization mass spectrometry with autoflexIII smartbeam (Bruker Daltonics Inc). UV/Vis spectra were recorded with a Shimadzu WV-2550 spectrophotometer. Fluorescence spectra were recorded on a Shimadzu RF-5301 fluorescence spectrophotometer. The concentration of the solution for the calculation of quantum yields was 10<sup>-5</sup> mol/L. Fluorescein in 0.1M NaOH aq was used as reference ( $\Phi r = 0.85$ ). Recycling preparative GPC purifi cations were carried out on a Shimadzu HPLC system, which consisted of a model SPD-20A tunable absorbance detector, a model RID-10A differential refractometer, an in-line degasser, a model LC-6AD Pump, a model CBM-20A controller, and a Shodex KF-802 preparative GPC column. The Shimadzu model LC-6AD pump was fitted with a 1.0 mL loop and a three directional recycling manifold that allowed for the product to be cycled back onto the column. All products were cycled over the column tow times before separation using THF as a solvent at a flow rate of  $5.0 \text{ mL min}^{-1}$ .

#### **Preparation of Cell Cultures**

HeLa cells were cultured in Dulbecco's modified Eagle's medium (DMEM medium, Invitrogen Corp) supplemented with 10% FBS, penicillin (100 units/ml), and streptomycin (100 ug/ml). All cells were maintained in a humidified atmosphere of 5/95 (v/v) of CO<sub>2</sub>/air at 37 C. The cells were passed for plated on 35 mm glass bottom poly-D-lysine coated Petri-dish for at least 24 h to enable adherence to the bottom.

#### Live cell imaging

HeLa cells were grown in Dulbecco's modified Eagle's medium (DMEM medium, Gibco) in a 35 mm glass bottom poly-D-lysine coated Petri-dish for at least 24 h to enable adherence to the bottom. The 200  $\mu$ l of 100  $\mu$ M Hanks' balanced salt solutions (HBSS) of MPFCP was added to the dish (final concentration of MPFCP is 20 $\mu$ M). After incubation at 37°C for 30 min, the cells were washed three times with PBS solutions. **CLSM** images were obtained using Olympus confocal laser scanning microscopy (Olympus Fluoview FV1000).

#### 2. Synthesis and characterization of MPFCP



Scheme S1 synthetic Scheme for Compound 3

(i) 5-Bromoisophthalic acid, DMF, 0 °C, EDCI, 0.5 h; compound 1, HOBt, rt, 24 h; (ii) [Pd(dppf)<sub>2</sub>Cl<sub>2</sub>], KOAc, bis(pinacolato)diboron, DMF, 90 °C, overnight.

#### 5-bromo-N<sup>1</sup>,N<sup>3</sup>-di(2,5,8,12,15,18-hexaoxanonadecan-10-yl)isophthalamide(2)

5-Bromoisophthalic acid (774 mg, 3.161 mmol) was dissolved in 100 mL of dry N,Ndimethylformamide (DMF), EDCI (1.5 g, 7.586 mmol) was added, and the mixture was stirred at 0°C for 30 min. Compound **1** (2.3 g, 7.651 mmol) , HOBt (1.2 g, 7.586 mmol) were added and the mixture was stirred at room temperature for 24 hours. The solvents were removed by reduced pressure, water was added to the reaction mixture, followed by extraction with  $CH_2Cl_2$ . The combined organic phase was washed with water, dried over  $Na_2SO_4$  and evaporated. The mixture was dried in a vacuum. The crude product was purified by silica gel column chromatography with  $CH_2Cl_2/CH_3CH_2OH$  (25:1) as eluent. After the solvent was removed by rotary evaporation, compound **2** was obtained as an oil liquid (1.39 g, 60%). The <sup>1</sup>H-NMR spctrum of compound **2** is shown in Fig.S6. <sup>1</sup>H-NMR (600MHz,CDCl<sub>3</sub>):  $\delta$  8.21(s, 1H), 8.15(s, 2H), 7.25(d, J=7.8Hz, 2H), 4.47 (m, 2H), 3.76 (m, 4H), 3.71-3.64 (m, 28H), 3.54 (s, 8H), 3.30(s, 12H).

# N<sup>1</sup>,N<sup>3</sup>-di(2,5,8,12,15,18-hexaoxanonadecan-10-yl)-5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2yl)isophthalamide (3)

A mixture of **2** (1.20 g, 1.610 mmol), KOAc (838.2 mg, 8.540 mmol), and bis(pinacolato)diboron (614.4 mg, 2.420 mmol) in dry DMF (23 mL) was placed in a 100 mL flask. After the mixture was stirred for 10 min, Pd(dppf)<sub>2</sub>Cl<sub>2</sub> (114.3mg, 0.140mmol) was added quickly. The mixture was stirred overnight at 90°C. After cooling to room temperature, the mixture was poured into water and extracted with dichloromethane ( $3 \times 60$  mL). The organic layer was washed with water ( $3 \times 100$  mL) and then dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. After the solvent was removed, the residue was purified by silica gel column chromatography (dichloromethane/ethanol = 20:1) to give compound **3** as a oil liquid (0.90 g, 66%).



Scheme S2 synthetic Scheme for MPFCP

( i ) POCl<sub>3</sub>, DMF, rt, 0.5 h, 45 °C, 20 h; ( ii ) 2,4-dimethylpyrrole, TFA, anhydrous  $CH_2Cl_2$ ,  $N_2$ , rt, 12 h; DDQ,  $N_2$ , 40 min; triethylamine, BF<sub>3</sub>·Et<sub>2</sub>O, 40 min;(iii) iodine monochloride, MeOH/CH<sub>2</sub>Cl<sub>2</sub>, rt, 5 min; (iv) compound 3, 2M K<sub>2</sub>CO<sub>3</sub>, Pd(PPh<sub>3</sub>)<sub>4</sub>, toluene, ethanol,  $N_2$ , 80 °C, 15 h; (v) 0.1M KOH aq, ethanol, rt, 12 h.

#### Compound 4a and 4b

For synthesis details of 4a and 4b see Roger Y. Tsien; Biochemistry. 1980, 19, 2396-2404.

#### **Compound 5a**

Compound **4a** (3.00g, 4.98mmol) was dissolved with stirring in 15 mL of dry DMF, The mixture was cooled in an ice bath and phosphorus oxychloride (3.8 mL) was added dropwise. The reaction mixture turned black almost immediately. Keep the room temperature for 30 minutes, The reaction mixture was stirred at 45 °C for 20 h. After cooling to room temperature, The reaction mixture was dissolved in 20 mL dichloromethane and poured onto crushed ice mixed with aqueous NaHCO<sub>3</sub>. The aqueous layer was extracted with dichloromethane (5 × 10 mL). The combined organic extracted with anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and avaporated. The residue was purified by silica gel column chromatography (cyclohexane/ethyl acetate = 4:1) to give compound **5a** as a pale yellow solid (1.66 g, 53%). The <sup>13</sup>C-NMR spetrum of compound **5a** is shown in Fig.S7.  $\delta$  <sup>13</sup>C-NMR (150MHz, CDCl<sub>3</sub>):  $\delta$  190.5, 171.6, 170.8, 150.1, 149.7, 145.2, 137.0, 132.0, 130.0, 126.5, 122.0, 119.3, 116.6, 114.3, 67.9,

67.4, 66.7, 61.2, 60.7, 53.7, 53.5, 25.6, 20.9, 14.0; Electrospray ionization mass spectrum is shown in Fig.S8. m/z Calcd for C<sub>32</sub>H<sub>43</sub>N<sub>2</sub>O<sub>11</sub>:631.28614, found:631.28590[M+H]+.

#### **Compound 5b**

Compound **4b** (3.00g, 5.10mmol) was dissolved with stirring in 45 mL of dry DMF, The mixture was cooled in an ice bath and phosphorus oxychloride (11 mL) was added dropwise. The reaction mixture turned black almost immediately. Keep the room temperature for 30 minutes, The reaction mixture was stirred at 45 °C for 20 h. After cooling to room temperature, The reaction mixture was dissolved in 20 mL dichloromethane and poured onto crushed ice mixed with aqueous NaHCO<sub>3</sub>. The aqueous layer was extracted with dichloromethane (5 × 20 mL). The combined organic extracted with anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and avaporated. The residue was purified by silica gel column chromatography (cyclohexane/ethyl acetate = 3:1) to give Compound **5b** as a white solid (2.59 g, 79%). The <sup>1</sup>H-NMR spetrum of **5b** is shown in Fig.S9. <sup>1</sup>H-NMR(600MHz, CDCl<sub>3</sub>):  $\delta$  9.82(s, 2H), 7.42(d, J=8.4Hz, 2H), 7.39(s, 2H), 6.80(d, J=8.4Hz, 2H), 4.34(s, 4H), 4.24(s, 8H), 4.10(m, 8H), 1.18(t, J=6.6Hz, 12H); Electrospray ionization mass spectrum is shown in Fig.S10. m/z Calcd for C<sub>32</sub>H<sub>41</sub>N<sub>2</sub>O<sub>12</sub>:645.26540, found:645.26523[M+H]+.

#### **Compound 6a**

To a solution of **5a** (1.20g, 1.90mmol) and 2,4-Dimethylpyrrole (543mg, 5.71mmol) in dry dichloromethane (DCM) (318 mL), 3 drop of trifluoroacetic acid (TFA) was added. The reaction mixture was stirred at room temperature under nitrogen for 12 h. DDQ (520 mg) was added in the mixture was stirred for 40 min at room temperature under nitrogen. Then triethylamine (10.0 mL) and borontrifluoride etherate (BF<sub>3</sub>·Et<sub>2</sub>O) (10.0 mL) were added and the mixture was stirred for 40 additional min. Water was added and the product extracted with CH<sub>2</sub>Cl<sub>2</sub>, the organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent was evaporated. The crude product was purified by silica gel column chromatography (dichloromethane/methanol = 75:1-50:1) to give compound **6a** as a drab solid (0.64 g, 40%). The <sup>1</sup>H-NMR spetrum of **6a** is shown in Fig.S11. <sup>1</sup>H-NMR (600MHz,CDCl<sub>3</sub>): $\delta$  6.91(d, J=7.8Hz,

1H), 6.81(t, J=4.8Hz, 3H), 6.71(t, J=8.4Hz, 2H), 6.00(s, 2H), 4.29(m, 4H), 4.23(s, 4H), 4.15(t, J=7.2Hz, 4H), 4.11(s, 4H), 4.10(t, J=7.2Hz, 4H), 2.57(s, 6H), 2.26(s, 3H), 1.51(s, 6H), 1.23-1.19(m, 12H);The <sup>13</sup>C-NMR spctrum of **6a** is shown in Fig.S12. <sup>13</sup>C-NMR(150MHz,CDCl<sub>3</sub>): δ 171.4, 171.3, 155.3, 150.8, 150.3, 143.1, 141.6, 140.2, 137.1, 132.3, 131. 7, 128.1, 122.3, 121.2, 121.1, 119.8, 119.1, 115.4, 113.2, 67.6, 67.2, 60.9, 60.5, 53.7, 53.6, 20.9, 14.6, 14.2, 14.1; MALDI-TOF spectrum is shown in Fig.S13. MALDI-TOF MS m/z Calcd for C<sub>44</sub>H<sub>55</sub>BF<sub>2</sub>N<sub>4</sub>O<sub>10</sub>:848.40, found:848.70[M+H]+.

#### Compound 6b

Prepared analogously to **6a**, The residue was purified by silica gel column chromatography (dichloromethane/methanol = 150:1-100:1) to give compound **6b** as a brown solid (0.11 g, 5%). The <sup>1</sup>H-NMR spetrum of **6b** is shown in Fig.S14. <sup>1</sup>H-NMR(600MHz,CDCl<sub>3</sub>): $\delta$  6.93(d, J=8.4Hz, 2H), 6.83(d, J=7.8Hz, 2H), 6.81(s, 2H), 5.99(s, 4H), 4.26(s, 4H), 4.19(s, 8H), 4.12(m, 8H), 2.57(s, 12H), 1.48(s, 12H), 1.22(t, J=7.2Hz, 12H); The <sup>13</sup>C-NMR spetrum of **6b** is shown in Fig.S15. <sup>13</sup>C-NMR(150MHz,CDCl<sub>3</sub>): $\delta$  171.0, 155.4, 150.8, 143.0, 141.3, 140.4, 131.6, 128.3, 121.6, 121.1, 119.5, 114.0, 67.6, 60.8, 53.8, 14.6, 14.5, 14.2; MALDI-TOF spectrum is shown in Fig.S16. MALDI-TOF MS m/z Calcd for C<sub>56</sub>H<sub>66</sub>B<sub>2</sub>F<sub>2</sub>N<sub>6</sub>O<sub>10</sub>:1080.50, found:1080.50[M+H]+.

#### **Compound 7a**

To a solution of 6a (600mg, 0.71mmol) in methanol/DCM (75 mL/25 mL), iodine monochloride (1.4 mL, 1M) in dichloromethane was added dropwise. The reaction mixture was stirred at room temperature for 5 min. After confirming the consumption of starting material by TLC, water was added and the product was extracted with DCM. The organic layer was washed with water and saturated NaCl aq., dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent was evaporated. The residue was purified by silica gel column chromatography (dichloromethane/methanol = 160:1-120:1) to give compound **7a** as a red solid (778 mg, 91%). The <sup>1</sup>H-NMR spetrum of **7a** is shown in Fig.S17. <sup>1</sup>H-NMR (600MHz, CDCl<sub>3</sub>):  $\delta$  6.92(d, J=8.4Hz, 1H), 6.89(d, J=8.4Hz, 1H), 6.76(s, 2H), 6.73(t, J=8.4Hz, 2H), 4.31(d, J=4.8Hz, 2H), 4.27(d, J=4.8Hz, 2H), 4.24(s, 4H), 4.16(s, 4H), 4.14-4.08(m, 8H), 2.65(s, 6H), 2.28(s,

3H), 1.51(s, 6H), 1.23(m, 12H); The <sup>13</sup>C-NMR spctrum of **7a** is shown in Fig.S18. <sup>13</sup>C-NMR(150MHz,CDCl<sub>3</sub>):δ 171.2, 171.1, 156.6, 150.9, 150.2, 145.4, 141.3, 140.7, 136.9, 132.5, 131.6, 127.5, 122.4, 121.0, 119.9, 119.2, 115.4, 113.0, 85.5, 67.6, 67.2, 62.8, 61.0, 60.6, 53.7, 20.9, 17.2, 16.0, 14.2, 14.1; MALDI-TOF spectrum is shown in Fig.S19. MALDI-TOF MS m/z Calcd for C<sub>44</sub>H<sub>53</sub>BF<sub>2</sub>I<sub>2</sub>N<sub>4</sub>O<sub>10</sub>:1100.19, found:1110.50[M+H]+.

#### Compound 7b

Prepared analogously to compound **7a** as a red solid (compound **7b**)(50 mg, 73%). The <sup>1</sup>H-NMR spctrum of **7b** is shown in Fig.S20. <sup>1</sup>H-NMR(600MHz, CDCl<sub>3</sub>): $\delta$  6.95(d, J=7.8Hz, 2H), 6.79(d, J=8.4Hz, 4H), 4.26(s, 4H), 4.20(s, 8H), 4.15(m, 8H), 2.66(s, 12H), 1.51(s, 12H), 1.24(t, J=7.2Hz, 12H); The <sup>13</sup>C-NMR spctrum of **7b** is shown in Fig.S21. <sup>13</sup>C-NMR(150MHz,CDCl<sub>3</sub>): $\delta$  170.9, 156.7, 151.0, 145.3, 141.1, 140.9, 131.6, 127.8, 121.5, 119.7, 113.7, 85.6, 67.8, 60.9, 53.8, 17.2, 16.0, 14.2; MALDI-TOF spectrum is shown in Fig.S22. MALDI-TOF MS m/z Calcd for C<sub>58</sub>H<sub>77</sub>BrN<sub>2</sub>O<sub>20</sub>:1584.08, found:1607.00[M+Na]+.

#### **Compound 8a**

In a 100 mL flask, **7a** (60mg, 0.0545mmol), **3** (125mg, 0.1636mmol) and Pd(PPh<sub>3</sub>)<sub>4</sub> (2 mg) were dissolved in freshly distilled toluene (30 ml). An aqueous solution of  $K_2CO_3$  (1.5 ml, 2M), ethanol(1.5 ml) was added and the mixture was heated at 65°C under nitrogen for 15 h. After cooling to room temperature, water was added and the product was extracted with CH<sub>2</sub>Cl<sub>2</sub>, the organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent was evaporated. The residue was purified by column chromatography on silica gel (dichloromethane/methanol = 35:1) to give compound **8a** as a red viscous liquid (124 mg, 80%). The <sup>1</sup>H-NMR spetrum of **8a** is shown in Fig.S23. <sup>1</sup>H-NMR(600MHz, CDCl<sub>3</sub>): $\delta$  8.24(s, 2H), 7.84(s, 4H), 7.26(d, J=8.4Hz, 4H) 6.93(d, J=7.8Hz, 1H), 6.88(t, J=9.6Hz, 2H), 6.78(d, J=8.4Hz, 1H), 6.69(s, 2H), 4.49(m, 4H), 4.30(d, J=6.6Hz, 4H), 4.20(s, 4H), 4.10(s, 4H), 4.09(m, 8H), 3.77(m, 8H), 3.70-3.64(m, 56H), 3.52(t, J=4.8Hz, 16H), 3.33(s, 24H), 2.53(s, 6H), 2.24(s, 3H), 1.44(s, 6H), 1.19(m, 12H); The <sup>13</sup>C-NMR spetrum of **8a** is shown in Fig.S24. <sup>13</sup>C-NMR(150MHz, CDCl<sub>3</sub>): $\delta$  171.1, 166.4,

166.1, 154.0, 151.0, 150.3, 142.7, 140.4, 139.8, 135.0, 134.7, 132.2, 132.1, 131.6, 130.3, 128.7, 127.9, 125.5, 124.1, 122.2, 121.1, 119.9, 119.3, 115.3, 113.0, 71.9, 70.6, 70.5, 69.5, 67.7, 67.3, 60.9, 60.6, 58.9, 49.4, 49.3, 29.7, 20.9, 14.2, 14.1, 13.3, 13.1; MALDI-TOF spectrum is shown in Fig.S25.
MALDI-TOF MS m/z Calcd for C<sub>112</sub>H<sub>171</sub>BF<sub>2</sub>N<sub>8</sub>O<sub>38</sub>:2285.18, found:2308.10[M+Na]+.

#### Compound 8b

Prepared analogously to compound **8a**, The residue was purified by Recycling preparative GPC with tetrahydrofuran (THF) as mobile phase to give a red viscous liquid (compound **8b**) (78 mg, 70%). The <sup>1</sup>H-NMR spetrum of 8b is shown in Fig.S26. <sup>1</sup>H-NMR(600MHz, CDCl<sub>3</sub>): $\delta$  8.17(s, 4H), 7.82(s, 8H), 7.16(d, J=7.8Hz, 8H), 6.92(d, J=8.4Hz, 2H), 6.87(d, J=10.2Hz, 4H), 4.47(m, 8H), 4.28(s, 4H), 4.16(s, 8H), 4.07(m, 8H), 3.77-3.74(m, 16H), 3.70-3.63(m, 112H), 3.52(t, J=4.8Hz, 32H), 3.32(d, J=2.4Hz, 48H), 2.52(s, 12H), 1.45(s, 12H), 1.17(t, J=7.2Hz, 12H); The <sup>13</sup>C-NMR spetrum of **8b** is shown in Fig.S27. <sup>13</sup>C-NMR(150MHz, CDCl<sub>3</sub>): $\delta$  170.9, 166.2, 154.1, 151.0, 142.6, 140.4, 139.8, 134.9, 134.6, 132.2, 131.6, 127.9, 124.0, 107.9, 107.6, 106.3, 71.8, 70.6, 70.4, 69.5, 68.0, 60.8, 58.9, 49.3, 29.1, 25.6, 23.9, 14.2, 13.3, 13.1; MALDI-TOF spectrum is shown in Fig.S28. MALDI-TOF MS m/z Calcd for C<sub>192</sub>H<sub>298</sub>B<sub>2</sub>F<sub>4</sub>N<sub>14</sub>O<sub>66</sub>:3956.09, found:3975.94[M+Na]+

#### MPFCP-1

To a solution of **8a** (40 mg, 17.6  $\mu$ mol) in methanol (4 mL), 0.1 M KOH aq. (12 mL) was added. The reaction mixture was stirred at room temperature for 12 h. The produce was dialyzed for three day against ultrapure water using Spectrumlabs dialysis membrane (molecular weight cutoff = 1000, Spectrum), then eluted through an Dowex® 50WX8-200 ion-exchange resin column (pre-washed with 18 Mohm Deionized water) to remove other cations. The solvents were removed by Freeze drying to give **MPFCP-1** as a red viscous liquid (34mg, 89%). The <sup>1</sup>H-NMR spetrum of **MPFCP-1** is shown in Fig.S29. <sup>1</sup>H-NMR(600MHz, DMSO-d<sub>6</sub>): $\delta$  8.42(d, J=7.8Hz, 4H), 8.35(s, 2H), 7.86(s, 4H), 7.21(s, 2H), 7.12(s, 2H), 7.04(s, 2H), 6.94(m, 2H), 6.65(t, J=7.8Hz, 2H), 4.31(m, 8H), 4.11(s, 4H), 3.98(s, 4H), 3.54-3.48(m, 80H), 3.19(d, J=7.8Hz, 24H), 2.46(s, 6H), 2.12(d, J=10.2Hz, 3H), 1.43(d, J=7.8Hz, 6H);

MALDI-TOF spectrum is shown in Fig.S30. MALDI-TOF MS m/z Calcd for  $C_{104}H_{155}BF_2N_8O_{38}$ :2174.05, found:2173.2[M-H]<sup>-</sup>.

#### MPFCP-2

To a solution of 8b (40 mg, 10  $\mu$ mol) in methanol (4 mL), 0.1 M KOH aq. (12 mL) was added. The reaction mixture was stirred at room temperature for 12 h. The produce was dialyzed for three day against ultrapure water using Spectrumlabs dialysis membrane (molecular weight cutoff = 1000, Spectrum, CA), then eluted through an Dowex® 50WX8-200 ion-exchange resin column (pre-washed with 18 Mohm Deionized water) to remove other cations. The solvents were removed by Freeze drying to give **MPFCP-2** as a red viscous liquid(28mg, 71%). The <sup>1</sup>H-NMR spectrum of **MPFCP-2** is shown in Fig.S31. <sup>1</sup>H-NMR(600MHz, DMSO-d<sub>6</sub>): $\delta$  8.41(d, J=7.8Hz, 8H), 8.33(s, 4H), 7.86(s, 8H), 7.01(s, 2H), 6.88(s, 2H), 4.30(m, 12H), 4.06(s, 4H), 3.53-3.47(m, 160H), 3.18(d, J=3.0Hz, 48H), 2.45(s, 12H), 1.44(s, 12H); MALDI-TOF spectrum is shown in Fig.S32. MALDI-TOF MS m/z Calcd for C<sub>184</sub>H<sub>282</sub>B<sub>2</sub>F<sub>4</sub>N<sub>14</sub>O<sub>66</sub>:3843.93, found:3842.9[M-H]<sup>-</sup>.

## **3** Supplementary Figures







Fig.S1 MPFCP-1 and MPFCP-2 probe with Ca2 + detection



**Fig.S2** Ca<sup>2+</sup>-UV-vis absorption spectra of **MPFCP** in the presence of free Ca<sup>2+</sup> at various concentrations (0, 0.017, 0.038, 0.065, 0.100, 0.150, 0.225, 0.351, 0.602, 1.35, 39 $\mu$ M) in 3-(N-morpholino)propanesulfonic acid (MOPS) buffer (30mM) containing KCl (100 mM) and ethyleneglycol tetraacetic acid (EGTA; 10mM) with the concentrations of probes 1 $\mu$ M at pH 7.2 and 22 °C.



Fig.S3 pH-dependent fluorescence emission spectra of MPFCP-1 (A) and MPFCP-2 (B)

**Table S1** $K_d$  values and optical properties of MPCFP and some previously reported Ca<sup>2+</sup> probes.

Compound	λmax (nm)	emission maximum (nm)		quantum efficiency( $\phi$ )		Fluorescence	Kd
	ε ( <u>M</u> -1cm-	Ca <sup>2+</sup> -free	Ca <sup>2+</sup> -bound	Ca <sup>2+</sup> -free	Ca <sup>2+</sup> -bound	(Fmax/Fmin)	(µM)
MPCFP- 1	520 2.6×10 <sup>4</sup>	545	545	0.0018 <sup>a</sup>	0.15	83	0.44
MPCFP- 2	525 8.0×10 <sup>4</sup>	549	550	0.0013ª	0.13	100	1.21
Fluo-4	491	nd <sup>b</sup>	516	nd <sup>b</sup>	0.14	120	0.35
CalciumG reen-1	506	531	531	nd <sup>b</sup>	0.75	14	0.19
<sup>a</sup> Estimated from the following calculation: $\varphi_{Ca2+-free} = \varphi_{Ca2+-bound} \times (F_{min}/F_{max})$ . <sup>b</sup> No date.							



**Fig.S4** Uptake of MPFCP into cell by flow cytometry. The cells were incubate with same concentration(20  $\mu$ M) of MPFCP for 2 h. The Black, green, magenta and cyan line were Control, MPFCP-0 (BODIPY alone), MPFCP-1 and MPFCP-2 respectively.



**Fig.S5** The bright-field images of Hela cells incubated by  $20\mu$ M MPFCP-2 (A), and the confocal images of HeLa cells incubated by  $20\mu$ M MPFCP-2 (B), the bright-field (C) and the confocal images (D) of HeLa cells incubated by  $20\mu$ M MPFCP-2 with ATP ( $100\mu$ M).

Part B: <sup>1</sup>H-NMR spectrum, <sup>13</sup>C NMR spectrum and MALDI-TOF

## spectrum



Fig.S7 <sup>13</sup>C-NMR spectrum(150MHz, CDCl<sub>3</sub>, 20 °C) of 5a



Fig.S8 Electrospray ionization mass spectrum of 5a



Fig.S9 <sup>1</sup>H-NMR spectrum(600MHz, CDCl<sub>3</sub>, 20 °C) of 5b



Fig.S10 Electrospray ionization mass spectrum of 5b



Fig.S13 MALDI-TOF spectrum of 6a







Fig.S16 MALDI-TOF spectrum of 6b



Fig.S17 <sup>1</sup>H-NMR spectrum(600MHz, CDCl<sub>3</sub>, 20 °C) of 7a





Fig.S19 MALDI-TOF spectrum of 7a



Fig.S21 <sup>13</sup>C-NMR spectrum(150MHz, CDCl<sub>3</sub>, 20 °C) of 7b



Fig.S22 MALDI-TOF spectrum of 7b







Fig.S24 <sup>13</sup>C-NMR spectrum(150MHz, CDCl<sub>3</sub>, 20 °C) of 8a



Fig.S25 MALDI-TOF spectrum of 8a



Fig.S28 MALDI-TOF spectrum of 8b



Fig.S29 <sup>1</sup>H-NMR spectrum(600MHz, DMSO-d<sub>6</sub>, 20 °C) of MPFCP-1



Fig.S30 MALDI-TOF spectrum of MPFCP-1



Fig.S31 <sup>1</sup>H-NMR spectrum(600MHz, DMSO-d<sub>6</sub>, 20 °C) of MPFCP-2



